

Scilab Textbook Companion for  
Materials Science and Engineering - A First  
Course  
by V. Raghavan<sup>1</sup>

Created by  
Vareesh Pratap  
B.Tech  
Mechanical Engineering  
Madan Mohan Malaviya University of Technology  
College Teacher  
None  
Cross-Checked by  
Aviral Yadav

June 2, 2016

<sup>1</sup>Funded by a grant from the National Mission on Education through ICT, <http://spoken-tutorial.org/NMEICT-Intro>. This Textbook Companion and Scilab codes written in it can be downloaded from the "Textbook Companion Project" section at the website <http://scilab.in>

# Book Description

**Title:** Materials Science and Engineering - A First Course

**Author:** V. Raghavan

**Publisher:** Prentice Hall, India

**Edition:** 5

**Year:** 2007

**ISBN:** 9788120324558

Scilab numbering policy used in this document and the relation to the above book.

**Exa** Example (Solved example)

**Eqn** Equation (Particular equation of the above book)

**AP** Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

# Contents

List of Scilab Codes	4
2 Equilibrium and Kinetics	5
3 Crystal Geometry and Structure Determination	7
4 Atomic Structure and Chemical Bonding	11
5 Structure of Solid	13
6 Crystal Imperfection	17
7 Phase Diagram	21
8 Diffusion in Solids	24
9 Phase Transformation	28
10 Elastic Anelastic and Viscoelastic Behaviour	32
11 Plastic Deformation and Creep in Crystalline Materials	34
12 Fracture	38
13 Oxidation and Corrosion	41
14 Conductors and Resistors	42
15 Semiconductors	44

<b>16 Magnetic Materials</b>	<b>45</b>
<b>17 Dielectric Materials</b>	<b>48</b>

# List of Scilab Codes

Exa 2.1	Calculate the entropy increase . . . . .	5
Exa 2.2	Calculation of fraction of atoms with energy equal to or greater than 1eV at temperatures . . . . .	5
Exa 3.2	Calculate effective number of lattice point in three cubic space lattice . . . . .	7
Exa 3.7	Determine interplanar spacing and miller indices . . . .	8
Exa 3.8	Determine structure and lattice parameter of material . . . . .	9
Exa 4.1	Calculate frequency and wavelength of radiation . . . . .	11
Exa 4.3	Reconcile the difference of energy . . . . .	12
Exa 4.4	Calculation of fraction of hydrogen bonds which breaks during ice melting . . . . .	12
Exa 5.1	Calculate packing efficiency and density of diamond . . . . .	13
Exa 5.3	Calculate the cation to anion ratio for an ideally close packed HCP crystal . . . . .	14
Exa 5.4	find the size of largest sphere that can fit into a tetrahedral void . . . . .	14
Exa 5.5	Find critical radius ratio for triangular coordination . . . . .	15
Exa 5.6	Calculate density of MgO . . . . .	15
Exa 6.1	Find equilibrium concentration of vacancies in metals at given temperature . . . . .	17
Exa 6.2	Compute the line energy of dislocation . . . . .	18
Exa 6.4	Calculation of down climb of crystal on heating . . . . .	18
Exa 6.5	Calculate surface energy of copper crystal of type 111 . . . . .	19
Exa 6.6	Compute the angle at the bottom of groove of a boundary . . . . .	20
Exa 7.1	Find degrees of freedom of a system of two components . . . . .	21
Exa 7.2	Find minimum number of component in system . . . . .	21
Exa 7.3	Calculate amount of pure water that can be extracted from sea water . . . . .	22

Exa 7.5	Calculate proeutectoid ferrite and eutectoid ferrite in 6 tenth percent steel . . . . .	22
Exa 8.1	Calculate the rate at which hydrogen escapes through the walls of the steel tank . . . . .	24
Exa 8.2	Calculate maximum time till which material can be kept at 550 degree Celsius . . . . .	24
Exa 8.3	Calculate minimum depth up to which post machining is to be done . . . . .	25
Exa 8.4	Calculate time required to get required boron concentration . . . . .	26
Exa 8.5	Calculate ratio of cross sectional areas . . . . .	27
Exa 9.1	Calculate the critical free energy of nucleation of ice from water and critical radius . . . . .	28
Exa 9.2	Calculate the change in delta f required to increase nucleation rate . . . . .	29
Exa 9.4	Calculate delta f of heterogeneous as a fraction of delta f of homogeneous . . . . .	30
Exa 9.6	Calculate the free energy change during recrystallization . . . . .	30
Exa 10.1	Estimate Youngs modulus of material . . . . .	32
Exa 10.2	Calculation of stress in fibers . . . . .	32
Exa 10.3	Estimate diffusion coefficient . . . . .	33
Exa 11.2	Calculate the stress required to move the dislocation at given temperature . . . . .	34
Exa 11.3	Calculate the dislocation density in copper . . . . .	35
Exa 11.4	Find the yield stress for a grain size of ASTM 9 . . . . .	36
Exa 11.5	Estimate the change in yield strength . . . . .	36
Exa 12.1	Estimate fracture strength . . . . .	38
Exa 12.2	Estimate the brittle fracture strength at low temperatures . . . . .	38
Exa 12.3	Estimate the temperature at which the ductility of brittle transition occurs at given strain rates . . . . .	39
Exa 13.2	Calculation of required quantity of magnesium . . . . .	41
Exa 14.1	Calculate energy difference . . . . .	42
Exa 14.2	Calculate conductivity of copper at 300 K . . . . .	42
Exa 14.3	Estimation of resistivity due to impurity scattering of 1 percent of Nickel in copper lattice . . . . .	43
Exa 15.1	Calculate intrinsic carrier density . . . . .	44

Exa 16.1	Calculate the net magnetic moment per iron atom in crystal . . . . .	45
Exa 16.2	Comparison of saturation temperatures . . . . .	46
Exa 16.4	Calculation of hysteresis loss . . . . .	46
Exa 16.5	Calculation of eddy current loss at normal voltage and frequency . . . . .	47
Exa 17.1	Calculation of relative dielectric constant . . . . .	48
Exa 17.2	Calculate the polarization of a BaTiO <sub>3</sub> crystal . . . . .	48



# Chapter 2

## Equilibrium and Kinetics

Scilab code Exa 2.1 Calculate the entropy increase

```
1 // Calculate the entropy increase
2 clc
3 del_h = 6.02 // Heat added in kJ/mol
4 t_m = 273.15 // mean temperature in kelvin
5 printf("\n Example 2.1")
6 del_s = del_h*1e3/t_m
7 printf("\n Increase in entropy is %.2f J mol-1 K-1
   ",del_s)
```

---

Scilab code Exa 2.2 Calculation of fraction of atoms with energy equal to or greater than 1eV at temperatures

```
1 // Calculation of fraction of atoms with energy
   equal to or greater than 1eV at temperatures
2 clc
3 E = 1 // energy in electron volt
4 e = 1.6e-19 // charge on electron
```

```
5 k = 1.38e-23 // constant
6 t1 = 300 // temperature in K
7 t2 = 1500 // temperature in K
8 printf("\n Example 2.2")
9 printf("\n\n Part A:")
10 n_N = exp(-(e/(k*t1)))
11 printf("\n Fraction of atoms with energy equal to or
    greater than 1eV at temperature %d K is %.2e ",
    t1,n_N) // numerical value of answer in book is 2
    e-17
12 printf("\n\n Part B:")
13 n_N = exp(-(e/(k*t2)))
14 printf("\n Fraction of atoms with energy equal to or
    greater than 1eV at temperature %d K is %.2e ",
    t2,n_N) // numerical value of answer in book is
    4.3e4
```

---

## Chapter 3

# Crystal Geometry and Structure Determination

**Scilab code Exa 3.2** Calculate effective number of lattice point in three cubic space lattice

```
1 // Calculate effective number of lattice point in
   three cubic space lattice
2 clc
3 sc_n = 1/8 // sharing of one lattice point in a unit
   cell
4 sc_N = 8 // Number of lattice points in Simple cubic
5 bcc_n_e = 1/4 // sharing of one edge lattice point in
   a BCC
6 bcc_N_e = 4 // Number of edge lattice point in a BCC
7 bcc_n_c = 1 // sharing of one body center lattice
   point in a BCC
8 bcc_N_c = 1 // Number of body center lattice point in
   a BCC
9 fcc_n_e = 1/8 // sharing of one corner lattice point
   in a FCC
10 fcc_N_e = 8 // Number of corner lattice point in a
   FCC
11 fcc_n_f = 1/2 // sharing of one face center lattice
```

```

    point in a FCC
12 fcc_N_f = 6// Number of face center lattice point in
    a FCC
13 printf("\n Example 3.2 ")
14 sc_net = sc_n*sc_N
15 bcc_net = bcc_n_e*bcc_N_e+bcc_n_c*bcc_N_c
16 fcc_net = fcc_n_e*fcc_N_e+fcc_n_f*fcc_N_f
17 printf("\n Effective number of lattice points are as
    :")
18 printf("\n\n Space lattice \t Abbreviation \t
    Effective number of lattice point in unit cell \n
    ")
19 printf("\n Simple cubic \t\tSC \t\t\t\t %d\n Body
    center cubic\tBCC \t\t\t\t %d\n Face centered
    cubic\tFCC \t\t\t\t %d ",sc_net,bcc_net, fcc_net)

```

---

### Scilab code Exa 3.7 Determine interplanar spacing and miller indices

```

1 // Determine Interplanar spacing and miller indices
2 clc
3 a = 3.16 // lattice parameter in angstrom
4 l1 = 1 // line number
5 l2 = 2 // line number
6 l3 = 3 // line number
7 l4 = 4 // line number
8 theta1 = 20.3 // angle for line 1
9 theta2 = 29.2// angle for line 2
10 theta3 = 36.7// angle for line 3
11 theta4 = 43.6// angle for line 4
12 n = 1 // order
13 lambda = 1.54 // wavelength in angstrom
14 printf("\n Example 3.7")
15 d1 = lambda/(2*sin(theta1*%pi/180))

```

```

16 d2 = lambda/(2*sin(theta2*pi/180))
17 d3 = lambda/(2*sin(theta3*pi/180))
18 d4 = lambda/(2*sin(theta4*pi/180))
19 x1 = a^2/d1^2
20 x2 = a^2/d2^2
21 x3 = a^2/d3^2
22 x4 = a^2/d4^2 // where x is function of h,k and l
23 printf("\n Interplanar spacing is %.3f angstrom ",
        x1) // answer in book is 2.220 angstrom
24 if floor(x1) == 2 then
25     printf("\n\n For a^2/d^2 = %d \t Reflection
        plane is {110}",x1)
26 end
27
28 if floor(x2) == 4 then
29     printf("\n For a^2/d^2 = %d \t Reflection
        plane is {200}",x2)
30 end
31
32 if floor(x3) == 6 then
33     printf("\n For a^2/d^2 = %d \t Reflection plane
        is {211}",x3)
34 end
35
36 if floor(x4) == 8 then
37     printf("\n For a^2/d^2 = %d \t Reflection plane
        is {220}",x4)
38 end

```

---

**Scilab code Exa 3.8** Determine structure and lattice parameter of material

```
1 // Determine structure and lattice parameter of
```

```

material
2  clc
3  d = 114.6 // diameter of power camera in angstrom
4  lambda = 1.54 // wavelength in angstrom
5  s1 = 86
6  s2 = 100
7  s3 = 148
8  s4 = 180
9  s5 = 188
10 s6 = 232
11 s7 = 272
12 printf("\n Example 3.8")
13 R = d/2 // Radius
14 if R==57.3 then
15     k = 1/4 // Bragg angle factor
16 end
17 a1 = (sin(s1*k*%pi/180))^2
18 a2 = (sin(s2*k*%pi/180))^2
19 a3 = (sin(s3*k*%pi/180))^2
20 a4 = (sin(s4*k*%pi/180))^2
21 a5 = (sin(s5*k*%pi/180))^2
22 a6 = (sin(s6*k*%pi/180))^2
23 a7 = (sin(s7*k*%pi/180))^2
24 c = 22 // constant to convert into integral number
25
26 printf("\n Within experimental error , values are as
in integral ratio are as: \n %d:%d:%d:%d:%d:%d:%d
",ceil(c*a1),ceil(c*a2),ceil(c*a3),ceil(c*a4),
ceil(c*a5),ceil(c*a6),ceil(c*a7))
27 printf("\n So, this structure is FCC and material is
copper with 3.62 angstrom lattice parameter")

```

---

## Chapter 4

# Atomic Structure and Chemical Bonding

Scilab code Exa 4.1 Calculate frequency and wavelength of radiation

```
1 // Calculate frequency and wavelength of radiation
2 clc
3 E = 1.64e-18 // energy difference between two states
   in J
4 h= 6.626e-34 // planks constant
5 c = 2.998e8 // speed of light in m/s
6 printf("\n Example 4.1")
7 nu = E/h
8 lambda = c/nu
9 printf("\n Frequency of emitted radiation is %.2e Hz
   ",nu)
10 printf("\n Wavelength of emitted radiation is %.2e m
   \n\t \tor\t\t %d angstrom",lambda,lambda*1e10)//
   answer in book is 1210 angstrom
```

---

**Scilab code Exa 4.3** Reconcile the difference of energy

```
1 // Reconcile the difference of energy
2 clc
3 e_a = 713 // enthalpy of atomization in kJ/mol
4 e_b = 347 // bond energy in kJ/mol
5 a = 4 // total number of atoms in single crystal
   structure
6 b = 2 // number of atoms in a bond
7 printf("\n Example 4.3")
8 k = a/b // effective number of bond per atom
9 e = k*e_b
10 printf("\n %d kJ should be the enthalpy of
   atomization of diamond", e)
11 printf("\n However, %d kJ is very close to %d kJ", e,
   e_a)
```

---

**Scilab code Exa 4.4** Calculation of fraction of hydrogen bonds which breaks during ice melting

```
1 // Calculation of fraction of hydrogen bonds which
   breaks during ice melting
2 clc
3 del_h = 6.02 // enthalpy of fusion in kJ/mol
4 n = 2 // number of hydrogen atom in 1 water atom
5 del_b = 20.5 // hydrogen bond energy in kJ/mol
6 printf("\n Example 4.4")
7 f = del_h/(n*del_b)
8 printf("\n Fraction of hydrogen bonds which broken
   is %.2f", f)
```

---



# Chapter 5

## Structure of Solid

Scilab code Exa 5.1 Calculate packing efficiency and density of diamond

```
1 // Calculate packing efficiency and density of
  diamond
2 clc
3 n_c = 1/8 // sharing of corner atom in a unit cell
4 N_c = 8 // Number of corner atoms in unit cell
5 n_b = 1 // sharing of body centered atom in a unit
  cell
6 N_b = 4 // Number of body centered atoms in unit
  cell
7 n_f = 0.5 // sharing of face centered atom in a unit
  cell
8 N_f = 6 // Number of face centered atoms in unit cell
9 a = 1 // let lattice parameter
10 m = 12 // mass of carbon
11 printf("\n Example 5.1")
12 printf("\n Part A:")
13 N = n_c*N_c+n_b*N_b+n_f*N_f // effective number of
  atoms
14 r = a*sqrt(3)/8
15 p_e = N*4/3*pi*r^3/a^3 // packing efficiency
16
```

```

17 printf("\n Packing efficiency of diamond is %.2f",
    p_e)
18 printf("\n\n Part B:")
19 a = 3.57 // lattice parameter in angstrom
20 d = m*1.66e-27*N/(a*1e-10)^3
21 printf("\n Density of diamond is %d Kg/m^3",d)//
    numerical answer in book is 3500
22 printf("\n Density of diamond is %.1f g/cm^3",d
    /1000)

```

---

**Scilab code Exa 5.3** Calculate the cation to anion ratio for an ideally close packed HCP crystal

```

1 // calculate the c/a ratio for an ideally close
    packed HCP crystal
2 clc
3 a = 1 // let
4 PR = a
5 printf("\ Example 5.3")
6 RT = a/sqrt(3)
7 PT = sqrt(PR^2-RT^2)
8 c_a = 2*PT/PR
9 // Calculations are made on the crystal structure
    drawn in book
10 printf("\n c/a ratio for an ideally close packed HCP
    crystal is %0.3f ",c_a)

```

---

**Scilab code Exa 5.4** find the size of largest sphere that can fit into a tetrahedral void

```

1 // find the size of largest sphere that can fit into
  a tetrahedral void
2 clc
3 r = 1 // let
4 a = 3/4
5 printf("\n Example 5.4")
6 pt = 2*sqrt(2/3)*r
7 s = a*pt-r // size of sphere
8 printf("\n Size of largest sphere that can fit into
  a tetrahedral void is %.3fr",s)

```

---

**Scilab code Exa 5.5** Find critical radius ratio for triangular coordination

```

1 // find critical radius ratio for triangular
  coordination
2 clc
3 theta = 60 // angle in degree
4 printf("\n Example 5.5")
5 r_c_a = (2/3*2*sin(theta*pi/180))-1 // ratio
  calculation
6 printf("\n Critical radius ratio for triangular
  coordination is %0.3f ",r_c_a)

```

---

**Scilab code Exa 5.6** Calculate density of MgO

```

1 // Calculate density of MgO
2 clc
3 r_mg = 0.78 // radius of magnesium cation in
  angstrom

```

```
4 r_o = 1.32 // radius of oxygen anion in angstrom
5 n = 4 // effective number of unit cell
6 m_o = 16 // mass of oxygen
7 m_mg = 24.3 // mass of magnesium
8 printf("\n Example 5.6")
9 a = 2*(r_mg+r_o)// lattice parameter
10 d = (m_o+m_mg)*1.66e-27*n/(a*1e-10)^3// density
11 printf("\n Density of MgO is %d Kg/m^3",d) // answer
    is 3610 kg/m^3
12 printf("\n Density of MgO is %0.2 f g/cm^3",d/1000)
```

---

# Chapter 6

## Crystal Imperfection

**Scilab code Exa 6.1** Find equilibrium concentration of vacancies in metals at given temperature

```
1 // Find equilibrium concentration of vacancies in
  metals at given temperature
2 clc
3 t1 = 0 // temperature in kelvin
4 t2 = 300 // temperature in kelvin
5 t3 = 900 // temperature in kelvin
6 R = 8.314 // universal gas constant
7 del_hf_al = 68 // Enthalpy of formation of aluminium
  crystal in KJ
8 del_hf_ni = 168 // Enthalpy of formation of nickel
  crystal in KJ
9 printf("\n Example 6.1")
10
11 printf("\n Equilibrium concentration of vacancies of
  aluminium at %dK is 0",t1)
12 n_N = exp(-del_hf_al*1e3/(R*t2))
13 printf("\n Equilibrium concentration of vacancies of
  aluminium at %dK is %.2e",t2,n_N) // answer in
  book is 1.45e-12
14 n_N = exp(-del_hf_al*1e3/(R*t3))
```

```

15 printf("\n Equilibrium concentration of vacancies of
    aluminium at %dK is %.2e",t3,n_N) // answer in
    book is 1.12e-4
16
17 printf("\n\n Equilibrium concentration of vacancies
    of Nickel at %dK is 0",t1)
18 n_N = exp(-del_hf_ni*1e3/(R*t2))
19 printf("\n Equilibrium concentration of vacancies of
    Nickel at %dK is %.2e",t2,n_N)
20 n_N = exp(-del_hf_ni*1e3/(R*t3))
21 printf("\n Equilibrium concentration of vacancies of
    Nickel at %dK is %.2e",t3,n_N) // answer in book
    is 1.78e-10

```

---

**Scilab code Exa 6.2** Compute the line energy of dislocation

```

1 // Compute the line energy of dislocation
2 clc
3 a = 2.87 // lattice parameter in angstrom
4 b= 2.49 // magnitude of burgers vector in angstrom
5 G = 80.2 // shear modulus in GN
6 printf("\n Example 6.2")
7 E = G*1e9*(b*1e-10)^2*1/2
8 printf("\n Line energy of dislocation is %.2e J m^-1
    ",E)

```

---

**Scilab code Exa 6.4** Calculation of down climb of crystal on heating

```

1 // calculation of down climb of crystal on heating

```

```

2  clc
3  a = 1e10 // total number of edge dislocation
4  N = 6.023e23 // Avogadro number
5  R = 8.314 // Universal gas constant
6  t1 = 0 // initial temperature in K
7  t2 = 1000 // Final temperature in K
8  del_hf = 100 // Enthalpy of vacancy formation in KJ
9  d = 2 // length of one step in angstrom
10 v = 5.5e-6 // volume of one mole crystal
11 printf("\n Example 6.4")
12 n = N*exp(-(del_hf*1e3)/(R*(t2-t1)))/v
13 k = 1/(d*1e-10) // atoms required for 1 m climb
14 b = n/(k*a) // average amount of climb
15 c = b*d*1e-10
16
17 printf("\n Average down climb of crystal is %.2em",c
    )

```

---

**Scilab code Exa 6.5** Calculate surface energy of copper crystal of type 111

```

1 // Calculate surface energy of copper crystal of
   type {111}
2  clc
3  E = 56.4 // bond energy in KJ
4  N_a = 6.023e23 // Avogadro s number
5  n = 12 // number of bonds
6  m = 3 // number of broken bonds
7  N = 1.77e19 // number of atoms in copper crystal of
   type {111} per m^2
8  printf("\n Example 6.5")
9  b_e = 1/2*E*1e3*n/N_a // bond energy per atom
10 e_b = b_e*m/n // energy of broken bond at surface
11 s_e = e_b*N // surface enthalpy of copper

```

```
12 printf("\n Surface enthalpy of copper is %0.2f J m
    ^-2",s_e)
13 printf("\n Surface enthalpy of copper is %d erg cm
    ^-2",s_e*1e3)
14 // Answer in book is 2490 erg cm^-2
```

---

**Scilab code Exa 6.6** Compute the angle at the bottom of groove of a boundary

```
1 // Compute the angle at the bottom of groove of a
    boundary
2 clc
3 Gamma_gb = 1 // let , energy of grain boundary
4 Gamma_s = 3* Gamma_gb// energy of free surface
5 printf("\n Example 6.6")
6 theta = 2*acos(1/2*Gamma_gb/Gamma_s)
7 printf("\n Angle at the bottom of groove of a
    boundary is %d degrees.",ceil(theta*180/%pi))
```

---



# Chapter 7

## Phase Diagram

**Scilab code Exa 7.1** Find degrees of freedom of a system of two components

```
1 // Find degrees of freedom of a system of two
  components
2 clc
3 c = 2 // number of components
4 printf("\\n Example 7.1")
5 for n = 1:4
6     p = (c-1) +2 // Total variables
7     f = c-n+2 // degree of freedom
8     printf("\\n\\n Degree of freedom for two
  components when \\n number of phases is %d is
  %d" ,n, f)
9 end
```

---

**Scilab code Exa 7.2** Find minimum number of component in system

```

1 // Find minimum number of component in system
2 clc
3 p = 4 // number of phases of system
4 f = 0 // number of degree of system
5
6 printf("\n Example 7.2")
7 C = f+p-1 // components number
8 printf("\n Minimum number of components in system is
    %d",C)

```

---

**Scilab code Exa 7.3** Calculate amount of pure water that can be extracted from sea water

```

1 // Calculate amount of pure water that can be
    extracted from sea water
2 clc
3 L = 23.3 // % composition of L
4 a = 3.5 // concentration of Nacl in sea water
5 ice = 0 // % composition of ice
6 printf("\n Example 7.3")
7 f_ice = (L-a)/(L-ice)
8 printf("\n Fractional amount of pure water that can
    be extracted from sea water is %0.2f",f_ice)

```

---

**Scilab code Exa 7.5** Calculate proeutectoid ferrite and eutectoid ferrite in 6 tenth percent steel

```

1 // Calculate proeutectoid ferrite and eutectoid
    ferrite in 0.6% steel

```

```
2 clc
3 a = 0 // limiting value
4 b = 0.8 // limiting value
5 c = 0.6 // percentage composition of carbon
6 f = 0.88 // fraction of ferrite in a eutectoid steel
7 printf("\n Example 7.5")
8 f_pro_alpha = (b-c)/(b-a)
9 f_perlite = 1 - f_pro_alpha
10 f_eut = f*f_perlite
11 printf("\n Composition of proeutectoid ferrite is %0
    .2f",f_pro_alpha)
12 printf("\n Composition of eutectoid ferrite is %0.2f
    ",f_eut)
```

---

# Chapter 8

## Diffusion in Solids

**Scilab code Exa 8.1** Calculate the rate at which hydrogen escapes through the walls of the steel tank

```
1 // calculate the rate at which hydrogen escapes
   through the walls of the steel tank
2 clc
3 t = 5 // thickness in mm
4 c = 10 // concentration
5 D = 1e-9 // diffusion coefficient
6 printf("\n Example 8.1")
7 j = D*c/(t*1e-3)
8 printf("\n Outward flux is %.0e kg m^-2 s^-1",j)
```

---

**Scilab code Exa 8.2** Calculate maximum time till which material can be kept at 550 degree Celsius

```
1 // Calculate maximum time till which material can be
   kept at 550 degree Celsius
2 clc
```

```

3 D_0 = 0.24e-4 // diffusion coefficient
4 Q = 121e3
5 R = 8.314 // Universal gas constant
6 T = 550 // temperature in Celsius
7 k = 0.2 // thickness of pure Al sheet in mm
8 d = 0.1 // penetration depth in mm
9 c_x = 0.4 // concentration in percentage
10 A = 2 // Constant in percentage
11 B = 2 // Constant in percentage
12 printf("\n Example 8.2")
13 x = d-k
14 D_cu_al = D_0*exp(-Q/(R*(T+273)))
15 k = (A-c_x)/B
16 if k ==0.8 then
17     z = 0.9 // from table
18 end
19 t = (x*1e-3)^2/(z^2*4*D_cu_al) // time in sec
20
21 printf("\n Material can be kept at %d degree Celsius
        for nearly %d minute",T,t/60) // answer in book
        is 100 min

```

---

**Scilab code Exa 8.3** Calculate minimum depth up to which post machining is to be done

```

1 // Calculate minimum depth up to which post
  machining is to be done
2 clc
3 D_0 = 0.7e-4 // diffusion coefficient
4 Q = 157 // Energy in kJ mol-1, considered from
  table 8.2
5 R = 8.314 // Universal gas constant
6 T = 950 // temperature in Celsius

```

```

7 c2 = 0.8 // concentration in percentage
8 cs = 0 // concentration in percentage
9 c_x = 0.6 // concentration in percentage
10 t = 4 // time in hours
11 a = 1 //let
12 printf("\n Example 8.3")
13 A = cs
14 B = c2-cs
15 D = D_0*exp(-Q*1e3/(R*(T+273)))
16 k = erf(((A-c_x)/B))*-1
17 if k >0.7 then
18     if k<0.712 then
19         z = 0.81 // from table
20     end
21
22 end
23 x = z*2*sqrt(D*t*3600)
24
25 printf("\n Depth up to which machining is required
        is nearly %.2f mm",x*1e3)
26 // numerical value of answer in book is 0.75

```

---

**Scilab code Exa 8.4** Calculate time required to get required boron concentration

```

1 // Calculate time required to get required boron
  concentration
2 clc
3 D = 4e-17 // diffusion coefficient
4 c1 = 0
5 cs = 3e26
6 c_x = 1e23 // number of atoms
7 x = 2e-6 // depth in m

```

```

8 printf("\n Example 8.4")
9 A = cs
10 B = cs - c1
11 k = (A-c_x)/B
12 if k >0.99966 then
13     if k< 0.9997 then
14         z = 2.55 // from table
15     end
16 end
17 t = x^2/(z^2*4*D) // time in sec
18
19 printf("\n Time required to get required boron
    concentration is %d sec",t) // answer in book is
    3845 sec

```

---

**Scilab code Exa 8.5** Calculate ratio of cross sectional areas

```

1 // Calculate ratio of cross sectional areas
2 clc
3 r = 10 // radius in mm
4 t = 4 // thickness in angstrom
5 printf("\n Example 8.5")
6 r = 2*pi*r*1e-3*t*1e-10/(%pi*(r*1e-3)^2)
7 printf("\n Ratio of cross sectional areas is %.0e ",
    r)

```

---

# Chapter 9

## Phase Transformation

**Scilab code Exa 9.1** Calculate the critical free energy of nucleation of ice from water and critical radius

```
1 // Calculate the critical free energy of nucleation
   of ice from water and critical radius
2 clc
3 del_t1 = 0 // temperature difference in degree
   Celsius
4 del_t2 = -5 // temperature difference in degree
   Celsius
5 del_t3 = -40 // temperature difference in degree
   Celsius
6 del_h = 6.02 // enthalpy of fusion in kJ/mol
7 T_m = 273 // mean temperature
8 Gamma = 0.076 // energy of ice water interface in J
   /m^2
9 v = 19 // molar volume of ice
10 printf("\\n Example 9.1")
11 printf("\\n Part A")
12 printf("\\n At del_t = %d, there is no supercooling.
   So there is no critical radius",del_t1)
13 printf("\\n\\n Part B")
14 del_f = 16/3*%pi*(Gamma)^3*T_m^2/((del_h*1e3*1e6/v)
```



```

    ^2*del_t2^2)
15 r = 2*T_m*Gamma/(-del_h*1e3*1e6/v*del_t2)
16 printf("\n Critical free energy of nucleation is %.1
    eJ",del_f)
17 printf("\n Critical radius is %d angstrom",ceil(r*1
    e10))
18 printf("\n\n Part C")
19 temp_r = del_t3/del_t2
20 del_f_ = del_f/temp_r^2
21 r_ = r/temp_r
22
23 printf("\n Critical free energy of nucleation is %
    .1eJ",del_f_)
24 printf("\n Critical radius is %d angstrom.", ceil(r_
    *1e10))

```

---

**Scilab code Exa 9.2** Calculate the change in delta f required to increase nucleation rate

```

1 // Calculate the change in del_f required to
    increase nucleation rate
2 clc
3 T= 300 // temperature in kelvin
4 R = 8.314 // universal gas constant
5 k = 2.303 // conversion factor
6 a1 = 1e42
7 a2 = 1e6 // nucleation rate
8 a3 = 1e10 // nucleation rate
9 printf("Example 9.2")
10 I1 = (log10(a1)-log10(a2))*k*R*T // exponent factor
11 I2 = (log10(a1)-log10(a3))*k*R*T // exponent factor
12 del_f = I1-I2 // difference
13 a = 10^(log10(a3)-log10(a2))

```

```

14
15 printf("\n A change of %d KJ mol-1 energy is
    required to increase nucleation factor from \n %
    .0e m-3 s-1 to %.0e m-3 s-1 ",ceil(del_f/1
    e3),a,a3)

```

---

**Scilab code Exa 9.4** Calculate delta f of heterogeneous as a fraction of delta f of homogeneous

```

1 // Calculate del_f_het as a fraction of del_f_homo
2 clc
3 Gamma_alpha_del = 0.5 // in J m-2
4 Gamma_alpha_beta = 0.5 // in J m-2
5 Gamma_beta_del = 0.01 // in J m-2
6
7 printf("\n Example 9.4")
8 theta = acos((Gamma_alpha_del -Gamma_beta_del)/
    Gamma_alpha_beta)
9 del_f_ratio = 1/4*(2-3*cos(theta)+(cos(theta))^3)
10
11
12 printf("\n del_f_het is %0.4fth fraction of
    del_f_homo.",del_f_ratio)

```

---

**Scilab code Exa 9.6** Calculate the free energy change during recrystallization

```

1 // Calculate the free energy change during
    recrystallization

```

```
2 clc
3 mu = 45.5e9
4 b = 2.55e-10
5 n1 = 1e9 // initial dislocation density
6 n2 = 1e13 // final dislocation density
7 printf("\n Example 9.6")
8 E = 1/2*mu*b^2*n2
9 del_g = E // as difference between initial and final
    dislocation energy is four order magnitude
10 printf("\n Free energy change during
    recrystallization is %d J m-3",-del_g)
11 // Numerical value of answer in book is 14800
```

---

# Chapter 10

## Elastic Anelastic and Viscoelastic Behaviour

Scilab code Exa 10.1 Estimate Youngs modulus of material

```
1 // Estimate young s modulus of material
2 clc
3 n = 1
4 m = 9
5 A = 7.68e-29 // Constant having unit J m
6 r_0 = 2.5e-10 // bonding distance in m
7 printf("\n Example 10.1")
8 B = A*r_0^8/9
9
10 Y = (90*B/(r_0)^11-2*A/(r_0)^3)/r_0
11
12 printf("\n Young s modulus of material is %d GN m
    ^-2",Y/1e9)
```

---

### Scilab code Exa 10.2 Calculation of stress in fibers

```
1 // Calculation of stress in fibers
2 clc
3 Y_f = 440
4 Y_m = 71
5 sigma_total= 100 // total load
6 printf("\n Example 10.2")
7 r = Y_f/Y_m
8 sigma_f = r*(sigma_total/0.7)/(1+r*3/7)
9 printf("\n Part A:")
10 printf("\n When load is applied parallel to fiber
    then, stress in fiber is %d MN m-2",sigma_f)
11
12 printf("\n\n Part B:")
13 printf("\n When load is applied perpendicular to
    fiber then, stress in fiber and matrix is same i.
    e. %d MN m-2",sigma_total)
```

---

### Scilab code Exa 10.3 Estimate diffusion coefficient

```
1 // Estimate diffusion coefficient
2 clc
3 t_r = 100 // relaxation time in s
4 d = 2.5 // distance in angstrom
5 printf("\n Example 10.3")
6 f = 1/t_r // jump frequency
7 D = (d*1e-10)2*f
8 printf("\n Diffusion coefficient is %.2e m2 s-1",D
    )
```

---

# Chapter 11

## Plastic Deformation and Creep in Crystalline Materials

**Scilab code Exa 11.2** Calculate the stress required to move the dislocation at given temperature

```
1 // Calculate the stress required to move the
   dislocation at given temperature
2 clc
3 b = 2 // burger vector in angstrom
4 v = 20*b^3 // activation volume
5 tau_pn = 1000 // P-N stress of crystal in MNm-2
6 k = 1.38e-23 // physical constant
7 t1 = 0 // temperature in K
8 t2 = 100 // temperature in K
9 t3 = 300 // temperature in K
10 t4 = 500 // temperature in K
11 printf("\\n Example 11.2")
12 printf("\\n\\n Part A:")
13 T = t1
14 tau_app = tau_pn - 40*k*T/(v*1e-30)
15 printf("\\n The stress required to move the
   dislocation at temperature %dK is %d MNm-2",T,
   tau_app)
```

```

16 printf("\n\n Part B:")
17 T = t2
18 tau_app = tau_pn - 40*k*T/(v*1e-30*1e6)
19 printf("\n The stress required to move the
        dislocation at temperature %dK is %d MNm-2",T,
        tau_app)
20 printf("\n\n Part C:")
21 T = t3
22 tau_app = tau_pn - 40*k*T/(v*1e-30*1e6)
23 if tau_app<0 then
24     printf("\n Stress to be applied is zero")
25     printf("\n The stress required to move the
            dislocation at temperature %dK entirely
            overcome by thermal fluctuations", T)
26 end
27 printf("\n\n Part D:")
28 T = t4
29 tau_app = tau_pn - 40*k*T/(v*1e-30*1e6)
30 if tau_app<0 then
31     printf("\n Stress to be applied is zero")
32     printf("\n The stress required to move the
            dislocation at temperature %dK entirely
            overcome by thermal fluctuations", T)
33 end

```

---

**Scilab code Exa 11.3** Calculate the dislocation density in copper

```

1 // Calculate the dislocation density in copper
2 clc
3 mu = 44 // shear modulus of copper in GN m-2
4 b = 2.55 // burgers vector in angstrom
5 tau = 35 // shear stress in MN m-2
6 printf("Example 11.3")

```

```

7 l = mu*1e9*b*1e-10/(tau*1e6)
8 rho = 1/l^2
9
10 printf("\n Dislocation density in copper is %.1e m
      ^-2",rho)
11 // Answer in book is 1e12 m^-2

```

---

**Scilab code Exa 11.4** Find the yield stress for a grain size of ASTM 9

```

1 // Find the yield stress for a grain size of ASTM 9
2 clc
3 sigma1 = 120 // initial yield strength of material
      in MNm^-2
4 sigma2 = 220 // Final yield strength of material in
      MN m^-2
5 d1 = 0.04 // initial diameter in mm
6 d2 = 0.01 // final diameter in mm
7 n = 9 // astm number
8 printf("Example 11.4")
9 k = (sigma2-sigma1)*1e6/(1/sqrt(d2*1e-3)-1/sqrt(d1*1
      e-3))
10 sigma_i = sigma1*1e6 -k/sqrt((d1*1e-3))
11 d = 1/sqrt(2^(n-1)*1e4/645)
12 sigma_y = sigma_i+k*(d*1e-3)^(-0.5)
13
14 printf("\n Yield stress for a grain size of ASTM 9
      is %d MN m^-2",ceil(sigma_y/1e6))

```

---

**Scilab code Exa 11.5** Estimate the change in yield strength



```
1 // Estimate the change in yield strength
2 clc
3 n1 = 1e6 // initial number of particles
4 n2 = 1e3 // final number of particle
5 printf("\n Example 11.5")
6 k = (n1/n2)^(1/3)
7 printf("\n Yield strength would have decreased to
   %d%% of its initial value.",100/k)
```

---

# Chapter 12

## Fracture

Scilab code Exa 12.1 Estimate fracture strength

```
1 // Estimate fracture strength
2 clc
3 c = 2 // crack of half length in micro meter
4 Y = 70 // young s modulus in GN m-2
5 Gamma = 1 // specific surface energy
6 printf("\n Example 12.1")
7 sigma_f = sqrt(2*Gamma*Y*1e9/(%pi*c*1e-6))/1e6
8 r = Y*1e3/sigma_f
9 printf("\n Fracture strength of %d MN m-2 is 1/%dth
    of young s modulus. ",ceil(sigma_f),ceil(r
    /100)*100)
10 printf("\n Thus Griffiths criterion bridges the gap
    between the \n observed and ideal strengths of
    brittle material")
```

---

Scilab code Exa 12.2 Estimate the brittle fracture strength at low temperatures

```

1 // Estimate the brittle fracture strength at low
  temperatures
2 clc
3 Gamma = 1.5 // specific surface energy in J/m^2
4 Y = 200 // Young s modulus in GN/m^2
5 c = 2 // half length of crack
6
7 printf(" \n Example 12.2")
8 sigma_f = sqrt(2*Gamma*Y*1e9/(%pi*c*1e-6))
9
10 printf(" \n Brittle fracture strength at low
  temperatures is %d MNm^-2 ",sigma_f/1e6) // answer
  in book is 310MNm^-2

```

---

**Scilab code Exa 12.3** Estimate the temperature at which the ductility of brittle transition occurs at given strain rates

```

1 // Estimate the temperature at which the ductility
  of brittle transition occurs at given strain
  rates
2 clc
3 Gamma = 2 // specific surface energy in J/m^2
4 Y = 350 // Young s modulus in GN/m^2
5 c = 2 // half length of crack
6 de_dt1 = 1e-2 // strain rate
7 de_dt2 = 1e-5 // strain rate
8 printf(" \n Example 12.3")
9 printf(" \n Part A:")
10 sigma_f = sqrt(2*Gamma*Y*1e9/(%pi*c*1e-6))
11 sigma_y = sigma_f/1e6
12 T = 173600/(sigma_y-20.6-61.3*log10(de_dt1)) //
  temperature calculation
13

```

```
14 printf("\n Transition temperature for strain rate %
    .0e s-1 is %d K",de_dt1,T)// answer in book is
    300 K
15 printf("\n\n Part B:")
16
17 T = 173600/(sigma_y-20.6-61.3*log10(de_dt2))//
    temperature calculation
18
19 printf("\n Transition temperature for strain rate %
    .0e s-1 is %d K",de_dt2,T)// answer in book is
    230 K
20 // Solution in book for two parts is divided into
    three parts
```

---

# Chapter 13

## Oxidation and Corrosion

Scilab code Exa 13.2 Calculation of required quantity of magnesium

```
1 // Calculation of required quantity of magnesium
2 clc
3 j = 15 // current density in mA m-2
4 m = 0.0243 // molar mass of magnesium
5 F = 96490 // farad charge
6 n = 2 // charge on ion
7 t = 10 // time in years
8 printf("\\n Example 13.2")
9 a = m*j*1e-3*(t*365*24*3600)/(n*F)
10 printf("\\n Amount of magnesium required is %0.1f kg
    m-2",a)
```

---

# Chapter 14

## Conductors and Resistors

Scilab code Exa 14.1 Calculate energy difference

```
1 // Calculate energy difference
2 clc
3 n_x1 = 1 // atomic level
4 n_y1 = 1 // atomic level
5 n_z1 = 1 // atomic level
6 n_x2 = 2 // atomic level
7 L = 10 // lattice parameter in mm
8 h = 6.626e-34 // plank constant
9 m_e = 9.109e-31 // mass of electron in kg
10 printf("\n Example 14.1")
11 E1 = h^2*(n_x1^2+n_y1^2+n_z1^2)/(8*m_e*(L*1e-3)^2)
12 E2 = h^2*(n_x2^2+n_y1^2+n_z1^2)/(8*m_e*(L*1e-3)^2)
13 E = E2-E1 // energy difference
14 printf("\n Energy difference is %.2e J ",E)
```

---

Scilab code Exa 14.2 Calculate conductivity of copper at 300 K

```

1 // Calculate conductivity of copper at 300 K
2 clc
3 tau = 2e-14 // collision time of electron in s
4 e = 1.602e-19 // charge on electron
5 m_e = 9.1e-31 // mass of electron in kg
6
7 printf("\n Example 14.2")
8 n = 6.023e23*8960/0.06354
9
10 sigma= n*e^2*tau/m_e
11 printf("\n Conductivity of copper at 300 K is %.1e
        ohm^-1 m^-1 ",sigma)

```

---

**Scilab code Exa 14.3** Estimation of resistivity due to impurity scattering of 1 percent of Nickel in copper lattice

```

1 // Estimation of resistivity due to impurity
  scattering of 1% of Nickel in copper lattice
2 clc
3 r_cu = 1.8e-8 // resistivity of pure copper in ohm-m
4 r_Ni_cu = 7e-8 //resistivity of copper 4% Ni in ohm-
  m
5 per1 = 4//impurity in percent
6 per2 = 1 // impurity in percent
7 printf("\n Example 14.3")
8 r = (r_Ni_cu-r_cu)*per2/per1 // resistivity of
  copper 1% Ni in ohm-m
9 printf("\n Resistivity due to impurity scattering of
  1 %% of Nickel in copper lattice is %.1e ohm-m",
  r)

```

---

# Chapter 15

## Semiconductors

Scilab code Exa 15.1 Calculate intrinsic carrier density

```
1 // Calculate intrinsic carrier density
2 clc
3 rho = 3000 // resistivity in ohm m
4 mu_e = 0.14
5 mu_h = 0.05
6 e = 1.602e-19 // charge on electron
7 printf("\n Example 15.1")
8 sigma = 1/rho
9 n = sigma/((mu_e+mu_h)*e)
10 printf("\n Intrinsic carrier density is %.3e m^-3",n
    )
```

---



# Chapter 16

## Magnetic Materials

**Scilab code Exa 16.1** Calculate the net magnetic moment per iron atom in crystal

```
1 // Calculate the net magnetic moment per iron atom
   in crystal
2 clc
3 a = 2.87 // lattice parameter in angstrom
4 n = 2 // number of atoms per unit cell
5 m = 1750 // Saturation magnetization in kAm-1
6 mu = 9.273e-24 // bohr magneton
7 printf("\\n Example 16.1")
8 m_atom = m*1e3*(a*1e-10)3 /n
9 mu_b = m_atom/mu
10
11 printf("\\n Net magnetic moment per iron atom in
   crystal is %.3e Am2",m_atom)
12 printf("\\n In unit of mu_b, Net magnetic moment per
   iron atom in crystal is %.1f ",mu_b)
```

---

### Scilab code Exa 16.2 Comparison of saturation temperatures

```
1 // Comparison of saturation temperatures
2 clc
3 t1 = 0 // temperature in kelvin
4 t2 = 300 // temperature in kelvin
5 m_net_Gd = 7 // net magnetic moment of gadolinium
6 m_net_Co = 1.7 // net magnetic moment of cobalt
7 t_c_Gd = 289 // curie temperature for Gd
8 printf("\n Example 16.2")
9 printf("\n Part A:")
10 if m_net_Gd > m_net_Co then
11     printf("\n At %d K, Net magnetic moment of
            gadolinium i.e. %d is greater than net magnetic
            moment of cobalt i.e. %.1f ", t1, m_net_Gd, m_net_Co
            )
12     printf("\n So, Gd will have higher saturation
            magnetization")
13 end
14 printf("\n\n Part B:")
15 if t_c_Gd < t2 then
16     printf("\n At temperature %d K, Gd is above its
            curie temperature of %dK", t2, t_c_Gd)
17     printf("\n Gd will be paramagnetic at %d K and
            will have negligible magnetization\n as
            compared to Co, which has higher curie
            temperature", t2)
18 end
```

---

### Scilab code Exa 16.4 Calculation of hysteresis loss

```
1 // Calculation of hysteresis loss
2 clc
```

```

3 v = 0.01 // volume in m^3
4 x = 1e-4 // axis intercept
5 y = 1e2 // axis intercept
6 a = 60000 // Hysteresis loop area
7 f = 50 // frequency in Hz
8 printf("\n Example 16.4")
9 e = x*y*a // Energy loss in one loop
10 E = e*v // energy loss in core in one cycle
11 P = E*f // Power loss
12 printf("\n Power loss due to hysteresis is %d W",P)

```

---

**Scilab code Exa 16.5** Calculation of eddy current loss at normal voltage and frequency

```

1 // Calculation of eddy current loss at normal
   voltage and frequency
2 clc
3 Total1 = 2300 // total iron loss in W at 440 V and
   50 Hz
4 Total2 = 750 // total iron loss in W at 220 V and 25
   Hz
5 printf("\n Example 16.5")
6 W_e = 1/2*(Total1-2*Total2)
7 printf("\n Eddy current loss at normal voltage and
   frequency is %dW",4*W_e)

```

---

# Chapter 17

## Dielectric Materials

Scilab code Exa 17.1 Calculation of relative dielectric constant

```
1 // calculation of relative dielectric constant
2 clc
3 l= 10 // length of capacitor in mm
4 b = 10 // width of capacitor in mm
5 d = 2 // distance of separation in mm
6 c = 1e-9 // capacitance in farad
7 epsilon_0 = 8.85e-12 // permittivity of free space
8 printf("\n Example 17.1")
9 epsilon_r = c*d*1e-3/(epsilon_0*l*1e-3*b*1e-3)
10
11 printf("\n Relative dielectric constant is %d",
        epsilon_r)
```

---

Scilab code Exa 17.2 Calculate the polarization of a BaTio3 crystal

```
1 // calculate the polarization of a BaTio3 crystal
2 clc
```

```

3 Ti_shift= 0.06 // shift of TI ion in angstrom
4 O_shift = 0.06 // shift of oxygen ion of side face
  in angstrom
5 o_shift = 0.08 //shift of oxygen ion of top and
  bottom faces in angstrom
6 O_charge = 2 // unit charge on oxygen ion of side
  face
7 o_charge = 2 // unit charge on oxygen ion of top
  and bottom faces
8 Ti_charge = 4 // unit charge on titanium ion
9 n_o = 2 // number of oxygen ion of side face
10 n_o = 1 // number of oxygen ion of top and bottom
  face
11 n_Ti = 1 // number of titanium ion
12 e = 1.6e-19 // amount of one unit charge in coulomb
13 printf("\n Example 17.2")
14 p_Ti = n_Ti*Ti_charge *e*Ti_shift*1e-10
15 p_o = n_o*O_charge*e*O_shift*1e-10
16 p_o = n_o*o_charge*e*o_shift*1e-10
17 Total = p_Ti+p_o+p_o
18 P = Total/(4.03*3.98^2*1e-30)
19 printf("\n Polarization of BaTiO3 crystal is %.2f Cm
  ^-2 ", P)

```

---