

Scilab Textbook Companion for  
Callister's Materials Science and Engineering  
(adopted By R. Balasubramaniam)  
by W. D. Callister<sup>1</sup>

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# Book Description

**Title:** Callister's Materials Science and Engineering (adopted By R. Balasubramaniam)

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Scilab numbering policy used in this document and the relation to the above book.

**Exa** Example (Solved example)

**Eqn** Equation (Particular equation of the above book)

**AP** Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

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## Chapter 3

# Fundamentals of structure of crystalline solids

Scilab code Exa 3.1 Location of point having specified coordinates

```
1  clc
2  // Given
3  a = 0.48 // let side of a unit cell
4  b = 0.46 // let side of a unit cell
5  c = 0.40 // let side of a unit cell
6  p_x = 1/4
7  p_y = 1
8  p_z = 1/2
9  printf("Example 3.1")
10 qa = a*p_x // distance of point along x axis from
    origin in mm
11 rb = b*p_y // distance of point along y axis from
    origin in mm
12 sc = c*p_z // distance of point along z axis from
    origin
13 printf("\n Point P (%.2f,%.2f,%.2f) corresponds to",
    p_x, p_y, p_z)
14 printf(" %.2fmm,%.2fmm,%.2fmm ", qa, rb, sc)
```

---

### Scilab code Exa 3.2 Specification of point coordinates

```
1
2 clc
3 //given that
4 // unit cell is BCC
5 printf("Example 3.2")
6 printf("\\n\\nPoint \\t Fractional Length\\t Point \\n")
7 printf("Number \\t x axix\\t y axix\\ z axix \\t
   Coordinate\\n")
8 n = 1 // serial number
9
10 for x = 0:1
11     for y = 0:1
12         for z = 0:1
13
14             printf("%d\\t   %d           %d           %d \\t %d
   %d %d \\n",n,x,y,z,x,y,z)
15             n = n+1
16         end
17     end
18 end
19 x = 0.5// x coordinate for center point
20 y = 0.5// y coordinate for center point
21 z = 0.5// y coordinate for center point
22 printf("%d\\t   %.1f           %0.1f           %.1f           %.1f %.1f %.1
   f \\n",n,x,y,z,x,y,z)
23 // sequence of point number is changed to make
   problemable
```

---

### Scilab code Exa 3.3 Specification of support Post Diameter

```
1
2 clc
3 // Given
4 a = 1 // let side of a unit cell
5 b = 1 // let side of a unit cell
6 c = 1 // let side of a unit cell
7 x_proj = 0.5*a // x coordinate of point
8 y_proj = 1*b // y coordinate of point
9 z_proj = 0*c // z coordinate of point
10
11 printf(" Example 3.3\n")
12 k = 1/x_proj
13 x_reduction = x_proj*k // reduction in x direction
14 y_reduction = y_proj*k // reduction in x direction
15 z_reduction = z_proj*k // reduction in x direction
16
17 printf(" \n Indices of direction are as [%d %d %d]\n"
    ,x_reduction ,y_reduction ,z_reduction)
```

---

### Scilab code Exa 3.4 Construction of specified crystallographic direction

```
1 clc
2 printf(" Example 3.4\n")
3 printf(" \n It is a diagram based theoretically
    solved example\n")
```

---

**Scilab code Exa 3.5** Determination of directional indices for a hexagonal unit cell

```
1
2
3 clc
4 // Given
5 u1 = 1 // Unit cell parameter
6 v1 = 1 // Unit cell parameter
7 w1 = 1 // Unit cell parameter
8
9 printf(" Example 3.5\n")
10 u_1 = 1/3*(2*u1-v1)
11 v_1 = 1/3*(2*v1-u1)
12 t_1 = -(u_1+v_1)
13 w_1 = w1
14 k = 1/u_1
15 u = u_1*k
16 v = v_1*k
17 w = w_1*k
18 t = t_1*k
19 printf(" \n Indices of direction are as [%d %d %d %d
    ]\n",u,v,t,w)
20 // while in notation negative value is indicated by
    a bar over it
```

---

**Scilab code Exa 3.6** Specification of support Post Diameter

```

1
2
3 clc
4 // Given
5 a = 1 // let side of a unit cell
6 b = 1 // let side of a unit cell
7 c = 1 // let side of a unit cell
8
9 printf(" Example 3.6\n")
10 x_proj = 0 // x coordinate of point
11 y_proj = -1*b // y coordinate of point
12 z_proj = 0.5*c // z coordinate of point
13 x_reduction = x_proj*a // reduction in x direction
14 y_reduction = 1/y_proj // reduction in x direction
15 z_reduction = 1/z_proj // reduction in x direction
16
17 printf(" \n Indices of direction are as [%d %d %d]\n"
        ,x_reduction,y_reduction,z_reduction)
18 // while in notation negative value is indicated by
        a bar over it

```

---

**Scilab code Exa 3.7** Construction of specified crystallographic plane

```

1
2 clc
3 printf(" Example 3.7 \n")
4 printf(" \n It is a diagram based theoretically
        solved example\n")

```

---

**Scilab code Exa 3.8** Determination of directional indices for a hexagonal unit cell

```
1
2
3 clc
4 // Given
5 u1 = 1 // Unit cell parameter
6 v1 = -1 // Unit cell parameter
7 w1 = 1 // Unit cell parameter
8
9 printf(" Example 3.8\n")
10 u_1 = 1/3*(2*u1-v1)
11 v_1 = 1/3*(2*v1-u1)
12 t_1 = -(u_1+v_1)
13 w_1 = w1
14 k = 1/u_1
15 u = u_1*k
16 v = v_1*k
17 w = w_1*k
18 t = t_1*k
19 printf("\n Indices of direction are as [%d %d %d %d
   ]\n",u,v,t,w)
20 // while in notation negative value is indicated by
   a bar over it
```

---

# Chapter 4

## Structure of crystalline solids

Scilab code Exa 4.1 Determination of FCC Unit Cell Volume

```
1
2
3 clc
4 //given that
5 R = 1 // let radius of an atom is unity
6 printf("Example 4.1\n")
7 //For FCC a=2*R*sqrt(2)
8
9 a=2*R*sqrt(2) //Edge Length
10 V=a^3 //Volume determination
11 printf("Volume of an FCC unit cell is %fR^3 \n\n
    ",V)
```

---

Scilab code Exa 4.2 Computation of the Atomic Packing Factor for FCC

```
1
2
```

```

3  clc
4  //given that
5  R = 1 // let radius of an atom to be unity
6  n=4 // no. of atoms for FCC are 4
7  printf("Example 4.2\n");
8  a=2*R*sqrt(2) // edge length for FCC
9  Vc=a^3 //Volume of cube
10 Vs=n*4*%pi*R^3/3 //Volume of sphere
11 APF=Vs/Vc //Atomic packing Fraction
12
13 printf(" \n Atomic packing factor is %.2f",APF)

```

---

#### Scilab code Exa 4.3 Theoretical Density Computation for Copper

```

1
2
3  clc
4  // given that
5  R=1.28D-08//Atomic radius in cm
6  A_Cu=63.5 //Atomic wt of copper
7  n=4 //For FCC
8  Na=6.023D23 //Avogadro no.
9  printf("Example 4.3\n")
10
11 a=2*R*sqrt(2)
12 Vc=a^3
13 rho=n*A_Cu/(Vc*Na)
14 printf(" \n Density of copper is %.2f g/cm^3.\n",rho)

```

---



**Scilab code Exa 4.4** Computation of minimum cation to anion radius ratio for a coordination number of 3

```
1 clc
2 //given that
3 theta = 30 // angle between a line joining centres
   of anion with their median in degree
4 printf("Example 4.4\n")
5 r_ratio = (1- cos(theta*pi/180))/ cos(theta*pi
   /180)
6 printf("\n Minimum cation to anion radius ratio is %
   .3f",r_ratio)
```

---

**Scilab code Exa 4.5** Ceramic crystal structure prediction

```
1 clc
2 // given that
3 r_fe2 = 0.077 // radius of Iron ion in nm
4 r_o2 = 0.14 // radius of oxygen ion in nm
5 r_ratio = r_fe2 /r_o2 // cation - anion radius ratio
   for FeO
6 printf("Example 4.5\n")
7 if r_ratio > 0.414 & r_ratio < 0.732 then
8     printf("\n As ratio lies between 0.414 and
   0.732 and coordination number 6 \n so it is
   of AX crystal structure.")
9 end
10
11 // coordination number 6 is taken from table
```

---

**Scilab code Exa 4.6** Theoretical density calculation for Sodium chloride

```
1
2
3 clc
4 //given that
5 n = 4 // number of crystals per unit cell
6 A_c = 22.99 // molar mass of cation i.e. sodium
7 A_a = 35.45 // molar mass of cation i.e. chlorine
8 r_c = 1.02e-8 // radius of sodium atom in cm
9 r_a = 1.81e-8 // radius of sodium atom in cm
10 N_a = 6.023e23 // Avogadro constant
11
12 printf("Example 4.6\n")
13 a = 2*(r_c+r_a)// edge length of Nacl molecule
14 V_c = a^3 // Volume of unit cell
15 rho = (n*(A_c+A_a))/(V_c*N_a)
16 printf(" \n Theoretical density of crystal is %0.2f g
    /cm^3 .",rho)
17 printf(" \n \n This result compares very favourable
    with \n experimental value of 2.16 g/cm^3 .")
```

---

**Scilab code Exa 4.7** Computation of density and percentage crystallinity of polyethylene

```
1
2
3 clc
```

```

4 // given that
5 n = 2 // number of repeated units within unit cell
6 A_c = 12.01 // molar mass of carbon
7 A_h = 1.008 // molar mass of hydrogen
8 a = 0.741 // edge length in x axis in nm
9 b = 0.494 // edge length in y axis in nm
10 c = 0.255 // edge length in z axis in nm
11 N_a = 6.023e23 // Avogadro constant
12 rho_s = 0.925 // density of branched polyethylene in
    g/cm^3
13 rho_a = 0.870 // density of totally amorphous
    polyethylene in g/cm^3
14 printf("Example 4.7\n")
15 printf(" \n Part A:")
16 A = 2*A_c+4*A_h // Molar mass of polyethylene
17 V_c = a*b*c*(1e-7)^3 // Volume of unit cell
18 rho_c = (n*A)/(V_c*N_a)
19
20 printf("\n Density of totally crystalline
    polyethylene is %0.3f g/cm^3 .",rho_c)
21 printf(" \n\n Part B:")
22 per_cry = (rho_c*(rho_s-rho_a))*100/(rho_s*(rho_c-
    rho_a))
23 printf("\n Percentage crystallinity is %0.1f%%",
    per_cry)

```

---

**Scilab code Exa 4.8** Inter planer angle and Diffraction Angle Computations

```

1
2
3 clc
4 //given that

```

```

5 a = 0.2866 // lattice parameter in nm
6 h = 2 // component of set of plane
7 k = 2 // component of set of plane
8 l = 0 // component of set of plane
9 n = 1 // order of diffraction
10 lambda = 0.1790 // wavelength of light in nm
11
12 printf("Example 4.8\n")
13 d_hkl=a/(sqrt(h^2+k^2+l^2))
14 theta=asind(n*lambda/(2*d_hkl))
15 printf("\n Answer A:")
16 printf("\n Interplanar spacing is %.4f nm.",d_hkl)
17 printf("\n\n Answer B:")
18 printf("\n Diffraction angle is %0.2f degree.\n",2*
    theta);
19 // Answer in book is 124.26 degree. It is so because
    of consideration of different number of
    significant figures in calculation.

```

---

# Chapter 5

## Imperfection in solids

**Scilab code Exa 5.1** Number of Vacancies Computation at a Specified temperature

```
1
2
3 clc
4 // given that
5 Na=6.023*10^23 //Avogadro No.
6 rho=8.4e6 //Density of Copper in g/m^3
7 A=63.5 //Atomic weight of Copper
8 Qv=0.9 //Activation energy in eV
9 k=8.62*10^-5 //Boltzmann Constant in eV/K
10 T=1000+273 //Temperature in K
11 printf(" Example 5.1\n");
12 N=Na*rho/A //No. of atomic site per cubic meter
13 Nv=N*exp(-Qv/(k*T))
14 printf(" \n Equilibrium number of vacancies/m^3 is %
    .1e for 1273K",Nv)
```

---

**Scilab code Exa 5.2** Derivation of composition conversion equation

```
1
2 clc
3 printf(" Example 5.2 \n")
4 printf(" \n It is a derivational problem \n")
```

---

**Scilab code Exa 5.3** Composition Conversion From weight percent to Atom percent

```
1
2
3 clc
4 // given that
5 C_Al=97 //Aluminium wt%
6 C_Cu=3 //Copper wt%
7 A_Al=26.98 //Atomic wt of Aluminium
8 A_Cu=63.55 //Atomic wt of Copper
9
10 printf(" Example 5.3\n")
11     CA1=C_Al*A_Cu*100/((C_Al*A_Cu)+(C_Cu*A_Al))
12     CCu=C_Cu*A_Al*100/((C_Cu*A_Al)+(C_Al*A_Cu))
13 printf(" \n Atomic %% of Al is %.1f%%",CA1);
14 printf(" \n Atomic %% of Cu is %.1f%%\n",CCu)
```

---

**Scilab code Exa 5.4** Computation of Number of Scotty Defect in KCl

```
1
2
```

```

3
4 clc
5 // given that
6 Na=6.023*10^23 //Avogadro No.
7 rho=1.955 //Density of KCl in g/cm^3
8 A_k= 39.10 //Atomic weight of potassium in g/mol
9 A_cl= 35.45 //Atomic weight of Chlorine in g/mol
10 Qs=2.6 //Activation energy in eV
11 k=8.62*10^-5 //Boltzmann Constant in eV/K
12 T=500+273 //Temperature in K
13
14 printf("Example 5.4\n")
15 A = A_k+A_cl // Molar mass of KCl in gram
16 N=Na*rho*1e6/A //No. of atomic site per cubic meter
17 Ns=N*exp(-Qs/(2*k*T))
18 printf(" \n Number of Schottky defects are %.2e
        defects/m^3." ,Ns)

```

---

**Scilab code Exa 5.5** Determination of possible point defect types in NaCl due to presence of Ca ions

```

1
2 clc
3 printf(" Example 5.5\n")
4 printf(" \n It is a theoretically solved example\n")

```

---

**Scilab code Exa 5.6** Computations of ASTM Grain Size Number of Grains per Unit Area

```

1
2
3 clc
4 // given that
5 N=45 //Number of grains per square inch
6 M=85 // magnification
7
8 printf("Example 5.6\n");
9 printf(" \n Part A");
10 n=(log(N)/log(2))+1 //calculation for grain size no.
      N=2^(n-1)
11 printf(" \n Grain size number is %.1f\n",n)
12 printf(" \n Part B")
13 Nm=(100/M)^2*2^(n-1)
14 printf(" \n At magnification of 85x\n")
15 printf(" Number of grains per inch square are %.2f\n
      ",Nm)
16 // answer in book is 62.6. It is because of rounding
      off at intermediate stages

```

---



# Chapter 6

## Diffusion

Scilab code Exa 6.1 Diffusion Flux Computation

```
1
2
3 clc
4 Ca=1.2 //Concentration at A in kg/m^3
5 Cb=0.8 //Concentration at B in kg/m^3
6 xa=5*10^-3//Position 1 in m
7 xb=10*10^-3//Position 2 in m
8 D=3*10^-11//Diffusion coefficient in m^2/s
9 printf("Example 6.1\n")
10 J=-D*(Ca-Cb)/(xa-xb)
11 printf("Diffusion flux is %1e kg/m^2-s",J)
```

---

Scilab code Exa 6.2 Nonsteady State Diffusion Time Computation I

```
1
2
3 clc
```

```

4 // given that
5 x=5*10^-4 // Position in m
6 D=1.6*10^-11 // Diffusion coefficient in m^2/s
7 Co=0.25 // Initial Concentration in wt%
8 Cs=1.2 // Surface concentration in wt%
9 Cx=0.8 // Concentration at any x in wt%
10 z1 = 0.35 // tabular data
11 z2 = 0.4 // tabular data
12 erf_z1 = 0.3794 // tabular data
13 erf_z2 = 0.4284 // tabular data
14 printf("Example 6.2\n")
15 C=1-((Cx-Co)/(Cs-Co))
16 z = (C-erf_z1)/(erf_z2-erf_z1) * (z2-z1) + z1 //
    Calculation by interpolation
17 t= ((x/(2*sqrt(D)))/z)^2 // calculation of time
18 printf("\n Time required is %d s or %.2f h\n",t,t
    /3600);
19 // Answer in book is 25400 sec or 7.1 h. It is
    because of considering different number of
    significant figure

```

---

### Scilab code Exa 6.3 Nonsteady State Diffusion Time Computation II

```

1
2
3
4 clc
5 // Given that
6 D_500=4.8*10^-14 // Diffusion coefficient at 500 C
7 D_600=5.3*10^-13 // Diffusion coefficient at 600 C
8 t_600=10 // Time in hours to diffuse
9 printf("Example 6.3\n")
10 t_500=D_600*t_600/D_500

```

```
11 printf("\n Time to diffuse at 500 degree Celsius is
    %.1f h\n",t_500)
```

---

#### Scilab code Exa 6.4 Diffusion Coefficient Determination

```
1
2
3 clc
4 //given that
5 T=550+273 // temperature of aluminium in K
6 D_0=1.2*10^-4 //Temperature independent
    preexponential in m^2/s
7 Q_d=131000 //Activation energy in J/mol-K
8 R=8.31 //Universal Gas constant
9 printf("Example 6.4\n");
10 D=D_0*exp(-Q_d/(R*T));
11
12 printf("\n Diffusion coefficient is %.1e m^2/s\n",D)
    ;
```

---

#### Scilab code Exa 6.5 Diffusion Coefficient Activation Energy and Preexponential Calculations

```
1
2
3 clc
4 //From graph log D ad 1/T are deducted
5 inv_T1=0.8*10^-3 //Reciprocal of temp. in K^-1
6 inv_T2=1.1*10^-3 //Reciprocal of temp. in K^-1
```

```

7 logD1=-12.4
8 logD2=-15.45
9
10 R=8.31 //Gas law Constant in J/mol-K
11
12 printf("Example 6.5\n")
13 Q_d=-2.3*R*(logD1-logD2)/(inv_T1-inv_T2)
14 printf("\n Activation energy is %d kJ/mol",Q_d/1000)
15 D_0=10^(logD2+(Q_d*inv_T2/(2.3*R)))//For calculating
    Preexponential factor
16 printf("\n Preexponential factor D_0 is %.1e m^2/s\n
    ",D_0)
17 // Answer in book is 5.2e-5 m^2/s. It is because of
    consideration of different number of significant
    figures

```

---

**Scilab code Exa 6.6** Computation of Diffusion Flux of carbon dioxide through a plastic beverage shelf life

```

1
2
3 clc
4 //given that
5 P_m = 2.3e-14 // permissibility coefficient of CO2
    through PET
6 P_1 = 400 // Pressure inside bottle in KPa
7 P_2 = 0.4 // Pressure outside bottle in KPa
8 A = 500 // Surface area of bottle in cm^2
9 x = 0.05 // wall thickness of bottle in cm
10 V = 750 // volume in cm^3
11 printf("\Example 6.6\n")
12 J = -P_m*(P_2-P_1)*1e3/x // calculation of diffusion
    flux

```

```
13 printf("\n Part A:")
14 printf("\n Diffusion flux is %0.1e cm^3 STP/(cm^2-s)
    ",J)
15 printf("\n\n Part B:")
16 V_co2 = J*A
17 t = V/V_co2 // calculation of self life
18 printf("\n Self life for bottle of pop is %d days (
    or about %d months).", t/(60*60*24),t
    /(60*60*24*30))
19 // Answer in book is 97 days. It is because of
    considering different number of significant
    figure
```

---

# Chapter 7

## Phase diagram

Scilab code Exa 7.1 Derive Lever rule

```
1 clc
2 printf(" Example 7.1 \n")
3 printf(" \n It is a derivational problem \n")
```

---

Scilab code Exa 7.2 Relative phase amount determination mass and volume fraction

```
1
2 clc
3 printf(" Example 7.2 \n")
4 printf(" \n It is a diagram based theoretically
    solved example \n")
```

---

**Scilab code Exa 7.3** Determination of phase present and computation of phase composition

```
1
2 clc
3 // given that
4 C1=40 // Overall alloy composition
5 Cb=98
6 Ca=10
7 rho_Sn=7.24 // density in g/cm^3 density of tin
8 rho_Pb=11.23 // density in g/cm^3 density of lead
9 Ca_Sn=10
10 Ca_Pb=90
11 Cb_Sn=98
12 Cb_Pb=2
13 printf("Example 7.3\n")
14
15 printf(" \n Part A:")
16
17 Wa=(Cb-C1)/(Cb-Ca)
18 Wb=(C1-Ca)/(Cb-Ca)
19
20 printf(" \n Mass fraction for alpha and beta phases
    is %.2f and %.2f respectively\n",Wa,Wb);
21 printf(" \n Part B:");
22
23 rho_a=100/((Ca_Sn/rho_Sn)+(Ca_Pb/rho_Pb));
24 rho_b=100/((Cb_Sn/rho_Sn)+(Cb_Pb/rho_Pb));
25
26 printf(" \n Density of alpha phase is : %.2f g/cm^3",
    rho_a);
27 printf(" \n Density of beta phase is : %.2f g/cm^3",
    rho_b);
28
29 Va=Wa/(rho_a*((Wa/rho_a)+(Wb/rho_b)));
30 Vb=Wb/(rho_b*((Wa/rho_a)+(Wb/rho_b)));
31 printf(" \n Volume fraction of alpha phase : %.2f",Va
    );
```

```
32 printf("\n Volume fraction of beta phase : %.2f",Vb)
    ;
```

---

#### Scilab code Exa 7.4 Determining ferrite and cementite phase

```
1
2
3 clc
4 // given that
5 C0=0.35 // given composition
6 C_a=0.022 // given composition
7 C_Fe3C=6.7 // given composition
8 C_p=0.76
9 printf("Example 7.4\n")
10
11 printf("\n Part A: ")
12 W_a=(C_Fe3C-C0)/(C_Fe3C-C_a)
13 W_Fe3C=(C0-C_a)/(C_Fe3C-C_a)
14
15 printf("\n Mass fraction of total ferrite phase is
    %.2f",W_a)
16 printf("\n Mass fraction of cementide phase is %.2f\n
    ",W_Fe3C)
17
18 printf("\n Part B:")
19 Wp=(C0-C_a)/(C_p-C_a)
20 W_a1=(C_p-C0)/(C_p-C_a)
21
22 printf("\n Mass fraction of Pearlite is %.2f",Wp)
23 printf("\n Mass fraction of proeutectoid ferrite is
    %.2f\n",W_a1)
24
25 printf("\n Part C:")
```



```
26
27 Wae=W_a-W_a1
28 printf("\n Mass fraction of eutectoid ferrite : %.3f
        \n",Wae)
29 // Answer in book is 0.39. It is due to considering
        different number of significant figure at
        intermediate steps
```

---

# Chapter 8

## Phase transformation

**Scilab code Exa 8.1** Computation of Critical Nuclear Radius and Activation energy

```
1
2
3 clc
4 // Given that
5 Hf=-1.16*10^9 // Latent heat of fusion in J/m^3
6 Y=0.132 // Surface energy in J/m^2
7 Tm=1064+273 // Melting point of gold in K
8 T=1064+273-230 // 230 is supercooling value in K
9 a=0.413*10^-9 // Unit Cell edge length in m
10 n = 4 // Number of atoms in a FCC unit cell
11 printf("Example 8.1\n");
12
13 printf("\n Part A");
14
15 r=-2*Y*Tm/(Hf*(Tm-T));
16
17 printf("\n Critical Radius is : %.2f nm\n",r/10^-9);
18
19 G=16*pi*Y^3*Tm^2/(3*Hf^2*(Tm-T)^2);
20
```

```

21 printf(" Activation free energy is : %.2e J\n",G) //
    Answer in book is 9.64e-19 J. It is due to
    approximation at intermediate stage
22
23 printf("\n Part B")
24 u_c=4*%pi*r^3/(3*a^3)
25
26 printf("\n Unit cells per paricle are : %d",u_c);
27 tot_uc = n*int(u_c) // Total no. of atoms per
    critical nucleus
28 printf("\n Total no. of atoms per critical nucleus
    are : %d\n",tot_uc);

```

---

**Scilab code Exa 8.2** Microstructure determination for three isothermal heat transfer

```

1 clc
2 printf(" Example 8.2 \n")
3 printf("\n It is a diagram based theoretically
    solved example\n")

```

---

# Chapter 9

## Mechanical properties of metals

Scilab code Exa 9.1 Elongation or Elastic Computation

```
1
2
3 clc
4 // given that
5 E=110*10^3 //Young's modulus of Copper in MPa
6 sigma=276 //Applied stress in MPa
7 l_o=305 //Original length in mm
8 printf("Example 9.1\n")
9
10 del_l=sigma*l_o/E //Deformation
11
12 printf("\n Elongation obtained is %.2f mm \n",del_l)
```

---

Scilab code Exa 9.2 Computation of Load to Produce Specified Diameter Change

```

1
2 clc
3 // given that
4 del_d=-2.5*10^-3 //Deformation in diameter in mm
5 d_0=10 //Initial diameter in mm
6 v=0.34 //Poisson ratio for brass
7 E=97*10^3 //Modulus of elasticity in MPa
8 printf("Example 9.2\n")
9
10 e_x=del_d/d_0 // Strain in x-direction
11 printf(" \n Strain in x-direction is %.1e",e_x)
12
13 e_z=-e_x/v
14 printf(" \n Strain in z-direction is %.2e",e_z)
15
16 sigma=e_z*E // Stress produced
17 F=sigma*pi*(d_0^2)/4
18
19 printf(" \n Applied force is %d N",F)
20 // Answer in book is 5600N. It is due to rounding
    off at intermediate steps

```

---

**Scilab code Exa 9.3** Mechanical property determination from stress stain curve

```

1
2 clc
3 // given that
4
5 sigma_2=150 // Stress at a point in MPa
6 sigma_1=0 // Stress at a point in MPa
7 epsilon_2=0.0016 // Strain at a point in MPa
8 epsilon_1=0 // Strain at a point in MPa

```

```

 9 d_0=12.8*10^-3 //Initial Diameter in m
10 sigma=450*10^6 //tensile strength in MPa
11 l0=250; //Initial length in mm
12 e=0.06; //strain
13 printf("Example 9.3\n")
14
15 printf("\n Part A")
16 E=(sigma_2-sigma_1)/(epsilon_2-epsilon_1) //Young's
    Modulus = stress/strain
17 printf("\n Modulus of elasticity is %.1f GPa \n
    which is very close to its true value 97 GPa",E
    /10^3)
18
19 printf("\n\n Part C")
20 A_0=%pi*d_0^2/4
21
22 F=sigma*A_0
23 printf("\n Maximum load sustained is %d N\n",floor(F
    /10)*10)
24 printf("\n Part D")
25 dl=e*l0
26 printf("\n Change in length is %d mm",dl);

```

---

#### Scilab code Exa 9.4 Ductile and true stress at fracture composition

```

1 clc
2 // given that
3
4 di=12.8 //Initial diameter in mm
5 df=10.7 //Final diameter in mm
6 sigma=460*10^6 //Tensile strength
7 printf("Example 9.4\n")
8

```

```

9  printf("\n Part A")
10
11 RA = ((di^2-df^2)/di^2)*100 //Ductility in terms of
    Reduction Area
12 printf("\n Percent reduction in area is %d%%\n",RA)
13
14 printf("\n Part B")
15 A_o=%pi*di^2*10^-6/4
16
17 F=sigma*A_o
18 Af=%pi*df^2/4
19 sigma_t=F/Af
20 printf("\n True stress at failure is %d MPa",sigma_t
    )
21 // Answer in book is 660. It is due to founding off
    at intermediate stage

```

---

### Scilab code Exa 9.5 Calculation of Strain Hardening Exponent

```

1
2
3  clc
4  //Given that
5
6  sigma_t=415 //True stress in MPa
7  et=0.1 //True strain
8  K=1035 // Bulk modulus in MPa
9
10 printf(" Example 9.5\n")
11
12 n=log(sigma_t/K)/log(et)
13
14 printf("\n Strain hardening coefficient is %.2f",n);

```

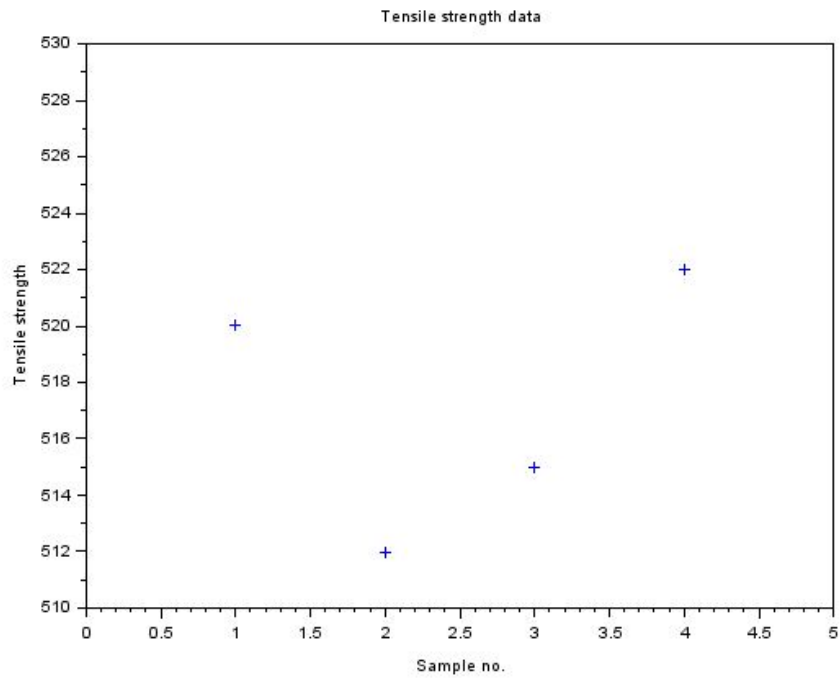


Figure 9.1: Average Computations

---

### Scilab code Exa 9.6 Average Computations

```
1
2
3 clc
4 // given that
5 n=[0 1 2 3 4 5]
6 T_S=[510 520 512 515 522 530]
```



```

7
8 printf(" Example 9.6 \n")
9
10 //First and Last points are arbitrary to plot the
    required points
11
12 printf("\n Part A:")
13 plot(n,T_S, '+')
14 xtitle('Tensile strength data','Sample no.','Tensile
    strength');
15
16 //Mean Tensile strength
17 i=2;
18 TS_mean=0;
19
20 for i=2:5
21     TS_mean=TS_mean+(T_S(i)/4);
22 end
23 printf("\n Average tensile strength is %d MPa.\n",
    TS_mean);
24 //Standard Deviation
25 printf("\n\n Part B:")
26 j=0
27 std=0
28
29 for i=2:5
30     std=std+((T_S(i)-TS_mean)^2/(4-1))
31 end
32
33 printf("\n Standard deviation is %.1f MPa\n",sqrt(
    std))

```

---

**Scilab code Exa 9.7** Specification of support Post Diameter

```
1
2
3 // Page no 334
4 clc
5 // Given
6 f = 220000 // Maximum load in N
7 sigma_y = 310 // Minimum yield strength in MPa
8 sigma_t = 565 // Tensile strength in MPa
9 N= 5 // Factor of safety
10 printf("Design Example 9.1\n")
11 sigma_w = sigma_y*1e6/N
12 d = 2*sqrt (f/(2*pi*sigma_w))// Calculation of
    diameter
13 printf("n Diameter of each of two rods should be %
    .1 f mm\n",d*1e3)
```

---

# Chapter 10

## Dislocation and strengthening mechanism

Scilab code Exa 10.1 Resolved shear stress and stress to initiate yielding computations

```
1
2
3
4 clc
5 u1 = 0 // Bravais index
6 v1 = 1// Bravais index
7 w1 = 0// Bravais index
8 u2 = 1// Bravais index
9 v2 = 1// Bravais index
10 w2 = 0// Bravais index
11 u3 = -1// Bravais index
12 v3 = 1// Bravais index
13 w3 = 1// Bravais index
14 tau_r1 = 30 // Critical resolved shear stress
15 sigma = 52// Tensile strength in MPa
16 printf("Example 10.1\n")
17 printf(" \n Part A:")
18 phi = acos((u1*u2+v1*v2+w1*w2)/sqrt((u1^2+v1^2+w1^2)
```

```

    *(u2^2+v2^2+w2^2))
19 lambda = acos((u3*u1+v3*v1+w3*w1)/sqrt((u1^2+v1^2+w1
    ^2)*(u3^2+v3^2+w3^2)))
20 tau_r = sigma*cos(phi)*cos(lambda)
21 printf("\n Resolved shear stress is %.1f MPa",tau_r)
22 // Answer in book is 21.3 MPa which is due to
    approximation
23 printf("\n\n Part B:")
24 sigma1 = tau_r1/(cos(phi)*cos(lambda))
25 printf("\n Applied tensile force to initiate
    yielding is %.1f MPa",sigma1)
26 // Answer in book is 73.4 MPa which is due to
    approximation

```

---

**Scilab code Exa 10.2** Tensile strength and ductility Determination for cold worked copper

```

1
2
3 clc
4 // Given that
5 d1 = 15.2 // Initial diameter in mm
6 d2 = 12.2 // Final diameter in mm
7 printf("Example 10.2\n")
8 per_CW = (d1^2 - d2^2)*100/d1^2
9 printf("\n The tensile strength is read directly
    from the curve for copper (figure 10.9b) \n as 340
    MPa From figure 10.19c, the Ductility at %0.1f
    CW is about 7%% EL.",per_CW)
10
11 // Some values are deduced from figures

```

---

**Scilab code Exa 10.3** Tensile strength and ductility Determination for cold worked copper

```
1
2
3 //page no 371
4 clc
5 // Given that
6 d1 = 6.4 // Initial diameter in mm for first drawing
7 sigma = 345 // tensile strength in MPa
8 e1 = 20 // ductility in percent
9 d2 = 5.1 // Final diameter in mm
10 per_cw = 21.5 // deformation
11 printf("Design Example 10.1\n")
12 per_CW = (d1^2 - d2^2)*100/d1^2
13 d0 = sqrt((d2^2*100)/(100-e1))
14 printf("\\n Theoretical %% Cold Work is %.1f",per_CW)
15 printf("\\n Original Diameter for second drawing is %
    .1f mm",d0)
16 // Answer in book is 5.8 mm which is due to
    approximation at intermediate steps
```

---

# Chapter 11

## Failure of metals

Scilab code Exa 11.1 Maximum Flaw Length Computation

```
1
2
3
4 clc
5 // given that
6
7 sigma=40*10^6 // in Pa Tensile stress
8 E=69*10^9 //Modulus of elasticity in pa
9 Ys=0.3 //Specific surface energy in N/m^2
10 printf(" Example 11.1\n");
11
12 a=2*E*Ys/(%pi*sigma^2) //Maximum length of surface
    flaw
13
14 printf(" \n Maximum length of surface flaw without
    fracture is %.1f micro meter \n",a*1e6);
```

---

### Scilab code Exa 11.2 Materials specification for a pressurized steel tank

```
1 clc
2 //page no 399
3 printf(" Design Example 11.1\n")
4 printf("\n It is a diagram and table based
   theoretically solved example\n")
```

---

### Scilab code Exa 11.3 Rupture Lifetime Prediction

```
1
2 //page no 421
3 clc
4 // Given that
5 T=800+273; // Ambient temperature in K
6 //stress is 140 MPa
7 L_M=24*10^3 // Larson - miller parameter
8 //From Graph of Fig. 8.32 Larson-Miller Parameter is
   deduced
9 printf("\n Design Example 11.2\n")
10 t=10^((L_M/T)-20)
11 printf("\n Time to rupture for a component is %d
   hours(%.1f days)\n",t,t/(24))
12 // Answer in book is 233 hours. it is because of
   approximation at intermediate stage
```

---

# Chapter 13

## Structure of polymers

**Scilab code Exa 13.1** Computation of Average molecular weights and degree of polymerization

```
1
2
3 clc
4 // Given that
5 m1 = 7500 // Mean molecular weight in g/mol
6 m2 = 12500 // Mean molecular weight in g/mol
7 m3 = 17500 // Mean molecular weight in g/mol
8 m4 = 22500 // Mean molecular weight in g/mol
9 m5 = 27500 // Mean molecular weight in g/mol
10 m6 = 32500 // Mean molecular weight in g/mol
11 m7 = 37500 // Mean molecular weight in g/mol
12 x1 = 0.05 // Mole fraction
13 x2 = 0.16 // Mole fraction
14 x3 = 0.22 // Mole fraction
15 x4 = 0.27 // Mole fraction
16 x5 = 0.20 // Mole fraction
17 x6 = 0.08 // Mole fraction
18 x7 = 0.02 // Mole fraction
19 w1 = 0.02 // weight fraction
20 w2 = 0.10 // weight fraction
```



```

21 w3 = 0.18 // weight fraction
22 w4 = 0.29 // weight fraction
23 w5 = 0.26 // weight fraction
24 w6 = 0.13 // weight fraction
25 w7 = 0.02 // weight fraction
26 m_c = 12.01 // molar mass of carbon in gram/mole
27 m_h = 1.01 // molar mass of hydrogen in gram/mole
28 m_cl = 35.45 // molar mass of chlorine in gram/mole
29
30 printf("\n Problem 13.1")
31 printf("\n Part A:")
32 M_n = m1*x1+m2*x2+m3*x3+m4*x4+m5*x5+m6*x6+m7*x7
33 printf("\n The number average molecular weight is %d
    gram/mol",M_n)
34 printf("\n\n Part B:")
35 m = 2*m_c+3*m_h+1*m_cl // Mass of repeating unit
36 DP = M_n/m
37 printf("\n The degree of polimerization is %d ",DP)
38 printf("\n\n Part C:")
39 M_w = m1*w1+m2*w2+m3*w3+m4*w4+m5*w5+m6*w6+m7*w7
40 printf("\n The weight average molecular weight is %d
    gram/mol",M_w)

```

---

# Chapter 15

## Composite materials

**Scilab code Exa 15.1** Property Determination for a glass fiber reinforced composite longitudinal direction

```
1
2
3 clc
4 // given that
5 v_f = 40 // volume percent of fibre in composite
6 E_f= 69 // Modulus of elasticity of fibre in GPa
7 v_m = 60 // volume percent of matrix in composite
8 E_m = 3.4 // Modulus of elasticity of matrix in GPa
9 a = 250 // cross sectional area in mm^2
10 sigma = 50 // Tensile stress in MPa
11 Fm = 1 // let
12 Ff = 13.5*Fm
13 printf("\\n Example 15.1")
14 printf("\\n Part A:")
15 E_c1 = (v_f*E_f+v_m*E_m)/100
16 printf("\\n Modulus of elasticity of composite is %0
    .0f GPa",E_c1)
17 printf("\\n\\n Part B:")
18 Fc = a*sigma
19 Fm = Fc/(Fm+Ff)
```

```

20 Ff = Fc - Fm
21 printf("\n Force supported by m is %d N \n Force
    supported by fibre is %d N",Fm,Ff)
22 // Answer in book is as Fm = 860 N and Ff = 11640.
    It is due to approximation.
23 printf("\n\n Part C")
24 a_m = v_m*a/100
25 a_f = v_f*a/100
26 sigma_m = Fm/a_m
27 sigma_f = Ff/a_f
28 epsilon_m = sigma_m/(E_m*1000)
29 epsilon_f = sigma_f/(E_f*1000)
30 printf("\n Strain for matrix phase is %0.2e\n",
    epsilon_m)
31 printf(" Strain for fibre phase is %.2e.\n Both are
    identical",epsilon_f)

```

---

**Scilab code Exa 15.2** Elastic Modulus Determination for a Glass Fiber Reinforced Composite Transverse Direction

```

1
2
3 clc
4
5 printf(" Example 15.2\n")
6 E_gf=69 // Elasticity of glass fibre in GPa
7 mf_gf=0.4 //Volume percentage of glass fibre
8 E_pr=3.4 // Elasticity of polyester resin in GPa
9 mf_pr=0.6 //Vol percentage of polyester resin
10
11 E_ct=E_pr*E_gf/((E_pr*mf_gf)+(E_gf*mf_pr)) //
    Calculation of modulus of elasticity in GPa
12

```

```
13 printf("\n In transverse direction , modulus of
    elasticity is %.1f GPa.\n" ,ceil(E_ct*10)/10)
```

---

### Scilab code Exa 15.3 Design a tubular composite shaft

```
1 //page no 563
2 clc
3 // Given
4 F = 1000 // Force in N
5 L = 1 // length in m
6 del_y = 0.35 // extension in mm
7 d_o = 70 // Outer diameter in mm
8 d_i = 50 // Innrer diameter in mm
9 V_f_max = 0.6 // Maximum allowable fiber Volume in
    cm fraction
10 Vf_glass = 0.945 // V_f for glass
11 Vf_C_standard = 0.293// V_f for carbon standard
    modulus
12 Vf_c_intermediate = 0.237// V_f for carbon
    intermediate modulus
13 Vf_c_high = 0.168 // V_f for carbon high modulus
14 d_epoxy = 1.14 // density of epoxy resin in g/cm^3
15 d_C_sm = 1.8 // density of carbon fiber (Standard
    modulus) in g/cm^3
16 d_C_im = 1.8 // density of carbon fiber (
    intermediate modulus) in g/cm^3
17 d_C_hm = 1.8 // density of carbon fiber (high
    modulus) in g/cm^3
18 C_im_cost = 70.00 // cost of carbon fiber (
    intermediate modulus) in USD/kg
19 C_sm_cost = 35.00 // cost of carbon fiber (Standard
    modulus) in USD/kg
20 C_hm_cost = 175.00 // cost of carbon fiber (high
```

```

    modulus) in USD/kg
21 d_epoxy = 1.14 // density of epoxy resin in g/cm^3
22 epoxy_cost = 9.00 // cost of epoxy resin in USD/kg
23
24 printf("Design Example 15.1 \n")
25 I = %pi/64* (1e-12*(d_o*1e-3)^4-(d_i*1e-3)^4)
26 E = 4*F*L^3/(3*%pi*del_y*1e-3*((d_o*1e-3)^4-(d_i*1e
    -3)^4)) // Required modulus of elasticity
27 printf("\n Part A:\n")
28 if Vf_glass < V_f_max then
29     printf("\n Glass, when embedded in epoxy matrix,
        meet the stipulated criteria. \n")
30 end
31 if Vf_C_standard < V_f_max then
32     printf("\n Carbon (standard modulus), when
        embedded in epoxy matrix, meet the stipulated
        criteria. \n")
33 end
34 if Vf_c_intermediate < V_f_max then
35     printf("\n Carbon (intermediate modulus), when
        embedded in epoxy matrix, meet the stipulated
        criteria.\n ")
36 end
37 if Vf_c_high < V_f_max then
38     printf("\n Carbon (high modulus), when embedded in
        epoxy matrix, meet the stipulated criteria.\n"
        )
39 end
40 printf("\n Part B:\n")
41 Vc = %pi*L*1e-6*(d_o^2 - d_i^2)/4
42 F_v_C_sm = Vc*Vf_C_standard*1e6 // Fiber Volume in
    cm^3 for carbon (Standard modulus)
43
44 F_m_C_sm = F_v_C_sm * d_C_sm/1000 // Fiber mass for
    carbon (Standard modulus) in kg
45
46 F_c_C_sm = F_m_C_sm * C_sm_cost // Fiber cost for
    carbon (Standard modulus) in USD

```

```

47
48 m_v_C_sm = Vc*(1-Vf_C_standard)*1e6 // Matrix Volume
      in cm^3 for carbon (Standard modulus)
49
50 m_m_C_sm = m_v_C_sm * d_epoxy/1000 // Matrix mass
      for carbon (Standard modulus) in kg
51
52 m_c_C_sm = m_m_C_sm * epoxy_cost // Matrix cost for
      carbon (Standard modulus) in USD
53
54 Total_c_C_sm = m_c_C_sm + F_c_C_sm // Total cost for
      carbon (Standard modulus) in USD
55
56 F_v_C_im = Vc*Vf_c_intermediate*1e6 // Fiber Volume
      in cm^3 for carbon (intermediate modulus)
57
58 F_m_C_im = F_v_C_im * d_C_im/1000 // Fiber mass for
      carbon (intermediate modulus) in kg
59
60 F_c_C_im = F_m_C_im * C_im_cost // Fiber cost for
      carbon (intermediate modulus) in USD
61
62 m_v_C_im = Vc*(1-Vf_c_intermediate)*1e6 // Matrix
      Volume in cm^3 for carbon (intermediate modulus)
63
64 m_m_C_im = m_v_C_im * d_epoxy/1000 // Matrix mass
      for carbon (intermediate modulus) in kg
65
66 m_c_C_im = m_m_C_im * epoxy_cost // Matrix cost for
      carbon (intermediate modulus) in USD
67
68 Total_c_C_im = m_c_C_im + F_c_C_im // Total cost for
      carbon (intermediate modulus) in USD
69
70 F_v_C_hm = Vc*Vf_c_high*1e6 // Fiber Volume in cm^3
      for carbon (high modulus)
71
72 F_m_C_hm = F_v_C_hm * d_C_hm/1000 // Fiber mass for

```

```

    carbon (high modulus) in kg
73
74 F_c_C_hm = F_m_C_hm * C_hm_cost // Fiber cost for
    carbon (high modulus) in USD
75
76 m_v_C_hm = Vc*(1-Vf_c_high)*1e6 // Matrix Volume in
    cm^3 for carbon (high modulus)
77
78 m_m_C_hm = m_v_C_hm * d_epoxy/1000 // Matrix mass
    for carbon (high modulus) in kg
79
80 m_c_C_hm = m_m_C_hm * epoxy_cost // Matrix cost for
    carbon (high modulus) in USD
81 Total_c_C_hm = m_c_C_hm + F_c_C_hm // Total cost for
    carbon (high modulus) in USD
82 printf(" Cost of Carbon (standard modulus) composite
    is:%.2f ",Total_c_C_sm) // whereas Value in
    table is 48.50 USD
83
84 printf("\n Cost of Carbon (intermediate modulus)
    composite is:%.2f ",Total_c_C_im)// whereas Value
    in table is 71.10 USD
85 printf(" \n Cost of Carbon (high modulus) composite
    is:%.2f ",Total_c_C_hm) // whereas Value in table
    is 115.00 USD
86
87 printf("\n\n The material of choice (i.e. least
    expensive) \n is standard modulus carbon fiber
    composite; the relatively low cost per unit mass
    of this fiber offsets its relatively low modulus
    of elasticity and \n required high Volume
    fraction.")

```

---

# Chapter 16

## Corrosion and degradation

Scilab code Exa 16.1 Determination of Electrochemical Cell Characteristics

```
1
2
3 clc
4 // Given that
5
6 V_Cd=-0.403 //Half Cell Potential of Cd++|Cd in
   volt
7 V_Ni=-0.250 //Half Cell Potential of Ni++|Ni volt
8 C_Ni=10-3
9 C_Cd=0.5
10 n=2 //Net electron exchange in Redox reaction
11 printf(" Example 16.1\n")
12 printf(" \n Part A:")
13 dV=V_Ni-V_Cd // Potential difference in volts
14 printf(" \n Standard Cell potential is : %.3f V\n",dV
   )
15 printf(" \n Part B:")
16
17 V=-dV-(0.0592*log10(C_Ni/C_Cd)/n)
18 printf(" \n Net EMF is : %.3f V\n",V)
```



```

19 printf("\n That is ,")
20
21 if V<0 then
22     printf("\t Cd is oxidised & Ni is reduced \n")
23 else
24     printf("\t Cd is reduced & Ni is oxidised\n")
25 end

```

---

### Scilab code Exa 16.2 Rate of Oxidation Computation

```

1
2 clc
3 // given that
4 VH_H2=0 // half cell voltage
5 VZn_Zn2=-0.763 // half cell voltage
6 iZn=10^-7 // current density in A/cm^2
7 iH2=10^-10 // current density in A/cm^2
8 beta_Zn=0.09
9 beta_H2=-0.08
10 n=2 //Exchange of 2 electrons
11 F=96500//Faradays constant
12
13 printf(" Example 16.2\n")
14 printf("\n Part A")
15 i_c=10^[(VH_H2-VZn_Zn2-(beta_H2*log10(iH2)))+(beta_Zn
    *log10(iZn)))/(beta_Zn-beta_H2)]
16 printf("\n Rate of oxidation with zinc is %.2e A/cm
    ^2",i_c)
17 r=i_c/(n*F)
18 printf("\n Rate of oxidation is %.2e mol/cm^2-s\n",
    floor(r*1e12)/1e12)
19 printf("\n Part B")
20 Vc=VH_H2+(beta_H2*log10(i_c/iH2))

```

```
21 printf("\n Corrosion potential is %.3f V\n",Vc)
```

---

# Chapter 17

## Electrical properties of materials

Scilab code Exa 17.1 Computation of the Room Temperature Intrinsic Carrier Concentration for Gallium Arsenide

```
1
2 clc
3 // Given that
4 sigma=10-6 // ( Electrical Conductivity in Ohm-m)
   -1
5 e=1.6*10-19 // Charge on electron in Coulomb
6 m_e=0.85 //Mobility of electron in m2/V-s
7 m_h=0.04 // Mobility of holes in m2/V-s
8
9 printf(" Example 17.1\n")
10 n_i=sigma/(e*(m_e+m_h))//n_i is Intrinsic carrier
   concentration
11
12 printf(" \n Intrinsic Carrier Concentration is %.1e m
   -3\n",n_i);
```

---

**Scilab code Exa 17.2** Electrical Conductivity Determination for Intrinsic Silicon at 150 degree C

```
1
2 clc
3 // Given that
4 e=1.6*10^-19 //Charge on electron in Coulomb
5 ni=4*10^19 // number of electron per unit volume
   for Si at 423 K (m^-3)
6 //Values of m_e and m_h are deduced from graphs at
   page No.689
7 m_e=0.06 //Mobility of electron in m^2/V-s
8 m_h=0.022 // Mobility of holes in m^2/V-s
9 printf(" Example 17.2\n")
10 sigma=ni*e*(m_e+m_h) // electrical conductivity
11
12 printf(" \n Electrical Conductivity is %.2f (Ohm-m)
   ^-1\n",sigma)
```

---

**Scilab code Exa 17.3** Room Temperature for Extrinsic Silicon

```
1
2 clc
3 // given that
4 n=10^23 // Carrier Concentration in m^-3
5 e=1.6*10^-19 //Charge on electron in Coulomb
6 //From graph 18.18 m_e is calculated corresponding
   to n=10^23
```

```

7 m_e=0.07 // Mobility of electron in m^2/V-s
8 m_e2=0.04 // Mobility of electron m^2/V-s
9 printf(" Example 17.3\n")
10 printf("\n Part A:\n ")
11 printf("\n Material is n-type \n ")
12 printf("\n Part B\n ")
13 sigma=n*e*m_e // electrical conductivity calculation
    for extrinsic n-type
14 printf("\n Conductivity is just %d (Ohm-m)^-1\n",
    sigma)
15 printf("\n Part C\n");
16 //From graph 18.19a m_e2 is calculated corresponding
    to 373 K
17 sigma2=n*e*m_e2
18 printf("\n Conductivity at T=373 K becomes %d (Ohm-m
    )^-1\n",sigma2);

```

---

#### Scilab code Exa 17.4 Hall Voltage Computation

```

1
2 clc
3 // given that
4 sigma=3.8*10^7 // Electrical Conductivity in (Ohm-m
    )^-1
5 m_e=0.0012 // Mobility of electron in m^2/V-s
6 I_x=25 // Current in Ampere(A)
7 d=15*10^-3 //Thickness in m
8 B_z=0.6 // Magnetic field in Tesla
9 printf("Example 17.4\n")
10 Rh=-m_e/sigma //Hall coefficient
11 printf("\n Hall coefficient is %.2e V-m/A-Tesla\n",
    Rh)
12 Vh=Rh*I_x*B_z/d

```

```
13 printf("\n Hall Voltage is %.2e V\n",Vh)
```

---

### Scilab code Exa 17.5 Computation of Capacitor properties

```
1
2 clc
3 // given that
4 A = 6.45e-4 // Area of plat in m^2
5 l = 2e-3 // separation between plates in m
6 epsilon_r = 6 //dielectric constant of material
7 epsilon_0 = 8.85e-12 // universal constant
8 V = 10 // Applied voltage in Volt
9 printf("\n Example 17.5")
10 printf("\n Part A:")
11 C = epsilon_0*epsilon_r*A/l // Capacitance of a
    parallel plat capacitor
12 printf("\n Capacitance of capacitor is %.2e F",C)
13 printf("\n\n Part B:")
14 Q = C*V // Stored charge calculation
15 printf("\n Stored charge in capacitor is %.2e C",Q)
16 printf("\n\n Part C:")
17 D = epsilon_0*epsilon_r*V/l // Dielectric
    displacement
18 printf("\n Dielectric displacement in capacitor is %
    .2e C/m^2",ceil(D*1e9)/1e9)
19 printf("\n\n Part D:")
20 P = D - epsilon_0*V/l // Polarisation
21 printf("\n Polarization is %.2e C/m^2", ceil(P*1e9)
    /1e9)
```

---

### Scilab code Exa 17.6 Acceptor impurity doping in Silicon

```
1
2
3
4 //page no 649
5 clc
6 // given that
7 p1 = 1e22 // Number of electrons per unit volume
8 e = 1.6e-19 // Charge on electron in coulomb
9 mu_h1 = 0.04 // concentration of holes mobility in m
    ^/Vs
10 sigma_d = 50 // Desired conductivity in (ohm-m)^-1
11 p2 = 1e21 // Number of electrons per unit volume
12 mu_h2 = 0.045 // concentration of holes mobility in m
    ^/Vs
13 p3 = 8e21 // Number of electrons per unit volume
14 mu_h3 = mu_h1 // concentration of holes mobility in m
    ^/Vs
15 N_a = 6.023e23 // Avogadro s constant
16 rho_si = 2.33e6 // density of silicon in g/m^3
17 A_si = 28.09 // molecular weight in g/mol
18 printf("\\n Design Example 17.1\\n")
19 sigma1 = p1*e*mu_h1
20 sigma2 = p2*e*mu_h2
21 sigma3 = p3*e*mu_h3
22 N_Si = N_a*rho_si/A_si
23 Ca = p3*100/(p3+N_Si)
24 printf("\\n Silicon material of p-type of
    conductivity %d (ohm-m)^-1 \\n must have %.2e%%
    doping material",sigma_d,Ca)
```

---





# Chapter 18

## Magnetic Properties

**Scilab code Exa 18.1** Calculation of saturation magnetization and flux density for Nickel

```
1
2 clc
3 // Given that
4 b_m=9.27*10^-24 // Bohr Magneton in ampere*m^2
5 Na=6.023*10^23 //Avogadro's Number
6 d=8.9*10^6 // density in g/m^3
7 uo=4*pi*10^-7 //Permittivity of free space
8 A=58.71 // Atomic weight of Nickel g/mol
9 printf("Example 18.1\n")
10 N=d*Na/A //No. of atoms per cubic meter
11 M=0.6*b_m*N //0.6= Bohr Magneton/atom
12 printf(" \n Part A:")
13 printf(" \n Saturation Magnetization is %.1e Ampere/m
    ",M)
14 M=0.6*b_m*N //0.6= Bohr Magneton/atom
15 printf(" \n \n Part B:")
16 B=uo*M
17 printf(" \n Saturation Flux Density is %.2f Tesla\n",
    B);
```

---

### Scilab code Exa 18.2 Calculation of saturation magnetization of Fe<sub>3</sub>O<sub>4</sub>

```
1
2
3 clc
4 // Given that
5 a=0.839*10^-9 //a is edge length in m
6 b_m=9.27*10^-24 // Bohr Magneton in ampere*m^2
7 n_b=8*4 //8 is no. of Fe++ ions per unit cell and
   4 is Bohr magnetons per Fe++ ion
8 printf("Example 18.2\n")
9 M=n_b*b_m/a^3 //M is Saturation magnetisation
10 printf(" \n Saturation Magnetization is %.1e Ampere/m
   \n",M)
```

---

### Scilab code Exa 18.3 Designing a cubic mixed ferrite magnetic material

```
1
2 //page no 692
3 clc
4 // Given that
5 Ms_Fe=5.25*10^5; //Required saturation
   Magnetisation in A/m
6 b_m=9.27*10^-24; //Bohr Magneton in ampere*m^2
7 a=0.839*10^-9; //a is edge length in m
8 M=5*10^5; //From previous question result
9 printf(" Design Example 18.1\n");
10 y=poly([0], 'y') // Defining X
```

```
11 nb=Ms_Fe*a^3/b_m;
12 // 'x' represent fraction of Mn++ that have
    substituted Fe++
13 n=roots(8*[5*y+4*(1-y)]-nb); //5 is Bohr magnetons
    per Fe++ ion
14                                     //4 is Bohr magnetons
                                        per Mn++ ion
15 printf("\n Replacing %.1f%% of Fe++ with Mn++ would
    produce \n the required saturation magnetisation
    of %.2e A/m\n",n*100,Ms_Fe);
```

---