

Scilab Textbook Companion for  
Applied Physics  
by P. K. Palanisamy<sup>1</sup>

Created by  
G Krishna Chaitanya  
B.Tech  
Computer Engineering  
CVSR College of Engineering  
College Teacher  
NA  
Cross-Checked by  
K. V. P. Pradeep

June 16, 2014

<sup>1</sup>Funded by a grant from the National Mission on Education through ICT, <http://spoken-tutorial.org/NMEICT-Intro>. This Textbook Companion and Scilab codes written in it can be downloaded from the "Textbook Companion Project" section at the website <http://scilab.in>

# Book Description

**Title:** Applied Physics

**Author:** P. K. Palanisamy

**Publisher:** Scitech Publications(india)

**Edition:** 8

**Year:** 2009

**ISBN:** 9788183710985

Scilab numbering policy used in this document and the relation to the above book.

**Exa** Example (Solved example)

**Eqn** Equation (Particular equation of the above book)

**AP** Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

# Contents

List of Scilab Codes	4
1 Bonding in Solids	8
2 Crystal Structures and X ray diffraction	11
3 Principles of Quantum Mechanics	24
4 Electron Theory of Metals	30
6 Dielectric Properties	35
7 Magnetic Properties	40
8 Semiconductors	45
9 Superconductivity	56
10 Lasers	58
11 Fiber Optics and Holography	60

# List of Scilab Codes

Exa 1.1	To determine whether the gaseous molecule is stable . . . . .	8
Exa 1.2	To calculate the energy required and seperation between ion pair . . . . .	9
Exa 1.3	To calculate the bond energy for NaCl molecule . . . . .	9
Exa 1.4	To calculate the cohesive energy of NaCl . . . . .	10
Exa 2.1	To calculate the maximum radius of the interstitial sphere . . . . .	11
Exa 2.2	To calculate the percent volume change . . . . .	11
Exa 2.3	To calculate the volume of the unit cell and density of Zinc . . . . .	12
Exa 2.4	To calculate the maximum radius of the sphere . . . . .	13
Exa 2.5	To calculate the density of diamond . . . . .	13
Exa 2.6	To calculate the distance between two adjacent atoms . . . . .	14
Exa 2.7	To calculate the density of Copper crystal . . . . .	14
Exa 2.8	To calculate the free volume per unit cell . . . . .	15
Exa 2.9	To sketch the crystal plane structures . . . . .	15
Exa 2.10	To sketch the crystal plane structures . . . . .	15
Exa 2.11	To calculate the number of atoms per square millimetre . . . . .	16
Exa 2.12	To calculate the interplanar spacing for the planes . . . . .	16
Exa 2.13	To calculate the ratio of the seperation between successive lattice planes . . . . .	17
Exa 2.14	To calculate the miller indices of a plane . . . . .	18
Exa 2.15	To calculate the wavelength of X rays and maximum order of diffraction possible . . . . .	18
Exa 2.16	To calculate the highest order for which Braggs reflection can be seen . . . . .	19
Exa 2.17	To calculate the interatomic spacing . . . . .	19
Exa 2.18	To calculate the glancing angle . . . . .	20
Exa 2.19	To calculate the distance between 110 planes . . . . .	20

Exa 2.20	To compare the density of lattice points . . . . .	20
Exa 2.21	To calculate the glancing angle . . . . .	21
Exa 2.22	To calculate the cube edge . . . . .	22
Exa 2.23	To calculate the cube edge . . . . .	22
Exa 3.1	To calculate the de Broglie wavelength . . . . .	24
Exa 3.2	To calculate the de Broglie wavelength . . . . .	24
Exa 3.3	To calculate the de Broglie wavelength . . . . .	25
Exa 3.4	To calculate the wavelength . . . . .	25
Exa 3.5	To calculate the uncertainty in momentum . . . . .	25
Exa 3.6	To calculate the lowest energy of an electron . . . . .	26
Exa 3.7	To calculate the energy of electron . . . . .	26
Exa 3.8	To calculate the wavelength . . . . .	27
Exa 3.9	To calculate the minimum energy . . . . .	27
Exa 3.10	To calculate the energy values . . . . .	28
Exa 3.11	To calculate the velocity and kinetic energy of electron . . . . .	28
Exa 3.12	To calculate the wavelength . . . . .	29
Exa 3.13	To calculate the spacing of crystal . . . . .	29
Exa 4.1	To calculate the density and mobility of electrons . . . . .	30
Exa 4.2	To calculate the mobility of electrons . . . . .	30
Exa 4.3	To calculate the relaxation time . . . . .	31
Exa 4.4	To calculate the free electron concentration mobility and drift velocity . . . . .	31
Exa 4.5	To calculate the lowest energy of an electron . . . . .	32
Exa 4.6	To evaluate the fermi function . . . . .	32
Exa 4.7	To calculate the temperature . . . . .	33
Exa 4.8	To calculate the temperature . . . . .	33
Exa 6.1	To calculate the energy stored in the condensor and polarizing the dielectric . . . . .	35
Exa 6.2	To calculate the ratio between electronic and ionic polarizability . . . . .	35
Exa 6.3	To calculate the difference in magnetic potential energy . . . . .	36
Exa 6.4	To calculate the dielectric constant of material . . . . .	37
Exa 6.5	To calculate the electronic polarizability . . . . .	37
Exa 6.6	To calculate the capacitance and charge on plates . . . . .	38
Exa 6.7	To calculate the electronic polarizability . . . . .	38
Exa 6.8	To calculate the resultant voltage across capacitors . . . . .	39
Exa 6.9	To calculate the dielectric displacement . . . . .	39
Exa 7.1	To calculate the relative permeability . . . . .	40

Exa 7.2	To calculate the relative permeability of material . . . .	40
Exa 7.3	To calculate the magnetisation and flux density . . . .	41
Exa 7.4	To calculate the magnetisation and flux density . . . .	41
Exa 7.5	To calculate the magnetic moment . . . . .	41
Exa 7.6	To calculate the change in magnetic moment . . . . .	42
Exa 7.7	To calculate the susceptibility . . . . .	42
Exa 7.8	To calculate the magnetic moment . . . . .	43
Exa 7.9	To calculate the temperature . . . . .	43
Exa 7.10	To calculate the magnetic moment and saturation magnetisation . . . . .	43
Exa 8.1	To calculate the resistivity . . . . .	45
Exa 8.2	To determine the position of Fermi level . . . . .	45
Exa 8.3	To calculate the concentration of intrinsic charge carriers	46
Exa 8.4	To calculate the resistivity . . . . .	46
Exa 8.5	To calculate the resistance . . . . .	47
Exa 8.6	To calculate the conductivity . . . . .	47
Exa 8.7	To calculate the intrinsic carrier density and conductivity	48
Exa 8.8	To calculate the forbidden energy gap . . . . .	49
Exa 8.9	To calculate the energy band gap . . . . .	49
Exa 8.10	To calculate the temperature . . . . .	50
Exa 8.11	To calculate the electron concentration . . . . .	50
Exa 8.12	To calculate the conductivity of intrinsic Silicon . . . .	51
Exa 8.13	To calculate the conductivity equilibrium hole concentration and position of Fermi level . . . . .	52
Exa 8.14	To calculate the diffusion coefficient of electrons . . . .	52
Exa 8.15	To calculate the Hall voltage . . . . .	53
Exa 8.16	To calculate the Hall coefficient . . . . .	53
Exa 8.17	To calculate the density and mobility of charge carriers	54
Exa 8.18	To calculate the magnitude of Hall voltage . . . . .	54
Exa 8.19	To calculate the value of $\mu$ and $n$ . . . . .	55
Exa 9.1	To calculate the critical field . . . . .	56
Exa 9.2	To calculate the critical current . . . . .	56
Exa 9.3	To calculate the penetration depth . . . . .	57
Exa 9.4	To calculate the critical temperature . . . . .	57
Exa 10.1	To calculate the relative population . . . . .	58
Exa 10.2	To calculate the divergence . . . . .	58
Exa 10.3	To calculate the spot size . . . . .	59
Exa 11.1	To calculate the numerical aperture . . . . .	60

Exa 11.2	To calculate the angle of acceptance . . . . .	60
Exa 11.3	To calculate the refractive index of the core . . . . .	61
Exa 11.4	To calculate the fractional index change . . . . .	61
Exa 11.5	To calculate the numerical aperture and acceptance angle	61



# Chapter 1

## Bonding in Solids

Scilab code Exa 1.1 To determine whether the gaseous molecule is stable

```
1 //To determine whether the gaseous molecule is
  stable
2 IE_1 = 502;      //first ionisation energy, kJ/mol
3 EA_B = -335;    //electron affinity for B atom, kJ/
  mol
4 e = 1.602*10^-19;
5 r = 0.3;        //inter ionic seperation, nm
6 r = r*10^-9;    //inter ionic seperation, m
7 N = 6.022*10^23*10^-3;
8 epsilon0 = 8.85*10^-12;    //permittivity of free
  space, C/N-m
9 E = (-e^2*N)/(4*%pi*epsilon0*r);    //
  electrostatic attraction energy, kJ/mol
10 printf("electrostatic attraction energy is %d kJ/mol
  ",E);
11 dE = IE_1+EA_B+E;    //net change in energy per
  mole
12 printf("net change in energy is %d kJ/mol",dE);
13 printf("since the net change in energy is negative,
  A+B- molecule will be stable");
14
```

```
15 //answer for net change, dE given in the book is
    wrong
```

---

**Scilab code Exa 1.2** To calculate the energy required and separation between ion pair

```
1 //To calculate the energy required and separation
  between ion pair
2 IP_K = 4.1;      //IP of K, eV
3 EA_Cl = 3.6;    //EA of Cl, eV
4 e = 1.602*10^-19;
5 epsilon0 = 8.85*10^-12;
6 delta_E = IP_K - EA_Cl;      //energy required, eV
7 printf("energy required to form K+Cl- ion pair is %f
  eV",delta_E);
8 //if their total energy is 0, delta_E = Ec
9 Ec = delta_E;
10 R = e/(4*pi*epsilon0*Ec);    //separation between
  ion pair, m
11 R = R*10^9;      //separation between ion pair,
  nm
12 printf("separation between ion pair is %5.2f nm",R);
```

---

**Scilab code Exa 1.3** To calculate the bond energy for NaCl molecule

```
1 //To calculate the bond energy for NaCl molecule
2 e = 1.602*10^-19;
3 EA = 3.65;      //electron affinity of Cl, eV
4 IP = 5.14;     //ionisation energy of Na, eV
5 epsilon0 = 8.85*10^-12;
6 r0 = 236;      //equilibrium distance, pm
7 r0 = r0*10^-12; //equilibrium distance, m
```

```

8 V = (-e^2)/(4*pi*epsilon0*r0);          //potential
   energy, J
9 V = V/e;          //potential energy, eV
10 Ue = V;
11 BE = -Ue - IP + EA;          //bond energy, eV
12 printf("bond energy for NaCl molecule is %5.2f eV",
   BE);

```

---

**Scilab code Exa 1.4** To calculate the cohesive energy of NaCl

```

1 //To calculate the cohesive energy of NaCl
2 e = 1.602*10^-19;
3 epsilon0 = 8.85*10^-12;
4 r0 = 0.281;          //equilibrium seperation, nm
5 r0 = r0*10^-9;          //equilibrium seperation, m
6 A = 1.748;          //Madelung constant
7 n = 9;          //born repulsive exponent
8 CE = (A*e^2)*(1-(1/n))/(4*pi*epsilon0*r0);          //
   cohesive energy, J
9 CE = CE/e;          //cohesive energy, eV
10 printf("cohesive energy of NaCl is %5.3f eV",CE);

```

---

## Chapter 2

# Crystal Structures and X ray diffraction

**Scilab code Exa 2.1** To calculate the maximum radius of the interstitial sphere

```
1 //To calculate the maximum radius of the
   interstitial sphere
2 //We know that body diagonal = 4*r = sqrt(3).a
3 //hence a = (4/sqrt(3)).r
4 //R = (a-2*r)/2
5 //R = ((4/(2*sqrt(3)))*r) - (2*r/2) = (2*r/sqrt(3))
   - r
6 //R = 1.155*r - r = 0.155*r
7 printf("The maximum radius of the interstitial
   sphere is 0.155*r");
```

---

**Scilab code Exa 2.2** To calculate the percent volume change

```
1 //To calculate the percent volume change
2 r_BCC = 1.258; //atomic radius, A
```

```

3 r_FCC = 1.292;           //atomic radius , A
4 a_BCC = (4*r_BCC)/sqrt(3); //In BCC, A
5 a_BCCm = a_BCC*10^-10;   //converting a from A to
    m
6 V_BCC = a_BCCm^3;       //volume of unit cell , m^3
7 n_BCC = ((1/8)*8)+1;    //number of atoms per unit
    cell
8 V1_BCC = V_BCC/n_BCC;   //volume occupied by 1
    atom, m^3
9 a_FCC = 2*sqrt(2)*r_FCC; //In FCC, A
10 a_FCCm = a_FCC*10^-10; //converting a from A to
    m
11 V_FCC = a_FCCm^3;      //volume of unit cell , m^3
12 n_FCC = ((1/2)*6) + ((1/8)*8); //number of
    atoms per unit cell
13 V1_FCC = V_FCC/n_FCC;  //volume occupied by 1
    atom, m^3
14 delta_V = (V1_BCC - V1_FCC)*100/V1_BCC; //
    change in volume in %
15 printf("decrease of volume during conversion from
    BCC to FCC is %5.1f percent",delta_V);

```

---

**Scilab code Exa 2.3** To calculate the volume of the unit cell and density of Zinc

```

1 //To calculate the volume of the unit cell and
    density of Zinc
2 a = 0.27;           //nearest neighbour distance , mm
3 c = 0.494;         //height of unit cell , mm
4 N = 6.023*10^26;   //avagadro number , k/mol
5 M = 65.37;         //atomic weight , kg
6 a_m = a*10^-9;     //nearest neighbour distance , m
7 c_m = c*10^-9;     //height of unit cell , m
8 V = 3*sqrt(3)*(a_m^2)*c_m/2; //volume of unit
    cell , m^3

```

```

9 //if rho is density then mass = V*rho
10 //V*rho = 6*M/N
11 rho = (6*M)/(N*V); //density , kg/m^3
12 printf("Volume of the unit cell in m^3 is");
13 disp(V);
14 printf("density of Zinc is %d kg/m^3",rho);
15
16 //answer for density given in the book is wrong

```

---

**Scilab code Exa 2.4** To calculate the maximum radius of the sphere

```

1 //To calculate the maximum radius of the sphere
2 //a = 4*r/sqrt(2) and R = (a/2) - r
3 //R = (4*r/(2*sqrt(2))) - r = (2*r/sqrt(2)) - r
4 //R = r*(sqrt(2)-1) = 0.414*r
5 printf("maximum radius of the sphere that can fit
   into the void is 0.414*r");

```

---

**Scilab code Exa 2.5** To calculate the density of diamond

```

1 //To calculate the density of diamond
2 a = 0.356; //cube edge of diamond, nm
3 aw = 12.01; //atomic weight of carbon in kg
4 N = 6.023*10^26; //avagadro's number
5 a_m = a*10^-9; //cube edge of diamond, m
6 n = 8/(a_m^3); //number of atoms/m^3
7 M = aw/N; //mass of 1 carbon atom, kg
8 rho = M*n;
9 printf("number of atoms per m^3 is");
10 disp(n);
11 printf("density of diamong is %d kg/m^3",rho);
12
13 //answer for density given in the book is wrong

```

---

**Scilab code Exa 2.6** To calculate the distance between two adjacent atoms

```
1 //To calculate the distance between two adjacent
  atoms
2 MW = 23+35.5;      //molecular weight of NaCl, gm
3 N = 6.023*10^23;   //avagadro number, mol-1
4 rho = 2.18;        //density of NaCl, gm/cm^3
5 M = MW/N;          //mass of NaCl molecule, gm
6 n = rho/M;         //number of molecules per unit volume
7 n = 2*n;           //since NaCl is diatomic, atoms/cm^3
8 //length of edge of unit cube is n*a
9 //volume V = n^3*a^3 = 1 cm^3
10 V = 1;            //volume of unit cube, cm^3
11 a = (V/n)^(1/3);   //distance between two
  adjacent atoms, cm
12 a = a*10^8;       //distance between two adjacent
  atoms, A
13 printf("distance between two adjacent atoms is %5.2 f
  A", a);
```

---

**Scilab code Exa 2.7** To calculate the density of Copper crystal

```
1 //To calculate the density of Copper crystal
2 AW = 63.5;         //atomic weight of Cu, gm/mol
3 N = 6.023*10^23;   //avagadro's number, mol-1
4 r = 1.278;         //atomic radius, A
5 n = 4;             //number of atoms in unit cell
6 r = r*10^-8;       //atomic radius, cm
7 M = AW/N;          //mass of each Copper atom, gm
8 a = 4*r/sqrt(2);    //lattice constant, cm
9 m = n*M;           //mass of unit cell, gm
```

```

10 rho = m/a^3;      //density of copper crystal , gm/cm
    ^3
11 printf("density of Copper crystal is %5.3f gm/cm^3",
    rho);

```

---

**Scilab code Exa 2.8** To calculate the free volume per unit cell

```

1 //To calculate the free volume per unit cell
2 r = 0.1249;      //atomic radius , nm
3 a = 4*r/sqrt(3); //lattice constant , nm
4 a_m = a*10^-9;   //lattice constant , m
5 V = a_m^3;       //volume of unit cell , m^3
6 PF = 0.68;       //packing factor for BCC
7 FV = 1 - PF;     //free volume
8 FV1 = FV*V;      //free volume per unit cell , m^3
9 printf("free volume per unit cell in m^3 is");
10 disp(FV1);

```

---

**Scilab code Exa 2.9** To sketch the crystal plane structures

```

1 //To sketch the crystal plane structures
2 //sketching is not possible

```

---

**Scilab code Exa 2.10** To sketch the crystal plane structures

```

1 //To sketch the crystal plane structures
2 //sketching is not possible

```

---



**Scilab code Exa 2.11** To calculate the number of atoms per square millimetre

```
1 //To calculate the number of atoms per square
   millimetre
2 //in (100) plane the total number of atoms are n
3 n = (1/4)*4;
4 //A = a^2. number of atoms per mm^2 is n/a^2
5 printf("number of atoms in (100) plane are %d",n);
6 printf("number of atoms per mm^2 is 1/a^2");
7 //in (110) plane, area is sqrt(2)*a*a = sqrt(2)*a^2
8 printf("number of atoms in (100) plane is 1");
9 printf("unit area contains 1/(sqrt(2)*a^2) = 0.707/a
   ^2 atoms/mm^2");
10 //in (111) plane, area = (1/2)*base*height = (1/2)*a
   *sqrt(2)*a*sqrt(2)*cosd(30)
11 x = cosd(30);
12 //area = (1/2)*a*sqrt(2)*a*sqrt(2)*x = 0.866*a^2 =
   0.58/a^2
13 n1 = (1/360)*60*3;
14 //number of atoms per unit area is 0.5/(0.866*a^2) =
15 printf("total number of atoms in (111) plane is %5.1
   f",n1);
16 printf("number of atoms per unit area is 0.58/a^2
   atoms/mm^2");
```

---

**Scilab code Exa 2.12** To calculate the interplanar spacing for the planes

```
1 //To calculate the interplanar spacing for the
   planes
2 h1 = 1;
3 k1 = 1;
4 l1 = 0; //for (110) plane
5 h2 = 2;
6 k2 = 1;
```

```

7 l2 = 2;          //for (212) plane
8 r = 0.1278;     //atomic radius , nm
9 r = r*10^-9;   //atomic radius , m
10 x1 = sqrt(h1^2+k1^2+l1^2);
11 a = 4*r/x1;    //nearest neighbouring distance , m
12 a = a*10^9;   //nearest neighbouring distance , nm
13 d_110 = a/x1;  //interplanar spacing for (110), nm
14 x2 = sqrt(h2^2+k2^2+l2^2);
15 d_212 = a/x2;  //interplanar spacing for (212), nm
16 printf("interplanar spacing for (110) is %5.4f nm",
          d_110);
17 printf("interplanar spacing for (212) is %5.4f nm",
          d_212);

```

---

**Scilab code Exa 2.13** To calculate the ratio of the separation between successive lattice planes

```

1 //To calculate the ratio of the separation between
  successive lattice planes
2 h1 = 1;
3 k1 = 0;
4 l1 = 0;    //for (100) plane
5 x1 = sqrt(h1^2+k1^2+l1^2);
6 h2 = 1;
7 k2 = 1;
8 l2 = 0;    //for (110) plane
9 x2 = sqrt(h2^2+k2^2+l2^2);
10 h3 = 1;
11 k3 = 1;
12 l3 = 1;   //for (111) plane
13 x3 = sqrt(h3^2+k3^2+l3^2);
14 //d = a/sqrt(h^2+k^2+l^2)
15 //d100:d110:d111 = a:a/sqrt(2):a/sqrt(3)
16 //d100:d110:d111 = 1:1/sqrt(2):1/sqrt(3) =
    1:0.71:0.58

```

```
17 printf("ratio of the separation between successive  
    lattice planes is 1:0.71:0.58");
```

---

**Scilab code Exa 2.14** To calculate the miller indices of a plane

```
1 //To calculate the miller indices of a plane  
2 //plane intercepts at a,b/2,3*c  
3 //therefore intercepts are (1 1/2 3)  
4 //reciprocal of the intercepts is (1/2 2 1/3)  
5 //thus miller indices are (3 6 1)  
6 printf("miller indices of the plane are (3 6 1)");
```

---

**Scilab code Exa 2.15** To calculate the wavelength of X rays and maximum order of diffraction possible

```
1 //To calculate the wavelength of X-rays and maximum  
    order of diffraction possible  
2 d = 0.282; //lattice spacing, nm  
3 n = 1; //first order  
4 theta = 8+(35*0.016666667); //glancing angle  
    in degrees  
5 d = d*10^-9; //lattice spacing, m  
6 lamda = 2*d*sind(theta)/n; //wavelength of X-  
    rays, m  
7 lamda_nm = lamda*10^9; //wavelength of X-rays,  
    nm  
8 theeta = 90; //maximum value possible in degrees  
9 n1 = 2*d*sind(theeta)/lamda; //maximum order  
    of diffraction possible  
10 printf("wavelength of X-rays is %5.4f nm", lamda_nm);  
11 printf("maximum order of diffraction possible is %d"  
    ,n1);
```

---

**Scilab code Exa 2.16** To calculate the highest order for which Bragg's reflection can be seen

```
1 //To calculate the highest order for which Bragg's
  reflection can be seen
2 lamda = 1.5;      //wavelength, A.U
3 d = 1.6;         //interplanar spacing, A.U
4 theta = 90;     //maximum glancing angle possible,
  degrees
5 n = 2*d*sind(theta)/lamda;    //maximum possible
  diffraction order
6 printf("maximum possible diffraction order is %d",n)
  ;
```

---

**Scilab code Exa 2.17** To calculate the interatomic spacing

```
1 //To calculate the interatomic spacing
2 h = 1;
3 k = 1;
4 l = 1;          //for (111) plane
5 theta = 30;    //glancing angle, degrees
6 n = 1;        //first order
7 x = sqrt(h^2+k^2+l^2);
8 lamda = 1.5418; //wavelength of X-rays, A
9 lamda = lamda*10^-10; //wavelength of X-rays, m
10 d = lamda/(2*sind(theta));
11 a = d*x;      //interatomic spacing, m
12 a = a*10^10;  //interatomic spacing, A
13 printf("the interatomic spacing is %5.3f A",a);
```

---

**Scilab code Exa 2.18** To calculate the glancing angle

```
1 //To calculate the glancing angle
2 d100 = 0.28; //lattice constant, nm
3 h = 1;
4 k = 1;
5 l = 0; //for (110) plane
6 n = 2; //second order
7 lamda = 0.071; //wavelength of X-rays, nm
8 lamda_m = lamda*10^-9; //wavelength of X-rays, m
9 d110 = d100/sqrt(h^2+k^2+l^2); //interatomic
   spacing. nm
10 d110 = d110*10^-9; //interatomic spacing. m
11 theta = asind(n*lamda_m/(2*d110));
12 printf("the glancing angle is %d degrees",theta);
```

---

**Scilab code Exa 2.19** To calculate the distance between 110 planes

```
1 //To calculate the distance between (110) planes
2 a = 0.38; //lattice constant, nm
3 h = 1;
4 k = 1;
5 l = 0; //for (110) plane
6 d = a/sqrt(h^2+k^2+l^2);
7 printf("the distance between (110) planes is %5.2f
   nm",d);
```

---

**Scilab code Exa 2.20** To compare the density of lattice points

```
1 //To compare the density of lattice points
2 //area of (110) plane is a*sqrt(2)*a = sqrt(2)*a^2
3 n = (1/4)*4; //number of atoms
4 theta = 30; //glancing angle
```

```

5 x = cosd(theta);
6 //area of (111) plane is (a/sqrt(2))*x*a*sqrt(2)
7 //hence area is (sqrt(3)/2)*a^2
8 n1 = 3*(1/6); //number of atoms
9 printf("area of (110) plane contains %d atom",n);
10 printf("density of lattice points is 1/(sqrt(2)*a^2)
    ");
11 printf("area of (111) plane contains %5.1f atom",n1)
    ;
12 //density of lattice points is (1/2)/(sqrt(3)*a^2/2)
13 printf("density of lattice points is 1/(sqrt(3)*a^2)
    ");
14 //density of lattice points (111) plane : (110)
    plane is 1/(sqrt(3)*a^2) : 1/(sqrt(2)*a^2) = sqrt
    (2):sqrt(3)
15 printf("density of lattice points (111) plane :
    (110) plane is sqrt(2):sqrt(3)");

```

---

**Scilab code Exa 2.21** To calculate the glancing angle

```

1
2 //To calculate the glancing angle
3 n = 2; //second order
4 h = 1;
5 k = 1;
6 l = 0; //plane (110)
7 lamda = 0.065; //wavelength of X-rays, nm
8 lamda_m = lamda*10^-9; //wavelength of X-rays, m
9 a = 0.26; //axial length, nm
10 a_m = a*10^-9; //axial length, m
11 x = sqrt(h^2+k^2+l^2);
12 theta = asind(n*lamda_m*x/(2*a_m)); //glancing
    angle, degrees
13 deg = int(theta); //glancing angle, degrees
14 t = 60*(theta-deg);

```

```

15 mint = int(t);           //glancing angle , minutes
16 printf("the glancing angle is %d degrees %d minutes"
    ,deg,mint);

```

---

**Scilab code Exa 2.22** To calculate the cube edge

```

1 //To calculate the cube edge
2 n = 1;    //first order
3 h = 1;
4 k = 1;
5 l = 1;    //for (111) plane
6 lamda = 1.54;    //wavelength , A.U
7 lamda_m = lamda*10^-10;    //wavelength , m
8 theta = 19.2;    //Bragg's angle , degrees
9 d = n*lamda_m/(2*sind(theta));    //interplanar
    spacing , m
10 a = d*sqrt(h^2+k^2+l^2);    //cube edge , m
11 a_AU = a*10^10;    //cube edge , A.U
12 printf("the cube edge is %5.3 f A.U" ,a_AU);

```

---

**Scilab code Exa 2.23** To calculate the cube edge

```

1 //To calculate the cube edge
2 n = 1;    //first order
3 h = 1;
4 k = 1;
5 l = 1;    //for (111) plane
6 lamda = 1.54;    //wavelength , A.U
7 lamda_m = lamda*10^-10;    //wavelength , m
8 theta = 19.2;    //Bragg's angle , degrees
9 d = n*lamda_m/(2*sind(theta));    //interplanar
    spacing , m
10 a = d*sqrt(h^2+k^2+l^2);    //cube edge , m

```

```
11 printf("the cube edge in m is");  
12 disp(a);
```

---



## Chapter 3

# Principles of Quantum Mechanics

Scilab code Exa 3.1 To calculate the de Broglie wavelength

```
1 //To calculate the de Broglie wavelength
2 c = 3*10^8; //velocity of light, m/sec
3 v = (1/10)*c; //velocity of proton, m/sec
4 m = 1.67*10^-27; //mass of proton, kg
5 h = 6.626*10^-34; //planck's constant
6 lamda = h/(m*v); //de Broglie wavelength, m
7 printf("de Broglie wavelength of proton in m is");
8 disp(lamda);
```

---

Scilab code Exa 3.2 To calculate the de Broglie wavelength

```
1 //To calculate the de Broglie wavelength
2 V = 400; //potential, V
3 lamda = 12.26/sqrt(V); //de Broglie wavelength,
    armstrong
4 printf("de Broglie wavelength of electron is %5.3f
    armstrong", lamda);
```

---

**Scilab code Exa 3.3** To calculate the de Broglie wavelength

```
1 //To calculate the de Broglie wavelength
2 m = 1.674*10^-27; //mass of neutron, kg
3 h = 6.626*10^-34; //planck's constant
4 e = 1.6*10^-19;
5 KE = 0.025; //kinetic energy, eV
6 E = KE*e; //kinetic energy, J
7 lamda = h/sqrt(2*m*E); //de Broglie wavelength,
  m
8 lamda_nm = lamda*10^9; //de Broglie wavelength,
  nm
9 printf("de Broglie wavelength is %5.3f nm",lamda_nm)
  ;
```

---

**Scilab code Exa 3.4** To calculate the wavelength

```
1 //To calculate the wavelength
2 V = 1600; //potential, V
3 lamda = 12.26/sqrt(V); //de Broglie wavelength,
  armstrong
4 printf("de Broglie wavelength of electron is %5.4f
  armstrong",lamda);
```

---

**Scilab code Exa 3.5** To calculate the uncertainty in momentum

```
1 //To calculate the uncertainty in momentum
2 delta_x = 0.2; //electron distance, armstrong
```

```

3 delta_x = delta_x*10^-10;      //electron distance ,
  m
4 h = 6.626*10^-34;      //planck's constant
5 delta_p = h/(2*pi*delta_x);    //uncertainty in
  momentum, kg.m/s
6 printf("uncertainty in momentum in kg m/s is");
7 disp(delta_p);

```

---

**Scilab code Exa 3.6** To calculate the lowest energy of an electron

```

1 //To calculate the lowest energy of an electron
2 n1 = 1;
3 n2 = 1;
4 n3 = 1;      //for lowest energy
5 e = 1.6*10^-19;
6 h = 6.62*10^-34;      //planck's constant
7 m = 9.1*10^-31;      //mass of electron , kg
8 L = 0.1;      //side of box, nm
9 L = L*10^-9;      //side of box, m
10 E1 = h^2*(n1^2+n2^2+n3^2)/(8*m*L^2);      //lowest
  energy , J
11 E1 = E1/e;      //lowest energy , eV
12 printf("lowest energy of electron is %5.1f eV",E1);

```

---

**Scilab code Exa 3.7** To calculate the energy of electron

```

1 //To calculate the energy of electron
2 n1 = 1;
3 n2 = 1;
4 n3 = 2;      //for level next to the lowest
5 e = 1.6*10^-19;
6 h = 6.62*10^-34;      //planck's constant
7 m = 9.1*10^-31;      //mass of electron , kg

```

```

8 L = 0.1;           //side of box, nm
9 L = L*10^-9;      //side of box, m
10 E1 = h^2*(n1^2+n2^2+n3^2)/(8*m*L^2);           //lowest
    energy, J
11 E1 = E1/e;        //lowest energy, eV
12 printf("energy of electron is %5.2f eV",E1);
13
14 //answer given in the book is wrong in the 2nd
    decimal

```

---

**Scilab code Exa 3.8** To calculate the wavelength

```

1 //To calculate the wavelength
2 m = 9.1*10^-31;      //mass of electron, kg
3 h = 6.626*10^-34;   //planck's constant
4 e = 1.6*10^-19;
5 E = 2000;           //energy, eV
6 E = E*e;           //energy, J
7 lamda = h/sqrt(2*m*E); //wavelength, m
8 lamda_nm = lamda*10^9; //wavelength, nm
9 printf("wavelength is %5.4f nm",lamda_nm);

```

---

**Scilab code Exa 3.9** To calculate the minimum energy

```

1 //To calculate the minimum energy
2 n = 1;           //for minimum energy
3 h = 6.626*10^-34; //planck's constant, J sec
4 m = 9.1*10^-31;  //mass of electron, kg
5 L = 4*10^-10;    //side of box, m
6 E1 = h^2*n^2/(8*m*L^2); //lowest energy, J
7 printf("energy of electron in J is");
8 disp(E1);
9

```

10 //answer given in the book is wrong

---

**Scilab code Exa 3.10** To calculate the energy values

```
1 //To calculate the energy values
2 n1 = 1;      //for ground state
3 n2 = 2;      //for 1st excited state
4 n3 = 3;      //for 2nd excited state
5 h = 6.626*10^-34; //planck's constant, J sec
6 m = 9.1*10^-31; //mass of electron, kg
7 L = 1*10^-10; //width, m
8 E1 = h^2*n1^2/(8*m*L^2); //energy in ground
   state, J
9 E2 = n2^2*E1; //energy in 1st excited state, J
10 E3 = n3^2*E1; //energy in 2nd excited state, J
11 printf("energy in ground state in J is");
12 disp(E1);
13 printf("energy in 1st excited state in J is");
14 disp(E2);
15 printf("energy in 2nd excited state in J is");
16 disp(E3);
```

---

**Scilab code Exa 3.11** To calculate the velocity and kinetic energy of electron

```
1 //To calculate the velocity and kinetic energy of
   electron
2 h = 6.626*10^-34; //planck's constant, J sec
3 m = 9.1*10^-31; //mass of electron, kg
4 e = 1.6*10^-19;
5 lamda = 1.66*10^-10; //wavelength, m
6 v = h/(m*lamda); //velocity of electron, m/sec
7 v_km = v*10^-3; //velocity of electron, km/sec
```

```

8 printf("velocity of electron is %d km/sec",v_km);
9 KE = (1/2)*m*v^2;      //kinetic energy , J
10 KE_eV = KE/e;        //kinetic energy , eV
11 printf("kinetic energy of electron is %5.2f eV",
    KE_eV);

```

---

**Scilab code Exa 3.12** To calculate the wavelength

```

1 //To calculate the wavelength
2 V = 15;      //potential , kV
3 V = V*10^3;  //potential , V
4 lamda = 12.26/sqrt(V);      //de Broglie wavelength ,
    armstrong
5 printf("wavelength of electron waves is %5.1f
    armstrong",lamda);

```

---

**Scilab code Exa 3.13** To calculate the spacing of crystal

```

1 //To calculate the spacing of crystal
2 V = 344;      //potential , V
3 lamda = 12.26/sqrt(V);      //de Broglie wavelength ,
    armstrong
4 lamda_m = lamda*10^-10;      //de Broglie wavelength ,
    m
5 n = 1;      //for 1st reflection maximum
6 theta = 60;      //glancing angle , degrees
7 d = n*lamda_m/(2*sind(theta));      //interatomic
    spacing , m
8 d = d*10^10;      //interatomic spacing , armstrong
9 printf("interatomic spacing of crystal is %5.4f
    armstrong",d);

```

---

## Chapter 4

# Electron Theory of Metals

**Scilab code Exa 4.1** To calculate the density and mobility of electrons

```
1 //To calculate the density and mobility of electrons
2 rho_s = 10.5*10^3; //density of silver , kg/m^3
3 NA = 6.02*10^26; //avagadro number, /k-mol
4 MA = 107.9; //atomic weight of silver
5 n = rho_s*NA/MA; //molar volume of silver
6 printf("density of electrons in silver is");
7 disp(n);
8 sigma = 6.8*10^7; //conductivity of silver , ohm
   -1 m-1
9 e = 1.6*10^-19;
10 mew = sigma/(n*e); //mobility of electrons , m^2/
   Vs
11 printf("mobility of electrons is %5.5f m^2/Vs",mew);
```

---

**Scilab code Exa 4.2** To calculate the mobility of electrons

```
1 //To calculate the mobility of electrons
2 d = 8.92*10^3; //density , kg/m^3
```

```

3 e = 1.6*10^-19;
4 m = 9.1*10^-31;      //mass of electron , kg
5 N = 6.02*10^26;      //avagadro's number per k-mol
6 AW = 63.5;          //atomic weight
7 rho = 1.73*10^-8;    //resistivity of copper , ohm-m
8 n = d*N/AW;         //number of cu atoms/m^3
9 mew = 1/(rho*n*e);   //mobility of electrons , m/
    Vs
10 printf("mobility of electrons is %f m/Vs",mew);
11 tow = m/(n*e^2*rho); //relaxation time , s
12 printf("relaxation time in sec is");
13 disp(tow);

```

---

**Scilab code Exa 4.3** To calculate the relaxation time

```

1 //To calculate the relaxation time
2 rho = 1.54*10^-8;    //resistivity , ohm-m
3 n = 5.8*10^28;      //conduction electrons per m^3
4 m = 9.108*10^-31;   //mass of electron , kg
5 e = 1.602*10^-19;
6 tow = m/(n*e^2*rho); //relaxation time , sec
7 printf("relaxation time of conduction electrons in
    sec is");
8 disp(tow);

```

---

**Scilab code Exa 4.4** To calculate the free electron concentration mobility and drift velocity

```

1 //To calculate the free electron concentration ,
    mobility and drift velocity
2 R = 0.06;           //resistance , ohm
3 D = 5;              //length of Al wire , m
4 e = 1.602*10^-19;

```



```

5 rho = 2.7*10^-8; //resistivity of Al, ohm-m
6 MA = 26.98; //atomic weight
7 NA = 6.025*10^26; //avagadro number, k/mol
8 rho_s = 2.7*10^3; //density, kg/m^3
9 n = 3*rho_s*NA/MA; //free electron concentration
    , electrons/m^3
10 printf("free electron concentration in electrons/m^3
    is");
11 disp(n);
12 mew = 1/(n*e*rho); //mobility, m/Vs
13 printf("mobility is %f m/Vs",mew);
14 I = 15; //current, A
15 E = I*R/D; //electric field, V/m
16 vd = mew*E; //drift velocity, m/s
17 printf("drift velocity is %f m/s",vd);

```

---

**Scilab code Exa 4.5** To calculate the lowest energy of an electron

```

1 //To calculate the lowest energy of an electron
2 n1 = 1;
3 n2 = 1;
4 n3 = 1; //for lowest energy
5 h = 6.62*10^-34; //planck's constant, Js
6 e = 1.6*10^-19;
7 m = 9.1*10^-31; //mass of electron, kg
8 L = 0.1; //side of box, mm
9 L = L*10^-9; //side of box, m
10 E1 = h^2*(n1^2+n2^2+n3^2)/(8*m*L^2); //lowest
    energy, J
11 E1 = E1/e; //lowest energy, eV
12 printf("lowest energy of electron is %5.1f eV",E1);

```

---

**Scilab code Exa 4.6** To evaluate the fermi function

```

1 //To evaluate the fermi function
2 //Fermi function  $F(E) = 1/(1+\exp((E-E_f)/(kT)))$ 
3 //given  $E-E_f = kT$ . therefore  $F(E) = 1/(1+\exp(1))$ 
4 F_E = 1/(1+exp(1));
5 printf("fermi function is %5.3 f",F_E);

```

---

**Scilab code Exa 4.7** To calculate the temperature

```

1
2 //To calculate the temperature
3 F_E = 10; //probability in percent
4 k = 1.38*10^-23;
5 e = 1.6*10^-19; //conversion factor
6 EF = 5.5; //fermi energy, eV
7 E = EF+(EF/100); //energy, eV
8 X = E-EF; //E-EF, eV
9 X = X*e; //E-EF, J
10 T = X/(k*log(F_E-1)); //temperature, K
11 printf("temperature is %5.1 f K",T);

```

---

**Scilab code Exa 4.8** To calculate the temperature

```

1 //To calculate the temperature
2 F_E = 0.01; //probability in percent
3 k = 1.38*10^-23;
4 e = 1.6*10^-19;
5 //let E-EF be X
6 X = 0.5; //E-EF, eV
7 kT = X/(2.303*log10((1-F_E)*100)); //value of kT,
    eV
8 T = kT*e/k; //temperature, K
9 printf("temperature is %d K",T);
10

```

11 //answer given in the book is wrong

---

## Chapter 6

# Dielectric Properties

**Scilab code Exa 6.1** To calculate the energy stored in the condensor and polarizing the dielectric

```
1 //To calculate the energy stored in the condensor
  and polarizing the dielectric
2 C = 2; //capacitance , micro-farad
3 C = C*10^-6; //capacitance , farad
4 V = 1000; //voltage applied , V
5 epsilon_r = 100; //permittivity
6 W = (C*V^2)/2; //energy stored in capacitor , J
7 printf("energy stored in capacitor is %d J",W);
8 C0 = C/epsilon_r; //capacitance removing the
  dielectric
9 W0 = C0*V^2/2; //energy stored without
  dielectric , J
10 E = 1-W0; //energy stored in dielectric , J
11 printf("energy stored in the dielectric is %5.2f J",
  E);
```

---

**Scilab code Exa 6.2** To calculate the ratio between electronic and ionic polarizability

```

1 //To calculate the ratio between electronic and
   ionic polarizability
2 epsilon_r = 4.94;
3 n_2 = 2.69;      //square of index of refraction
4 alpha_i = 0;    //at optical frequencies
5 //(epsilon_r-1)/(epsilon_r+2) = N*(alpha_e+alpha_i)
   /(3*epsilon0)
6 X = (epsilon_r-1)/(epsilon_r+2);
7 //epsilon_r = n^2. therefore (n^2-1)/(n^2+2) = N*
   alpha_e/(3*epsilon0)
8 Y = (n_2-1)/(n_2+2);
9 //N*(alpha_e+alpha_i)/N*alpha_e = X/Y
10 //let alpha = alpha_i/alpha_e
11 alhai_e = (X/Y)-1;    //ratio between electronic
   ionic and electronic polarizability
12 printf("ratio between electronic ionic and
   electronic polarizability is %5.4f",alhai_e);
13 alhae_i = 1/alhai_e;    //ratio between
   electronic and ionic polarizability
14 printf("ratio between electronic and ionic
   polarizability is %5.3f",alhae_i);

```

---

**Scilab code Exa 6.3** To calculate the difference in magnetic potential energy

```

1 //To calculate the difference in magnetic potential
   energy
2 epsilon_r = 2.56;
3 tan_delta = 0.7*10^-4;
4 f = 1;        //frequency , MHz
5 A = 8;        //area , cm^2
6 A = A*10^-4;    //area , m^2
7 d = 0.08;     //diameter , mm
8 d = d*10^-3;   //diameter , m
9 epsilon_rdash = epsilon_r*tan_delta;

```

```

10 epsilon0 = 8.85*10^-12;
11 omega = 2*%pi*10^6;
12 Rp = d/(omega*epsilon0*epsilon_rdash*A);          //
    parallel loss resistance , ohm
13 Rp = Rp*10^-6;          //parallel loss resistance , Mega
    ohm
14 Cp = A*epsilon0*epsilon_r/d;          //capacitance ,
    farad
15 printf("parallel loss resistance is %5.3f ohm",Rp);
16 printf("capacitance in Farad is");
17 disp(Cp);

```

---

**Scilab code Exa 6.4** To calculate the dielectric constant of material

```

1 //To calculate the dielectric constant of material
2 N = 3*10^28;          //density , atoms/m^3
3 alpha_e = 10^-40;    //electronic polarizability ,
    Farad-m^2
4 epsilon0 = 8.854*10^-12;
5 epsilon_r = 1+(N*alpha_e/epsilon0);          //
    dielectric constant of material
6 printf("dielectric constant of material is %5.3f",
    epsilon_r);

```

---

**Scilab code Exa 6.5** To calculate the electronic polarizability

```

1 //To calculate the electronic polarizability
2 epsilon0 = 8.854*10^-12;
3 epsilon_r = 1.0000684;          //dielectric constant
4 N = 2.7*10^25;          //density , atoms/m^3
5 alpha_e = epsilon0*(epsilon_r-1)/N;          //
    electronic polarizability , Fm^2
6 printf("electronic polarizability in Fm^2 is");

```

```
7 disp(alpha_e);
```

---

**Scilab code Exa 6.6** To calculate the capacitance and charge on plates

```
1 //To calculate the capacitance and charge on plates
2 epsilon0 = 8.85*10^-12;
3 V = 100; //potential, V
4 A = 100; //area, cm^2
5 A = A*10^-4; //area, m^2
6 d = 1; //plate separation, cm
7 d = d*10^-2; //plate separation, m
8 C = epsilon0*A/d; //capacitance, farad
9 printf("capacitance of capacitor in F is");
10 disp(C);
11 Q = C*V; //charge on plates
12 printf("charge on plates in coulomb is");
13 disp(Q);
```

---

**Scilab code Exa 6.7** To calculate the electronic polarizability

```
1 //To calculate the electronic polarizability
2 N = 6.02*10^26; //avagadro number
3 d = 2050; //density, kg/m^3
4 AW = 32; //atomic weight of sulphur
5 n = N*d/AW; //number of atoms/m^3
6 epsilon_r = 3.75; //relative dielectric constant
7 epsilon0 = 8.55*10^-12;
8 alpha_e = ((epsilon_r-1)/(epsilon_r+2))*3*epsilon0/n
; //electronic polarizability, Fm^2
9 printf("electronic polarizability in Fm^2 is");
10 disp(alpha_e);
```

---

**Scilab code Exa 6.8** To calculate the resultant voltage across capacitors

```
1 //To calculate the resultant voltage across
  capacitors
2 Q = 2*10^-10;      //charge, coulomb
3 d = 4;            //plate seperation, mm
4 d = d*10^-3;      //plate seperation, m
5 epsilon_r = 3.5;   //dielectric constant
6 A = 650;          //area, mm^2
7 A = A*10^-6;      //area, m^2
8 epsilon0 = 8.85*10^-12;
9 V = Q*d/(epsilon0*epsilon_r*A);      //voltage
  across capacitors, V
10 printf("resultant voltage across capacitors is %5.2f
  V",V);
```

---

**Scilab code Exa 6.9** To calculate the dielectric displacement

```
1 //To calculate the dielectric displacement
2 V = 10;           //potential, V
3 d = 2*10^-3;      //plate seperation, m
4 E = V/d;          //electric field, V/m
5 epsilon_r = 6;
6 epsilon0 = 8.85*10^-12;
7 D = epsilon0*epsilon_r*E;      //dielectric
  displacement, C/m^2
8 printf("dielectric displacement in C/m^2 is");
9 disp(D);
```

---



# Chapter 7

## Magnetic Properties

**Scilab code Exa 7.1** To calculate the relative permeability

```
1 //To calculate the relative permeability
2 M = 1.4;           //field , T
3 H = 6.5*10^-4;    //magnetic field , T
4 chi = M/H;        //susceptibility
5 mew_r = 1+chi;    //relative permeability
6 printf("relative permeability of iron is %d",mew_r);
7
8 //answer given in the book is wrong
```

---

**Scilab code Exa 7.2** To calculate the relative permeability of material

```
1 //To calculate the relative permeability of material
2 M = 3300;         //magnetisation , amp/m
3 H = 220;          //field strength , amp/m
4 mew_r = (M/H)+1; //relative permeability
5 printf("relative permeability of material is %d",
        mew_r);
```

---

**Scilab code Exa 7.3** To calculate the magnetisation and flux density

```
1 //To calculate the magnetisation and flux density
2 H = 10^6; //magnetic field intensity , amp/m
3 chi = 1.5*10^-3; //susceptibility
4 mew0 = 4*pi*10^-7;
5 M = chi*H; //magnetisation , A/m
6 printf("magnetisation of material is %d A/m",M);
7 B = mew0*(M+H); //flux density , T
8 printf("flux density is %5.4f T",B);
9
10 //answer given in the book is wrong
```

---

**Scilab code Exa 7.4** To calculate the magnetisation and flux density

```
1 //To calculate the magnetisation and flux density
2 H = 10^4; //magnetic field intensity , amp/m
3 chi = 3.7*10^-3; //susceptibility
4 mew0 = 4*pi*10^-7;
5 M = chi*H; //magnetisation , A/m
6 printf("magnetisation of material is %d A/m",M);
7 B = mew0*(M+H); //flux density , Wb/m^2
8 printf("flux density is %5.4f Wb/m^2",B);
```

---

**Scilab code Exa 7.5** To calculate the magnetic moment

```
1 //To calculate the magnetic moment
2 I = 500; //current , mA
3 I = I*10^-3; //current , A
```

```

4 d = 10;          //diameter , cm
5 r = d/2;        //radius , cm
6 r = r*10^-2;    //radius , m
7 A = 2*%pi*r^2;  //area , m^2
8 mew_m = I*A;    //magnetic moment, Am^2
9 printf("magnetic moment associated with the loop is
    %f Am^2",mew_m);

```

---

**Scilab code Exa 7.6** To calculate the change in magnetic moment

```

1 //To calculate the change in magnetic moment
2 r = 5.29*10^-11; //radius of orbit , m
3 B = 2;           //field applied , T
4 e = 1.602*10^-19; //charge of electron , coulomb
5 m = 9.108*10^-31; //mass of electron , kg
6 mew_ind = e^2*r^2*B/(4*m); //change in magnetic
    moment
7 printf("change in magnetic moment in Am^2 is");
8 disp(mew_ind);

```

---

**Scilab code Exa 7.7** To calculate the susceptibility

```

1 //To calculate the susceptibility
2 chi_1 = 2.8*10^-4; //susceptibility
3 T1 = 350;         //temperature , K
4 T2 = 300;         //temperature , K
5 //chi = C/T where C is curie constant
6 chi_2 = chi_1*T1/T2; //susceptibility at 300 K
7 printf("susceptibility at 300 K is %f",chi_2);

```

---

**Scilab code Exa 7.8** To calculate the magnetic moment

```
1 //To calculate the magnetic moment
2 d = 8906; //density , kg/m^3
3 n = 6.025*10^26; //avagadro number
4 AW = 58.7; //atomic weight
5 N = d*n/AW; //number of atoms/m^3
6 Bs = 0.65; //magnetic induction , Wb/m^2
7 mew0 = 4*pi*10^-7;
8 mew_m = Bs/(N*mew0); //magnetic moment, Am^2
9 mewB = 9.27*10^-24;
10 mew_m = mew_m/mewB; //magnetic moment, mewB
11 printf("the magnetic moment of Ni is %5.2f mewB",
    mew_m);
```

---

**Scilab code Exa 7.9** To calculate the temperature

```
1 //To calculate the temperature
2 H = 2; //magnetic field , Wb/m^2
3 mew = 9.4*10^-24;
4 k = 1.38*10^-23;
5 //np = C*n0*exp(mew*H/(k*T))
6 //na = C*n0*exp(-mew*H/(k*T))
7 // np/na = exp(mew*H/(k*T))/exp(-mew*H/(k*T)) = exp
    (2*mew*H/(k*T))
8 //given np/na = 2. therefore exp(2*mew*H/(k*T)) = 2
9 T = 2*mew*H/(k*log(2)); //temperature , K
10 printf("temperature is %5.1f K",T);
```

---

**Scilab code Exa 7.10** To calculate the magnetic moment and saturation magnetisation

```

1 //To calculate the magnetic moment and saturation
  magnetisation
2 AW = 157.26;      //atomic weight
3 d = 7.8*10^3;    //density , kg/m^3
4 A = 6.025*10^26; //avagadro number
5 mew0 = 4*pi*10^-7;
6 N = d*A/AW;     //number of atoms 1 kg contains
7 g = N/10^3;     //number of atoms 1 g contains
8 mew_B = 7.1;    //bohr magneton
9 mew_m = 9.27*10^-24;
10 mew_mg = g*mew_B*mew_m; //magnetic moment per
   gram, Am^2
11 printf("magnetic moment per gram is %5.2f Am^2",
   mew_mg);
12 Bs = N*mew0*mew_m; //saturation magnetisation ,
   Wb/m^2
13 printf("saturation magnetisation is %5.4f Wb/m^2",Bs
   );
14
15 //answer for saturation magnetisation given in the
   book is wrong

```

---

# Chapter 8

## Semiconductors

Scilab code Exa 8.1 To calculate the resistivity

```
1 //To calculate the resistivity
2 ni = 2.37*10^19; //intrinsic carrier density ,
   per m^3
3 mew_e = 0.38; //electron mobility , m^2/Vs
4 mew_h = 0.18; //hole mobility , m^2/Vs
5 e = 1.6*10^-19;
6 sigma_i = ni*e*(mew_e+mew_h); //conductivity ,
   ohm-1 m-1
7 rho = 1/sigma_i; //resistivity , ohm m
8 printf("resistivity is %5.3f ohm m",rho);
```

---

Scilab code Exa 8.2 To determine the position of Fermi level

```
1 //To determine the position of Fermi level
2 Eg = 1.12; //band gap, eV
3 k = 1.38*10^-23;
4 T = 300; //temperature , K
5 e = 1.6*10^-19;
```

```

6 m0 = 1;          //for simplicity assume value of m0 to
  be unity
7 mh = 0.28*m0;
8 me = 0.12*m0;
9 EF = (Eg/2)+(3*k*T*log(mh/me)/(4*e));          //
  position of Fermi level , eV
10 printf("position of Fermi level is %5.3f eV from the
  top of valence band",EF);

```

---

**Scilab code Exa 8.3** To calculate the concentration of intrinsic charge carriers

```

1 //To calculate the concentration of intrinsic charge
  carriers
2 T = 300;          //temperature , K
3 pi = 22/7;       //value of pi
4 e = 1.6*10^-19;
5 m = 9.109*10^-31;      //mass of electron , kg
6 k = 1.38*10^-23;      //boltzmann's constant
7 h = 6.626*10^-34;     //planck's constant
8 Eg = 0.7;          //band gap, eV
9 Eg = Eg*e;        //band gap, J
10 A = (2*pi*m*k*T/h^2)^(3/2);
11 B = exp(-Eg/(2*k*T));
12 ni = 2*A*B;       //concentration of intrinsic charge
  carriers per m^3
13 printf("concentration of intrinsic charge carriers
  per m^3 is");
14 disp(ni);

```

---

**Scilab code Exa 8.4** To calculate the resistivity

```

1 //To calculate the resistivity

```

```

2 ni = 2.4*10^19;      //intrinsic carrier density ,
    per m^3
3 mew_e = 0.39;      //electron mobility , m^2/Vs
4 mew_h = 0.19;      //hole mobility , m^2/Vs
5 e = 1.6*10^-19;
6 sigma_i = ni*e*(mew_e+mew_h);      //conductivity ,
    ohm-1 m-1
7 rho = 1/sigma_i;      //resistivity , ohm m
8 printf("resistivity is %5.3f ohm m",rho);

```

---

**Scilab code Exa 8.5** To calculate the resistance

```

1 //To calculate the resistance
2 ni = 2.5*10^19;      //intrinsic carrier density ,
    per m^3
3 mew_e = 0.39;      //electron mobility , m^2/Vs
4 mew_h = 0.19;      //hole mobility , m^2/Vs
5 e = 1.6*10^-19;
6 sigma_i = ni*e*(mew_e+mew_h);      //conductivity ,
    ohm-1 m-1
7 w = 1;      //width , mm
8 w = w*10^-3;      //width , m
9 t = 1;      //thickness , mm
10 t = t*10^-3;      //thickness , m
11 A = w*t;      //area , m^2
12 l = 1;      //length , cm
13 l = l*10^-2;      //length , m
14 R = 1/(sigma_i*A);      //resistivity , ohm m
15 printf("resistance of intrinsic Ge rod is %5.1f ohm"
    ,R);

```

---

**Scilab code Exa 8.6** To calculate the conductivity



```

1 //To calculate the conductivity
2 m = 9.109*10^-31; //mass of electron , kg
3 k = 1.38*10^-23; //boltzmann constant
4 pi = 22/7; //value of pi
5 h = 6.626*10^-34; //planck's constant
6 C = 2*(2*pi*m*k/h^2)^(3/2);
7 T = 300; //temperature , K
8 e = 1.6*10^-19;
9 Eg = 1.1; //energy gap, eV
10 mew_e = 0.48; //electron mobility, m^2/Vs
11 mew_h = 0.013; //hole mobility, m^2/Vs
12 ni = C*T^(3/2)*exp(-Eg*e/(2*k*T)); //intrinsic
    carrier density per m^3
13 sigma_i = ni*e*(mew_e+mew_h); //conductivity,
    ohm-1 m-1
14 printf("conductivity is %f ohm-1 m-1",sigma_i);
15
16 //answer given in the book is wrong

```

---

**Scilab code Exa 8.7** To calculate the intrinsic carrier density and conductivity

```

1 //To calculate the intrinsic carrier density and
    conductivity
2 m = 9.109*10^-31; //mass of electron , kg
3 k = 1.38*10^-23; //boltzmann constant
4 pi = 22/7; //value of pi
5 h = 6.626*10^-34; //planck's constant
6 T = 300; //temperature , K
7 e = 1.6*10^-19;
8 Eg = 0.7; //energy gap, eV
9 ni = 2*(2*pi*m*k*T/h^2)^(3/2)*exp(-Eg*e/(2*k*T));
    //intrinsic carrier density per m^3
10 printf("intrinsic carrier density per m^3 is");
11 disp(ni);

```

```

12 mew_e = 0.4;           //electron mobility , m^2/Vs
13 mew_h = 0.2;           //hole mobility , m^2/Vs
14 sigma = ni*e*(mew_e+mew_h);           //conductivity , ohm
    -1 m-1
15 printf("conductivity is %5.2f ohm-1 m-1",sigma);
16
17 //answer given in the book is wrong

```

---

**Scilab code Exa 8.8** To calculate the forbidden energy gap

```

1 //To calculate the forbidden energy gap
2 rho = 2.12;           //resistivity , ohm m
3 sigma = 1/rho;       //conductivity , ohm-1 m-1
4 e = 1.6*10^-19;
5 pi = 22/7;           //value of pi
6 m = 9.109*10^-31;    //mass of electron , kg
7 k = 1.38*10^-23;     //boltzmann constant
8 h = 6.626*10^-34;    //planck 's constant
9 mew_e = 0.36;        //electron mobility , m^2/Vs
10 mew_h = 0.17;       //hole mobility , m^2/Vs
11 ni = sigma/(e*(mew_e+mew_h));           //intrinsic
    carrier density per m^3
12 C = 2*(2*pi*m*k/h^2)^(3/2);
13 T = 300;           //temperature , K
14 //let exp(Eg/(2*k*T)) be a
15 a = (C*T^(3/2))/ni;
16 //Eg/(2*k*T) = log(a) and Eg = 2*k*T*log(a)
17 Eg = 2*k*T*log(a)/e;           //forbidden energy gap , eV
18 printf("forbidden energy gap is %5.3f eV",Eg);
19
20 //answer given in the book is wrong

```

---

**Scilab code Exa 8.9** To calculate the energy band gap

```

1 //To calculate the energy band gap
2 rho_2 = 4.5;          //resistivity at 20C
3 rho_1 = 2;           //resistivity at 32C
4 T1 = 20;            //temperature, C
5 T1 = T1+273;        //temp, K
6 T2 = 32;            //temp, C
7 T2 = T2+273;        //temp, K
8 k = 8.616*10^-5;
9 dy = log10(rho_2)-log10(rho_1);
10 dx = (1/T1)-(1/T2);
11 Eg = 2*k*dy/dx;     //energy band gap, eV
12 printf("energy band gap is %5.3f eV",Eg);
13
14 //answer given in the book is wrong

```

---

**Scilab code Exa 8.10** To calculate the temperature

```

1 //To calculate the temperature
2 Eg = 1;             //band gap, eV
3 e = 1.602*10^-19;
4 Eg = Eg*e;         //band gap, J
5 me = 1;            //for simplicity assume me to be unity
6 mh = 4*me;         //effective mass of holes is 4 times
   of electrons
7 E_Ef = 10/100;     //fermi level shift, eV
8 E_Ef = E_Ef*e;     //fermi level shift, J
9 k = 1.38*10^-23;   //boltzmann constant
10 //E_Ef = 3*k*T*log(mh/me)/4
11 T = 4*E_Ef/(3*k*log(mh/me)); //temperature, K
12 printf("temperature is %d K",T);

```

---

**Scilab code Exa 8.11** To calculate the electron concentration

```

1 //To calculate the electron concentration
2 Na = 5*10^23; //atoms of boron
3 Nd = 3*10^23; //arsenic atoms
4 p = 2*(Na-Nd)/2; //hole concentration per m^3
5 ni = 2*10^16; //intrinsic charge carriers per m
    ^3
6 n = ni^2/p; //electron concentration per m^3
7 printf("electron concentration per m^3 is");
8 disp(n);

```

---

**Scilab code Exa 8.12** To calculate the conductivity of intrinsic Silicon

```

1 //To calculate the conductivity of intrinsic Silicon
2 ni = 1.5*10^16; //intrinsic charge carriers per
    m^3
3 e = 1.6*10^-19;
4 mew_e = 0.13; //electron mobility, m^2/Vs
5 mew_h = 0.05; //hole mobility, m^2/Vs
6 sigma = ni*e*(mew_e+mew_h); //conductivity, ohm
    -1 m-1
7 printf("conductivity is %f ohm-1 m-1",sigma);
8 AW = 28.1; //atomic weight of Si, kg
9 d = 2.33*10^3; //density of Si, kg/m^3
10 N = 6.02*10^26; //avagadro number
11 Nd = d*N/AW; //impurity atoms per m^3
12 Nd = Nd/10^8; //extent of 10^8 Si atoms
13 p = ni^2/Nd; //hole concentration per m^3
14 sigma_ex = Nd*e*mew_e; //conductivity, ohm-1 m
    -1
15 printf("conductivity if donor type impurity is added
    is %5.2f ohm-1 m-1",sigma_ex);
16 Na = Nd;
17 n = ni^2/Na; //electron concentration per m^3
18 sigma_EX = Na*e*mew_h; //conductivity, ohm-1 m
    -1

```

```
19 printf("conductivity if acceptor type impurity is
    added is %5.2f ohm-1 m-1",sigma_EX);
```

---

**Scilab code Exa 8.13** To calculate the conductivity equilibrium hole concentration and position of Fermi level

```
1 //To calculate the conductivity , equilibrium hole
    concentration and position of Fermi level
2 ni = 1.5*10^16;      ////intrinsic charge carriers
    per m^3
3 e = 1.6*10^-19;
4 mew_e = 0.135;      //electron mobility , m^2/Vs
5 mew_h = 0.048;      //hole mobility , m^2/Vs
6 sigma = ni*e*(mew_e+mew_h);      //conductivity , ohm
    -1 m-1
7 printf("conductivity is %f ohm-1 m-1",sigma);
8 Nd = 10^23;          //phosphorus atoms per m^3
9 p = ni^2/Nd;        //hole concentration per m^3
10 printf("hole concentration per m^3 is");
11 disp(p);
12 sigma_ex = Nd*e*mew_e;      //conductivity , ohm-1 m
    -1
13 k = 1.38*10^-23;      //boltzmann constant
14 T = 300;             //temperature , K
15 //EF = (Eg/2)+(3*k*T*log(mew_e/mew_h)/4)
16 X = 3*k*T*log(mew_e/mew_h)/(4*e);
17 //EF = (Eg/2)+X
18 printf("EF = Eg/2 + %5.2f",X);
19 printf("Fermi level will be %5.2f eV above intrinsic
    level",X);
```

---

**Scilab code Exa 8.14** To calculate the diffusion coefficient of electrons

```

1 //To calculate the diffusion coefficient of
  electrons
2 mew_e = 0.19;          //electron mobility , m^2/Vs
3 k = 1.38*10^-23;     //boltzmann constant
4 T = 300;             //temperature , K
5 e = 1.6*10^-19;
6 Dn = mew_e*k*T/e;    //diffusion coefficient , m^2/s
7 printf("diffusion coefficient of electrons is %f m
  ^2/s" ,Dn);

```

---

**Scilab code Exa 8.15** To calculate the Hall voltage

```

1 //To calculate the Hall voltage
2 RH = 3.66*10^-4;     //Hall coefficient , m^3/coulomb
3 I = 10^-2;           //current , amp
4 B = 0.5;             //magnetic field , Wb/m^2
5 t = 1;               //thickness , mm
6 t = t*10^-3;        //thickness , m
7 VH = RH*I*B/t;      //Hall voltage , V
8 VH = VH*10^3;       //Hall voltage , mV
9 printf("Hall voltage developed is %5.2f mV" ,VH);

```

---

**Scilab code Exa 8.16** To calculate the Hall coefficient

```

1 //To calculate the Hall coefficient
2 Vy = 37;             //voltage , micro-V
3 Vy = Vy*10^-6;      //voltage , V
4 t = 1;               //thickness , mm
5 t = t*10^-3;        //thickness , m
6 Bz = 0.5;           //flux density , Wb/m^2
7 Ix = 20;            //current , mA
8 Ix = Ix*10^-3;     //current , A
9 RH = Vy*t/(Ix*Bz); //Hall coefficient , C-1 m^3

```

```

10 printf("Hall coefficient of semiconductor in C-1 m^3
        is");
11 disp(RH);

```

---

**Scilab code Exa 8.17** To calculate the density and mobility of charge carriers

```

1 //To calculate the density and mobility of charge
  carriers
2 RH = -7.35*10^-5;      //Hall coefficient , m^3/C
3 e = 1.6*10^-19;
4 sigma = 200;          //conductivity , ohm-1 m-1
5 n = -1/(RH*e);        //density , m^3
6 printf("density of charge carriers in m^3 is");
7 disp(n);
8 mew = sigma/(n*e);    //mobility , m^2/Vs
9 printf("mobility of charge carriers is %5.4f m^2/Vs"
        ,mew);

```

---

**Scilab code Exa 8.18** To calculate the magnitude of Hall voltage

```

1 //To calculate the magnitude of Hall voltage
2 I = 50;              //current , A
3 B = 1.5;             //magnetic field , T
4 e = 1.6*10^-19;
5 n = 8.4*10^28;       //free electron concentration ,
  electron/m^3
6 t = 0.5;            //thickness , cm
7 t = t*10^-2;        //thickness , m
8 VH = I*B/(n*e*t);   //hall voltage , V
9 VH = VH*10^6;        //hall voltage , micro-V
10 printf("magnitude of Hall voltage is %5.3f micro-V" ,
        VH);

```

---

**Scilab code Exa 8.19** To calculate the value of  $\mu_{ew}$  and  $n$

```
1 //To calculate the value of  $\mu_{ew}$  and  $n$ 
2 RH = 3.66*10^-4; //Hall coefficient , m^3/C
3 e = 1.6*10^-19;
4 n = 1/(RH*e);
5 rho_n = 8.93*10^-3; //resistivity , ohm m
6 printf("value of n per m^3 is");
7 disp(n);
8 mew_e = RH/rho_n;
9 printf("value of  $\mu_{ew}$  is %5.3f m^2/Vs" ,mew_e);
```

---



# Chapter 9

## Superconductivity

Scilab code Exa 9.1 To calculate the critical field

```
1 //To calculate the critical field
2 Tc = 3.7; //critical temperature, K
3 Hc_0 = 0.0306; //critical field, T
4 T = 2; //temperature, K
5 Hc_2 = Hc_0*(1-(T/Tc)^2); //critical field, T
6 printf("critical field at 2K is %5.5f T",Hc_2);
```

---

Scilab code Exa 9.2 To calculate the critical current

```
1 //To calculate the critical current
2 T = 4.2; //temperature, K
3 d = 1; //diameter, mm
4 d = d*10^-3; //diameter, m
5 Tc = 7.18; //critical temperature, K
6 H0 = 6.5*10^4; //critical field, A/m
7 Hc = H0*(1-(T/Tc)^2); //critical field at 2K, A/m
8 ic = %pi*d*Hc; //critical current, A
9 printf("critical current for lead is %5.2f A",ic);
```

```
10
11 //answer given in the book is wrong
```

---

**Scilab code Exa 9.3** To calculate the penetration depth

```
1 //To calculate the penetration depth
2 lamda_T = 750; //penetration depth of mercury,
   Armstrong
3 T = 3.5; //temperature, K
4 Tc = 4.12; //critical temperaturure, K
5 lamda_0 = lamda_T*((1-(T/Tc)^4))^(1/2); //
   penetration depth, Armstrong
6 printf("penetration depth at 0K is %d armstrong",
   lamda_0);
```

---

**Scilab code Exa 9.4** To calculate the critical temperature

```
1 //To calculate the critical temperature
2 T1 = 3; //temperature, K
3 T2 = 7.1; //temperature, K
4 lamda_T1 = 396; //penetration depth, armstrong
5 lamda_T2 = 1730; //penetration depth, armstrong
6 A = (((lamda_T2/lamda_T1)^2)*T2^4) - T1^4;
7 B = ((lamda_T2/lamda_T1)^2)-1;
8 Tc = (A/B)^(1/4); //critical temperature, K
9 printf("critical temperature for lead is %5.3f K",Tc
   );
```

---

# Chapter 10

## Lasers

Scilab code Exa 10.1 To calculate the relative population

```
1 //To calculate the relative population
2 c = 3*10^8; //speed of light , m/sec
3 h = 6.6*10^-34; //planck's constant
4 e = 1.6*10^-19;
5 T = 300; //temperature , K
6 K = 8.61*10^-5;
7 lamda = 6943; //wavelength , armstrong
8 lamda = lamda*10^-10; //wavelength , m
9 // let E2 - E1 be E
10 E = h*c/lamda; //energy , J
11 E = E/e; //energy , eV
12 //let population ratio N2/N1 be N
13 N = exp(-E/(K*T));
14 printf("relative population of 2 states is");
15 disp(N);
16
17 //answer given in the book is wrong
```

---

Scilab code Exa 10.2 To calculate the divergence

```

1 //To calculate the divergence
2 a2 = 6; //spot diameter , mm
3 a2 = a2*10^-3; //spot diameter , m
4 a1 = 4; //spot diameter , mm
5 a1 = a1*10^-3; //spot diameter , m
6 d2 = 2; //distance from laser , m
7 d1 = 1; //distance from laser , m
8 theta = (a2-a1)/(2*(d2-d1)); //divergence ,
   radian
9 theta = theta*10^3; //divergence , milli radian
10 printf("divergence is %d milli radian",theta);

```

---

**Scilab code Exa 10.3** To calculate the spot size

```

1 //To calculate the spot size
2 n = 1; //for air
3 lamda = 650; //wavelength , nm
4 lamda = lamda*10^-9; //wavelength , m
5 bs = 1; //beam size , mm
6 bs = bs*10^-3; //beam size , m
7 fl = 1; //focal length of lens , mm
8 fl = fl*10^-3; //focal length of lens , m
9 tan_theta = fl/(2*bs); //value of tan_theta
10 theta = atand(tan_theta);
11 NA = n*sind(theta);
12 ss = 0.6*lamda/NA; //spot size , m
13 ss = ss*10^6; //spot size , micro metre
14 printf("spot size is %5.3f micro metre",ss);
15
16 //answer given in the book is wrong

```

---

# Chapter 11

## Fiber Optics and Holography

Scilab code Exa 11.1 To calculate the numerical aperture

```
1 //To calculate the numerical aperture
2 n1 = 1.55; //refractive index of core
3 n2 = 1.50; //refractive index of cladding
4 NA = sqrt(n1^2 - n2^2);
5 printf("numerical aperture is %5.3f",NA);
```

---

Scilab code Exa 11.2 To calculate the angle of acceptance

```
1 //To calculate the angle of acceptance
2 n1 = 1.563; //refractive index of core
3 n2 = 1.498; //refractive index of cladding
4 NA = sqrt(n1^2 - n2^2); //numerical aperture
5 alpha_i = asind(NA); //angle of acceptance,
degrees
6 deg = int(alpha_i); //angle in degrees
7 t = 60*(alpha_i-deg);
8 mint = int(t); //angle in minutes
9 printf("the angle of acceptance is %d degrees %d
minutes",deg,mint);
```

10  
11 //answer given in the book is wrong

---

**Scilab code Exa 11.3** To calculate the refractive index of the core

```
1 //To calculate the refractive index of the core
2 delta = 0.05; //difference in refractive indices
   of core and cladding
3 NA = 0.39; //numerical aperture
4 n1 = NA/sqrt(2*delta); //refractive index of
   core
5 printf("refractive index of the core is %5.4f",n1);
```

---

**Scilab code Exa 11.4** To calculate the fractional index change

```
1 //To calculate the fractional index change
2 n1 = 1.563; //refractive index of core
3 n2 = 1.498; //refractive index of cladding
4 delta = (n1-n2)/n1; //fractional index change
5 printf("fractional index change is %5.4f",delta);
```

---

**Scilab code Exa 11.5** To calculate the numerical aperture and acceptance angle

```
1 //To calculate the numerical aperture and acceptance
   angle
2 n1 = 1.48; //refractive index of core
3 n2 = 1.45; //refractive index of cladding
4 NA = sqrt(n1^2 - n2^2); //numerical aperture
5 printf("numerical aperture is %5.4f",NA);
```

```
6 alpha_i = asind(NA);      //angle of acceptance ,
    degrees
7 deg = int(alpha_i);      //angle in degrees
8 t = 60*(alpha_i-deg);
9 mint = int(t);          //angle in minutes
10 printf("the angle of acceptance is %d degrees %d
    minutes",deg,mint);
11
12 //answer given in the book is wrong
```

---