

Scilab Textbook Companion for  
Coulson And Richardson's Chemical  
Engineering, Volume 2  
by J. M. Coulson, J. F. Richardson, J. R.  
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# Book Description

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Scilab numbering policy used in this document and the relation to the above book.

**Exa** Example (Solved example)

**Eqn** Equation (Particular equation of the above book)

**AP** Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

# Contents

List of Scilab Codes	5
1 Particulate Solids	8
2 Particle size reduction and enlargement	18
3 Motion of particles in a fluid	22
4 Flow of fluids through granular beds and packed columns	29
5 Sedimentation	31
6 Fluidisation	35
7 Liquid filtration	43
8 Membrane Separation Processes	52
9 Centrifugal Separations	56
10 Leaching	59
11 Distillation	65
12 Absorption of gases	98
13 Liquid liquid extraction	106
14 Evaporation	112

15 Crystallisation	126
16 Drying	134
17 Adsorption	144
18 Ion Exchange	151

# List of Scilab Codes

Exa 1.1	Surface mean diameter . . . . .	8
Exa 1.2	Surface and mass distribution curve and surface mean diameter . . . . .	9
Exa 1.3	variation of mixing index with time . . . . .	11
Exa 1.4	minimum apparent density for seperation . . . . .	12
Exa 1.5	efficiency of collection for dust . . . . .	13
Exa 1.6	Overall efficiency of collector . . . . .	14
Exa 1.7	Estimation of particle size . . . . .	16
Exa 2.1	Consumption of energy . . . . .	18
Exa 2.2	Maximum size of the particle . . . . .	19
Exa 2.3	Proposed modifications in ball mill . . . . .	20
Exa 3.1	terminal velocity . . . . .	22
Exa 3.2	Estimation of galena . . . . .	23
Exa 3.3	terminal velocity . . . . .	24
Exa 3.4	Approximate distance travelled . . . . .	26
Exa 3.5	maximum size of crystals . . . . .	27
Exa 4.1	pressure calculation . . . . .	29
Exa 5.1	Minimum area of thickener . . . . .	31
Exa 5.2	Sedimentation velocity and solids flux . . . . .	32
Exa 5.3	Rate of deposition and maximum flux . . . . .	33
Exa 6.1	minimum fluidising velocity . . . . .	35
Exa 6.2	fluidisation and transport of particles . . . . .	36
Exa 6.3	Voidage of the bed . . . . .	37
Exa 6.4	slope of adsorption isotherm . . . . .	39
Exa 6.5	Coefficient of heat transfer between gas and the particles . . . . .	39
Exa 6.6	Volumetric fraction of the bed carrying out evaporation . . . . .	41
Exa 7.1	volume of filtrate collected per cycle . . . . .	43
Exa 7.2	Effect on optimum thickness of the cake . . . . .	45

Exa 7.3	Time taken to produce 1 m <sup>3</sup> of filtrate and pressure in this time . . . . .	46
Exa 7.4	Speed of rotation for maximum throughput . . . . .	47
Exa 7.5	Optimum filtration time for maximum throughput . . . . .	48
Exa 7.6	Thickness of cake produced . . . . .	49
Exa 7.7	Increase in the overall throughput of the press . . . . .	50
Exa 8.2	Area of membrane and average flux . . . . .	52
Exa 8.3	Minimum number of membrane modules required . . . . .	53
Exa 9.1	Value of capacity factor . . . . .	56
Exa 9.3	Time taken to produce filtrate . . . . .	57
Exa 10.1	Time required for solute to dissolve . . . . .	59
Exa 10.2	Rate of feed of neutral water to the thickeners . . . . .	60
Exa 10.3	Required number of thickeners . . . . .	61
Exa 10.4	Number of ideal stages required . . . . .	62
Exa 10.5	Number of theoretical stages required . . . . .	63
Exa 11.1	Mole fraction calculation . . . . .	65
Exa 11.2	Saturated Pressure calculation . . . . .	66
Exa 11.3	Vapour phase composition of a mixture . . . . .	67
Exa 11.4	Boiling point of equimolar mixture . . . . .	68
Exa 11.5	Dew point of a equimolar mixture . . . . .	70
Exa 11.6	Composition of vapour and liquid . . . . .	71
Exa 11.7	Number of theoretical plates needed . . . . .	72
Exa 11.8	The Mc Cabe Thiele method . . . . .	74
Exa 11.9	Number of plates required at total reflux . . . . .	76
Exa 11.10	Heat required . . . . .	77
Exa 11.11	Number of theoretical stages required . . . . .	78
Exa 11.12	Amount of distillate . . . . .	79
Exa 11.13	Heat required and average composition . . . . .	80
Exa 11.14	Ideal plates required . . . . .	81
Exa 11.15	Minimum reflux ratio . . . . .	88
Exa 11.16	Minimum reflux ratio . . . . .	92
Exa 11.17	Number of theoretical plates required . . . . .	94
Exa 11.18	Xchange in n with R . . . . .	95
Exa 11.19	Optimum reflux ratio . . . . .	96
Exa 11.20	Plate efficiency for the given data . . . . .	97
Exa 12.1	Overall liquid film coefficient . . . . .	98
Exa 12.2	Mass transfer coefficient . . . . .	99
Exa 12.3	Overall transfer units required . . . . .	101

Exa 12.4	Height of transfer units and number of transfer units .	102
Exa 12.5	Height of the tower . . . . .	102
Exa 12.6	specific steam consumption . . . . .	104
Exa 13.1	Composition of final raffinate . . . . .	106
Exa 13.3	Overall transfer coefficient . . . . .	108
Exa 13.4	Surface mean droplet size . . . . .	109
Exa 13.5	Number of overall transfer units in raffinate phase . .	110
Exa 14.1	Heat surface required . . . . .	112
Exa 14.2A	Forward feed . . . . .	114
Exa 14.2B	Backward feed . . . . .	118
Exa 14.3	Efficiency of the compressor . . . . .	119
Exa 14.4	Quantity of additional stream required . . . . .	120
Exa 14.5	Capacity and economy of the system . . . . .	121
Exa 14.6	Method to drive the compressor . . . . .	123
Exa 14.7	Optimum boiling time . . . . .	124
Exa 15.1	Supersaturation ratio . . . . .	126
Exa 15.2	Increase in solubility . . . . .	127
Exa 15.3	Theoretical yield . . . . .	128
Exa 15.4	Yield of Sodium acetate . . . . .	128
Exa 15.5	Length of crystalliser . . . . .	130
Exa 15.6	Crystal production rate . . . . .	131
Exa 15.7	Vapour pressure . . . . .	132
Exa 15.8	Mass sublimation rates . . . . .	132
Exa 16.1	Time taken to dry the solids . . . . .	134
Exa 16.2	Time taken to dry the solids . . . . .	135
Exa 16.3	Mass flow rate of dry air . . . . .	136
Exa 16.4	Approximate drying time . . . . .	138
Exa 16.5	Proposed diameter and length . . . . .	139
Exa 16.6	Specified diameter of the bed . . . . .	142
Exa 17.1	Comparison of estimates with the geometric surfaces .	144
Exa 17.2	Applicability of various equilibrium theories . . . . .	145
Exa 17.3	Length of the bed . . . . .	147
Exa 17.5	Moving bed adsorption design . . . . .	149
Exa 18.1	Prediction of time t against xs values . . . . .	151
Exa 18.2	Concentration of HNO <sub>3</sub> in solution . . . . .	152



# Chapter 1

## Particulate Solids

Scilab code Exa 1.1 Surface mean diameter

```
1 clear all;
2 clc;
3 printf("\n Example 1.1");
4 //Given size analysis of a powdered material
5 d=[1,101]; //diameter of the powdered particles
6 x=[0,1]; //mass fractions of the particles
7 plot2d(d,x,style=2,rect=[0,0,120,1])
8 xtitle("size analysis of powder", "particle size (um)"
9         , "mass fraction (x)")
9 d=100*x+1; // from the given plot
10 //calculation of surface mean diameter
11 function [ds]=surface_mean_diameter(x0,x1)
12     ds=1/(integrate('1/(100*x+1)', 'x', x0, x1))
13     funcprot(0)
14 endfunction
15 ds=surface_mean_diameter(0,1); //deduced surface mean
16 printf("\n The surface mean diameter is %fum", ds);
```

---

**Scilab code Exa 1.2** Surface and mass distribution curve and surface mean diameter

```
1 clear;
2 clc;
3 printf("\n Example 1.2");
4 //from given differential eq we get these functions
5 //particle number distribution for the size range
   0-10um
6
7
8 //n=0.5*d^2;
9 //const of integration is0 since at n=0,d=0
10
11 //particle number distribution for the size range
   10-100um
12 //n=83-(0.33*(10^(5))*d^(-3))
13 //c2=83,since at d=10um,n=50
14
15 //number distribution plot for the powdered material
   of size range 0-100um
16 function [n]= number_distribution(d)
17     if(d<=10) then
18         n=0.5*d^2;
19     else
20         n=83-(0.33*(10^(5))*d^(-3));
21     end
22     funcprot(0)
23 endfunction
24 d=0;
25 while(d<=100)
26     n=number_distribution(d);
27     plot(d,n,"+");
28     d=d+1;
29 end
30 xtitle("number_distribution_plot","diameter(um)","
   number distribution");
31 ps=[0 6.2 9.0 10.0 11.4 12.1 13.6 14.7 16.0 17.5
```

```

    19.7 22.7 25.5 31.5 100];
32 function [n1]=difference(i)
33 //ps=[0 6.2 9.0 10.0 11.4 12.1 13.6 14.7 16.0 17.5
    19.7 22.7 25.5 31.5 10];
34 //according to the given particle sizes particle
    sizes are in um
35     n1=number_distribution(ps(i+1))-
        number_distribution(ps(i));
36     funcprot(0);
37 endfunction
38 function [da]=average(i)
39     da= (ps(i+1)+ps(i))/2;
40     funcprot(0);
41 endfunction
42 tot_n1d12=0;
43 tot_n1d13=0;
44 i=1;
45 for i=1:14
46     tot_n1d12=tot_n1d12+difference(i)*(average(i)
        )^2;
47     tot_n1d13=tot_n1d13+difference(i)*(average(i)
        )^3;
48 end
49 printf("\n tot_n1d12 =%d \n tot_n1d13=%d",tot_n1d12,
    tot_n1d13);
50 function [s]=surface_area(j)
51     s=(difference(j)*(average(j))^2)/tot_n1d12;
52     funcprot(0);
53 endfunction
54 su=0;
55 j=0;
56 xset('window',1);
57
58 plot(0,0,"o-");
59 for j=1:14
60     su=su+surface_area(j);
61     plot(ps(j+1),su,"o-");
62 end

```

```

63 xtitle("surface area and mass distribution plot","
        diameter(um)","surface area or mass distribution"
        );
64 //mass distribution plot
65 function [x]=mass_distribution(k)
66     x=(difference(k)*(average(k))^3)/tot_n1d13;
67     funcprot(0);
68 endfunction
69 ma=0;
70 k=0;
71 plot(0,0,"+");
72 for k=1:14
73     ma=ma+mass_distribution(k);
74     plot(ps(k+1),ma,"+");
75 end
76 //evaluating surface mean diameter
77 function [d]=surface_mean_diameter(l)
78     e=0;
79     for l=1:14
80         n=(mass_distribution(l)/average(l));
81         e=e+n;
82     end
83 d=1/e;
84     funcprot(0);
85 endfunction
86 printf("\nthe surface mean diameter is: %fum",
        surface_mean_diameter());

```

---

**Scilab code Exa 1.3** variation of mixing index with time

```

1 clear all;
2 clc;
3 printf("\n Example 1.3");
4 p=0.20; //components analysed represents 20 percent
        of the mixture by mass

```

```

5 //for a completely unmixed system
6 so=p*(1-p);
7 //for a completely random mixture :
8 n=100; //Each of the sample removed contains 100
    particles
9 sr=p*(1-p)/n;
10 s=[0.025 0.006 0.015 0.018 0.019];
11 time_secs=[30 60 90 120 150];
12 printf("\n degree of mixing is :\n")
13 function [b]=degree_of_mixing()
14 for i=1:5
15     b(i)=(so-s(i))/(so-sr);
16     disp(b(i)); //b is the degree of mixing
17 end
18     return b;
19 funcprot(0)
20 endfunction
21 plot2d(time_secs,degree_of_mixing(),style=3)
22 xtitle("degree of mixing curve","time_secs","
    degree_of_mixing")
23 //plot of sample variance vs time(secs)
24 xset('window',1)
25 plot2d(time_secs,s,style=2)
26 xtitle("sample variance curve","time_secs","sample
    variance")
27 //from the graph the maxima is at 60 secs

```

---

**Scilab code Exa 1.4** minimum apparent density for separation

```

1 //minimum size of the particle in the mixture of
    quartz and galena(mm)
2 clear all;
3 clc;
4 printf("\n Example 1.4");
5

```

```

6 //maximum size of the particle(mm)
7 d_max=0.065;
8 //minimum size of the particle(mm)
9 d_min=0.015;
10 //density of quartz(kg/m^3)
11 p_quartz=2650;
12 //density of galena (kg/m^3)
13 p_galena=7500;
14 //minimum density of the particle which will give
    this seperation
15 //When stoke's law is applied the required density
    is as given below
16 function [d]=stoke_required_density()
17     p=poly([0], 'p');
18     d=roots((p-7500)-(p-2650)*(d_max/d_min)^2);
19     funcprot(0);
20 endfunction
21 d=stoke_required_density();
22 printf("\n required density is = %d kg/m^3",d);
23 //When Newton's law is applied then the required
    density is as given below
24 function [e]=newton_required_density()
25     r=poly([0], 'r');
26     e=roots((r-7500)-(r-2650)*(d_max/d_min));
27     funcprot(0);
28 endfunction
29 e=newton_required_density();
30 printf("\nrequired density is by newton law =%d kg/m
    ^3",e);

```

---

**Scilab code Exa 1.5** efficiency of collection for dust

```

1 clear all;
2 clc;
3 printf("\n Example 1.5");

```

```

4 //efficiency of the collector for different size
  ranges
5 efficiency_1=45; //in percentage for the size range
  of 0–5um
6 efficiency_2=80; //in percentage for the size range
  of 5–10um
7 efficiency_3=96; //in percentage for the size range
  greater than 10um
8
9 //mass percent of the ndust for various size range
10 mass_1=50; //in percentage for the size range of 0–5
  um
11 mass_2=30; //in percetage for the size range of 5–10
  um
12 mass_3=20; //in percentage for the size range
  greater than 10um
13 // on the basis of 100kg dust
14 mass_retained_1=0.45*50; //mass retained (kg) in the
  size range of 0–5um
15 mass_retained_2=0.80*30; //mass retained (kg) in the
  size range of 5–10um
16 mass_retained_3=0.96*20; //mass retained (kg) in the
  size range greater than10um
17 overall_efficiency=0.45*50+0.80*30+0.96*20;
18 printf("\n the overall efficiency is =%f",
  overall_efficiency);

```

---

#### Scilab code Exa 1.6 Overall efficiency of collector

```

1 clear all;
2 clc;
3 printf("\n Example 1.6");
4 //To calculate mass flow of the dust emitted
5 mass_1=10; //in percentage in the size range of 0–5um
6 mass_2=15; //in percentage in the size range of 5–10

```

```

    um
7 mass_3=35; //in percentage in the size range of 10–20
    um
8 mass_4=20; //in percentage in the size range of 20–40
    um
9 mass_5=10; //in percentage in the size range of 40–80
    um
10 mass_6=10; //in percentage in the size range of
    80–160um
11 effieicny_1=20; //in percentage in the size range of
    0–5um
12 efficiency_2=40; //in percentage in the size range of
    5–10um
13 efficiency_3=80; //in percentage in the size range of
    10–20um
14 efficiency_4=90; //in percentage in the size range of
    20–40um
15 efficiency_5=95; //in percentage in the size range of
    40–80um
16 efficiency_6=100; //in percentage in the size range
    of 80–160um
17 dust_burden=18; //in g/m^3 at the entrance
18 //taking 1m^3 as the basis of calculation
19 total_mass_retained
    =18*(0.1*0.20+0.15*0.40+0.35*0.80+0.2*0.9+0.1*0.95+0.1*1)
    ;
20 printf("\ntotal mass retained =%fg",
    total_mass_retained);
21 total_efficiency=(total_mass_retained/18)*100;
22 printf("\ntotal efficiency is =%f",total_efficiency)
    ;
23 total_mass_emitted=18-total_mass_retained;
24 printf("total mass emitted is:%fg",
    total_mass_emitted);
25 t=18*(0.1*0.80+0.15*0.60+0.35*0.20);
26 printf("\ntotal mass emitted less than 20um is %fg",
    t);
27 e=t*100/total_mass_emitted;

```



```

28 printf("\nThe efficiency of particles emitted is %f"
    ,e);
29 //gas flow is 0.3m^3/sec
30 f=0.3*total_mass_emitted;
31 printf("\nmass flow rate is:%fkg/sec",f);

```

---

### Scilab code Exa 1.7 Estimation of particle size

```

1 clear;
2 clc;
3 printf("\n Example 1.7");
4 Ai=(%pi/4)*(0.075)^2;//cross sectional area at the
    gas inlet in m^2
5 do=0.075;//gas outlet diameter in m
6 p=1.3;//gas density in kg/m^3
7 Z=1.2;//height of the seperator in m
8 dt=0.3;//seperator diameter in m
9 v=1.5;//gas entry velocity in m/sec
10 G=(Ai*v*p);//mass flow rate of the gas in kg/sec
11 printf("\n cross sectional area at the gas inlet is
    %fm^2",Ai);
12 printf("\ngas outlet diameter is %fm",do);
13 printf("\ngas density is %fkg/m^3",p);
14 printf("\nheight of the seperator is %fm",Z);
15 printf("\nseperator diameter is%fm",dt);
16 printf("\nmass flow rate of the gas is %fkg/sec",G);
17 function [u]=terminal_vel()
18     u=0.2*(Ai)^2*(do)*p*9.8/(%pi*Z*(dt)*G);//
    velocity is in m/sec
19     funcprot(0);
20 endfunction
21 u=terminal_vel();
22 printf("\nthe terminal velocity of the smallest
    particle retained by the seperator =%fm/sec",u);
23 function [d]=particle_diameter(u)

```

```
24     u=terminal_vel();
25     n=0.018*10^(-3); //viscosity in mNs/m^2
26     ps=2700; //density of the particle in kg/m^3
27     d=((u*18*n)/(9.8*(ps-p)))^(0.5); //particle size
        in um
28     funcprot(0);
29 endfunction
30 u=terminal_vel();
31 d=particle_diameter(u);
32 do=d*10^6;
33 printf("\n particle diameter by the stoke law is
        %fum",do);
```

---

## Chapter 2

# Particle size reduction and enlargement

Scilab code Exa 2.1 Consumption of energy

```
1 clear;
2 clc;
3 printf("\n Example 2.1");
4 //Computing energy required in particle size
   reduction by Rittinger 's law
5 //energy required in crusing is given by E =Kr.fc
   ((1/L2)-(1/L1))
6 //Given : Energy required in crushing particles from
   50mm to 10mm is      13.0kw/(kg/sec).
7
8 Cr=13.0*(50/4); //Cr = Kr*fc
9
10 //Energy required to crush the particles from 75mm
   to 25mm
11 E = Cr*((1/25)-(1/75));
12 printf("\n The energy required in crushing the
   materials from average particle size of 75mm to
   25mm by Rettingers law is %fkj/kg",E)
13
```

```

14 //Computing the energy required for crushing by Kick
    's law
15 //E = (Kk*fc)*ln(L1/L2) by Kick's law
16
17 Ck = (13.0)/log(50/10); //Ck = Kk*fc
18
19 //Energy required to crush the material from 75mm to
    25mm
20 Ek = Ck*log(75/25);
21 printf("\nThe energy required for crushing the
    material from average particle size of by Kicks
    law 75mm to 25mm is %fkj/kg \n",Ek);
22 printf("\n The size range required is that for
    coarse crushing and Kicks law more closely
    relates the energy required for plastic
    deformation before fracture occurs so the energy
    calculated as that by Kicks law will be taken as
    the more reliable result\n");

```

---

### Scilab code Exa 2.2 Maximum size of the particle

```

1 clear;
2 clc;
3 printf("\n Example 2.2");
4 //Calculating the maximum size of the particle that
    can be fed to the rollers.
5 //Given angle of nip = 31degree
6 //Given diameter of the crushing rolls = 1m
7 //Distance between the crushing rolls is 12.5mm
8 r1 = 0.5; //size of crushing rolls is in meters
9 b = 0.00625; //Distance between the crushing rolls is
    0.0125mm
10 r2 = (r1 + b)/(cos((%pi/180)*15.5))-0.5;
11 printf("\nThe maximum size of the particles which
    should be fed to the rollers : %d mm",r2*10^3);

```

```

12
13 //Calculating the throughput at 2.0 Hz when the
    actual capacity of the machine is 12%.
14 //Working face of the rolls are 0.4m long
15 //bulk density of the feed is 2500kg/m3
16 printf("\nThe cross sectional area for flow is %.3f
    m2",0.0125*0.4);
17 printf("\nThe volumetric flow rate is %.2f m3/sec"
    ,2.0*0.005);
18 printf("\nThe actual throughput is %d kg/sec"
    ,0.010*12*2500/100);

```

---

### Scilab code Exa 2.3 Proposed modifications in ball mill

```

1 clear;
2 clc;
3 printf("\n Example 2.3");
4 //Given diameter of the ball mill is 1.2meters
5 //Speed of rotation is 0.80Hz
6
7 //for small particles effective radius is 0.6meters
8 //critical speed of the rotation
9 g=9.80; //acceleration due to gravity is in m
    ^2/sec.
10 r=0.6; //effective radius of rotation is in
    meters.
11 w = sqrt(g/r);
12 printf("\nThe critical speed of the rotation is %.2f
    rad/sec",w);
13 f=w/(2*%pi); //f is the frequency of the rotation
    and is in Hz
14 printf("\nThe critical frequency of the rotation is
    %fHz\n",f);
15 optimum_frequency = 0.6*f;
16 printf("The optimum frequency of the rotation is %.2

```

```
    f Hz", optimum_frequency);
17 printf("\nGiven frequency of the rotation is 0.80Hz\n
    ");
18 printf("The frequency of the rotation should be
    halved\n");
19 printf("Therefore the optimal frequency is half the
    critical frequency");
```

---

# Chapter 3

## Motion of particles in a fluid

Scilab code Exa 3.1 terminal velocity

```
1 clear all;
2 clc;
3 printf("\n Example 3.1");
4     d = 0.00040; //Diameter of the particle in m
5     p1 = 820; //Density of the fluid in kg/m^3
6     meu = 0.01; //Viscosity of the fluid in N s/m
           ^2
7     p2 = 7870; //Density of steel in kg/m^3
8     g = 9.81; //Acceleartion due to gravity in m
           /sec^2
9 //Computation of terminal velocity of a spherical
   particle
10 function [x]=galileo_number()
11
12     x = (2*d^3*(p2-p1)*p1*g)/(3*(meu)^2); //x = (Ro/
           pu^2)*Re^2
13     funcprot(0);
14 endfunction
15 x = galileo_number();
16 printf("\n The value of (Ro/pu^2)/Re^2 is %f\n",x);
17
```

```

18
19 //From table 3.4 log(x) corresponds to log(Re)=0.222
20 Re = 1.667;
21 function [u]=teminal_velocity()
22     u=(Re*meu)/(p1*d);
23     funcprot(0);
24 endfunction
25 u = teminal_velocity();
26 printf("\n The terminal velocity of the steel ball
    is %.3f m/sec or %.0f mm/sec",u,u*1000);

```

---

### Scilab code Exa 3.2 Estimation of galena

```

1 clear;
2 clc;
3 printf("\n Example 3.2");
4 u_water = 5*10^(-3); //The flow velocity of the
    water in m/sec
5 p_galena = 7500; //The density of galena is in
    kg/m^3
6 p_limestone = 2700; //The density of limestone is
    in kg/m^3
7 viscosity = 0.001; //The viscosity of water in N
    s/m^2
8
9 //calculating maximum value of reynold's number
    considering 5mm particle size
10 Re_max = (u_water*1000*0.0001)/(viscosity);
11 printf("\n The maximum permissible value of Re is %f
    ",Re_max);
12
13 //maximum particle size of galena which will be
    carried away by water
14 d = sqrt((u_water*(18*viscosity))/((7500-1000)*9.81)
    );

```



```

15 printf("\nmaximum particle size of galena which will
    be carried away by water is %.1f um",d*10^(6));
16
17 //maximum particle size of limestone which will be
    carried away by water
18 d1 = sqrt((u_water*(18*viscosity))/((2700-1000)
    *9.81));
19 printf("\nmaximum particle size of limestone which
    will be carried away by water is %.1f um",d1
    *10^(6));
20
21
22 //From the given data 43% galena and 74% limestone
    will be removed .
23 //Given that in the feed there is 20% galena and 80%
    limestone
24 //Assuming 100g feed
25 printf("\n\nIn the overflow:");
26 printf("\nAmount of galena is %fg",(20*0.43));
27 printf("\nAmount of limestone is %fg",(80*0.74));
28 printf("\nconcentration of galena is %.1f per cent"
    ,(20*0.43*100)/(20*0.43+80*0.74));
29 printf("\n\nconcentration of galena is %fper cent"
    ,(80*0.74*100)/(20*0.43+80*0.74));
30 printf("\n\nIn the underflow:")
31 printf("\nconcentration of galena is %.1f percent"
    ,(20*(1-0.43)*100)/(20*(1-0.43)+80*0.26))
32 printf("\nconcentration of limestone is %.1f per
    cent" ,(80*0.26*100)/(20*0.57+80*0.26))

```

---

### Scilab code Exa 3.3 terminal velocity

```

1 clear;
2 clc;
3 printf("\n Example 3.3");

```

```

4 min_area =6*10(-6); //minimum area of mica
   plates in m2
5 max_area =6*10(-4); //maximum area of mica
   plates in m2
6 p_oil = 820; //density of the oil in kg/m
   ^3
7 Viscosity = 0.01; //Viscosity is in N s/m2
8 p_mica = 3000; //Density of mica in kg/m3
9
10 printf("\n smallest particles
           largest particles");
11 printf("\nA: %fm2
           %fm2
           ",min_area,max_area);
12 printf("\ndp: %fm
           %fm
           ",sqrt(4*min_area/(%pi)),sqrt(4*max_area/(%pi)));
13 printf("\ndp3:%f*10(-8)m3
           %f*10(-5)m3
           ",sqrt(4*min_area/(%pi))(3)*10(8),sqrt(4*
           max_area/(%pi))(3)*10(5));
14 printf("\nv: %f*109m3
           %f*107m3
           ",0.285*sqrt(4*min_area/(%pi))(3)*10(9),sqrt(4*
           max_area/(%pi))(3)*(107)*0.0285);
15 printf("\nk: %f
           %f
           ",0.285,0.0285);
16
17 x1 = (4*0.285/(%pi*0.012))*(3000-820)
   *(820*2.103*10(-8)*9.81);
18 x2 = (4*0.0285/(%pi*0.012))*(3000-820)
   *(820*2.103*10(-5)*9.81);
19 printf("\n(Ro/ρu2)Reo2 is %d for the smallest
   particles and %d for the largest particles",x1,x2
   );
20 //From table 3.4 Re for smallest particle is 34.9
   and that for the largest particle is 361

```

```

21 u1 = 34.9*0.01/(820*2.76*10(-3));
22 printf("\nTerminal velocity for the smallest
    particle is %.3f m/sec",u1);
23 u2 = 361*0.01/(820*2.76*10(-2));
24 printf("\nTerminal velocity for the largest particle
    is %.3f m/sec",u2);
25 printf("\n\n Thus it is seen that all the particles
    settle at approximately the same velocity");

```

---

#### Scilab code Exa 3.4 Approximate distance travelled

```

1 clear;
2 clc;
3 printf("\n Example 3.4");
4 v_particle = 6; //velocity of the particle in m/sec
5 v_water = 1.2; //velocity of the water in m/sec
6 v_rel = v_particle - v_water;//relative velocity of
    particles relative to the fluid in m/sec
7 Re1 = 6*10(-3)*v_rel*1000/(1*10(-3)); //Re1 is the
    reynold's no.
8 printf("\nReynold no. is %d",Re1);
9
10 //When the particle has been retarded to a velocity
    such that Re=500
11 ydot = (v_rel*500)/Re1;
12 printf("\nParticle velocity is %.3f m/sec\n",ydot);
13 c = 0.33/(6*10(-3))*(1000/2500);
14 f = sqrt((3*6*10(-3))*(2500-1000)*9.81)/1000);
15 function [y]=Fa(t)
16     y = (-1/22)*(log(cos(0.517*22*t) + 4.8/0.517*sin
        (0.517*22*t)));
17     funcprot(0);
18 endfunction
19
20 function [yd]=deriv(t)

```

```

21     yd= -0.083+(0.517*(9.28*cos(11.37*t) - sin
        (11.37*t))/(cos(11.37*t) + 9.28*sin(11.37*t))
        );
22     funcprot(0);
23 endfunction
24
25 function[ydd]=double_deriv(t)
26     ydd = -0.517*(11.37)^2*(9.28*cos(11.37*t) - sin
        (11.37*t))/(cos(11.37*t) + 9.28*sin(11.37*t))
        ;
27     funcprot(0);
28 endfunction
29
30
31 told = 0;
32 while 1
33     tnew = told - deriv(told)/double_deriv(told);
34     if (tnew == told) then
35         y = Fa(told);
36         d = y;
37         printf("\nThe distance moved with speed less
                than 0.083m/sec is %.3fm",d);
38         t=told;
39         printf("\n The time taken by particle to
                move this distance is %.3fsec",t);
40         break;
41     end
42     told = tnew;
43 end
44
45
46 printf("\nThe distance moved by the particle
        relative to the walls of the plant %.3fm",1.2*t -
        d);

```

---

Scilab code Exa 3.5 maximum size of crystals

```
1 clear;
2 clc;
3 printf("\n Example 3.5");
4 //rate of dissolution of salt
5 function [x]=dissolution(d)
6     x = (3*10^(-6))-(2*10^(-4)*3.406*10^(5)*d^2);
7     funcprot(0)
8 endfunction
9
10 //rate of falling of the particle in stokes law
    region
11 function [y]=rate_h(d)
12     y = 3.406*d^(2)/(-3*10^(-6)-68.1*d^2); //y is in
        m/sec
13     funcprot(0);
14 endfunction
15
16 printf("\n By trial and error the solution for d is
    0.88 mm");
17 printf("\n The rate of dissolution is %f ",
    dissolution(8.8*10^(-4)));
18 printf("\n The rate of falling of the particle is %f
    m/sec",rate_h(8.8*10^(-4)));
```

---

# Chapter 4

## Flow of fluids through granular beds and packed columns

Scilab code Exa 4.1 pressure calculation

```
1 clear;
2 clc;
3 printf("\n Example 4.1");
4 //Calculating modified reynold's no.
5 a = 800; //it is in m^2/m^3
6 Product_rate = 0.5; //it is in g/sec
7 Reflux_ratio = 8;
8 Vapour_rate = 4.5; //it is in g/sec
9 G = (4.5*10^(-3))/((%pi/4)*(0.1^2)); // units are in
    kg/m^2.sec
10 meu = 0.02*10^(-3); //units are in Ns/
    m^2
11 e = 0.72;
12
13 Re1 = G/(800*0.28*0.02*10^(-3));
14 printf("\n The modified reynolds no. is %d",Re1);
15
16 x = 4.17/Re1 + 0.29; //x = R1/(pu1^2)
17 printf("\n The value of R1/(pu1^2) is %f",x);
```

```
18
19 l = 16*0.15;           //in meters
20 //Solving the integral integral of(pdP)from Pc toPs
    =(R1/ρu2)*S*(1-e)*G2*l/e3
21
22 Pc = poly([0], 'Pc');
23 y = roots(151.3-(4.73*10(-5)*(80002-Pc2)));
24 printf("\n The value of Pc is %dN/m2",y(1));
```

---

# Chapter 5

## Sedimentation

Scilab code Exa 5.1 Minimum area of thickener

```
1 clear;
2 clc;
3 printf("\n Example 5.1");
4 //Basis 1 kg of solids
5 feed_rate_solid = 1.33; //Mass rate of feed of
   solids in kg/sec
6 U = 1.5; //Mass rate of solids in
   the underflow in kg/sec
7 Y = [5.0 ; 4.2; 3.7 ; 3.1; 2.5];
8 printf("\n concentration(Y) (kg water/kg solids):\n"
   );
9 printf("%.1 f\n",Y);
10 printf("\n water to overflow (Y-U) (kg water/kg
   solids):\n");
11 O = Y - 1.5; //Amount of water to overflow in kg
   water/kg solids
12 printf("\n %.1 f\n",O);
13 Uc = [2.00*10(-4);1.20*10(-4);0.94*10(-4)
   ;0.70*10(-4);0.50*10(-4)];
14 printf("\n sedimentation rate uc (m/sec):\n");
15 printf("%f \n",Uc);
```



```

16 X
    =[1.75*10^4;2.25*10^4;2.34*10^4;2.29*10^4;2.00*10^4];
    //X = (Y-U)/Uc
17 printf(" \n\n(Y-U)/Uc (s/m):\n");
18 printf("\n %d\n",X);
19 z = max(X); //prints the maximum value of X
20 printf("\nMaximum value of (Y-U)/Uc = %ds/m",z);
21
22 //Calculating the require darea of the thickener
23 A = z*1.33/1000; //1.33 is the mass feed rate of
    solids in kg/sec
24 //1000 is the density of water in kg/m^3
25 printf("\n The required area of the thickener is :%
    .2fm^2\n",A);

```

---

### Scilab code Exa 5.2 Sedimentation velocity and solids flux

```

1 clear;
2 clc;
3 printf("\n Example 5.2");
4 //Area of the tank required to give an underflow
    concentration of 1200kg/m^3 for a feed rate of 2
    m^3/min
5
6 //Initial height of slurry in the tank
7 H =
    [900;800;700;600;500;400;300;260;250;220;200;180];
8 uc =
    [13.4;10.76;8.6;6.6;4.9;3.2;1.8;1.21;1.11;0.80;0.60;0.40];
9 i=1;
10 while i<13
11     c(i)=200*900/H(i);
12     x(i)=1000*(1/c(i)-1/1200);

```

```

13     sed(i) = c(i)*uc(i)/(1000*60);
14     y(i)= uc(i)*10^(-3)/((1/c(i)-1/1200)*60);
15     z(i) = 1/y(i);
16     i=i+1;
17 end
18 printf("\nH(mm)");
19 printf("\n%d",H);
20 printf("\n c(kg/m^3):\n");
21 printf("%d\n",c);
22 printf("Sedimentation flux(kg.s/m^2):\n");
23 printf("%.4f\n",x);
24 printf("uc/(1/c-1/1200)\nkg.sec/m^2:\n");
25 printf("%.4f\n",y);
26 printf("1000*(1/c-1/cu)\nmm^3/kg*10^3\n");
27 printf("%.3f \n",x);
28 printf("\n\n(1/c-1/1200)/uc\n m^2.kg/sec\n");
29 printf("%.1f\n",z);
30 m1=max
    ([18.7;20.1;21.3;22.7;23.8;26.0;27.8;30.3;30.0;29.2;27.8;25.0])
    ;
31 printf("\n\nthe maximum value of (1/c-1/1200)/uc is
    %.1f m^2*kg/s",m1);
32 A = 2*200*30.3/60;
33 printf("\n The area required is A = Qc[(1/c-1/cu)/uc
    ]max = %dm^2",A)

```

---

### Scilab code Exa 5.3 Rate of deposition and maximum flux

```

1 clear;
2 clc;
3 printf("\t Example 5.3 ");
4 //Assumption: Resistance force F on an isolated
    sphere is given by Stoke's law:F = 3*pi(meu)d*u
5
6 C = poly([0], 'C');

```

```

7 x=roots(-4.8*C+(1-C));
8 printf("\n concentration is: %.3f",x);
9
10 //terminal falling velocity u can be calculated by
    force balance
11 //u = d^2*g/(18*meu)*(ps-p)
12 function [u]=terminal_velocity()
13     d = 10^(-4); //diameter is in meters
14     g = 9.81; //acceleration due to gravity is
        in m/sec^2
15     meu = 10^(-3); //viscosity is in N.s/m^2
16     ps = 2600; //density is in kg/m^3
17     p = 1000; //density is in kg/m^3
18
19     u = (d^2)*g*(ps-p)/(18*meu);
20     funcprot(0);
21 endfunction
22
23
24 function [si]=si_max()
25     u=terminal_velocity()
26     printf("\n The terminal falling velocity is %.5f
        m/sec",u);
27     si=u*x*(1-x)^(4.8);
28     funcprot(0);
29 endfunction
30 si = si_max();
31 printf("\nThe maximum value is %f*10^(-4) m^3/m^2sec
        ",si*10^4)

```

---

# Chapter 6

## Fluidisation

Scilab code Exa 6.1 minimum fluidising velocity

```
1 clear;
2 clc;
3 printf("\n Example 6.1");
4 //Calculating minimum fluidisation velocity
5
6 //Calculating Galileo number
7 function [Ga]=Galileo_number()
8     d = 3*10^(-3); //particle size is in meters
9     p = 1100;      //density of liquid is in kg/m^3
10    ps = 4200;     //density of spherical particles
11                    is in kg/m^3
12    g = 9.81;      //acceleration due to gravity is
13                    in m/sec^2
14    u = 3*10^(-3); //viscosity is in Ns/m^2
15    Ga = d^3*p*(ps-p)*g/u^2;
16    funcprot(0);
17 endfunction
18 Ga = Galileo_number();
19 printf("\nGalileo number = %f*10^5",Ga*10^(-5));
20
21 //Calculating Re mf
```

```

20 Remf = 25.7*(sqrt(1+5.53*10(-5)*(1.003*10(5)))-1);
21 printf("\nValue of Remf is %d",Remf);
22
23 umf = Remf*(3*10(-3))/(3*10(-3)*1100);
24 printf("\nminimum fluidisation velocity is %.1f mm/
    sec",umf*1000);

```

---

### Scilab code Exa 6.2 fluidisation and transport of particles

```

1 clear;
2 clc;
3 printf("\n Example 6.2");
4 //Calculating voidage by considering eight closely
    packed spheres of diameter d in a cube of size 2d
5 printf("\n (a)");
6 function [e]=voidage()
7     d = 1*10(-4); //diameter is in meters
8     meu = 3*10(-3); //viscosity is in Ns/m(2)
9     ps = 2600; //density is in kg/m(3)
10    p = 900; //density is in kg/m(3)
11    e = [8*d(3)-8*(%pi/6)*d(3)]/(8*d(3));
12    funcprot(0);
13 endfunction
14 e = voidage();
15 printf("\nvoidage = %.2f",e);
16
17 //Calculating minimum fluidisation mass flow rate
18
19 function [Gmf]=min_fluidis_vel()
20    e = voidage();
21    d = 1*10(-4); //diameter is in meters
22    meu = 3*10(-3); //viscosity is in Ns/m(2)
23    ps = 2600; //density is in kg/m(3)
24    p = 900; //density is in kg/m(3)
25    g = 9.81; //acceleration due to gravity

```

```

        is in m/sec^2
26     Gmf = 0.0055*(e)^(3)/(1-e)*(d^2)*p*(ps-p)*g/meu;
27     funcprot(0);
28 endfunction
29 Gmf = min_fluidis_vel();
30 printf("\nminimum fluidisation velocity is %.3f kg/m
    ^2sec",Gmf);
31
32
33 printf("\n (b)");
34 function [u]=terminal_velocity()
35     e = voidage();
36     d = 1*10^(-4); //diameter is in meters
37     meu = 3*10^(-3); //viscosity is in Ns/m^2
38     ps = 2600; //density is in kg/m^3
39     p = 900; //density is in kg/m^3
40     g = 9.81; //acceleration due to gravity
        is in m/sec^2
41     u = d^(2)*g*(ps-p)/(18*meu);
42     funcprot(0);
43 endfunction
44 printf("\nterminal velocity is %.4fm/sec",
    terminal_velocity());
45
46 //Reynolds no for this Terminal velocity is
47 Re = (10^(-4)*0.0031*900)/(3*10^(-3));
48 printf("\nReynlds no =%.3f",Re);
49 printf("\nThe required mass flow rate is %.2f kg/m^2
    sec",terminal_velocity()*900);

```

---

### Scilab code Exa 6.3 Voidage of the bed

```

1 // to calculate voidage of the bed
2 clear;
3 clc;

```

```

4 printf("\n Example 6.3");
5 function [Ga]=Galileo_number()
6     d = 4*10(-3); //particle size is in meters
7     p = 1000; //density of water is in kg/m3
8     ps = 2500; //density of glass is in kg/m3
9     g = 9.81; //acceleration due to gravity is
        in m/sec2
10    u = 1*10(-3); //viscosity is in Ns/m2
11    Ga = d3*p*(ps-p)*g/u2;
12    funcprot(0);
13 endfunction
14 printf("\nGalileo number = %.2f*105",Galileo_number
        ()*10(-5));
15
16 function [Re]=Reynolds_no()
17     Ga = Galileo_number();
18     Re = (2.33*Ga(0.018)-1.53*Ga(-0.016))(13.3);
19     funcprot(0);
20 endfunction
21 printf("\n The Reynolds no is %d",Reynolds_no());
22 v = Reynolds_no()*(1*10(-3))/(0.004*1000);
23 printf("\nvelocity = %.2f m/sec",v);
24
25 n = poly([0], 'n');
26 z = roots((4.8-n)-0.043*(Galileo_number())(0.57)*(n
        -2.4));
27 printf("\nvalue of n is %.2f",z);
28
29 //voidage at a velocity is 0.25m/sec
30 e=0.1;
31 while 1
32     enew = e -((0.25/0.45)-e(2.42))/(-2.42*e1.42);
33     if (enew == e) then
34         printf("\nVoidage is %.3f",e);
35         break;
36     end
37     e=enew;
38 end

```

---

**Scilab code Exa 6.4** slope of adsorption isotherm

```
1 clear;
2 clc;
3 printf("\n Example 6.4");
4 t = 250:250:2000; //time is in secs
5 y =
    [0.00223;0.00601;0.00857;0.0106;0.0121;0.0129;0.0134;0.0137];

6
7 i =1;
8 yo = 0.01442;
9 while i<9
10     z(i) = y(i)/yo;
11     y(i)=1-z(i);
12     x(i)=log(y(i));
13     i=i+1;
14 end
15 xtitle("slope of adsorption isotherm","Time(sec)","
    log(1-(y/yo))")
16 plot(t,x,"o-")
17 printf("\nFrom the graph the value of slope is %f"
    ,-0.00167);
18 Gm = 0.679*10^(-6); //units are in kmol/sec
19 W = 4.66; //units are in gram
20 b = poly([0], 'b');
21 s = roots(-0.00167*4.66*b+0.679*10^(-6));
22 printf("\n b = %.4f kmol/kg",s*10^3);
```

---

**Scilab code Exa 6.5** Coefficient of heat transfer between gas and the particles



```

1 clear;
2 clc;
3 printf("\n Example 6.5");
4 gas_flow_rate =0.2;           //units are in kg/m^2
5 c = 0.88;                     //specific heat capacity
    of air is kj/kg K
6 viscosity = 0.015*10^(-3); //viscosity is in Ns/m^2
7 d = 0.25*10^(-3);           //particle size is in
    meters
8 k = 0.03;                     //thermal conductivity is
    in W/m K
9 e = 0.57;                     //e is voidage
10
11 T = [339.5;337.7;335.0;333.6;333.3;333.2]; //
    temperature is in kelvins
12 deltaT = T - 333.2;
13 h = [0;0.64;1.27;1.91;2.54;3.81];
14 xtitle("temperature rise as a function of bed height
    ", "height above bed support(mm)", "deltaT(K)");
15 plot(h,deltaT, 'o-');
16
17 //Area under the curve gives the value of the heat
    teansfer integral =8.82mm K
18
19 q = 0.2*0.88*(339.5-332.2);
20 printf("\n heat transferred = %.2f kw/m^2 of bed
    cross sectional area",q);
21
22 //Assuming 1m^3 volume
23 Vp = (1-0.57);               //Volume of particles is in m^3
24 printf("\n Volume of particles is %.2f m^3",Vp)
25 v1 = (%pi/6)*d^3;           //Volume of 1 particle in m^3
26 printf("\n Volume of 1 particle is %.2f*10^(-12) m^3
    ",v1*10^(12));
27 printf("\n number of particles is %.2f*10^(10)",Vp/
    v1*10^(-10));
28
29 x =poly([0], 'x');

```

```

30 h = roots(1100 - x*(1.03*10^4)*(8.82*10^(-3)));
31 printf("\n heat transfer coefficient = %.1f W/m^2",h
    );
32
33 //Nu = 0.11Re^(1.28)
34 Re = (0.2*0.25*10^(-3))/(0.015*10^(-3));
35 h1 = 0.11*(Re)^(1.28)*k/d;
36 printf("\n h = %.1f W/m^2 K",h1);

```

---

**Scilab code Exa 6.6** Volumetric fraction of the bed carrying out evaporation

```

1 clear;
2 clc;
3 printf("\n Example 6.6");
4 cp = 0.85; //specific
    heat capacity of the air
5 h = [0 0.625 1.25 1.875 2.5 3.75]; //height in mm
6 T=[339.5 337.7 335.0 333.6 333.3 333.2]; //
    temperature in K
7 deltaT = T - 333.2; //
    temperature difference in kelvins
8 plot(h,deltaT,"o-");
9 xtitle("deltaT as a function of bed height", "Height
    above bed support z(mm)", "Temperature difference
    deltaT (K)");
10 //From the plot area under the curve is 6.31 K mm
11 sp = (6/(0.25*10^(-3)))*(0.5); //sp is surface area
    per unit volume in m^2/m^3
12 G = 0.2; //in kg/m^2sec
13 Cp = 850; //Cp is in J/kg K
14 h1 = poly([0], 'h1');
15 s = roots(0.2*850*6.3-h1*1.2*10^4)*6.31*10^(-3));
16 printf(" \n Coefficient for heat transfer between

```

```

    the gas and the particles= %.1fW/m^2 K",s);
17
18 printf("\n Let the evaporation rate be 0.1 kg/sec at
    a temp difference = 50 degK");
19 mdot = 0.1;           //evaporation rate is 0.1 kg/sec
20 Latent_heat = 2.6*10^(6);
21 printf("\n The heat flow = %.1f*10^5 W",mdot*
    Latent_heat*10^(-5));
22 A=(2.6*10^5)/(14.1*50);
23 printf("\n the effective area of the bed A = %d m^2"
    ,A);
24 printf("\n The surface area of the bed = %d m^2"
    ,0.1*1.2*10^4);
25 printf("\n hence the fraction of the bed = %.2f"
    ,369/1200);

```

---

# Chapter 7

## Liquid filtration

Scilab code Exa 7.1 volume of filtrate collected per cycle

```
1 clear;
2 clc;
3 printf("\n Example 7.1");
4 //In the leaf filter filtration is at const pressure
   from the start
5 //V^2 + 2ALV/v = 2(-deltaP)A^2t/(ruv)
6
7 //In the filter press ,a volume V1 of filtrate is
   obtained under const rate conditions in time t1,
   and filtration is then carried out at constant
   pressure.
8 //V1^2 + 2ALV1/v = 2(-deltaP)A^2t1/(ruv)
9 //and (V^2 - V1^2) + 2AL/ (V - V1) = 2( - P )A
   ^2/ r (t - t1)
10
11 //for the leaf filter
12 t2 = 300; //t2 is in secs
13 V2 = 2.5*10^(-4); //V2 is in m^3
14 t3 = 600; //t3 is in secs
15 V3 = 4*10^(-4); //V3 is in m^3
16 A = 0.05; //A is in m^2
```

```

17 deltaP = -7.13*10^(4); //it is in N/m^2
18 //putting these values in above eq
19
20 a = [2*7.13*10^(4)*0.05^(2)*300 -2*0.05*2.5*10^(-4)
      ;2*7.13*10^(4)*0.05^(2)*600 -2*0.05*4*10^(-4)];
21 b = [(2.5*10^(-4))^2;(4*10^(-4))^2];
22 x = inv(a)*b;
23 y = [1/x(1);x(2)];
24 printf("\n L/   =%f*10^(-3)    and r   = %f*10^(11)"
      ,y(2)*10^3,y(1)*10^(-11));
25
26 //for the filter press
27 V1 = poly([0], 'V1');
28 s = roots(V1^2 + (2.16*y(2)*V1)-(4*10^(5)*2.16^2)/y
      (1)*180);
29 printf("\n the value of V1 = %fm^3",s(2));
30
31 //For a constant pressure period (t - t1)=900secs
32 //Calculating the total volume of filtrate
33 V = poly([0], 'V');
34 d = roots((V^2-3.33*10^(-4))+(1.512*10^(-2)*(V
      -1.825*10^(-2))-5.235*10^(-6)*900));
35 printf("\n The value of V = %.3f m^3",d(2));
36
37 f = (4*10^(5)*(2.16)^2)/(7.13*10^(11)*(6.15*10^(-2)
      + 2.16*3.5*10^(-3)));
38 printf("\n The final rate of filtration is %.2f
      *10^(-5) m^3/sec",f*10^(5));
39
40 // Assuming viscosity of the filtrate is the same
      as that of the wash-water
41 rw_400 = (0.25)*f;
42 printf("\n Rate of washing at 400 kN/m2 = %.1f
      *10^(-6) m^3/sec",rw_400*10^(6));
43
44 rw_275 = rw_400*(275/400);
45 printf("\n Rate of washing at 275 kN/m^2 = %.1f
      *10^(-6) m^3/sec",rw_275*10^(6));

```

```
46 printf("\n Thus the amount of wash-water passing in
    600s = %.3f m^3",600*rw_275);
```

---

### Scilab code Exa 7.2 Effect on optimum thickness of the cake

```
1 clear;
2 clc;
3 printf("\n Example 7.2");
4 //The slurry contains 100kg whiting/m^3 of water
5 printf("\n Volume of 100 kg whiting = %f m^3"
    ,100/3000);
6 printf("\n Volume of cake = %f m^3",0.0333/0.6);
7 printf("\n Volume of liquid in cake = %f m^3"
    ,0.05556*0.4);
8 printf("\n Volume of filtrate = %.3f m^3", (1-0.0222)
    );
9 printf("\n volume of cake/volume of filtrate v = %f"
    ,0.05556/0.978);
10 A = 10^(-4); //area in sq meters
11 deltaP = -1.65*10^(5); //P is in pascals
12 l = 0.01; //length is in meters
13 vol_flow_rate = 2*10^(-8); //Volume flow
    rate is in m^3/sec
14 u = 10^(-3); //viscosity is in Ns/m^2
15
16 r = poly([0], 'r');
17 r1 = roots((10^4)*(2*10^(-8)*r) -1.65*10^(5)/(10^(-5)
    ));
18 printf("\n r = %.2f*10^(13)/m^2",r1*10^(-13));
19
20 function [Lopt]=optimum()
21     Lopt = 1.161*10^(-3)*(900)^(0.5); //t = 900
    secs
22     funcprot(0);
23 endfunction
```

```

24 printf("\n optimum frame thickness = %.1f mm",2*
    optimum()*1000);
25
26 //total cycle time = 1.015L^2 + 900
27 //rate of cake production R = L/(1.015L^2 + 900)
28
29 L = poly([0], 'L');
30 L1 = roots(1.025*10^(6)*L^2 + 900 - 2.050*10^(6)*L
    ^2);
31 printf("\n Frame thickness = %.2f mm",2*L1(1)*10^3)
    ;

```

---

**Scilab code Exa 7.3** Time taken to produce 1 m<sup>3</sup> of filtrate and pressure in this time

```

1 clear;
2 clc;
3 printf("\n Example 7.3");
4 V = 0.094; //volume in m^3
5 deltaP = -3530; //P is in kN/m^2
6
7 //At t = 1105 secs
8 V1 = 0.166; //V is in m^3
9 deltaP1 = -5890; //P is in kN/m^2
10
11 a = [2.21*10^(6) -0.094;6.51*10^(6) -0.166];
12 b = [0.0088;0.0276];
13 x = inv(a)*b;
14 y = [x(2);x(1)];
15 printf("\n LA/v =%f A^2/ r v = %f*10^(-7)",y
    (1),y(2)*10^7);
16 printf("\n For the full size plant:");
17 printf("\n LA/v = %f A^2/ r v=%f*10^(-7)",10*y
    (1),y(2)*10^8);
18

```

```

19 //Solving LHS of the integral
20 LHS = integrate('b+0.154+2.31', 'b', 0, 1);
21 //Equating LHS = RHS
22 t = LHS/(3.46*10^(-3));
23 printf("\n t = %d secs", t);
24 printf("\n deltaP = %dkN/m^2", (1+0.154)
        /(4.64*10^(-7)*857));

```

---

#### Scilab code Exa 7.4 Speed of rotation for maximum throughput

```

1 clear;
2 clc;
3 printf("\n Example 7.4");
4 a = [2*84300*0.02^(2)*60
      -2*0.02*0.0003;2*84300*0.02^(2)*120
      -2*0.02*0.00044];
5 b = [0.0003^2;0.00044^2];
6 x = inv(a)*b;
7 y = [x(2);1/x(1)];
8 printf("\n L/v = %f          ruv = %f*10^(10)", y(1), y
        (2)*10^(-10));
9 printf("\n Area of filtering surface = %f m^2", 4*(
        %pi));
10 printf("\n Bulk volume of cake deposited =%.3f m^3/
        revolution", 4*(%pi)*0.005);
11
12 V = sqrt(1*10^(-6)*143^2);
13 printf("\n V = %.3f m^3", V);
14
15 t =poly([0], 't');
16 t1 = roots(0.141^2 +2*2.19*10^(-3)
            *0.141-2*84300*(4*(%pi))^(2)*t/(3.48*10^10));
17 printf("\n t = %f secs", t1);
18 printf("\n time for 1 revolution =%.1f secs", t1/0.4)
        ;

```



```

19 printf("\n speed = %.3fHz",0.4/t1);
20 printf("\n rate of filtrate production w = %.2f kg/
    sec",143/67.3)
21 printf("\n mass of slurry S =%.1f kg/sec",1.66*2.11)
    ;

```

---

### Scilab code Exa 7.5 Optimum filtration time for maximum throughput

```

1 clear;
2 clc;
3 printf("\n Example 7.5");
4 V1 = 0.00025; //V is in m^3
5 t = 300; //t is in secs
6 a = [7.14*10^(-6) 2.86*10^(-4);11.42*10^(-6)
    2.86*10^(-4)];
7 b = [1.2*10^(6);1*10^(6)];
8 x = inv(a)*b;
9
10 //for the plate and frame filter
11 B1 = x(1)/(2*2.2^2*413*10^3);
12 B2 = x(2)/(2.2*413*1000);
13
14 printf("\n r v = %d\n",x(1));
15 printf("\n r l = %d",x(2));
16 printf("\n B1= %f B2= %f",B1,B2);
17 printf("\n the filtration time for maximum
    throughput is:");
18 t1 = 21.6*10^3;
19 t0= t1 +B2*(t1/B1)^(0.5);
20 printf("\n t = %f secs",t0);
21 V = (t1/B1)^(0.5);
22 printf("\n V= %f m^3",V);
23 printf("\nMean rate of filtration is: %.2f *10^-6 m
    ^3/s", (V/(t1+t0))/10^-6);

```

---

### Scilab code Exa 7.6 Thickness of cake produced

```
1 clear;
2 clc;
3 printf("\n Example 7.6");
4 A=0.6*0.6*%pi;           //in m^2
5 rate=1.25*10^-4;        // in m^3/s
6
7 v_w=0.2/(3*10^3);
8 v_f=10^-3-v_w;
9
10 v=v_w/v_f;
11 v_rate=rate*v;
12 w=360*0.2;
13
14 t=v_rate*w/A;
15 printf("\nThickness of cake produced is : %.1f mm",t
        /10^-4);
16 K = poly([0], 'K');
17 K1 = roots((1.25*10^(-4)*360)^2-K*(6.5*10^(4)
        *(0.36*(%pi))^(2)*72));
18 printf("\n The value of K is %.2f*10^(-10)",K1
        *10^(10));
19
20 //Filter press
21 //Using a filter press with n frames of thickness b
    m the total time, for one complete cycle of the
    press =(tf+120n+240),where tf is the time during
    which filtration is occurring
22 //overall rate of filtration = Vf/(tf + 120n + 240)
23
24 // Vf = 0.3^(2)*n*b/0.143
25 //tf = 2.064*10^5 b^2
26
```

```

27 b = poly([0], 'b');
28 b1 = roots(b^2 - 0.0458*b - 0.001162);
29 printf("\n The thickness is %.4f m", b1(1));
30
31 function [n]=number_of_plates()
32
33     n = (0.030 + 25.8*b1(1)^2)/(0.629*b1(1)-0.015);
34     funcprot(0);
35 endfunction
36 n = number_of_plates();
37 printf("\n The minimum number of plates required is
    %d", ceil(n));
38
39 d = poly([0], 'd');
40 d1 = roots(ceil(n)*(0.629*d-0.015)-0.030-25.8*d^2);
41 printf("\n The sizes of frames which will give
    exactly the required rate of filtration when six
    are used are %f mm", d1*10^3);
42 printf("\n\n\n Thus any frame thickness between 47
    and 99 mm will be satisfactory. In practice ,50 mm
    (2 in) frames would probably be used.")

```

---

**Scilab code Exa 7.7** Increase in the overall throughput of the press

```

1 clear;
2 clc;
3 printf("\n Example 7.7");
4 //Case 1
5
6         //dV/dt = A^2(-deltaP)/vru(V + AL/v
           ) = a/V+b
7         //For constant rate filtration:
8         //Vo/to = a/Vo + b
9         //Vo^2 + bVo = ato
10        //For constant pressure filtration

```

```

11          // 0.5(V^2 - Vo^2)+b(V-Vo)=a(t-to)
12          // to=600s , t-to=3600s , Vo=V/4
13          // V^2/16 +bV/4 = 600a
14          // 0.5(V^2 - V^2/16)+b(V-V/4)=3600a
15          // 3600a = (15/32)V^2 +3/4(bV) =
              3/8(V^2) + 3/2(bV)
16          // b = V/8
17          // a = (V^2/16 + V^2/32)/600 =
              (3/19200)V^2
18          // Total cycle time = 900 + 4200 =
              5100secs
19          // Filtration rate = V/5100 =
              0.000196V
20
21 // Case 2
22          // V1/t1 = a/ (V1 +b/4)=Vo/to=a/(Vo+b
              )
23          // 0.5*(49/64V^2 - V1^2)+b/4(7/8V-V1)
              =a(t-t1)
24          // V/2400 = (3/19200)V^2/(V1+V/32)
25          // t1 = (to/Vo)V1
26          t1 = 600/(1/4)*(11/32);
27          printf("\n t1 = %dsecs",t1);
28          //Substituting gives
29          deltaT = (19200/3)*(784-121+34)/2048;
30          printf("\n t -t1 = %d secs",deltaT);
31          Cycle_time = 180+900+t1+deltaT;
32          printf("\n cycle time = %d secs",Cycle_time);
33          Increase = (0.000214 - 0.000196)/(0.000196)*100;
34          printf("\n Increase in filtration rate is %.1f per
              cent",Increase);

```

---

# Chapter 8

## Membrane Separation Processes

Scilab code Exa 8.2 Area of membrane and average flux

```
1 clear;
2 clc;
3 printf("\n Example 8.2");
4
5 //From the gel polarisation model:
6     //J = (1/A)(dV/dt) = hD ln(Cg/Cf)
7     //Cf = Co(Vo/V)
8     //where Co and Vo are the initial concentration
9     //and volume, respectively and Cf and V are the
10    //values at subsequent times
11    //Combining these eq gives
12    //dV/dt = A(hDln(Cg/Co)-hDln(Vo/V))
13 V = [10 5 3 2 1];
14 y = [9.90 13.64 18.92 27.30 112.40];
15 plot(V,y)
16 xtittle("Area under the curve is 184.4", "Volume(m^3)",
17         "(J - hDln(Vo/V))^( -1)");
```

```

17 //(b)
18 Jo = 0.04*log(250/20);
19 printf("\n Jo = %.3f m/h", Jo);
20 Jf = 0.04*log(250/200);
21 printf("\n Jf = %f m/h", Jf);
22 Jav = Jf + 0.27*(0.101-0.008);
23 printf("\n Jav = %f m/h", Jav);
24 //For the removal of 9m^3 filtrate in 4 hours
25 Area = (9/4)/Jav;
26 printf("\n Area = %fm^2", Area);

```

---

### Scilab code Exa 8.3 Minimum number of membrane modules required

```

1 clear;
2 clc;
3 printf("\n Example 8.3");
4 //It is assumed that Q0 is the volumetric flowrate
   of feed
5 // Q2 the volumetric flowrate of concentrate
6 //C0 the solute concentration in the feed
7 // C2 the solute concentration in the concentrate
8 // F the volumetric flowrate of membrane permeate
9 // A the required membrane area.
10 // It is also assumed that there is no loss of
   solute through the membrane.
11 C1 = 3;
12 while 1
13     C1new = C1 -(0.04-0.02*log(30/C1))/(C1^(-1)/50);
14     if C1new == C1 then
15         break;
16     end
17     C1 = C1new;
18 end
19     printf("\n C1 = %d kg/m^3", C1);
20 printf("\n below this concentration the membrane

```

```

    flux is 0.04 m/h");
21
22 //This does not pose a constraint for the single
    stage as the concentration of solute C2 will be
    that of the final concentrate, 20 kg/m3.
23 //Conservation of solute gives:QoCo = Q2C2
24 //A fluid balance gives : Qo = F + Q2
25 //Combining these eq and substituting Known values:
26 A = (2.438/0.02)/log(30/20);
27 printf("\n A = %d m^2",A);
28 //The tubular membranes to be used are available as
    30 m^2modules.
29 printf("\n the no of required modules are %d ",A/30)
    ;
30
31 //Part(b)
32
33 //Conservation of solute gives = QoCo = Q1C1 = Q2C2
34 //A fluid balance on stage 2 gives Q1 = Q2 + F2
35 //A fluid balance on stage 1 gives Q1 = Q2 +F2
36 //Substituting given values in above eqns
37 //2.5 = 1.25/C1 + 0.02 A1ln(30/C1)
38 function [A1]=a(C1)
39     A1 = (2.5-1.25/C1)/(0.02*log(30/C1));
40     funcprot(0);
41 endfunction
42 function [A2]=b(C1)
43     A2 = (1.25/C1 - 0.0625)/0.00811
44     funcprot(0);
45 endfunction
46
47 printf("\n The procedure is to use trial and error
    to estimate the value of C1 that gives the
    optimum values of A1 and A2");
48 printf("\n If C1 = 5kg/m^3 then A1 = %d m^2 and A2 =
    %d m^2",a(5),b(5));
49 printf("\n an arrangement of 3 modules    1 module
    is required.");

```

```
50 printf("\n\n\n If C1 = 4 kg/m^3 then A1 = %dm^2 and
    A2 = %dm^2",a(4),b(4));
51 printf("\n an arrangement of 2 modules    1 module
    is almost sufficient.");
52 printf("\n\n\n If C1 = 4.5 kg/m^3 then A1 = %dm^2
    and A2 = %d m^2",a(4.5),b(4.5));
53 printf("\n an arrangement of 2 modules    1 module
    which meets the requirement");
54 printf("\n\n This arrangement requires the minimum
    number of modules.");
```

---



# Chapter 9

## Centrifugal Separations

Scilab code Exa 9.1 Value of capacity factor

```
1 clear;
2 clc;
3 printf("\n Example 9.1");
4 d_particle = 5;           //particle size is in um
5 p = 1000;                //density of water in kg/m^3
6 ps = 2800;              //density of solids in kg/m
   ^3
7 viscosity = 10^(-3);    //viscosity is in Ns/m^2
8 uo = ((d_particle*10^(-6))^2)*(ps-p)*9.81/(18*
   viscosity);
9 printf("\n Terminal falling velocity of particles of
   diameter = %.2 f m/sec",uo*10^5);
10 Q = 0.25;              //volumetric flow rate is in
   m^3/sec
11 printf("\n E = %.2 f*10^(4) m^2", (Q/uo)*10^(-4));
12
13 printf("\n For coal-in-oil mixture");
14 uo1 = 0.04/(Q/uo);
15 printf("\n uo = %.2 f*10^-6 m/sec",uo1*10^6);
16
17 d = sqrt((18*0.01*uo1)/((1300-850)*9.81));
```

```
18 printf("\n d = %d um", (d/3)*10^6);
```

---

### Scilab code Exa 9.3 Time taken to produce filtrate

```
1 clear;
2 clc;
3 printf("\n Example 9.3");
4 //In the filter press
5 //  $V^2 + 2(AL/v)V = 2(-\text{deltaP})A^2*t/(ruv)$ 
6
7 l = 0.025;           //l is in meters
8 L = 0.003;           //L is in meters
9 deltaP = 350;        //it is in N/m^2
10 t = 3600;           //t is in secs
11
12 //  $x = v/ru$ 
13 x = poly([0], 'x');
14 x1 = roots(0.025^2 + 2*0.003*0.025 - 2*3.5*10^(5)
15         *3600*x);
16 printf("\n the value of ru/v = %.2f*10^12", (1/x1)
17         *10^(-12));
18
19 //In the centrifuge
20 R = 0.15;           //R is in meters
21 H = 0.20;           //H is in meters
22 V = 0.00225;        //V is in m^3
23 r = poly([0], 'r');
24 r1 = roots(%pi*(R^2 - r^2)*H-V);
25 printf("\n Value of ro = %f mm", r1(1)/2);
26 printf("\n angular frequency = %.1frad/s", (r1(1)
27         /2*10^3)*2*(%pi));
28
29 //  $(R^2 - r^2)(1+2L/R)+2r^2\ln(r/R) = 2vtpw^2/ru(R^2-ro$ 
30 //  $^2)$ 
31 t = poly([0], 't');
```

```
28 t1 = roots((R^2 - r1(1)^2)*(1+2*(L/R))+2*(r1(1)^2)*
    log(r1(1)/R)-2*t*1000*408.4^(2)/(3.25*10^12)*(R
    ^2-(r1(1)/2)^2));
29 printf("\n time required = %f secs",t1);
```

---

# Chapter 10

## Leaching

Scilab code Exa 10.1 Time required for solute to dissolve

```
1 clear
2 clc;
3 printf("\n Example 10.1");
4 //For the pilot scale vessel
5 c = (2.5*75)/100;           //in kg/m^3
6 cs = 2.5;                 //in kg/m^3
7 V = 1.0;                  //V is in m^3
8 t = 10;                   //t is in secs
9
10 //1.875 = 2.5(1-e^(-kA/b*100))
11 // x =kA/b
12 x = -log(1-1.875/2.5)/10;
13
14 //For the full scale vessel
15 c = (500*28/100)/100;
16 printf("\n C = %fkg/m^3",c);
17 cs = 2.5;                 //cs is in kg/m^3
18 V = 100;                  //V is in m^3
19 t = -log(1-1.4/2.5)*(100/0.139); //t is in secs
20 printf("\n t = %d secs",t);
```

---

Scilab code Exa 10.2 Rate of feed of neutral water to the thickeners

```
1 clear;
2 clc;
3 printf("\n Example 10.2");
4
5 //If x1, x2, x3 are the solute: solvent ratios in
   thickeners 1, 2, and 3, respectively, the
   quantities of CaCO3, NaOH, and water in each of
   the streams can be calculated for every 100 kg of
   calcium carbonate
6
7 //Since the final underflow must contain only 1 per
   cent of NaOH
8 function [f]=F(x)
9     f(1)=(300*x(3))/100 - 0.01;
10    f(2)=300*(x(2)-x(3))/x(4) - x(3);           //Wf =
        x(4)
11    f(3)=300*(x(1)-x(3))/x(4) - x(2);
12    f(4)=(80-300*x(3))/(600+x(4))-x(1);
13    funcprot(0);
14 endfunction
15 //An initial guess
16 x = [0.1 0.1 0.1 0.1];
17 y = fsolve(x,F);
18 printf("\n x1 = %f  x2 = %f  x3 = %f Wf = %f",y(1),y
   (2),y(3),y(4));
19
20 printf("\n Thus the amount of water required for
   washing 100 kg CaCO3 is %f kg",y(4));
21 printf("\n The solution fed to reactor contains 0.25
   kg/s Na2CO3.This is equivalent to 0.236 kg/s
   CaCO3,and hence the actual water required is %.2f
   kg/sec",y(4)*0.236/100);
```

---

### Scilab code Exa 10.3 Required number of thickeners

```
1 clear;
2 clc;
3 printf("\n Example 10.3");
4
5 //part 1
6 //Solvent in underflow from final washing thickener
   = 50 kg/s.
7 //The solvent in the overflow will be the same as
   that supplied for washing (200 kg/s).
8 //Solvent discharged in overflow/Solvent discharged
   in underflow= forthe washing thickeners.
9 //Liquid product from plant contains 54.9 kg of salt
   in 195 kg of solvent.
10 //This ratio will be the same in the underflow from
   the first thickener.
11 printf("\n the material fed to the washing
   thickeners consists of 100 kg TiO2, 50 kg solvent
   and %d kg salt",50*(54.9/195));
12
13 //m =n+1
14 m = log(421)/log(4);
15 printf("\n The required number of thickeners for
   washing are %f",m);
16
17 //Part 2
18 //From an inspection of the data, it is seen that Wh
   = 0.30 + 0.2Xh.
19 //Thus: Sh = WhXh = 0.30Xh + 0.2X2h= 5Wh^2      1.5Wh
20 //Considering the passage of unit quantity of TiO2
   through the plant, then:
21 Ln = 0; wn = 2; Xn = 0;
22 //since 200 kg/s of pure solvent is used.
```

```

23 Sn = 0.001;
24 Wn = 0.3007;
25 So = 0.55;
26 Wo = 1.00;
27 //X1 = (Ln+1 + S0      Sn)/(wn+1 + W0      Wn)
28 X1 = (0+0.55-0.001)/(2+1-0.3007);
29 printf("\n concentration in the first thickener is
        %f",X1);
30
31 W1 = 0.30+0.2*0.203;
32 printf("\n W1 = %f",W1);
33 S1 = (0.3406*0.203);
34 printf("\n S1 = %f",S1);
35 X2 = (0.0691 - 0.001)/(1.7 + 0.3406);
36 printf("\n X2 = %f",X2);
37 W2 = (0.30+0.2*0.0334);
38 printf("\n W2 = %f",W2);
39 X3 = (0.01025-0.001)/(1.7+0.3067);
40 printf("\n X3 = %f",X3);
41 printf("\n W3 = 0.30089");
42 printf("\n S3 = %f",0.0013);
43 printf("\n W4 = %f and S4 = %f",0.30003,0.000045);
44 printf("\n Thus S4 is less than Sn and therefore 4
        thickeners are required.");

```

---

#### Scilab code Exa 10.4 Number of ideal stages required

```

1 clear;
2 clc;
3 printf("\n Example 10.4");
4 //Since the seeds contain 20 per cent of oil
5 xAo=0.2;
6 xBo=0.8;
7 printf("\n xAo = %.1f and xBo = %.1f",xAo,xBo);
8 //The final solution contains 50 per cent oil

```

```

9  yA1=0.5;
10 yS1=0.5;
11 printf("\n yA1 = %.1f and yS1 = %.1f",yA1,yS1);
12 //The solvent which is used for extraction is pure
    and hence
13 ySn1=1;
14 //1 kg of insoluble solid in the washed product is
    associated with 0.5 kg of solution and 0.025 kg
    oil.
15 xAn = 0.0167;
16 xBn = 0.6667;
17 xSn = 0.3166;
18
19 //The mass fraction of insoluble material in the
    underflow is constant and equal to 0.667. The
    composition of the underflow is therefore
    represented, on the diagram Figure 10.22, by a
    straight line parallel to the hypotenuse of the
    triangle with an intercept of 0.333 on the two
    main axes.
20
21 //The difference point is now found by drawing in
    the two lines connecting x0 and y1 and xn and yn
    +1.
22
23 //The graphical construction described in the text
    is then used and it is seen from Figure 10.22
    that xn lies in between x4 and x5.
24 printf("\n Thus 5 thickeners are adequate and for
    the required degree of extraction");

```

---

#### Scilab code Exa 10.5 Number of theoretical stages required

```

1  clear;
2  clc;

```



```

3 printf("\n Example 10.5");
4
5 //On the basis of 100 kg untreated solids
6 //In the underflow feed:
7 //0.35 kg oil is associated with each kg of
  exhausted livers .
8 printf("\n mass of livers fed=%d kg containing %d kg
  oil" ,100/(1+0.35) ,100-74);
9 xA = 0.26;
10 printf("\n xA = %.2 f" ,xA);
11 xs = 0;
12 //In the overflow feed , pure ether is used
13 ys =1.0;
14 xs = 0;
15 //Recovery = 90 per cent
16 printf("\n
  Oil          Ether")          Exhausted livers
17 printf("\n Underflow feed          %d
          %d          -" ,74 ,26);
18 printf("\n Overflow feed          -
          -          %d" ,50);
19 printf("\n Underflow product          %d
          %d          e(say)" ,74 ,2.6);
20 printf("\n Overflow product          -
          %d          50-e" ,23.4);
21 printf("\n In the underflow product:");
22 printf("\n the ratio(oil/exhausted livers) = %.3 f kg
  /kg" ,2.6/74);
23 printf("\n Ratio(ether/exhausted livers) = %.3 f kg/
  kg" ,0.306);
24 printf("\n e = %.1 f kg" ,0.306*74);
25 //In the overflow product:
26 printf("\n The mass of ether = %.1 f kg" ,50-22.6);
27 printf("\n yA = %.2 f" ,23.4/(23.4+27.4));
28 printf("\n ys = %.2 f" ,1-23.4/(23.4+27.4));
29 printf("\n ");

```

---

# Chapter 11

## Distillation

Scilab code Exa 11.1 Mole fraction calculation

```
1 //Example 11.1
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.1\n");
7
8 p0_A=106; //Vapour pressure of n-heptane in kN/m^2
9 p0_B=73.7; //Vapour pressure of toluene in kN/m^2
10 P=101.3; //Total pressure in kN/m^2
11
12 xA=(P-p0_B)/(p0_A-p0_B);
13
14 yA=p0_A*xA/P;
15
16 printf("\nMole fraction in liquid phase is : %.3f",
17        xA);
18 printf("\nMole fraction in vapour phase is : %.3f\n",
19        yA);
20
21 //End
```

---

**Scilab code Exa 11.2** Saturated Pressure calculation

```
1 //Example 11.2
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.2\n");
7
8 Pc=4700; //Critical pressure in kN/m^2
9 Tc=508.1; //critical temperature in K
10 p1=100.666; //in kN/m^2
11 T1=329.026; //in K
12 T=350.874; //in K
13
14 Tr1=T1/Tc;
15 Pr1=p1/Pc;
16 printf("\nTr1 = %f \nPr1 = %f\n",Tr1,Pr1);
17
18 c5=-35+(36/Tr1)+(42*log(Tr1))-(Tr1^6);
19 c2=((0.315*c5)-log(Pr1))/((0.0838*c5)-log(Tr1));
20 c1=0.0838*(3.758-c2);
21 printf("\nc5 = %.4f \nc2 = %.4f \nc1 = %.4f\n",c5,c2
    ,c1);
22
23 k9=-35*c1;
24 k10=-36*c1;
25 k11=(42*c1)+c2;
26 k12=-c1;
27
28 printf("\nk9 = %.3f \nk10 = %.3f \nk11 = %.4f",k9,
    k10,k11);
29 printf("\nk12 = %.5f\n",k12);
30
```

```

31 Tr=T/Tc;
32 Pr=exp(k9-(k10/Tr)+(k11*log(Tr))+(k12*Tr^6));
33 p0=Pc*Pr;
34 printf("\nPr = %f \n\nP0 = %.2 f kN/m^2\n",Pr ,p0);
35
36 //End

```

---

### Scilab code Exa 11.3 Vapour phase composition of a mixture

```

1 //Exaple 11.3
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.3\n");
7
8 k1_B=6.90565;
9 k2_B=1211.033;
10 k3_B=220.79;
11
12 k1_T=6.95334;
13 k2_T=1343.943;
14 k3_T=219.377;
15
16 t=338-273; //in Degree celsius
17 P=101.3; //in kN/m^2
18 xB=0.5;
19 xT=0.5;
20
21 function [p0]=antoine(k1,k2,k3,T)
22     p0=10^(k1-(k2/(T+k3)));
23     funcprot(0)
24 endfunction
25
26 p0_B=antoine(k1_B,k2_B,k3_B,t)*101.325/760;

```

```

27 p0_T=antoine(k1_T,k2_T,k3_T,t)*101.325/760;
28
29 printf("\nP0_B = %.1 f kN/m^2\nP0_T = %.1 f kN/m^2\n",
        p0_B,p0_T);
30
31 pB=xB*p0_B;
32 pT=xT*p0_T;
33 printf("\npB = %.2 f kN/m^2 \npT = %.3 f\n",pB,pT);
34
35 p=pB+pT;
36 yB=pB/p;
37 yT=pT/p;
38 printf("\nyB = %.3 f \nyT = %.3 f\n",yB,yT);
39
40 //End

```

---

#### Scilab code Exa 11.4 Boiling point of equimolar mixture

```

1 //Example 11.4
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.4\n");
7
8 function [p0]=antoine(k1,k2,k3,T)
9     p0=10^(k1-(k2/(T+k3-273)));
10     funcprot(0)
11 endfunction
12
13 k1_B=6.90565;
14 k2_B=1211.033;
15 k3_B=220.79;
16
17 k1_T=6.95334;

```

```

18 k2_T=1343.943;
19 k3_T=219.377;
20
21 xB=0.5;
22 xT=0.5;
23
24 printf("\n\tT(K)");
25 T=[373 353 363 365 365.1];
26 disp(T);
27
28 i=1;
29
30 while i<6
31     p0_B(i)=antoine(k1_B,k2_B,k3_B,T(i))
           *101.325/760;
32     p0_T(i)=antoine(k1_T,k2_T,k3_T,T(i))
           *101.325/760;
33     pB(i)=xB*p0_B(i);
34     pT(i)=xT*p0_T(i);
35     p(i)=pB(i)+pT(i);
36     i=i+1;
37 end
38 printf("\n\tp0 B");
39 disp(p0_B);
40 printf("\n\tp0 T");
41 disp(p0_T);
42 printf("\n\tpB");
43 disp(pB);
44 printf("\n\tpT");
45 disp(pT);
46 printf("\n\tpB+pT");
47 disp(p);
48
49 //since total pressure at 365.1 K is nearly same as
   101.3 kPa
50 printf("\nBoiling temperature is %f K",T(5));
51
52 //End

```

---

**Scilab code Exa 11.5** Dew point of a equimolar mixture

```
1 //Example 11.5
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.5\n");
7
8 function [p0]=antoine(k1,k2,k3,T)
9     p0=10^(k1-(k2/(T+k3-273)));
10     funcprot(0)
11 endfunction
12
13 k1_B=6.90565;
14 k2_B=1211.033;
15 k3_B=220.79;
16
17 k1_T=6.95334;
18 k2_T=1343.943;
19 k3_T=219.377;
20
21 //Since total pressure is 101.3 kPa, pB=pT
22 pB=50.65;
23 pT=50.65;
24
25 printf("\n\tT(K)");
26 T=[373.2 371.2 371.7 371.9 372];
27 disp(T);
28
29 i=1;
30
31 while i<6
32     p0_B(i)=antoine(k1_B,k2_B,k3_B,T(i))
```

```

        *101.325/760;
33     p0_T(i)=antoine(k1_T,k2_T,k3_T,T(i))
        *101.325/760;
34     xB(i)=pB/p0_B(i);
35     xT(i)=pT/p0_T(i);
36     x(i)=xB(i)+xT(i);
37     i=i+1;
38 end
39
40 printf("\n\tp0 B")
41 disp(p0_B);
42 printf("\n\tp0 T");
43 disp(p0_T);
44 printf("\n\txB");
45 disp(xB);
46 printf("\n\txT");
47 disp(xT);
48 printf("\n\tx = xB+xT");
49 disp(x);
50
51 //Since last value is closer to 1, 372 K is the
    required dew point
52 printf("\nDew point can be taken as %f K",T(5));
53
54 //End

```

---

### Scilab code Exa 11.6 Composition of vapour and liquid

```

1 //Example 11.6
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.6\n");
7

```



```

8 //Fractional vapourisation    f=V/F
9 f=0.25;
10
11 slope=(1-f)/-f;
12
13 //from fig 11.9, a construction is made
14 printf("\nFor slope = %d",slope);
15 printf("\nx=0.42 \t y=0.63\n");
16
17 //from fig. 11.10,temperature corresponding to x
    =0.42 is noted
18 printf("\nfor x=0.42 \t T=366.5");
19
20 //End

```

---

#### Scilab code Exa 11.7 Number of theoretical plates needed

```

1 clear all;
2 clc;
3 printf('Example 11.7'); //Example 11.7
4 // Find Number of theoretical plates needed and the
    position of entry for the feed
5
6 F = 100; //Feed [kmol]
7
8 function [f]=Feed(x)
9     f(1)=x(1)+x(2)-100; //Overall
    mass Balance
10     f(2)=0.9*x(1)+.1*x(2)-(100*.4); //A
    balance on MVC, benzene
11     funcprot(0)
12 endfunction
13 x = [50 50];
14 product = fsolve(x,Feed);
15

```

```

16 //Using notation of figure 11.13
17 Ln = 3*product(1);
18 Vn = Ln + product(1);
19
20 //Reflux to the plate
21 Lm = Ln + F;
22 Vm = Lm - product(2);
23
24 //Equilibrium Composition
25 xt = .79;    yt = .9;
26 //From Top eqm line
27 yt1 = (Ln/Vn)*xt + (product(1)/Vn);
28 xt1=.644;    //Thus from Eqm curve for yt1
29 //From Top eqm line
30 yt2 = (Ln/Vn)*xt1 + (product(1)/Vn);
31 xt2=.492;    //Thus from Eqm curve for yt2
32 //From Top eqm line
33 yt3 = (Ln/Vn)*xt2 + (product(1)/Vn);
34 xt3=.382;    //Thus from Eqm curve for yt3
35 //From II Eqm Line
36 yt4 = (Lm/Vm)*xt3 - (product(2)/Vm)*.1;
37 xt4=.2982;    //Thus from Eqm curve for yt4
38 //From II Eqm Line
39 yt5 = (Lm/Vm)*xt4 - (product(2)/Vm)*.1;
40 xt5=.208;    //Thus from Eqm curve for yt5
41 //From II Eqm Line
42 yt6 = (Lm/Vm)*xt5 - (product(2)/Vm)*.1;
43 xt6=.120;    //Thus from Eqm curve for yt6
44 //From II Eqm Line
45 yt7 = (Lm/Vm)*xt6 - (product(2)/Vm)*.1;
46 xt7=.048;    //Thus from Eqm curve for yt7
47
48 //Equilibrium Data
49 y=[0 yt7 yt6 yt5 yt4 yt3 yt2 yt1 yt];
50 x=[0 xt7 xt6 xt5 xt4 xt3 xt2 xt1 xt];
51 //Top Equilibrium Line equation 11.35
52 x1 = linspace(0, .79, 100);
53 y1 = (Ln/Vn)*x1 + (product(1)/Vn);

```

```

54 //Equilibrium Line equation 11.37
55 x2 = linspace(0.048, .44, 100);
56 y2 = (Lm/Vm)*x2 - (product(2)/Vm)*.1;
57 clf();
58 plot(x,y,x1,y1,x2,y2);
59 xtitle("Lewis-Sorel Method", "Mole fraction of C6H6
        in Liquid (x)", "Mole Fraction C6H6 in Vapor (y)"
        );
60 legend("Equilirium Plot", "Top Eqm Line", "Bottom
        Eqm Line");
61 printf("\n\n As the least point on equilibrium Line
        xt=7 correspond to reboiler, and there will be
        seven plates");
62
63
64 //END

```

---

#### Scilab code Exa 11.8 The Mc Cabe Thiele method

```

1 clear all;
2 clc;
3 printf('Example 11.8 '); //Example 11.8
4 // Find Number of theoretical plates needed and the
   position of entry for the feed by mccabe thiele
   method
5
6 F = 100; //Feed [kmol]
7
8 function [f]=Feed(x)
9     f(1)=x(1)+x(2)-100; //Overall
   mass Balance
10    f(2)=0.9*x(1)+.1*x(2)-(100*.4); //A
   balance on MVC, benzene
11    funcprot(0)
12 endfunction

```

```

13 x = [50 50];
14 product = fsolve(x,Feed);
15
16 //Using notation of figure 11.13
17 Ln = 3*product(1);
18 Vn = Ln + product(1);
19
20 //Reflux to the plate
21 Lm = Ln + F;
22 Vm = Lm - product(2);
23
24 //Equilibrium Data
25 y=[0 .127 .252 .379 .498 .594 .708 .818 .9 1];
26 x=[0 .048 .12 .208 .298 .382 .492 .644 .79 1];
27 //Diagnol Line
28 y3 = [0 1];
29 x3 = [0 1];
30 //Top Equilibrium Line equation 11.35
31 x1 = linspace(0,.985,100);
32 y1 = (Ln/Vn)*x1 + (product(1)/Vn);
33 //Equilibrium Line equation 11.37
34 x2 = linspace(0.048,.44,100);
35 y2 = (Lm/Vm)*x2 - (product(2)/Vm)*.1;
36 clf();
37 //Setting initial point A x = .985 at top eqm line
38 xm = [.985 .965 .965 .92 .92 .825 .825 .655 .655 .44
        .44 .255 .255 .125 .125 .048];
39 ym = [.985 .985 .965 .965 .92 .92 .825 .825 .655
        .655 .44 .44 .255 .255 .125 .125];
40 xp = [.985 .965 .92 .825 .655 .44 .255 .125 .048];
41 yp = [.985 .965 .92 .825 .655 .44 .255 .125 .048];
42 plot(x,y,x3,y3,x1,y1,x2,y2,xm,ym);
43 xtitle("McCabe Thiele Method", "Mole fraction of
        C6H6 in Liquid (x)", "Mole Fraction C6H6 in Vapor
        (y)");
44 legend("Equilibrium Plot", "Diagnol Line", "Top Eqm
        Line", "Bottom Eqm Line",5);
45 xset('window',1);

```

```

46 for(i=2:8)
47 plot(xp(i),yp(i),"o-");
48 xtitle("Equilibrium plot","mole fraction C6H6 in
         liquid(x)","mole fractionC6H6 in vapour(y)");
49 end
50 printf("\n\n The Number of stages are then counted
         highlighted points that is number of plates
         required as 7");
51
52
53 //END

```

---

**Scilab code Exa 11.9** Number of plates required at total reflux

```

1 //Example 11.9
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.9\n");
7
8 xA_d=0.9;
9 xA_s=0.1;
10 xB_d=1-xA_d;
11 xB_s=1-xA_s;
12
13 a=2.4;
14 xd=0.9;
15 xf=0.4;
16 xw=0.1;
17
18 n=(log(xA_d*xB_s/(xB_d*xA_s))/log(a))-1;
19 printf("\nNo. of theoretical plates in column is :
         %d",n);
20

```

```

21 Rm=((xd/xf)-(a*((1-xd)/(1-xf))))/(a-1);
22 printf("\nMinimum reflux ratio is %.2f\n",Rm);
23
24 yf=0.61;
25 printf("\nUsing Graphical construction with yf=0.61\n
n");
26 Rmin=(xd-yf)/(yf-xf);
27 printf("Minimum reflux ratio is %.2f\n",Rmin);
28
29 //End

```

---

#### Scilab code Exa 11.10 Heat required

```

1 //Example 11.10
2
3 clear;
4 clc;
5
6 printf("\tExample 11.10\n");
7
8 //From material balance
9 // D+W=1
10 // 0.995D+0.1W=1*3
11
12 A=[1 1;0.995 0.1];
13 B=[1;3];
14 Rm = (1952-1547)/(1547-295);
15 printf("\n Rm = %.3f",Rm);
16 NA = 1.08*405;
17 printf("\n Since the actual reflux is 8 pre cent
above the minimum NA = 1.08*NmA = %.3f",NA);
18 N = 5/0.6;
19 printf("\n Number of plates to be required are %.3f
",5/0.6);
20

```

```

21 Qb_W = 582 - (-209);
22 printf("\n Heat input to the boiler per unit mass of
    bottom product is %.3f", Qb_W);
23 printf("\n Heat input to the boiler = %.3f kW"
    ,791*0.78);
24 printf("\n Condenser duty = %d kW" ,(1984-296)*0.22);

```

---

### Scilab code Exa 11.11 Number of theoretical stages required

```

1 //Example 11.11
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.11\n");
7
8 // F is feed
9 // D is distillate
10 // W is waste
11 // S is sidestream
12
13 F=100;
14 S=10;
15
16 //Mass fractions of CCl4 in various streams
17 xf=0.5;
18 xd=0.95;
19 xw=0.05;
20 xs=0.8;
21
22 // D + W = 100-10
23 // 0.95D + 0.00W = 50-8
24 A=[1, 1; 0.95, 0.05];
25 B=[90; 42];
26 x=inv(A)*B;

```

```

27 printf("\nD = %.1f      W = %.1f\n",x(1),x(2));
28
29 disp("From the enthalpy data and the reflux ratio ,
      the upper pole point M is located as shown in
      Figure.");
30
31 disp("Points F and S are located ,such that FS/FF =
      10.");
32
33 disp("MF is Joined and extended to cut NS^A at O,
      the immediate pole point.");
34 disp("The number of stages required is then obtained
      from the figure and");
35 printf("13 theoretical stages are required");
36
37 //End

```

---

#### Scilab code Exa 11.12 Amount of distillate

```

1 clear;
2 clc;
3 printf("\n Example 11.12");
4 R = [0.85 1.0 1.5 2.0 3.0 4.0]; //Reflux ratio
5 xd = 0.75; //top
      concentration of alcohol
6 xs = [0.55 0.50 0.37 0.20 0.075 0.05]; //From the
      graph fig.11.35 page-596
7 Db(1) = 0;
8
9 printf("\n      R          Fi          xs
      Db      ");
10 i=1;
11 while i<=6
12     Fi(i) = xd/(R(i) + 1);
13     if i>1 then

```



```

14     Db(i) = 100*(xs(1)-xs(i))/(xd-xs(i));
15     end
16     printf("\n %.2 f          %.3 f          %.2 f          %
        .1 f",R(i),Fi(i),xs(i),Db(i));
17     i=i+1;
18 end
19 plot(R,Db);
20 xtitle("","Reflux ratio(R)","Product Db (kmol)");
21 printf("\n The area under the Db vs R curve is given
        by 96 kmol");
22 Hav = 4000;          //average latent heat in kJ/
        kmol
23 Qr = 96*Hav/1000;
24 printf("\n Heat to be supplied to provide the reflux
        ,Qr is approximately %.1 f MJ",Qr);
25 printf("\n Heat to be supplied to provide the reflux
        per kmol of product is then %.2 f MJ",380/71.4);
26 printf("\n Total heat = %.2 f MJ/kmol product"
        ,5.32+4.0);

```

---

### Scilab code Exa 11.13 Heat required and average composition

```

1 //Example 11.13
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.13\n");
7
8 xs=[0.55;0.5;0.425;0.31;0.225;0.105];
9 xd=[0.78;0.775;0.77;0.76;0.75;0.74];
10 differ=xd-xs;
11 for i=1:6
12     reci(i)=1/(xd(i)-xs(i));
13 end

```

```

14
15 m=[xs xd differ reci];
16 printf("\n      xs      xd      (xd-xs)      1/(xd-xs)\n
      ");
17 disp(m);
18 plot(xs,reci);
19 xtitle('Graphical integration ','xs ','1/xd-xs'));
20
21 //Area under the curve is calculated and is found to
      be 1.1
22
23 //logdiv = S1/S2 = area under the curve
24 logdiv=1.1;
25 S1=100;          //Assume
26 div=exp(logdiv);
27 S2=S1/div;
28 Db=S1-S2;       //Product obtained
29 amt=xs(1)*S1-(xs(6)*S2);
30 avg=amt/Db;
31
32 printf("\nAverage composition is %.2f kmol\n",avg);
33
34 L=4000;         //latent heat
35 R=2.1;
36
37 h=R*L;
38 printf("Heat required to produce reflux per kmol :
      %d kJ\n",h);
39
40 //End

```

---

#### Scilab code Exa 11.14 Ideal plates required

```

1 clear;
2 clc;

```

```

3 xdo = 0.98;           //per cent of ortho top
  product
4 xwo = 0.125;         //per cent of ortho bottom
  product
5
6 function [f]=product(x)
7     f(1) = 100 - x(1) - x(2);           //x(1) is D and x
      (2) is W
8     f(2) = 60 - x(1)*xdo - x(2)*xwo;
9     funcprot(0);
10 endfunction
11 x = [0,0];
12 y = fsolve(x,product)
13 printf("\n D = %.2f kmol & W = %.2f kmol",y(1),y(2))
    ;
14
15 printf("\n Let us assume that the distillate
    contains 0.6 mole per cent meta and 1.4 mole per
    cent para");
16 printf("\n Component          Feed
          Distillate
          Bottoms          ");
17 printf("\n          (kmol)    (mole per cent)    (
    kmol)    (mole per cent)    (kmol)    (mole per cent
    ) ");
18 printf("\n Ortho  %.3f      %.2f          %.2f
          %.2f          %.2f          %.2f
          ",60,60,y(1)*0.98,98,y(2)*0.125,12.5);
19 printf("\n Meta  %.3f      %.2f
          %.2f          %.2f
          %.2f          %.2f          "
    ,4,4,y(1)*0.006,0.6,y(2)*0.083,8.3);
20 printf("\n Para  %.3f      %.2f          %
    .2f          %.2f          %.2f          %.2
    f          ",36,36,y(1)*0.014,1.4,y(2)*0.792,79.2);
21
22 ao = 1.7;           //relative volatility of ortho
    relative to para

```

```

23 am = 1.16;           //relative volatility of meta
    relative to para
24 ap =1;              //relative volatility of para
    w.r.t. to itself
25 xso = 0.125;
26 xsm = 0.083;
27 xsp = 0.792;
28 xwo = 0.125;
29 xwp = 0.083;
30 xwm = 0.792;
31 yso = ao*xso/(ao*xso+ap*xsp+am*xsm);
32 ysm = am*xsm/(ao*xso+ap*xsp+am*xsm);
33 ysp = ap*xsp/(ao*xso+ap*xsp+am*xsm);
34 //Equations of operating lines
35 //Above the feed point
36 Ln = 5*y(1);        //Liquid downflow
37 Vn = 6*y(1);        //Vapour up
38 //Assuming the feed is liquid at its boiling point
39 F = 100;            //feed
40 Lm = Ln+F;          //liquid downflow
41 Vm = Lm-y(2);       //Vapour up
42 x1o = poly([0], 'x1o');
43 x11 = roots(yso - (Lm/Vm)*x1o + (y(2)/Vm)*xwo);
44 x1p = poly([0], 'x1p');
45 x12 = roots(ysp - (Lm/Vm)*x1p + (y(2)/Vm)*xwp);
46 x1m = poly([0], 'x1m');
47 x13 = roots(ysm - (Lm/Vm)*x1m + (y(2)/Vm)*xwm);
48 x1 = [x11 x13 x12];
49 ax1 = [ao*x11 am*x13 ap*x12];
50 y1 = [ax1(1)/(ax1(1)+ax1(2)+ax1(3)) ax1(2)/(ax1(1)+
    ax1(2)+ax1(3)) ax1(3)/(ax1(1)+ax1(2)+ax1(3))];
51 x2o = poly([0], 'x2o');
52 x21 = roots(y1(1) - (Lm/Vm)*x2o + (y(2)/Vm)*xwo);
53 x2p = poly([0], 'x2p');
54 x22 = roots(y1(3) - (Lm/Vm)*x2p + (y(2)/Vm)*xwp);
55 x2m = poly([0], 'x2m');
56 x23 = roots(y1(2) - (Lm/Vm)*x2m + (y(2)/Vm)*xwm);
57 x2 = [x21 x23 x22];

```

```

58 printf("\n          plate compositions below the
      feed plate");
59 printf("\n Component          xs
      axs          ys          x1
      ax1          y1          x2");
60 printf("\n          o          %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xso, ao*xso, yso, x1(1)
      , ax1(1), y1(1), x2(1));
61 printf("\n          m          %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xsm, am*xsm, ysm, x1(2)
      , ax1(2), y1(2), x2(2));
62 printf("\n          p          %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xsp, ap*xsp, ysp, x1(3)
      , ax1(3), y1(3), x2(3));
63 printf("\n          %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xso+xsm+xsp, ao*xso+
      am*xsm+ap*xsp, yso+ysm+ysp, x1(1)+x1(2)+x1(3), ax1
      (1)+ax1(2)+ax1(3), y1(1)+y1(2)+y1(3), x2(1)+x2(2)+
      x2(3));
64
65 ax2 = [ao*x2(1) am*x2(2) ap*x2(3)];
66 y2 = [ax2(1)/(ax2(1)+ax2(2)+ax2(3)) ax2(2)/(ax2(1)+
      ax2(2)+ax2(3)) ax2(3)/(ax2(1)+ax2(2)+ax2(3))];
67 x3o = poly([0], 'x3o');
68 x31 = roots(yso - (Lm/Vm)*x3o + (y(2)/Vm)*xwo);
69 x3p = poly([0], 'x3p');
70 x32 = roots(ysp - (Lm/Vm)*x3p + (y(2)/Vm)*xwp);
71 x3m = poly([0], 'x3m');
72 x33 = roots(ysm - (Lm/Vm)*x3m + (y(2)/Vm)*xwm);
73 x3 = [x31 x33 x32];
74
75 ax3 = [ao*x3(1) am*x3(2) ap*x3(3)];
76 y3 = [ax3(1)/(ax3(1)+ax3(2)+ax3(3)) ax3(2)/(ax3(1)+
      ax3(2)+ax3(3)) ax3(3)/(ax3(1)+ax3(2)+ax3(3))];

```

```

77 x4o = poly([0], 'x4o');
78 x41 = roots(yso - (Lm/Vm)*x4o + (y(2)/Vm)*xwo);
79 x4p = poly([0], 'x4p');
80 x42 = roots(yso - (Lm/Vm)*x4p + (y(2)/Vm)*xwp);
81 x4m = poly([0], 'x4m');
82 x43 = roots(yso - (Lm/Vm)*x4m + (y(2)/Vm)*xwm);
83 x4 = [x41 x43 x42];
84
85 ax4 = [ao*x4(1) am*x4(2) ap*x4(3)];
86 y4 = [ax4(1)/(ax4(1)+ax4(2)+ax4(3)) ax4(2)/(ax4(1)+
      ax4(2)+ax4(3)) ax4(3)/(ax4(1)+ax4(2)+ax4(3))];
87 x5o = poly([0], 'x5o');
88 x51 = roots(yso - (Lm/Vm)*x5o + (y(2)/Vm)*xwo);
89 x5p = poly([0], 'x5p');
90 x52 = roots(yso - (Lm/Vm)*x5p + (y(2)/Vm)*xwp);
91 x5m = poly([0], 'x5m');
92 x53 = roots(yso - (Lm/Vm)*x5m + (y(2)/Vm)*xwm);
93 x5 = [x51 x53 x52];
94
95 ax5 = [ao*x5(1) am*x5(2) ap*x5(3)];
96 y5 = [ax5(1)/(ax5(1)+ax5(2)+ax5(3)) ax5(2)/(ax5(1)+
      ax5(2)+ax5(3)) ax5(3)/(ax5(1)+ax5(2)+ax5(3))];
97 x6o = poly([0], 'x6o');
98 x61 = roots(yso - (Lm/Vm)*x6o + (y(2)/Vm)*xwo);
99 x6p = poly([0], 'x6p');
100 x62 = roots(yso - (Lm/Vm)*x6p + (y(2)/Vm)*xwp);
101 x6m = poly([0], 'x6m');
102 x63 = roots(yso - (Lm/Vm)*x6m + (y(2)/Vm)*xwm);
103 x6 = [x61 x63 x62];
104
105 ax6 = [ao*x6(1) am*x6(2) ap*x6(3)];
106 y6 = [ax6(1)/(ax6(1)+ax6(2)+ax6(3)) ax6(2)/(ax6(1)+
      ax6(2)+ax6(3)) ax6(3)/(ax6(1)+ax6(2)+ax6(3))];
107 x7o = poly([0], 'x7o');
108 x71 = roots(yso - (Lm/Vm)*x7o + (y(2)/Vm)*xwo);
109 x7p = poly([0], 'x7p');
110 x72 = roots(yso - (Lm/Vm)*x7p + (y(2)/Vm)*xwp);
111 x7m = poly([0], 'x7m');

```

```

112 x73 = roots(ysm - (Lm/Vm)*x7m + (y(2)/Vm)*xwm);
113 x7 = [x71 x73 x72];
114 printf("\n Component          ax2
      y2              x3          ax3
                        y3              x4          ax4");
115 printf("\n      o              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", ax2(1), y2(1), x3(1),
      ax3(1), y3(1), x4(1), ax4(1));
116 printf("\n      m              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xsm, am*xsm, ysm, x1(2)
      , ax1(2), y1(2), x2(2));
117 printf("\n      p              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", xsp, ap*xsp, ysp, x1(3)
      , ax1(3), y1(3), x2(3));
118
119 printf("\n Component          y4
      x5              ax5          y5
                        x6              ax6          y6");
120 printf("\n      o              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(1), x5(1), ax5(1),
      y5(1), x6(1), ax6(1), y6(1));
121 printf("\n      m              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(2), x5(2), ax5(2),
      y5(2), x6(2), ax6(2), y6(2));
122 printf("\n      p              %.3 f
      %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(3), x5(3), ax5(3),
      y5(3), x6(3), ax6(3), y6(3));
123
124
125 ax7 = [ao*x7(1) am*x7(2) ap*x7(3)];
126 y7 = [ax7(1)/(ax7(1)+ax7(2)+ax7(3)) ax7(2)/(ax7(1)+
      ax7(2)+ax7(3)) ax7(3)/(ax7(1)+ax7(2)+ax7(3))];

```

```

127 x8o = poly([0], 'x8o');
128 x81 = roots(yso - (Ln/Vn)*x8o + (y(2)/Vn)*xwo);
129 x8p = poly([0], 'x8p');
130 x82 = roots(ySp - (Ln/Vn)*x8p + (y(2)/Vn)*xwp);
131 x8m = poly([0], 'x8m');
132 x83 = roots(ySm - (Ln/Vn)*x8m + (y(2)/Vn)*xwm);
133 x8 = [x81 x83 x82];
134
135 ax8 = [ao*x8(1) am*x8(2) ap*x8(3)];
136 y8 = [ax8(1)/(ax8(1)+ax8(2)+ax8(3)) ax8(2)/(ax8(1)+
        ax8(2)+ax8(3)) ax8(3)/(ax8(1)+ax8(2)+ax8(3))];
137 x9o = poly([0], 'x9o');
138 x91 = roots(yso - (Ln/Vn)*x9o + (y(2)/Vn)*xwo);
139 x9p = poly([0], 'x9p');
140 x92 = roots(ySp - (Ln/Vn)*x9p + (y(2)/Vn)*xwp);
141 x9m = poly([0], 'x9m');
142 x93 = roots(ySm - (Ln/Vn)*x9m + (y(2)/Vn)*xwm);
143 x9 = [x91 x93 x92];
144
145 printf("\n Component          x7
        ax7          y7          x8
        ax8          y8          x9");
146 printf("\n
        o          %.3f          %.3f          %.3f          %
        .3f          %.3f          %.3f", x7(1), ax7(1), y7(1),
        x8(1), ax8(1), y8(1), x9(1));
147 printf("\n
        m          %.3f          %.3f          %.3f          %
        .3f          %.3f          %.3f", x7(2), ax7(2), y7(2),
        x8(2), ax8(2), y8(2), x9(2));
148 printf("\n
        p          %.3f          %.3f          %.3f          %
        .3f          %.3f          %.3f", x7(3), ax7(3), y7(3),
        x8(3), ax8(3), y8(3), x9(3));
149
150 printf("\n Component          x7
        ax7          y7          x8
        ax8          y8          x9");

```



```

151 printf("\n      o          %.3 f
          %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(1), x5(1), ax5(1),
      y5(1), x6(1), ax6(1), y6(1));
152 printf("\n      m          %.3 f
          %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(2), x5(2), ax5(2),
      y5(2), x6(2), ax6(2), y6(2));
153 printf("\n      p          %.3 f
          %.3 f          %.3 f          %.3 f          %
      .3 f          %.3 f          %.3 f", y4(3), x5(3), ax5(3),
      y5(3), x6(3), ax6(3), y6(3));

```

---

#### Scilab code Exa 11.15 Minimum reflux ratio

```

1 clear all;
2 clc;
3 printf("\n Example 11.15");
4
5 printf("\n Dew point calculation");
6 xd1 = 0.975;          //n-C4 light distillate
7 xd2 = 0.025;          //C5 heavy key distillate
8 Td = 344;             //temperature in kelvins
9 K1 = 1.05;            //Equilibrium constant
      calculation for n-C4 at 344 K
10 K2 = 0.41;           //Equilibrium constant
      calculation for C5 at 344K
11 //By a dew point calculation
12 //sum(xd)=sum(xd/K)
13 printf("\n Component      xd          Td = 344
      K");
14 printf("\n          K
      xd/K ");
15 printf("\n n-C4          %.3 f          %.2 f
      %.3 f", xd1, K1, xd1/K1);

```

```

16 printf("\n C5           %.3 f           %.2 f
    %.3 f", xd2, K2, xd2/K2);
17 printf("\n           %.1 f
    %.3 f", xd1+xd2, (xd1/K1)+(xd2/K2));
18
19 K11 = 1.04;
20 K21 = 0.405;
21 //Calculation for xd at 343 K
22 x = poly([0], 'x');
23 x1 = roots(x/K11 + (1-x)/K21);
24 printf("\n
    Td = 343 K");
25 printf("\n
    K    xd/K");
26 printf("\n
    %.3 f    %.3 f", K11, x1/K11);
27 printf("\n
    %.3 f    %.3 f", K21, (1-x1)/K21);
28 printf("\n
    %.3 f", x1/K11+(1-x1)/K21);
29 printf("\n At 343 K    K1 = 1.04    K2 = 0.405
    from fig.11.39");
30
31 printf("\n\n\n Estimation of still temperature Ts");
32 //sum(xw) = sum(K*xw)
33 K31 = 3.05;           //equillibrium const at 419 K
34 K32 = 1.6;           //equillibrium const at 419 K
35 K33 = 0.87;          //equillibrium const at 419 K
36 K34 = 0.49;          //equillibrium const at 419 K
37 xw1 = 0.017;         //mass fraction of n-C4
38 xw2 = 0.367;         //mass fraction of C5
39 xw3 = 0.283;         //mass fraction of C6
40 xw4 = 0.333;         //mass fraction of C7

```

```

41
42 printf("\n At Ts = 416 K equilibrium constants are
      from fig.11.39");
43 printf("\n n-C4          C5          C6          C7")
      ;
44 printf("\n  %.2f          %.2f          %.2f          %.2f"
      ,K31 ,K32 ,K33 ,K34);
45 printf("\n\n      at Ts = 416 K");
46 printf("\n n-C4          C5          C6          C7")
      ;
47 printf("\n  %.3f          %.3f          %.3f          %.3f"
      ,xw1*K31 ,xw2*K32 ,xw3*K33 ,xw4*K34);
48 printf("\n Sum of Kxw = %d" ,xw1*K31+xw2*K32+xw3*K33+
      xw4*K34);
49 printf("\n Hence the still temperature Ts = 416 K");
50 printf("\n\n\n Calculation of feed condition");
51 printf("\n Component      xf          Tb = 377K
      Tb = 376 K");
52 printf("\n
      K          Kxf ");
53 xf1 = 0.40;
54 xf2 = 0.23;
55 xf3 = 0.17;
56 xf4 = 0.20;
57 Kb1 = 1.80;          //equilibrium constants at
      377 K for n-C4
58 Kb2 = 0.81;          //equilibrium constants at
      377 K for C5
59 Kb3 = 0.39;          //equilibrium constants at
      377 K for C6
60 Kb4 = 0.19;          //equilibrium constants at
      377 K for C7
61 Kb11 = 1.78;          //equilibrium constants at
      377 K for n-C4
62 Kb21 = 0.79;          //equilibrium constants at
      377 K for C5
63 Kb31 = 0.38;          //equilibrium constants at
      377 K for C6

```

```

64 Kb41 = 0.185; //equilibrium constants at
    377 K for C7
65 printf("\n n-C4          %.2 f      %.2 f      %.3 f
    %.2 f      %.3 f", xf1, Kb1, xf1*Kb1, Kb11, xf1*Kb11);
66 printf("\n C5          %.2 f      %.2 f      %.3 f
    %.2 f      %.3 f", xf2, Kb2, xf2*Kb2, Kb21, xf2*Kb21);
67 printf("\n C6          %.2 f      %.2 f      %.3 f
    %.2 f      %.3 f", xf3, Kb3, xf3*Kb3, Kb31, xf3*Kb31);
68 printf("\n C7          %.2 f      %.2 f      %.3 f
    %.2 f      %.3 f", xf4, Kb4, xf4*Kb4, Kb41, xf4*Kb41);
69 printf("\n
    %.3 f", xf1*Kb1+xf2*Kb2+xf3*Kb3+xf4*Kb4
    , xf1*Kb11+xf2*Kb21+xf3*Kb31+xf4*Kb41);
70
71 //Calculation of pinch temperatures
72 printf("\n\n\n The upper pinch temperature ,Tn = %d K
    ", 343+0.33*(416-343));
73 printf("\n The lower pinch temperature ,Tm = %d K"
    , 343+0.67*(416-343));
74
75 //Calculation of approximate minimum reflux ratio.
76 printf("\n\n\n");
77 printf("\n Component          Tn = 367 K          Tm = 391 K
    xfh      axfh");
78 printf("\n
    ");
79 printf("\n n-C4          %.2 f          %.2 f
    ", 2.38, 2.00);
80 printf("\n C5          %.2 f          %.2 f
    ", 1.00, 1.00);
81 printf("\n C6          %.3 f          %.3 f
    %.2 f      %.3 f", 0.455, 0.464, 0.17, 0.077);
82 printf("\n C7          %.3 f          %.3 f
    %.2 f      %.3 f", 0.220, 0.254, 0.20, 0.044);
83 printf("\n
    %.3 f", 0.077+0.044);
84 rf = xf1/xf2;

```

```

85 printf("\n rf = %.3 f",rf);
86 xn4 = rf/[(1+rf)*(1+0.121)];
87 printf("\n xn4 = %.3 f",xn4);
88 xn5 = xn4/rf;
89 printf("\n xn5 = %.3 f",xn5);
90 Rm = [1/(2.38-1)]*(0.975/0.563) -2.38*(0.025/0.325);
91 printf("\n Rm = %.2 f",Rm);
92
93 //The streams in the column
94 D = 40;
95 Ln = D*Rm;
96 Vn = Ln+D;
97 F = 100;
98 Lm = Ln + F;
99 W = 60;
100 Vm = Lm - W;
101 Ratio = Lm/W;
102 printf("\n Ln = %.1 f kmol",44.8);
103 printf("\n Vn = %.1 f kmol",84.8);
104 printf("\n Lm = %.1 f kmol",144.8);
105 printf("\n Vn = %.1 f kmol",84.4);
106 printf("\n Lm/W = %.2 f",Ratio);
107 //Check on minimum reflux ratio
108 //xn = xd/(a-1)Rm
109 xn = xd1/[(2.38-1)*Rm];
110 printf("\n For n-C4..... xn = %.3 f",xn);
111 xn1 = 1-xn;
112 printf("\n For n-C5... xn = %.3 f",xn1);
113 printf("\n Temperature check for upper pinch gives
      sum of K*xn = ");
114 sumKxn = 1.62*xn +0.68*xn1;
115 printf("%.3 f",sumKxn);

```

---

Scilab code Exa 11.16 Minimum reflux ratio

```

1 //Example 11.16
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.16\n");
7
8 //Mole fractions
9 xf=[0.40 0.35 0.25];
10 xd=[0.534 0.453 0.013];
11 xw=[0 0.04 0.96];
12
13 //Amount of feed, product, bottom in kmol
14 F=[40 35 25];
15 D=[40 34 1];
16 W=[0 1 24];
17
18 //roots of equation
19 theta=[1.15 1.17];
20
21 //relative volatility
22 alpha=[2.7 2.22 1];
23
24 //Underwoods 1st equation for q=1
25 sums=[0 0];
26 for i=1:2
27     for j=1:3
28         sums(i)=sums(i)+(alpha(j)*xf(j)/(alpha(j)-
29             theta(i)));
30     end
31 end
32 printf("\nFrom Underwoods 1st eq\n");
33 printf("The value of 1-q at theta = 1.15 and 1.17
34     are");
35 disp(sums);
36
37 //Underwoods 2nd equation for minimum reflux ratio
38 sum2=0;

```

```

37 for l=1:3
38     sum2=sum2+(alpha(1)*xd(1)/(alpha(1)-theta(2)));
39 end
40
41 Rm=sum2-1;
42 printf("\nMinimum Reflux ratio is %.3f\n",Rm);
43
44 //End

```

---

Scilab code Exa 11.17 Number of theoretical plates required

```

1 //Example 11.17 Fenske's Equation
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.17\n");
7
8 //From previous question data
9 xA_d=0.453;
10 xB_d=0.013;
11
12 xA_s=0.04;
13 xB_s=0.96;
14
15 alpha_av=2.22;
16
17 //By Fenske Equation for no. of plates
18
19 n=((log(xA_d*xB_s/(xA_s*xB_d)))/log(alpha_av))-1;
20
21 printf("\nMinimum no. of plates are %f or %d\n",n,n)
22     ;
23 //End

```

---

**Scilab code Exa 11.18** Xchange in n with R

```
1 //Example 11.18
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.18\n");
7
8 R=[1 2 5 10];
9 Rm=0.83;
10
11 nm=8.5-1;
12
13 //X are points on X-axis of graph
14 for i=1:4
15     X(i)=(R(i)-Rm)/(R(i)+1);
16 end
17
18 //Values at Y-axis for corresponding values of X-
    axis in graph are given by
19 Y=[0.55 0.32 0.15 0.08];
20
21 //where  $Y=(n+1)-(nm+1)/(n+2)$ 
22
23
24 for i=1:4
25     n=poly([0], 'x');
26     N(i)=roots(((n+1)-(nm+1))-(Y(i)*(n+2)));
27 end
28
29 printf("\n\nThe values of R and n are\n");
30 for i=1:4
31     printf("\n\t%d \t %.2f",R(i),N(i));
```



```

32 end
33
34 printf("\n\nThe change in R and n can be seen as
      above\n");
35
36 //End

```

---

### Scilab code Exa 11.19 Optimum reflux ratio

```

1 clear all;
2 clc;
3 printf("\n Example 11.19");
4 //Data from fig. 11.42
5 a = [0 0.02 0.04 0.06 0.08 0.1 0.2 0.4 0.6 0.8 1.0];
6 b = [0.75 0.62 0.60 0.57 0.55 0.52 0.45 0.30 0.18
      0.09 0];
7 //a = (R-Rm)/(R+1)
8 //b = [(n+1)-(nm+1)]/(n+2)
9 R = [0.92 1.08 1.25 1.75 2.5 3.5 5.0 7.0 9.0];
10 n = [28.6 22.8 16.9 13.5 11.7 10.5 9.8 9.2 8.95];
11 plot(n,R);
12 xtitle("Plot of R vs n", "n", "R");
13 printf("\n Derivative calculated from the graph");
14 d = [110.0 34.9 9.8 3.8 1.7 0.6 0.4 0.2 0.05];
15 i=1;
16 while i <=9
17     s = R(i)+1 - (n(i)+7.72)/d(i);
18     if s <=0.0001 then
19         Ropt = R(i);
20         printf("\n Ropt = %.2f", Ropt);
21         break;
22     end
23     i=i+1;
24 end
25 printf("\n R is approximately %.1f percent of the

```

```
minimum reflux condition",1.25/0.866666666*100);
```

---

**Scilab code Exa 11.20** Plate efficiency for the given data

```
1 //Example 11.20
2
3 clear all;
4 clc;
5
6 printf("\tExample 11.20\n");
7
8 //Mole fraction
9 xf=[0.2 0.3 0.2 0.3];
10
11 //Viscosity at mean tower temp. in mNs/m^2
12 uL=[0.048 0.112 0.145 0.188];
13
14 //Viscosity of water in mNs/m^2
15 uw=1;
16
17 sums=0;
18 for i=1:4
19     sums=sums+(xf(i)*uL(i)/uw);
20 end
21
22 //Efficiency by DRICKAMER and BRADFORD
23 E=0.17-(0.616*log10(sums));
24
25 printf("\nEfficiency is %.2f",E);
26
27 //End
```

---

# Chapter 12

## Absorption of gases

Scilab code Exa 12.1 Overall liquid film coefficient

```
1 clear;
2 clc;
3 printf("\n Example 12.1");
4 //Overall liquid transfer coefficient KLa = 0.003
   kmol/s.m^3(kmol/m^3)
5
6 //(1/KLa)=(1/kLa)+(1/HkGa)
7 // let (KLa)=x
8 x = 0.003;
9 overall = 1/x;
10
11 //For the absorption of a moderately soluble gas it
   is reasonable to assume that the liquid and gas
   phase resistances are of the same order
   of magnitude, assuming them to be equal.
12 //(1/KLa)=(1/kLa)+(1/HkGa)
13 //let 1/kLa = 1/HkGa = y
14 y = (1/(2*x));
15 z = (1/y); //z is in kmol/s m^3(kmol/m
   ^3)
16 printf("\n For SO2:");
```

```

17 printf("\n      kGa = %f kmol/s m^3(kN/m^2)",z/50);
18 printf("\n For NH3:");
19 d_SO2 = 0.103;           //diffusivity at 273K for
    SO2 in cm^2/sec
20 d_NH3 = 0.170;         //diffusivity at 273K for
    NH3 in cm^2/sec
21 printf("\n      kGa = %f kmol/s m^3(kN/m^2)",(z/50)*(
    d_NH3/d_SO2)^0.56);
22 printf("\n For a very soluble gas such as NH3, kGa =
    KGa.");
23 printf("\n For NH3 the liquid-film resistance will
    be small, and:");
24 printf("\n      kGa =KGa = %fkmol/s m^3(kN/m^2)"
    ,(z/50)*(d_NH3/d_SO2)^0.56);

```

---

### Scilab code Exa 12.2 Mass transfer coefficient

```

1 clear;
2 clc;
3 printf("\n Example 12.2");
4 G = 2;           //air flow rate
    in kg/m^2.sec
5 Re = 5160;      //Re stands for
    Reynolds number
6 f = 0.02;      //friction factor
    = R/ρu^2
7 d_SO2 = 0.116*10^(-4); //diffusion
    coefficient in m^2/sec
8 ν = 1.8*10^(-5); //viscosity
    in mNs/m^2
9 ρ = 1.154;     //density is in kg
    /m^3
10
11 //(hd/u) (Pm/P) (u/ρ/d_SO2) ^ 0.56 = BRe^(-0.17)=jd
12

```

```

13 function [x]=a1()
14     x = (v/(p*d_SO2))^(0.56);
15     funcprot(0);
16 endfunction
17
18 //jd = f =R/pu^2
19 function [y]=a2()
20     y = f/a1();
21     funcprot(0);
22 endfunction
23 //G = pu
24     u = (G/p);           //u is in m/sec
25
26 function [x1]= a3()
27     x1 = a2()*u;
28     funcprot(0);
29 endfunction
30
31 function [d]=d1()
32     d = Re*v/(G);
33     funcprot(0);
34 endfunction
35
36 printf("\n d = %f mm",d1());
37 R = 8314;           //R is in m^3(N/m^2)/K kmol
38 T = 298;           //T is in Kelvins
39 function [kG]=kG1()
40     kG = a3()/(R*T);
41     funcprot(0);
42 endfunction
43
44 printf("\n kG = %.2 f*10^(-8) kmol/m^2 sec (N/m^2)",kG1
    ()*10^(8));
45 printf("\n kG = %.2 f*10^(-4) kg SO2/m^2 sec (kN/m^2)",
    kG1()*10^(7)*64);

```

---

### Scilab code Exa 12.3 Overall transfer units required

```
1 clear;
2 clc;
3 printf("\n Example 12.3");
4 //as the system are dilute mole fractions are
   approximately equal to mole ratios.
5 //At the bottom of the tower
6 y1 =0.015;           //mole fraction
7 G = 1;              //Gas flow rate is in kg/m^2
   sec
8 //At the top of the tower
9 y2 = 0.00015;      //mole fraction
10 x2 = 0;
11 L = 1.6;           //liquid flow rate is in kg/m
   ^2.sec
12 Lm = 1.6/18;       //liquid flow rate is in kmol/m
   ^2.sec
13 Gm = 1.0/29;       //gas flow rate is in kmol/m^2.
   sec
14 x1 = poly([0], 'x1');
15 x11 = roots(Gm*(y1-y2)-Lm*(x1));
16 printf("\n      x1 = %f", x11);
17 function [ye1]=henry_law(x)
18     ye1 = 1.75*x;
19     funcprot(0);
20 endfunction
21 bottom_driving_force = y1 - henry_law(x11);
22 function [lm]=log_mean()
23     lm = (bottom_driving_force-y2)/log(
       bottom_driving_force/y2);
24     funcprot(0);
25 endfunction
26 NoG = (y1-y2)/log_mean();
```

```
27 NoL = NoG*1.75*(Gm/Lm);
28 printf("\n      NoL =%.2 f",NoL);
```

---

#### Scilab code Exa 12.4 Height of transfer units and number of transfer units

```
1 clear;
2 clc;
3 printf("\n Example 12.4");
4 Gm = 0.015;
5 KGa = 0.04;
6 top = 0.0003;
7 Y1 = 0.03;
8 Ye = 0.026;
9 bottom = (Y1-Ye);
10 log_mean_driving_force = (bottom-top)/log(bottom/top
    );
11 Z = poly([0], 'Z');
12 Z1 = roots(Gm*(Y1-top)-KGa*log_mean_driving_force*Z)
    ;
13 printf("\n      Z =%.2 fm",Z1);
14 HoG = Gm/KGa;
15 printf("\n Height of transfer unit HoG = %.3fm",HoG
    );
16 printf("\n Number of transfer units NoG = %f",ceil
    (7.79/0.375));
```

---

#### Scilab code Exa 12.5 Height of the tower

```
1 clear;
2 clc;
3 printf("\n Example 12.5");
4 y1 = 0.10;
5 Y1 = 0.10/(1-0.10);
```

```

6 y2 = 0.001;
7 Y2 = y2;
8 mass_flowrate_gas = 0.95;          //mass flow rate in
   kg/m^2.sec
9 mass_percent_air = (0.9*29/(0.1*17+0.9*29))*100;
10 mass_flowrate_air = (mass_percent_air*
   mass_flowrate_gas); //in kg/m^2.sec
11 Gm = (mass_flowrate_air/29);
12 Lm = 0.65/18;                    //Lm is in kmol/m^2.sec
13 //A mass balance between a plane in the tower where
   the compositions are X and Y and the top of the
   tower gives:
14
15 //Y = 1.173X+0.001
16 X = 0:0.001:0.159;
17 Y = 1.173*X+0.001;
18 plot(X,Y);
19
20 x=[0.021 0.031 0.042 0.053 0.079 0.106 0.159];
21
22 PG = [12 18.2 24.9 31.7 50.0 69.6 114.0];
23 i=1;
24 while i<8
25     Y1(i) = PG(i)/(760-PG(i));
26     i=i+1;
27 end
28 plot(x,Y1);
29 //xlabel("Area under the curve is 3.82m", "kmolNH3/
   kmolH2O", "kmolNH3/kmol air");
30 xtitle("Operating and equilibrium lines", "kmol NH3/
   kmol H2O", "kmol NH3/kmol air");
31 //If the equilibrium line is assumed to be straight,
   then:
32 //Gm(Y2      Y1) = KGaZdeltaPlm
33
34 //Top driving force
35 deltaY2 = 0.022;
36 //Bottom driving force

```



```

37 deltaY1 = 0.001;
38 deltaYlm = 0.0068;
39 Z = (0.0307*0.11)/(0.001*0.688);
40 printf("The height of the tower is %.2f meters",Z);

```

---

**Scilab code Exa 12.6** specific steam consumption

```

1 clear;
2 clc;
3 printf("\n Example 12.6");
4 printf("\n Number of theoretical plates = %d"
        ,(30*0.3));
5
6 //At the bottom of the tower:
7 //Flowrate of steam = Gm (kmol/m^2s)
8 //Mole ratio of pentane in steam = Y1, and
9 //Mole ratio of pentane in oil = X1
10 X1 = 0.001;
11
12 //At the top of the tower:
13 //exit steam composition = Y2
14 //inlet oil composition = X2
15 X2 = 0.06;
16 //flowrate of oil = Lm (kmol/m^2.sec)
17
18 //The minimum steam consumption occurs when the exit
    steam stream is in equilibrium with the inlet
    oil, that is when:
19 //The equilibrium relation for the system may be
    taken as  $Y_e = 3.0X$ , where  $Y_e$  and  $X$  are expressed
    in mole ratios of pentane in the gas and liquid
    phases respectively.
20 Ye2 = X2*3;
21 //Lmin(X2      X1) = Gmin(Y2      Y1)
22 //If Y1 = 0, that is the inlet steam is pentane-free

```

```

    , then:
23 ratio_min = (X2 - X1)/Ye2;
24
25
26 //The operating line may be fixed by trial and error
    as it passes through the point (0.001, 0), and 9
    theoretical plates are required for the
    separation. Thus it is a matter of selecting the
    operating line which, with 9 steps, will give X2
    = 0.001 when X1 = 0.06. This is tedious but
    possible, and the problem may be better solved
    analytically since the equilibrium line is
    straight.
27
28 //let x = 1/A
29 //for a stripping operation
30 LHS = (X2-X1)/(X2);
31 printf("\n LHS = %f",LHS);
32 x = poly([0], 'x');
33 x1 = roots((x^10-x)-LHS*(x^(10)-1));
34 printf("\n 1/A = %.2 f", x1(9));
35 //A = Lm/mGm
36 printf("\n Gm/Lm = %.3 f", x1(9)/3);
37 printf("\n Actual/minimum Gm/Lm = %.2 f", 0.457/0.328)
    ;
38
39 //If (actual Gm/Lm)/(min Gm/Lm) = 2,
40 printf("\n Actual Gm/Lm = %.3 f", .328*2);
41 x2 = 3*(0.656);
42 printf("\n 1/A = mGm/Lm = %.3 f", 3*0.656);
43
44 //y = (1.968)^(N+1)
45 y = poly([0], 'y');
46 y1 = roots(0.983*(y-1)-(y-1.968));
47 N = log(y1)/log(1.968)-1;
48 printf("\n The actual number of plates are : %.1f",
    ceil(N/0.3));

```

---

# Chapter 13

## Liquid liquid extraction

Scilab code Exa 13.1 Composition of final raffinate

```
1 clear all;
2 clc;
3 printf("\n Example 13.1");
4 printf("\n      (a) Countercurrent operation");
5 //(a) Countercurrent operation
6 S = 1.6*10^(-4);           //Solvent flow rate
   in m^3/sec
7 mass_flowrate = (S*800);   //mass flow rate is
   in kg/sec
8
9 //Considering the solution, 400cm3/s = 4 10 4 m
   ^3/sec containing, say, a m^3/sec A and (5 10
   4 a) m3/sec B.
10 //Thus mass flow rate of A = 1200a kg/sec
11 //and mass flow rate of B = (4*10^(-4)-a)*1000 =
   (0.4-1000a)kg/sec
12 //a total of:           (0.4+200a) kg/sec
13 C = poly([0], 'C');
14 C1 = roots(0.1*(0.4+200*C)-1200*C);
15 printf("\n Concentration of the solution is %.2f
   *10^(-5) m^3/sec", C1*10^5);
```

```

16 printf("\n mass flow rate of A = %.3f kg/sec" ,1200*
    C1);
17 printf("\n mass flow rate of B =%.3f kg/sec"
    ,0.4+200*C1);
18 printf("\n Ratio of A/B in the feed , Xf = %.3f kg/kg
    " ,0.041/0.366);
19
20 X = [0.05 0.10 0.15];
21 Y = [0.069 0.159 0.258];
22 plot(X,Y);
23 xtitle("Equilibrium curve", "kg A/kg B", "kg A/kg S");
24 //From The curve:
25 slope = 0.366/0.128;
26 printf("\n Slope of the equilibrium line is %.2f",
    slope);
27
28 //Since pure solvent is added,  $Y_{n+1} = Y_4 = 0$  and a
    line of slope 2.86 is drawn in such that stepping
    off from  $X_f = 0.112$  kg/kg to  $Y_4 = 0$  gives
    exactly three stages. When  $Y_4 = 0$ ,  $X_n = X_3 =$ 
    0.057 kg/kg
29 printf("\n The composition of final raffinate is
    0.057kg A/kg B");
30
31 printf("\n\n\n          (b) Multiple contact");
32 printf("\n Stage 1");
33 printf("\n In this case %.4f kg/sec" ,0.128/3);
34 //and from the equilibrium curve, the extract
    contains 0.18 A/kg S and (0.18      0.0427) =
    0.0077 kg/s A.
35 printf("\n Thus raffinate from stage 1 contains %.4f
    kg/sec Aand %.3f kg/sec B" ,(0.041-0.0077) ,0.366)
    ;
36 X1 = 0.0333/0.366;
37 printf("\n X1 = %.3f kg/kg" ,0.0333/0.366);
38
39 printf("\n Stage 2");
40 //the extract contains 0.14 kg A/kg S

```

```

41 printf("\n The extract contains %.4f kg/sec A"
    ,0.14*0.0427);
42 //Thus: the raffinate from stage 2 contains (0.0333
    0.0060) = 0.0273 kg/s A and 0.366 kg/s B
43 X2 = (0.0273/0.366);
44 printf("\n X2 = %.3f kg/kg",X2);
45
46 printf("\n Stage 3");
47 //the extract contains 0.114 kg A/kg S
48 printf("\n The extract contains %.4f kg/sec A/kg S"
    ,0.114*0.0427);
49 printf("\n Thus the raffinate contains %.4f kg/sec A
    and %.3f kg/sec B", (0.0273-0.0049),0.366);
50 printf("\n The composition of final raffinate = %3f
    kg A/kg B",0.0224/0.366);

```

---

### Scilab code Exa 13.3 Overall transfer coefficient

```

1 clear all;
2 clc;
3 printf("\n Example 13.3");
4 //From the equilibrium relationship
5 CB1 = (0.0247*0.685);
6 printf("\n CB1* = %.4f kmol/m^3",CB1);
7 CB2 = (0.0247*0.690);
8 printf("\n CB2* = %.4f kmol/m^3",CB2);
9
10 //Thus the driving force at the bottom:
11 deltaC1 = (0.0169-0.0040);
12 printf("\n deltaC1 =%.4f kmol/m^3",deltaC1);
13 //Driving force at the top
14 deltaC2 = (0.0170-0.0115);
15 printf("\n deltaC2 = %.4f kmol/m^3",deltaC2);
16 function [x]= log_mean_driving_force()
17     x = (deltaC1 - deltaC2)/log((deltaC1)/deltaC2);

```

```

18     funcprot(0);
19 endfunction
20 printf("\n log mean driving force is given by
        deltaClm = %.4f kmol/m3",log_mean_driving_force
        ());
21 KBa = (4.125*10(-8))/(log_mean_driving_force()
        *0.0063);
22 printf("\n KBa = %.1f*10(-4) kmol/sec m3(kmol/m3)
        ",KBa*104);
23 HoB = (1.27*10(-3))/KBa;
24 printf("\n HoB = %.2f meters",HoB);

```

---

#### Scilab code Exa 13.4 Surface mean droplet size

```

1 clear all;
2 clc;
3 printf("\n Example 13.4");
4 diameter = [2 3 4 5 6];
5 number = [30 120 200 80 20];
6
7 function [x] = Sum_d1cube()
8     sum = 0;
9     i = 1;
10    while (i <= 5)
11        sum = sum + number(i)*(diameter(i))^3;
12        i = i+1;
13    end
14    x = sum;
15    funcprot(0);
16 endfunction
17
18 function [y]=sum_disquare()
19     sum1 = 0;
20     j=1;
21     while(j<=5)

```

```

22         sum1 = sum1 + number(j)*(diameter(j))^2;
23         j= j+1;
24     end
25     y = sum1;
26     funcprot(0);
27 endfunction
28
29 function [z]= ds()
30     z = Sum_d1cube()/sum_d1square();
31     funcprot(0);
32 endfunction
33 printf("\n Mean droplet size = %.2f mm",ds());

```

---

**Scilab code Exa 13.5** Number of overall transfer units in raffinate phase

```

1 clear all;
2 clc;
3 printf("\n Example 13.5");
4 CSA = (%pi/4)*(0.075)^2; //cross
    sectional area is in m^2
5 V = (0.0044*3); //volume of
    packing is in m^3
6 C = 0.01; //concentration
    is in kg/kg
7 printf("\n mass of acid transferred to the ether %.4
    f kg/m^2.sec or %f kg/sec",0.05*(0.01-0)
    ,0.005*0.0044);
8 printf("\n Acid in the aqueous feed = %.2f kg/m^3.
    sec",0.25*0.04);
9 printf("\n Acid in the raffinate = %.3f kg/m^2.sec"
    ,0.01-0.005);
10 printf("\n Concentration of acid in the raffinate =
    %.2f kg/kg",0.005/0.25);
11 printf("\n At the top of the column");
12 CR2 = 0.040; //Concentration is in kg/

```

```

kg
13 CR22 = 0.040*0.3;           //Concentration is in kg/
kg
14 deltaC2 = (0.012-0.010);
15 printf("\n deltaC2 = %.3f kg/kg",deltaC2);
16 printf("\n\n At the bottom of the column");
17 CR1 = 0.20;                 //Concentration is in kg/
kg
18 CR11 = 0.020*0.3           //Concentration is in kg/
kg
19 deltaC1 = 0.006 -0;        //Concentration is in kg/
kg
20 printf("\n deltaC1 = %.3f kg/kg",deltaC1);
21 deltaCRlm = (0.006-0.002)/log(0.006/0.002);
22 printf("\n Logarithmic driving force is : %.4f kg/kg
",deltaCRlm);
23 KRa = 0.000022/(0.01333*deltaCRlm);
24 printf("\n KRa = %.3f kg/m^3.sec(kg/kg)",KRa);
25 printf("\n Height of an overall transfer unit = %.2f
m",0.25/KRa);
26 printf("\n The number of overall transfer units = %
.2f ",3/0.54);

```

---



# Chapter 14

## Evaporation

Scilab code Exa 14.1 Heat surface required

```
1 clear;
2 clc;
3 printf("\n Example 14.1");
4 //Assuming that the steam is dry and saturated at
   205 kN/m2, then from the Steam Tables in the
   Appendix, the steam temperature = 394 K at which
   the total enthalpy = 2530 kJ/kg.
5
6 //At 13.5 kN/m2, water boils at 325 K and, in the
   absence of data on the boiling point elevation,
   this will be taken as the temperature of
   evaporation, assuming an aqueous solution. The
   total enthalpy of steam at 325 K is 2594 kJ/kg.
7
8 //Thus the feed, containing 10 per cent solids, has
   to be heated from 294 to 325 K at which
   temperature the evaporation takes place.
9
10 printf("\n mass of dry solids = %.1f kg/sec"
   , (7*10/100));
11 x = poly([0], 'x');
```

```

12 x1 = roots(0.7*100-50*(0.7+x));
13 printf("\n x = %.1f kg/sec",x1);
14 printf("\n Water to be evaporated = %.1f kg/sec"
    ,(7-0.7)-0.7);
15 printf("\n Summarising");
16 printf("\n Stream          Solids          Liquid
    Total  ");
17 printf("\n          (kg/s)          (kg/s)
    (kg/s)  ");
18 printf("\n Feed          %.1f          %.1f
    %.1f",x1,7-x1,x1+7-x1);
19 printf("\n Product          %.1f          %.1f
    %.1f",x1,x1,x1+x1);
20 printf("\n Evaporation          %.1f
    %.1f",7-x1-x1,7-2*x1);
21 //Using a datum of 273K
22 q_entering = (7*3.76)*(294-273);
23 printf("\n Heat entering with the feed = %.1f kW",
    q_entering);
24 q_leaving = (1.4*3.14)*(325-273);
25 printf("\n Heat leaving with the product = %.1f kW",
    q_leaving);
26 printf("\n Heat leaving with the evaporated water =
    %d kW",5.6*2594);
27 printf("\n Heat transferred from the steam = %d kW"
    ,14526+228.6-552.7);
28 printf("\n The enthalpy of the condensed steam
    leaving at 352.7 K = %.1f kJ/kg "
    ,4.18*(352.7-273));
29 printf("\n The heat transferred from 1 kg steam = %
    .1f kJ/kg",2530-333.2);
30 printf("\n Steam required = %.2f kg/s "
    ,14202/2196.8);
31
32 //s the preheating of the solution and the sub-
    cooling of the condensate represent but a small
    proportion of the heat load, the temperature
    driving force may be taken as the difference

```

```

    between the temperatures of the condensing steam
    and the evaporating water, or:
33 printf("\n deltaT = %d deg K ",394-325);
34 printf("\n Heat transfer area ,A = %.1f m^2"
    ,14202/(3*69));

```

---

#### Scilab code Exa 14.2A Forward feed

```

1 clear;
2 clc;
3 printf("\n Example 14.2a");
4 //Temperature of dry saturated steam at 205 kN/m2 =
   394 K.
5 //At a pressure of 13 kN/m2 (0.13 bar), the boiling
   point of water is325 K, so that the
   totaltemperature difference &T = (394      325) =
   69deg K.
6
7
8 //First Approximation.
9
10 //Assuming that: U1*deltaT1 = U2*deltaT2 = U3*
   deltaT3
11 //then substituting the values of U1, U2 and U3 and
   &T = 69 deg K gives:
12 printf("\n deltaT1 = %f deg K" ,13);
13 printf("\n deltaT2 = %f deg K" ,20);
14 printf("\n deltaT2 = %f degK" ,36);
15
16 //Since the feed is cold , it will be necessary to
   have a greater value of T1 than given by this
   analysis. It will be assumed that T1 = 18 deg K,
   T2 = 17 deg K, T3 = 34 deg K.
17
18 //For steam to 1: T0 = 394 K and 0 = 2200 kJ/kg

```

```

19 //For steam to 2: T1 = 376 K and 1 = 2249 kJ/kg
20 //For steam to 3: T2 = 359 K and 2 = 2293 kJ/kg
21 //          T3 = 325 K and 3 = 2377 kJ/kg
22
23 //Assuming that the condensate leaves at the steam
    temperature, then heat balances across each
    effect may be made as follows:
24
25 //Effect 1 :
26 //2200 D0 = 4      4.18(376      294) + 2249 D1
27
28 //Effect 2:
29 //2249 D1 + (4      D1) 4.18(376      359) = 2293 D2
30
31 //Effect 3:
32 //2293 D2 + (4      D1      D2) 4.18(359      325) =
    2377 D3
33
34 printf("\n          Solids          liquor
          Total ");
35 printf("\n          (kg/s)          (kg/s)
          (kg/s) ");
36 printf("\nFeed          %.1 f          %.1 f
          %.1 f ",0.4,3.6,4.0);
37 printf("\nProduct          %.1 f          %.1 f
          %.1 f ",0.4,0.4,0.8);
38 printf("\nEvaporation          %.1 f
          %.1 f ",3.6-0.4,4.0-0.8);
39
40 //D1 + D2 + D3 = 3.2 kg/s
41 D = [2177.94 -2293 0; -4.18*(359-325)
    2293-4.18*(359-325) -2377;1 1 1];
42 C = [-4*4.18*(376-359); -4*4.18*(359-325); 3.2];
43 D1 = inv(D)*C;
44 printf("\n ans = %.3 f kg/s", D1);
45
46 D0 = (4*4.18*(376-294)+2249*D1(1))/2200;
47 A1 = D0*2200/(3.1*18);

```

```

48 printf("\n A1 = %.1f m^2",A1);
49 A2 = D1(1)*2249/(2*17);
50 printf("\n A2 = %.1f m^2",A2);
51 A3 = D1(2)*2293/(1.1*34);
52 printf("\n A3 = %.1f m^2",A3);
53 printf("\n These three calculated areas are
    approximately equal, so that the temperature
    differences assumed may be taken as nearly
    correct");
54
55 //let , deltaT1 = x
56 //deltaT2 = y
57 //deltaT3 = z
58 X = [3.1 -2.0 0;0 2.0 -1.1;1 1 1];
59 A = [0;0;69];
60 T = inv(X)*A;
61 printf("\n deltaT1 = %d deg K",T(1));
62 printf("\n deltaT2 = %d deg K",T(2));
63 printf("\n deltaT3 = %d deg K",T(3));
64
65 //ghting the temperature differences to allow for
    the fact that the feed enters at ambient
    temperature gives:
66 //deltaT1 = 18 deg K delta T2 = 18 degK delta T3 = 33
    deg K
67 Steam_temp = 394;
68 T1 = Steam_temp-18;
69 printf("\n T1 = %d K",T1);
70 T2 = T1 - 18;
71 printf("\n T2 = %d K",T2);
72 T3 = T2 - 33;
73 printf("\n T3 = %d K",T3);
74
75 //The total evaporation (D1 + D2 + D3) is obtained
    from a material balance:
76 printf("\n
    Solids          liquor
    Total ");
77 printf("\n
    (kg/s)          (kg/s)

```

```

                                (kg/s) ");
78 printf("\nFeed          %.1f          %.1f
          %.1f  ",0.4,3.6,4.0);
79 printf("\nProduct      %.1f          %.1f
          %.1f  ",0.4,0.4,0.8);
80 printf("\nEvaporation          %.1f
          %.1f  ",3.6-0.4,4.0-0.8);
81
82 //Assuming, as an approximation, equal evaporation
    in each effect, or D1 = D2 = D3 = 1.07 kg/s, then
    the latent heat of flash vaporisation in the
    second effect is given by:
83 q = 4.18*(4-1.07)*(376-358);
84 printf("\n latent heat of flash vaporisation in the
    second effect is = %.1f kW",q);
85 printf("\n latent heat of flash vaporisation in the
    third effect is:= %.1f kW ",4.18*(4-2*1.07)
    *(358-325));
86 printf("\n At 394 K, the latent heat = 2200 kJ/kg");
87 printf("\n At 325 K, the latent heat = 2377 kJ/kg");
88 printf("\n Mean value,      = 2289 kJ/kg");
89
90 printf("\n Values ofT1,T2T3 are now chosen by trial
    and error to give equal values of A in each
    effect, as follows");
91 printf("\n deltaT1      A1      deltaT2      A2
    deltaT3          A3");
92 printf("\n (deg K)      (m^2)      (deg K)      (m^2)      (deg K
    )          m^2");
93 printf("\n      18          64.2          18          61.4          33
    66.9 ");
94 printf("\n      19          60.5          17          65.0          33
    66.9");
95 printf("\n      18          64.2          17.5          63.1          33.5
    65.9");
96 printf("\n      18          64.2          17          65.0          34
    64.9");
97

```

```

98 printf("\n Steam consumption = %.2f kg/s",3580/2289)
    ;
99 printf("\n Economy = %.1f kg/kg",3.2/1.56);

```

---

### Scilab code Exa 14.2B Backward feed

```

1 clear;
2 clc;
3 printf("\n Example 14.2b");
4 deltaT1 = 20; //temperature is in deg K
5 deltaT2 = 24; //temperature is in deg K
6 deltaT3 = 25; //temperature is in deg K
7 To = 394; //temperature is in deg K
8 T1 = 374; //temperature is in deg K
9 T2 = 350; //temperature is in deg K
10 T3 = 325; //temperature is in deg K
11 latent_heat0 = 2200; //latent heat in kJ/kg
12 latent_heat1 = 2254; //latent heat in kJ/kg
13 latent_heat2 = 2314; //latent heat in kJ/kg
14 latent_heat3 = 2377; //latent heat in kJ/kg
15
16 //Effect 3:
17 // 2314 D2 = 4 4.18(325 294) +
    2377 D3
18
19 //Effect 2:
20 // 2254 D1 = (4 D3) 4.18(350 325)
    + 2314 D2
21
22 //Effect 1:
23 // 2200 D0 = (4 D3 D2) 4.18(374
    350) + 2254 D1
24
25 //(D1 + D2 + D3) = 3.2 kg/s
26

```

```

27 D = [0 2314 -2377;2254 -2314 4.18*(350-325);1 1 1];
28 C = [4*4.18*(325-294);4*4.18*(350-325);3.2];
29 D1 = inv(D)*C;
30 printf("\n D1 = %.3 f kg/s",D(1));
31 printf("\n D2 = %.3 f kg/s",D(2));
32 printf("\n D3 = %.3 f kg/s",D(3));
33 A1 = 1.387*latent_heat0/(2.5*20);
34 A2 = 1.261*latent_heat1/(2*24);
35 A3 = 1.086*latent_heat2/(1.6*25);
36 printf("\n A1 = %.1 f m^2",A1);
37 printf("\n A2 = %.1 f m^2",A2);
38 printf("\n A3 = %.1 f m^2",A3);

```

---

### Scilab code Exa 14.3 Efficiency of the compressor

```

1 clear;
2 clc;
3 printf('\n Example 14.3');
4 He = 2690; //He is the enthalpy of entrained
    steam in kJ/kg
5 H4 = ((1*2780)+(1.6*2690))/2.60;
6
7 //Again assuming isentropic compression from 101.3
    to 135 kN/m2, then:
8 H3 = 2640; //in kJ/kg (from the chart)
9 n = (1.0+1.6)*(2725-2640)/[1.0*(2780-2375)];
10 printf("\n      = %.2 f ",n);
11 printf("\n This value is low, since in good design
    overall efficiencies approach 0.75  0.80.
    Obviously the higher the efficiency the greater
    the entrainment ratio or the higher the saving in
    live steam");
12 Pe = 101.3; //pressure of entrained vapour in
    kN/m^2
13 discharge_P = 135;//discharge pressure in kN/m^2

```



```

14 printf("\n the required flow of live steam = 0.5 kg/
    kg entrained vapour.");
15 printf("\n In this case the ratio is (1.0/1.6) =
    0.63 kg/kg");

```

---

#### Scilab code Exa 14.4 Quantity of additional stream required

```

1 clear;
2 clc;
3 printf("\n Example 14.4");
4 //Making a mass balance , there are two inlet streams
  -
5 //the additional steam , say Gx kg/s
6 //the sea water feed , say Gy kg/s.
7
8 //The two outlet streams are
9 //distilled water product , 0.125 kg/s
10 //the concentrated sea water , 0.5 Gy kg/s
11
12 //Thus: (Gx + Gy) = (0.125 + 0.5 Gy) or (Gx + 0.5 Gy
    ) = 0.125
13
14 //At 650 kN/m2, the total enthalpy of the steam =
    2761 kJ/kg.
15
16 //Thus the energy in this stream = 2761Gx kW
17 //sea water enters at 344 K
18 //enthalpy of feed = [Gy 4.18(344 273)] =
    296.8Gy kW
19 printf("\n The enthalpy of the product is 439*0.125
    = 54.9 kW");
20
21 //Making a balance
22 //(E + 2761Gx + 296.8Gy) = (211.3Gy + 54.9)
23 //(E + 2761Gx + 85.5Gy) = 54.9

```

```

24 //where E is the power supplied to the compressor
25 //(E + 2590Gx) = 33.5
26
27 //For a single-stage isentropic compression
28 //the work done in compressing a volume V1 of gas at
    pressure P1 to a volume V2 at pressure P2 is
    given by
29 P1 = 101.3;           //pressure is in kN/m^2
30 P2 = 120;           //pressure is in kN/m^2
31 dsteam_P1 = (18/22.4)*(273/374.1); //density of
    steam at 101.3 kN/m^2 and 374.1 K
32 //V1 = 0.5Gy/dsteam_P1 = 0.853 Gy m^3/sec
33 E = poly([0], 'E');
34 E1 = roots(0.5*E-(P1*0.853)/(1.3-1)*[(P2/P1)
    ^(0.3/1.3)-1]);
35 printf("\n E = %.1f kW/(kg/sec)", E1);
36
37 printf("\n As the compressor is 50 per cent
    efficient then E = %.1f kW/(kg/sec)", E1/0.5);
38 //E = 46*0.5*Gy = 23.0Gy kW
39 Gx = poly([0], 'Gx');
40 Gx1 = roots(2761*Gx - 54.9 - 4.72*(33.5-2590*Gx));
41 printf("\n Gx = %.3f kg/sec", Gx1);

```

---

#### Scilab code Exa 14.5 Capacity and economy of the system

```

1 clear;
2 clc;
3 printf("\n Example 14.5");
4 //Properties of the ejector
5 e1 = 0.95;           //nozzle efficiency
6 e2 = 0.80;           //efficiency of momentum
    transfer
7 e3 = 0.90;           //efficiency of compression
8 P1 = 650;           //pressure of live stream

```

```

9 P2 = 101.3;           //pressure of entrained steam
10 H1 = 2970; //The enthalpy of the live steam at 650 kN
    /m2 and (435 + 100) in kJ/kg
11 H2 = 2605; //the enthalpy after isentropic expansion
    from 650 to 101.3 kN/m2, using an
    enthalpy entropy chart, in kJ/kg
12 x2 = 0.97;           //dryness fraction
13
14 printf("\n The enthalpy of the steam after actual
    expansion to 101.3 kN/m^2 is %d kJ/kg", e1*(H1 -
    H2));
15 LatentHeat = 2258;           //latent heat at 101.3 kN/m
    ^2
16 //dryness after expansion but before entrainment x21
    is given by:
17 //((x21 - x2) = (1 - e1)(H1 - H2)
18 x21 = poly([0], 'x21');
19 x22 = roots((x21-x2)*LatentHeat-(1-e1)*(H1-H2));
20 printf("\n dryness after expansion but before
    entrainment x21 is = %.3 f", x22);
21
22 //If x23 is the dryness after expansion and
    entrainment, then:
23 //((x23 - x2) = (1 - e3)(H1 - H2)
24 x23 = poly([0], 'x23');
25 x24 = roots((x23-x2)*LatentHeat-(1-e3)*(H1-H2));
26 printf("\n the dryness after expansion and
    entrainment is %.2 f", x24);
27
28 P3 = 205;           //discharge pressure in kN/m^2
29 //Assuming that the steam at the discharge pressure
    P3 = 205 kN/m2 is also saturated
30 x3 = 1;
31 H3 = 2675; //enthalpy of the mixture at the start of
    compression in the diffuser section at 101.3 kN/
    m2
32
33 //Again assuming the entrained steam is also

```

```

    saturated, the enthalpy of the mixture after
    isentropic compression in the diffuser from 101.3
    to 205 kN/m2 is H4
34 H4 = 2810;           //in kJ/kg
35 m = {[ (H1-H2)/(H4-H3)]*e1*e2*e3-1}; //m = m2/m1
36 printf("\n m2/m1 = %.2f kg vapour entrained/kg live
    stream",m);
37 printf("\n Thus with a flow of 0.14 kg/s live steam,
    the vapour entrained at 101.3 kN/m2 is %.2f kg/
    sec",0.14*m);
38 printf('\n Total saturated steam = %.2f kg/sec at
    205 kN/m^2",0.14+0.14*m);
39 printf("\n The economy of the steam = %.2f
    ",0.214/0.14);
40 printf("\n The capacity in terms of throughput of
    solution Gf = %.3f kg/sec",0.214+0.025);

```

---

#### Scilab code Exa 14.6 Method to drive the compressor

```

1 clear;
2 clc;
3 printf("\n Example 14.6");
4 //(a) Diesel engine
5 printf("\n (a) Diesel engine");
6
7 printf("\n For 1 kg evaporation, ammonia circulated
    = 2.28 kg");
8 printf("\n Work done in compressing the ammonia = %d
    MJ/kg",150*2.28);
9 printf("\n For an output of 1 MJ, the engine
    consumes 0.4 kg fuel.");
10 printf("\n fuel consumption = %.3f kg/kg water
    evaporated",0.4*0.342);
11 printf("\n cost = %.5f euro/kg water evaporated "
    ,0.02*0.137);

```

```

12
13 printf("\n (b) Turbine");
14
15 printf("\n The work required is 0.342 MJ/kg
    evaporation");
16 //Therefore with an efficiency of 70 per cent:
17 printf("\n energy required from steam = %.3f MJ/kg "
    ,0.342*100/70);
18 printf("\n Enthalpy of steam saturated at 700 kN/m2
    = 2764 kJ/kg.");
19 printf("\n Enthalpy of steam saturated at 101.3 kN/
    m2 = 2676 kJ/kg");
20 printf("\n thus : energy from steam = %d kg/kg"
    ,2764-2676);
21 printf("\n steam required = %.2f kg/kg evaporation "
    ,0.489/0.088);
22 printf("\n Cost = %.4f euro/kg water evaporated"
    ,0.01*5.56/10);
23 printf("\n Hence the Diesel engine would be used for
    driving the compressor");

```

---

#### Scilab code Exa 14.7 Optimum boiling time

```

1 clear;
2 clc;
3 printf("\n Example 14.7");
4 //(a) Maximum throughput
5
6 printf("\n (a) Maximum throughput");
7 t_bopt = (15*10^3)+(2/(7*10^(-5)))*(7*10^(-5)
    *0.2*15*10^3)^0.5;
8 printf("\n The boiling time to give maximum heat
    transfer and hence maximum throughput is %.1f
    ksecs ",t_bopt*10^(-3));
9

```

```

10 Qb = (2*40*40)*(7*10^(-5))*[((7*10^(-5))*2.81*10^4)
    +0.2)^0.5-0.2^0.5];
11 printf("\n The heat transferred during boiling is %
    .2f*10^7 kJ",Qb);
12 printf("\n the water vaporated = %.2f*10^4 kg"
    ,(4.67*10^(7)/2300)*10^(-4));
13 printf("\n Rate of evaporation during boiling = %.3f
    kg/secs" ,(2.03*10^4)/(2.81*10^4));
14 printf("\n Mean rate of evaporation during the cycle
    = %.3f kg/secs" ,2.03*10^(4)/[(2.8*10^4)
    +(15*10^3)]);
15 printf("\n cost = %.1f euro/cycle" ,((2.81*10^(4)*18)
    /1000)+600);
16
17
18 printf("\n (b) Minimum cost");
19 t_bopt1 = (600/0.018)+[2*(7*10^(-5)*0.2*600*0.018)
    ^0.5]/(7*10^(-5)*0.018);
20 printf("\n The boiling time to give minimum cost is
    %.1f ksecs" ,t_bopt1*10^(-3));
21 Qb1 = [(2*40*40)/(7*10^(-5))]*[(7*10^(-5))*5.28*10^4
    + 0.2)^0.5-0.2^0.5];
22 printf("\n The heat transferred during one boiling
    period is %.2f*10^7 kJ",Qb1*10^(-7))
23 printf("\n The water evaporated = %.2f*10^4 kg",Qb1
    /2300);
24 printf("\n Rate of evaporation = %.3f kg/secs"
    ,(3.03/5.28));
25 printf("\n Mean rate of evaporation during the cycle
    = %.2f kg/secs" ,(3.03*10^4)/[(5.28*10^4)
    +(15*10^3)]);
26 printf("\n cost of one cyce = %.1f euro" ,5.28*10^(4)
    *0.018+600);

```

---

# Chapter 15

## Crystallisation

Scilab code Exa 15.1 Supersaturation ratio

```
1 clear;
2 clc;
3 printf("\n Example 15.1");
4 printf("\n For concentrations in kg sucrose/kg water
   :");
5 c = 2.45; //concentration is in kg/kg
6 printf("\n c = %.2 f kg/kg",c);
7 c1= 2.04; //concentration is in kg/kg
8 printf("\n c1 = %.2 f kg/kg",c1);
9 S = c/c1;
10 printf("\n S = %.2 f",S);
11
12 printf("\n\n For concentrations in kg sucrose/kg
   water:")
13 co = c/(c+1); //concentration is in kg/kg
   solution
14 printf("\n co = %.3 f kg/kg solution",co);
15 co1 = c1/(c1 + 1); //concentration is in kg/kg
   solution
16 printf("\n co1 = %.3 f kg/kg solution",co1);
17 S = co/co1;
```

```
18 printf("\n S = %.2f ",S);
```

---

### Scilab code Exa 15.2 Increase in solubility

```
1 clear;
2 clc;
3 printf("\n Example 15.2");
4 printf("\n For barium sulphate");
5
6 function [x] = increase_barium(d)
7     y = exp((2*233*0.13)/(2*8314*298*4500*d));
8     x = y-1;
9     funcprot(0);
10 endfunction
11 i=1;
12 d = [0.5 0.05 0.005];
13 while(i<=3)
14     printf("\n Increase in solubility for barium
15         sulphate fpr particle size %f um = %f per
16         cent",d(i)*2,(increase_barium(d(i)*10^(-6))
17         *100));
18     i = i+1;
19 end
20
21 printf("\n For Sucrose");
22 function [a] = increase_sucrose(d1)
23     b = exp((2*342*0.01)/(1*8314*298*1590*d1));
24     a = b-1;
25     funcprot(0);
26 endfunction
27 j=1;
28 while(j<=3)
29     printf("\n Increase in solubility for barium
30         sulphate fpr particle size %f um = %f per
31         cent",d(j)*2,(increase_sucrose(d(j)*10^(-6))
```



```

        *100));
27     j = j+1;
28 end

```

---

### Scilab code Exa 15.3 Theoretical yield

```

1 clear;
2 clc;
3 printf("\n Example 15.3");
4 R = 322/142;
5 //The initial concentration
6 c1 = 1000/5000;           //Concentration is in
    kg Na2SO4/kg water
7 printf("\n The initial concentration c1 = %.1f kg
    Na2SO4/kg water",c1);
8 //The solubility
9 c2 = 9/100;             //solubility is in Kg
    Na2SO4/kg water
10 printf("\n The solubility c2 = %.2f Kg Na2SO4/kg
    water",c2);
11 printf("\n Initial mass of water,w1 = 5000 kg and
    the water lost by evaporation E = %f kg/kg"
    ,2/100);
12 printf("\n yield y = %d kg Na2SO4.10H2O" ,(5000*2.27)
    *[0.2-0.09*(1-0.02)]/[1-0.09*(2.27-1)]);

```

---

### Scilab code Exa 15.4 Yield of Sodium acetate

```

1 clear;
2 clc;
3 printf("\n Example 15.4");
4 //Given data:
5 //Heat of crystallisation , q = 144 kJ/kg trihydrate

```

```

6 //Heat capacity of the solution , Cp = 3.5 kJ/kg deg
  K
7 //Latent heat of water at 1.33 kN/m2,    = 2.46 MJ/
  kg
8 //Boiling point of water at 1.33 kN/m2 = 290.7 K
9 //Solubility of sodium acetate at 290.7 K, c2 =
  0.539 kg/kg water
10
11 //Equilibrium liquor temperature
12 T = 290.7+11.5;
13 printf("\n Equilibrium liquor temperature is%.1f K",
  T);
14
15 //Initial concentration
16 c1 = 40/(100-40);
17 printf("\n Initial concentration c1 = % .3f kg/kg
  water",c1);
18
19 //Final concentration
20 c2 = 0.539;
21 printf("\n Final concentration ,c2 = %.3f kg/kg water
  ",c2);
22
23 //Ratio of molecular masses
24 printf("\n Ratio of molecular masses ,R = %.2f"
  ,136/82);
25
26 E = {144*1.66*(0.667-0.539)+3.5*(353-302.2)
  *(1+0.667)*[1-0.539*(1.66-1)
  ]}/{2460*[1-0.539*(1.66-1)]-(144*1.66*0.539)};
27 printf("\n\n\n E = %.3f kg/kg water originally
  present",E);
28 printf("\n yeild y = %.3f kg/sec", (0.56*(100-40)
  /100)*1.66*[0.667-0.539*(1-0.153)
  ]/[1-0.539*(1.66-1)]);

```

---

### Scilab code Exa 15.5 Length of crystalliser

```
1 clear;
2 clc;
3 //The molecular mass of hydrate/molecular mass of
  anhydrate
4 R = 380/164;
5 printf("\n The molecular mass of hydrate/molecular
  mass of anhydrate ,R = %.2f",R);
6 //It will be assumed that the evaporation is
  negligible and that E = 0.
7 c1 = 0.23;
8 printf("\n The initial concentration , c1 = %.2f kg/
  kg solution or %.2f kg/kg water",c1,c1/(1-c1));
9 c2 = 15.5;
10 printf("\n The final concentration , c2 = %.1f kg/kg
  water or 0.155 kg/kg water",c2);
11 w1 = 0.77;
12 printf("\n In 1 kg of the initial feed solution ,
  there is 0.23 kg salt and 0.77 kg water and hence
  w1 = %.2f kg",w1);
13 y = 2.32*0.77*[0.30-0.155*(1-0)]/[1-0.155*(2.32-1)];
14 printf("\n yeild y = %.2f kg",y);
15 printf("\n In order to produce 0.063 kg/s of
  crystals , the required feed is: %.3f kg/sec"
  ,1*0.063/0.33);
16 q = 0.193*3.2*(313-298);
17 printf("\n The heat required to cool the solution %
  .1f kW",q);
18 q1 = 0.063*146.5;
19 printf("\n Heat of crystallisation = %.1f kW",q1);
20 printf("\n total heat loss = %.1f kW",q+q1);
21 printf("\n Assuming counter current flow");
22 deltaT1 = (313-293);
```

```

23 printf("\n deltaT1 = %d deg K",deltaT1);
24 deltaT2 = 298 - 288;
25 printf("\n deltaT2 = %d deg K",deltaT2);
26 deltaTlm = (deltaT1-deltaT2)/log(deltaT1/deltaT2);
27 printf("\n The logarithmic mean temperature is %.1f
      deg K",deltaTlm)
28
29 A = (q+q1)/(0.14*14.4);
30 printf("\n The heat transfer area required , A= Q/U
31 deltaTm = %.1f m^2",A);
32 printf("\n Length of heat exchanger required = %.1f"
      ,A)
33
34 printf("\n\n\n\n In practice 3 lengths , each of 3 m
      length would be specified.");

```

---

#### Scilab code Exa 15.6 Crystal production rate

```

1 clear;
2 clc;
3 printf("\n Example 15.6");
4 down_time = (4/0.00017);
5 printf("\n Draw down time = %d secs",down_time);
6 printf("\n\n\n          (a)");
7 printf("\n from a mass balance the total mass of
      solids is :")
8 cs = (6*0.6*2660*10^13)*(10^(-8)*23530)^4;
9 printf("\n cs = %d kg/m^3",cs);
10 printf("\n\n\n          (b)");
11 printf("\n The production rate = %.3f kg/sec",cs
      *0.00017);
12
13 printf("\n The crystal population decreases
      exponentially with size ")

```

```

14 x = exp((-100*10^(-6))/(10^(-8)*23530));
15 printf("%d percent",x*100);
16 printf("\n Thus 34 percent have been discharged by
    the time they reach 100 um");
17 printf("\n\n\n          (c)");
18 //If (100      90) = 10 per cent of the nuclei remain
    and grow to >100 m , then
19 t = poly([0], 't');
20 t1 = roots(t - 10^(4)/log(1/0.10));
21 printf("\n tr = %d secs",t1);
22 Q = poly([0], 'Q');
23 Q1 = roots(Q + (4/4343-0.00017));
24 printf("\n Qf = %f m^3/sec",-Q1);

```

---

#### Scilab code Exa 15.7 Vapour pressure

```

1 clear;
2 clc;
3 printf("\n Example 15.7");
4 l = poly([0], 'l');
5 l1 = roots(log(780/220) - [1*(463-433)
    ]/[8314*463*433]);
6 printf("\n v = %d kj/kmol",l1);
7 P = poly([0], 'P');
8 P1 = roots(P-220/exp(70340*(433-393)/(8.314*433*393)
    ));
9 printf("\n P = %d kN/m^2",P1);

```

---

#### Scilab code Exa 15.8 Mass sublimation rates

```

1 clear;
2 clc;
3 printf("\n Example 15.8");

```

```
4 printf("\n Vaporisation stage :");
5 Pt = 101.5;
6 Ps = 1.44;
7 Sv = 0.56*(138/29)*(1.44/(101.5-1.44));
8 printf("\n Sv = %.3f kg/sec",Sv);
9
10 printf("\n\n COndensation stage: ")
11 Ptc = 100;
12 Psc = 0.0023;
13 S = 0.56*(138/29)*(0.0023/(100-0.0023));
14 printf("\n S = %f kg/sec",S);
```

---

# Chapter 16

## Drying

Scilab code Exa 16.1 Time taken to dry the solids

```
1 clear;
2 clc;
3 printf("\n Example 16.1");
4 //For the first drying operation
5 w1 = 0.25; //in kg/kg
6 w = 0.10; //in kg/kg
7 wc = 0.15; //in kg/kg
8 we = 0.05; //in kg/kg
9 f1 = w1-we; //in kg/kg
10 fc = wc - we; //in kg/kg
11 f = w - we; //in kg/kg
12
13 //x = mA
14 function [x]=TotalDryingTime(t)
15     x = (1/t)*[(f1-fc)/fc + log(fc/f)];
16     funcprot(0);
17 endfunction
18 printf("\n      mA = %.3f kg/sec",TotalDryingTime(15)
19     );
20 //For the second drying operation
```

```

21 w12 = 0.30;           //in kg/kg
22 w2 = 0.08;           //in kg/kg
23 wc2 = 0.15;          //in kg/kg
24 we2 = 0.05;          //in kg/kg
25 f12 = w12 - we2;     //in kg/kg
26 fc2 = wc2 - we2;     //in kg/kg
27 f2 = w2 - we2;       //in kg/kg
28
29 t1=(1/TotalDryingTime(15))*[(0.25-0.10)/0.10 + log
    (0.10/0.03)];
30 printf("\n The total drying time is then %.1f ksec
    or %.2f hr ",t1,t1*1000/3600);

```

---

### Scilab code Exa 16.2 Time taken to dry the solids

```

1 clear;
2 clc;
3 printf("\n Example 16.2");
4 E = [1 0.64 0.49 0.38 0.295 0.22 0.14];
5 J = [0 0.1 0.2 0.3 0.5 0.6 0.7];
6 plot(J,E,rect=[0,1,0,1]);
7 xtitle("Plot for drying data","J = kt/L^2","E");
8
9 //For the 10 mm strips
10 mi = (0.28 - 0.07); //Initial free moisture
    content in kg/kg
11 mf = (0.13-0.07); //Final free moisture
    content in kg/kg
12 //at
13 t = 25; //time is in ksecs
14 E = (0.06/0.21);
15 //at E = 0.286 ,J = 0.52 from the plot of given data
    and J = kt/L^2
16 k = poly([0], 'k');
17 k1 = roots(0.52 - (k*t)/(10/2)^2);

```



```

18 printf("\n k = %f",k1);
19
20 //for the 60 mm planks
21 m1i = (0.22 - 0.07);           //Initial free moisture
    content in kg/kg
22 m1f = (0.13 - 0.07);           //Final free moisture
    content in kg/kg
23 E = (m1f/m1i);
24 //at E = 0.20 from yhe plot of the given data J =
    0.63 and J = kt/L^2
25 t1 = 0.63*(60/2)^(2)/k1;
26 printf("\n t1 = %d ksecs or %.1f days",t1,(t1
    *1000/(3600*24)));

```

---

### Scilab code Exa 16.3 Mass flow rate of dry air

```

1 clear;
2 clc;
3 printf("\n example 16.3");
4 //(a) Air
5 //G kg/s dry air enter with 0.006G kg/s water vapour
    and hence the heat content of this stream=
6 //[(1.00G) + (0.006G 2.01)](385 273) = 113.35
    G kW
7
8 //(b) Wet solid
9 //0.125 kg/s enter containing 0.40 kg water/kg wet
    solid , assuming the moisture is expressed on a
    wet basis.
10 flowWater = (0.125*0.40);           //in kg/sec
11 flowDrySolid = (0.125-0.050);       //in kg/sec
12 //Hence heat content of this stream
13 q = [(0.050*4.18)+(0.075*0.88)]*(295-273);
14 printf("\n The heat content of this stream = %.2f kW
    ",q);

```

```

15
16 //Heat out
17 //(a) Air
18 //Heat in exit air = [(1.00 G) + (0.006 G    2.01)
    ](310    273) = 37.45G kW.
19 fd = 0.075;           //mass flow rate of dry
    solids in kg/sec
20 w = 0.05*0.075/(1+0.05); //water in the dried
    solids leaving in kg/secs
21 we = (0.050 - w);    //The water evaporated
    into gas stream in kg/secs
22
23 //Assuming evaporation takes place at 295 K,then:
24 qout = 0.0464*[2.01*(310-295)+2449+4.18*(295-273)];
25 printf("\n Heat in the water vapour = %.1f kW",qout)
    ;
26
27 //the total heat in this stream = (119.30 + 37.45G)
    kW.
28 //(b) Dried solids
29
30
31 //The dried solids contain 0.0036 kg/s water and
    hence heat conten    t of this stream is:
32 q1 = [(0.075*0.88)+(0.0036*4.18)/(305-273)];
33 printf("\n The dried solids contain 0.0036 kg/s
    water and hence heat content of this stream is
    : %.2f kW",q1);
34
35
36 //(c) Losses
37 //These amount to 20 kJ/kg dry air or 20m kW.
38 //Heat Balance
39 G = poly([0], 'G');
40 G1 = roots(113.35*G + 6.05 - 119.30 - 37.45*G - 2.59
    - 20*G);
41 printf("\n G = %.2f kg/secs",G1);
42 printf("\n Water in the outlet stream %.4f kg/secs"

```

```

    ,0.006*2.07+0.0464);
43 printf("\n The humidity H = %.4f kg/kg dry air"
    ,0.0588/2.07);

```

---

#### Scilab code Exa 16.4 Approximate drying time

```

1 clear;
2 clc;
3 printf("\n Example 16.4");
4 //In 100 kg of feed
5
6 //mass of water =
7 mw = 100*30/100; //mass of water
8 //mass of dry solids =
9 md = 100-30; //mass of dry
10 //solids
11 //and:
12 //For b kg water in the dried solids: 100b/(b + 70)
13 // = 15.5
14 b = poly([0], 'b');
15 b1 = roots(100*b - 15.5*(b+70));
16 printf("\n water in the product ,b = %.1f kg",b1);
17 //Initial water content
18 w1 = 30/70; //Intial moisture content
19 // in kg/kg dry solids
20 //Final moisture content
21 w2 = (12.8/70); //Final moisture content
22 // in kg/kg dry solids
23 //water to be removed
24 w3 = (30-12.8); //water to be removed in
25 // kg

```

```

24 //Surface of drying
25 S = (0.03*70);           //Surface for drying in m
    ^2
26 rate = (0.0007*2.1);    //Rate of drying in kg/sec
27
28 //As the final moisture content is above the
    critical value, all the drying is at this
    constant rate and the time of drying is:
29 t = 17.2/0.00147;
30 printf("\n As the final moisture content is above
    the critical value, all the drying is at this
    constant rate and the time of drying is: %d ksecs
    or %.2f hr",t,t/3600);

```

---

#### Scilab code Exa 16.5 Proposed diameter and length

```

1 clear;
2 clc;
3 printf("\n Example 16.5");
4 //Inlet air temperature
5 Tair = 400;           //Inlet air temperature in
    kelvins
6 H = 0.01;           //Humidity is in kg/kg dry
    air
7 //*therefore wet bulb temperature =
8 Twetbulb = 312;    //Inlet wet-bulb temperature
9 NTU = 1.5;         //Number of transfer units
10
11 //Then for adiabatic drying the outlet air
    temperature,To is given by
12 To = poly([0], 'To');
13 To1 = roots(exp(1.5)*(To-312)-(400-312));
14 printf("\n For adiabatic drying the outlet air
    temperature = %.1f K",To1);
15

```

```

16 //Solids outlet temperature will be taken to be
    maximum allowable,325K
17 //From the steam tables in the Appendix, the latent
    heat of vaporisation of water at 312 K is2410 kJ/
    kg. Again from steam tables, the specific heat
    capacity of water vapour = 1.88 kJ/kg K and that
    of the solids will be taken as 2.18 kJ/kg K.
18
19 //For a mass flow of solids of 0.35 kg/s and inlet
    and outlet moisture contents of 0.15 and 0.005 kg
    /kg dry solids respectively, the mass of water
    evaporated = 0.35(0.15 - 0.005) = 0.0508 kg/s.
20
21 //For unit mass of solids, the heat duty includes:
22 //Heat to the solids
23 qsolids = 2.18*(325-300); //heat to
    solids in kJ/kg
24 //Heat to raise the moisture to the dew point
25 qdew = (0.15*4.187*(312-300)); //in kJ/kg
26 //Heat of vaporisation
27 qvap = 2410*(0.15-0.005); //in kJ/kg
28 //Heat to raise remaining moisture to solids
    outlet temperature
29 qremaining = (0.05*4.187)*(325-312);
30 //Heat to raise evaporated moisture to the
    air outlet temp.
31 qevapo = (0.15-0.005)*1.88*(331.5-312);
32 qtotal = qsolids + qdew + qvap + qremaining +
    qevapo;
33 printf("\n Total heat = %.1f kJ/kg or %d kW",
    qtotal, qtotal*0.35);
34
35 //The humid heat of entering air is 1.03 kJ/kg K
36 //G1 (1 +H1) = Q/Cp1(T1 - T2)
37 //where: G1 (kg/s) is the mass flowrate of
    inlet air,
38 //H1 (kg/kg) is the humidity of inlet air,
39 //Q (kW) is the heat duty,

```

```

40 //Cp1 (kJ/kg K) is the humid heat of inlet
    air
41 //and: T1 and T2 (K) are the inlet and outlet
    air temperatures respectively.
42 G1 = poly([0], 'G1');
43 G = roots(G1*(1+0.01) -146/(1.03*(400-331.5)))
    ;
44 printf("\n Mass flow rate of inlet air = %.2f
    kg/secs", G);
45 printf("\n Mass flow rate of dry air ,Ga = %
    .2f kg/secs", G/1.01);
46 printf("\n the humidity of the outlet air H2
    = %.4f kg/kg" ,0.01+0.0508/2.05);
47
48 //At a dry bulb temperature of 331.5 K, with a
    humidity of 0.0347 kg/kg, the wet-bulb
    temperatureof the outlet air , from Figure 13.4 in
    Volume 1, is 312 K, the same as the inlet , which
    is the case for adiabatic drying.
49
50 //The dryer diameter is then found from the
    allowable mass velocity of the air and the
    entering air flow and for a mass velocity of 0.95
    kg/m^2.secs , the cross sectional area of the
    dryer is
51 printf("\n The cross sectional area of the dryer is
    %.2f m^2" ,2.07/0.95);
52 printf("\n The equivalent diameter of the dryer = %
    .2f m" ,[(4*2.18)/%pi]^0.5);
53
54 //With a constant drying temperature of 312 K:
55 //at the inlet
56 deltaT1 = 400-312; //inlet
    temperature is in deg K
57 //at the outlet
58 deltaT2 = 331.5-312; //outlet
    temperature is in deg K
59 Tlogmean = (deltaT1 - deltaT2)/log(deltaT1/

```

```

        deltaT2);
60     printf("\n Logarithmic mean temperature
           difference = %.1f deg K",Tlogmean);
61 //The length of the dryer , L is then:  $L = Q / (0.0625$ 
            $D G^{(0.67)} * T_m)$ 
62     //where D (m) is the diameter
63     //and G(kg/m2.secs) is the air mass velocity.
64     L = 146/[0.0625*(%pi)*1.67*(0.95)^(0.67)*45.5];
65     printf("\n The length of the dryer = %.1f m",L);
66     printf("\n Length/diameter ratio = %d "
           ,10.1/1.67);

```

---

#### Scilab code Exa 16.6 Specified diameter of the bed

```

1  clear;
2  clc;
3  printf("\n Example 16.6");
4  H = 0.036; //Humidity is in kg/kg at
           811 K
5  //Taking R as 90 per cent and P as 101.3 kN/m2, then
           , for assumed values of Tb of 321, 333 and 344 K
6     //Pw = 13,20 and 32 kN/m2, respectively
7     //G = 27.8, 12.9 and 6.02 kg/s, respectively.
8
9  //for Tb = 321, 333 and 344 K,
10 //G = 7.16, 7.8 and 7.54 kg/s respectively.
11 Tb = [321 333 344];
12 G1 = [27.8 12.9 6.02]; //Temperature is in
           kelvins
13 G = [7.16 7.8 7.54]; //flow rate in kg/
           secs
14 plot2d(Tb,G,style=3);
15 plot2d(Tb,G1,style=2);
16 xtitle("Temperature vs Flow rate", "Temperature Tb(K)
           ", "Flow rate G(kg/secs)");

```

```
17
18
19 //Plotting G against Tb for each equation on the
    same axis , then
20 Go = 8.3; //Gas flow rate is
    in kg/secs
21 Tb = 340; //temperature is
    in Kelvins
22 uf = 0.61; //velocity is in m
    /secs
23
24 D = sqrt(340*(8.3+(1.58*1.26))/(278*0.61));
25 printf("\n D = %.2 f m",D);
```

---



# Chapter 17

## Adsorption

Scilab code Exa 17.1 Comparison of estimates with the geometric surfaces

```
1 //Example 17.1
2 clear;
3 clc;
4 printf("\n Example 17.1");
5 //For 1 m3 of pellet with a voidage    , then
6     //Number of particles = (1          )/(    /6)(15
7         10^  9 )^3
8     //Surface area per unit volume = (1          )
9         (15    10  9 )2/(    /6)(15    10^  9 )^3
10    // = 6(1          )/(15    10^(  9 )) m^2/m^3
11    //1 m^3 of pellet contains 2290(1-    )kg
12    solid and hence:
13    specific_surface = 6/(15*10^(-9)*2290);
14    printf("\n Specific surface = %.3f*10^5 m^2/
15        kg ",specific_surface*10^(-5));
16 // (a) Using the BET isotherm
17
18     //y = (P/Po)/V *10^(-6)
19     y = [1500 2660 3576 4283 4613 4615];
20     //y1 = (P/Po)/V *(1-(P/Po))
```

```

17     y1 = [1666 3333 5109 7138 9226 11358];
18     //x = P/Po
19     x = [0.1 0.2 0.3 0.4 0.5 0.6];
20     plot2d(x,y,style = 2);
21     xtitle("","P/Po","(P/Po)/(V *10^(-6)) or (P/
22     Po)/(V *(1-(P/Po)))");
23     plot2d(x,y1,style=3);
24     legend("Langmuir isotherm","(BET) isotherm")
25     ;
26     //intercept , 1/VB = 300, and slope , (B      1)
27     //VB = 13,902
28     B = (13902/300)+1;
29     printf("\n B = %.2f ",B);
30     V = 1/(300*47.34);
31     //Total surface area
32     S = [(70.4*10^(-6)*808*6.2*10^(26)
33     *0.162*10^(-18))]/28;
34     printf("\n Total surface area = %.3f *10^5 m
35     ^2/kg",S*10^(-5));
36
37 // (b) Using the Langmuir form of the isotherm
38
39 //Slope 1/V = 13,902
40 V = 1/13902;
41 printf("\n V = %.1f*10^(-6)m^3/kg",V*10^6);
42 //which agrees with the value of the BET
43 isotherm
44 //It may be noted that areas calculated from the
45 isotherm are some 20 per cent greater than the
46 geometric surface , probably due to the
47 existence of some internal surface within the
48 particles

```

---

**Scilab code Exa 17.2** Applicability of various equilibrium theories

```

1 //Example 17.2
2 clear;
3 clc;
4 printf("\n Example 17.2");
5 //(a)
6 //For n = 1,
7 //(P/P0)/V= (P/P0)V1+ 1/B2V1
8 //where V is the gas phase volume equivalent to
   the amount adsorbed
9 //x = (P/Po)
10 x = [0.05 0.10 0.15 0.20 0.25 0.30 0.35 0.40];
11 //y = (P/Po)/V in kg/m^3
12 y = [0.76 1.35 1.85 2.27 2.66 2.94 3.21 3.42];
13 //y1 = (P/Po)/(V*(1-P/Po)) in kg/m^3
14 y1 = [0.80 1.50 2.18 2.88 3.55 4.20 4.94 5.73];
15 plot(x,y,"o-");
16 plot(x,y1,"+-");
17 xtitle("","(P/Po)","(P/Po)/V or (P/Po)/(V*(1-P/
   Po))");
18 legend("B.E.T.", "Langmuir");
19
20 //The data, which are plotted as (P /P 0)/V
   against P/P0 may be seen to conform to a
   straight line only at low values of P/P0,
   suggestin g that more than one layer of
   molecules isadsorbed.
21 Slope = 12.56;
22 V1 = 1/12.56;
23 //The surface area occupied by this absorbed
   volume
24 S = V1*6.02*10^(26)*0.162*10^(-18)/24;
25 printf("\n S = %d m^2/kg",S);
26
27
28 //(b)
29 //P/P0/V*(1 P/P0)= 1/V1*B2+( B 2 1)/(V1*B2)*(P
   /P 0)
30 //y2 = (P/Po)

```

```

31 //x2 = 1V^2
32 y2 = [0.05 0.10 0.15 0.20 0.25 0.30 0.35 0.40 0.50
        0.60 0.87 0.80];
33 x2 = [230 183 152 129 113 96 84 73 53 37 26 20];
34 xset('window',1);
35 plot(x2,y2,"o-");
36 xtitle("Harkins-Jura Plot", "1/V^2(kg^2/m^6)", "P/Po
        ");

```

---

### Scilab code Exa 17.3 Length of the bed

```

1 //Example 17.3
2 clear;
3 clc;
4 printf("\n Example 17.3");
5 //For case (a):
6 //Cs = 10C which represents a linear isotherm.
7 //All concentrations move at the same velocity. If
  z0=0 at t = 0 for a ll concentrations, the
  adsorption wave propagates as a step change
  from the inlet to the outlet concentration.
8 //f (Cs ) = 10
9 //u = 1 10 ^(-4) /[( /4)*(0.15^(2)* )] m/s
10 //where is the intergranular voidage = 0.4
11 e = 0.4;
12 m = e/(1-e);
13 printf("\n m=%f",m);
14 t = 3600; //time is in secs
15 z = [(4*10^(-4))/(%pi*(0.15^2)*0.4)
        ]*[3600/(1+10*(0.6/0.4))];
16 printf("\n z = %.2 f m",z);
17
18
19 //It may be noted that, when the adsorption wave
  begins to emerge from the bed, the bed is

```

```

    saturated in equilibrium with the inlet
    concentration.
20 //Hence:  $uA tC_0 = zA [ C_0 + (1 - C_0/C_s)Cs ]$ 
21
22 //For case (b):
23 //Cs = 3C0.3 which represents a favourable isotherm.
24 //As C increases , f(C) decreases and points of
    higher concentrations are predicted to move a
    greater distance in a given time than lower
    concentrations. It is not possible for points of
    higher concentrationsto overtake lower
    concentrations , and if  $z_0=0$  for all
    concentrations ,the adsorption wave will propagate
    as a step change similar to case a.
25
26 //Hence:  $z = ut / [1+(1/m)(Cs/Co) ]$ 
27  $Co = 0.003;$  //in kmol/m3
28  $z = [(4*10^{(-4)}*5/((\%pi)*0.4*0.15^2))$ 
     $]*[3600/(1+(0.6/0.4)*(3/Co^{0.7}))];$ 
29 printf(" \n z = %.2 f m",z);
30
31 //For case (c):
32 //Cs =  $(10^{(4)})*(C^2)$  which represents an
    unfavourable isotherm.
33 //f(C)= $2* 10^4* C.$ 
34 //As C increases , f (C) increases such that , in a
    given time , z for lower concentrations is greater
    than for higher concentrations. Following the
    progress of the breakpoint concentration ,C =
    0.003 kmol/m3, then:
35 //f(0.003) = 60
36  $z_1 = [4*10^{(-4)}/(\%pi*0.15^{(2)}*0.4)$ 
     $]*[3600/(1+(0.60/0.40)*60)];$ 
37 printf(" \n For case (c)");
38 printf(" \n z = %.2 f m",z1);
39
40
41 //At breakpoint ,the bed is far from saturated and:

```

```

42 saturation = 100*[(4*10(-4)*3600)/(%pi*(0.15(2)
    *0.4))]/[0.55*(1+(0.6/0.4)*(9/0.03))];
43 printf("\n saturation = %.1f per cent",saturation)
    ;

```

---

### Scilab code Exa 17.5 Moving bed adsorption design

```

1 clear;
2 clc;
3 printf("\n Example 17.5");
4 //let x = [f(C)]mean
5 x = (8.4*10(-2)*1266)/(2.67*10(-3)*1.186);
6
7 //e is porosity
8 e = 0.47;
9 m = e/(1-e);
10 printf("\n m = %.2 f",m);
11
12 //The velocity uc with which the adsorption wave
    moves through the column may be obtained from
    equation 17.79
13 uc = 6.2*10(-6);
14 density = 1266; //density is in kg/m
    ^3
15 Gs = uc*(1-e)*density;
16 printf("\n Gs = %.2 f*10(-3) kg/m2.sec ",Gs*10(3));
17
18
19 //From overall balance:
20 //We have given eq: Gs1(0.084-0)=0.129(0.00267)
21 //On solving above eq :
22 Gs1 = poly([0], 'Gs1');
23 Gs2 = roots(Gs1*(0.084-0)-0.129*(0.00267));
24 printf("\n Gs1 = %.2 f*10(-3) kg/m2.sec",Gs2
    *1000);

```

```

25
26 //Part(b)
27 //time ta for the adsorption zone to move its own
    length za is given by:
28 za = 0.185;
29 ta =(za/uc)/3600;
30 printf("\n ta = %.1f hrs",ta);
31
32 //The time taken for a point at a distance z into
    the zone to emerge is given by:
33 //t = (z1/za)ta
34 yr = [0.0001 0.0002 0.0006 0.0010 0.0014 0.0018
    0.0022 0.0024];
35 yre = [0.00005 0.0001 0.00032 0.00062 0.00100
    0.00133 0.00204 0.00230];
36 I=0;
37 I1 = 0;
38 i=1;
39 printf("\nI= %.1f",I);
40 while i <=8
41     y(i)=1/(yr(i)-yre(i));
42     if i>1 then
43         I = I + (yr(i)-yr(i-1))*(y(i)+y(i-1))/2;
44         printf("\n I=%.1f",I);
45     end
46     i=i+1;
47 end
48 z_za = [0 0.137 0.362 0.473 0.560 0.674 0.852
    1.000];
49 t = (z_za)*ta;
50 j=1;
51 while j<=8
52     printf("\ntime = %.1f h",t(j));
53     j=j+1;
54 end

```

---

# Chapter 18

## Ion Exchange

Scilab code Exa 18.1 Prediction of time t against xs values

```
1 clear;
2 clc;
3 printf("\n Example 18.1");
4 Cse=0.132;          //in kg/kg
5 Cs = [0.091 0.097 0.105 0.113 0.125 0.128 0.132];
6 C = Cs/Cse;
7 C1 = 1-C;
8 C2 = 1-C^2;
9 t = [2 4 10 20 40 60 120];
10 //xset('window',1);
11 //
12 plot(t,C1,t,C2);
13 xtitle('1-(Cs/Cs* vs t (min)', '1-(Cs/Cs*)', 't (min)');
14 legend("1-(Cs/Cs*)", "1-(Cs/Cs*)^2");
15
16 //From the plot  $\frac{d^2C}{dt^2} = 0.043$ 
17 //For a pellet of twice the radius, that is  $r = 2r_i$ 
18 Slope = -0.043/4;
19 printf("\n Slope = %.3f",Slope);
20
21 //Thus, when the radius = 2r_i
```



```

22 function [x]=equation1(t)
23     x = 1-(6/(%pi)^2)*exp(-Slope*t);
24     funcprot(0);
25 endfunction
26
27 //CS/CS*=[1      exp(   D R /tri^2)]^0.5
28 // DR/ri^2 = 0.04
29 //For a pellet twice the size
30
31 function [x1]=equation2(t)
32     x1 = [1-exp(-0.01*t)]^0.5;
33     funcprot(0);
34 endfunction
35
36 printf("\n t (min)                Cs(kg/kg)          ");
37 printf("\n                equation(i)
38     equation(ii)  ");
38 t = [4 20 60];                //t is in min
39 i=1;
40 while i<=3
41 printf("\n %f                %f
42     %f                ",t(i),Cse*equation1(t(i)),Cse*equation2(
43     t(i)));
42 i=i+1;
43 end

```

---

### Scilab code Exa 18.2 Concentration of HNO<sub>3</sub> in solution

```

1 clear;
2 clc;
3 printf("\n Example 18.2");
4 c_NaNO3 = 2;                //per cent by mass
5 printf("\n Concentration of NaNO3 is %.3f kg/m^3", (
6     c_NaNO3/85)*(103/100));

```

```

7 //HNO3 (Molecular weight = 63 kg/kmol)
8 //Concentration = p per cent
9 //Concentration = (10p/63)(1030/1000)= 0.163p kg/m3
10 //In the solution: xNa+ = 0.242/(0.242 + 0.163p)
11 //For univalent ion exchange
12 //yNa+/(1 - yNa+) = KNa+H + [xNa+ /(1 - xNa+)]
13
14 yNa = 0.1;
15 K_NaH = 2/1.3;
16 p = poly([0], 'p');
17 p1 = roots((0.1/0.9)*[0.163*p*(0.242+0.163*p)
    ]-1.5*[0.242*(0.242+0.163*p)]);
18 printf("\n p = %d percent", p1(1));

```

---