

Scilab Textbook Companion for
Thermal Engineering
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Book Description

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Scilab numbering policy used in this document and the relation to the above book.

Exa Example (Solved example)

Eqn Equation (Particular equation of the above book)

AP Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

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Chapter 1

Fuels and Combustion

Scilab code Exa 1.1 Minimum mass of air required

```
1 //Chapter -1, Illustration 1, Page 15
2 //Title: Fuels and Combustion
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 C=0.91; //Percentage composition of Carbon
9 H=0.03; //Percentage composition of Hydrogen
10 O=0.02; //Percentage composition of Oxygen
11 N=0.008; //Percentage composition of Nitrogen
12 S=0.008; //Percentage composition of Sulphur
13
14 //CALCULATIONS
15 m=(11.5*C)+(34.5*(H-(O/8)))+(4.3*S); //Mass of air
    per kg of coal in kg
16
17 //OUTPUT
18 mprintf('Minimum mass of air per kg of coal is %3.2f
```

```

    kg',m)
19
20
21
22
23
24 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.2 Theoretical volume of air required

```

1 //Chapter-1, Illustration 2, Page 16
2 //Title: Fuels and Combustion
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 C=0.86; //Percentage composition of Carbon
9 H=0.12; //Percentage composition of Hydrogen
10 O=0.01; //Percentage composition of Oxygen
11 S=0.01; //Percentage composition of Sulphur
12 v=0.773; //Specific volume of air at N.T.P in (m^3)/
    kg
13
14 //CALCULATIONS
15 m=(11.5*C)+(34.5*(H-(O/8)))+(4.3*S); //Theoretical
    mass of air per kg of coal in kg
16 vth=m*v; //Theoretical volume of air at N.T.P per kg
    fuel in (m^3)/kg of fuel
17
18 //OUTPUT
19 printf('Theoretical volume of air at N.T.P per kg

```

```

    fuel is %3.2f (m^3)/kg of fuel ',vth)
20
21
22
23
24
25 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.3 Minimum quantity of air and Total mass of products of combustion

```

1 //Chapter -1, Illustration 3, Page 16
2 //Title: Fuels and Combustion
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 C=0.78; //Percentage composition of Carbon
9 H=0.06; //Percentage composition of Hydrogen
10 O=0.078; //Percentage composition of Oxygen
11 N=0.012; //Percentage composition of Nitrogen
12 S=0.03; //Percentage composition of Sulphur
13
14 //CALCULATIONS
15 m=(11.5*C)+(34.5*(H-(O/8)))+(4.3*S); //Minimum
    quantity of air required in kg
16 mt=((11*C)/3)+(9*H)+(2*S)+(8.32*N); //Total mass of
    products of combustion in kg
17
18 //OUTPUT
19 printf('Minimum quantity of air required for

```

```

    complete combustion is %3.2f kg \n Total mass of
    products of combustion is %3.3f kg',m,mt)
20
21
22
23
24
25 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.4 Mass of dry flue gas

```

1 //Chapter –1, Illustration 4, Page 17
2 //Title: Fuels and Combustion
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 C=0.84; //Percentage composition of Carbon
9 H=0.09; //Percentage composition of Hydrogen
10 C02=0.0875; //Volumetric composition of CO2
11 C0=0.0225; //Volumetric composition of CO
12 O2=0.08; //Volumetric composition of Oxygen
13 N2=0.81; //Volumetric composition of Nitrogen
14 M1=44; //Molecular mass of CO2
15 M2=28; //Molecular mass of CO
16 M3=32; //Molecular mass of O2
17 M4=28; //Molecular mass of N2
18
19 //CALCULATIONS
20 c1=C02*M1; //Proportional mass of CO2
21 c2=C0*M2; //Proportional mass of CO

```

```

22 c3=O2*M3; //Proportional mass of O2
23 c4=N2*M4; //Proportional mass of N2
24 c=c1+c2+c3+c4; //Total proportional mass of
    constituents
25 m1=c1/c; //Mass of CO2 per kg of flue gas in kg
26 m2=c2/c; //Mass of CO per kg of flue gas in kg
27 m3=c3/c; //Mass of O2 per kg of flue gas in kg
28 m4=c4/c; //Mass of N2 per kg of flue gas in kg
29 d1=m1*100; //Mass analysis of CO2
30 d2=m2*100; //Mass analysis of CO
31 d3=m3*100; //Mass analysis of O2
32 d4=m4*100; //Mass analysis of N2
33 m=((3*m1)/11)+((3*m2)/7); //Mass of carbon in kg
34 md=C/m; //Mass of dry flue gas in kg
35
36 //OUTPUT
37 mprintf('Mass of dry flue gases per kg of coal burnt
    is %3.1f kg',md)
38
39
40
41
42
43 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.5 Minimum air required and Mass of air actually supplied and Amount of excess air supplied

```

1 //Chapter -1, Illustration 5, Page 17
2 //Title: Fuels and Combustion
3 //
    =====
4 clc

```

```

5 clear
6
7 //INPUT DATA
8 C=0.624; //Percentage composition of Carbon
9 H=0.042; //Percentage composition of Hydrogen
10 O=0.045; //Percentage composition of Oxygen
11 CO2=0.13; //Volumetric composition of CO2
12 CO=0.003; //Volumetric composition of CO
13 O2=0.06; //Volumetric composition of Oxygen
14 N2=0.807; //Volumetric composition of Nitrogen
15 M1=44; //Molecular mass of CO2
16 M2=28; //Molecular mass of CO
17 M3=32; //Molecular mass of O2
18 M4=28; //Molecular mass of N2
19 mw=0.378; //Mass of H2O in kg
20
21 //CALCULATIONS
22 m=(11.5*C)+(34.5*(H-(O/8))); //Minimum air required
    in kg
23 c1=CO2*M1; //Proportional mass of CO2
24 c2=CO*M2; //Proportional mass of CO
25 c3=O2*M3; //Proportional mass of O2
26 c4=N2*M4; //Proportional mass of N2
27 c=c1+c2+c3+c4; //Total proportional mass of
    constituents
28 m1=c1/c; //Mass of CO2 per kg of flue gas in kg
29 m2=c2/c; //Mass of CO per kg of flue gas in kg
30 m3=c3/c; //Mass of O2 per kg of flue gas in kg
31 m4=c4/c; //Mass of N2 per kg of flue gas in kg
32 d1=m1*100; //Mass analysis of CO2
33 d2=m2*100; //Mass analysis of CO
34 d3=m3*100; //Mass analysis of O2
35 d4=m4*100; //Mass analysis of N2
36 mC=((3*m1)/11)+((3*m2)/7); //Mass of carbon in kg
37 md=C/mC; //Mass of dry flue gas in kg
38 mact=(md+mw)-(C+H+O); //Actual air supplied per kg of
    fuel in kg
39 me=mact-m; //Mass of excess air per kg of fuel in kg

```

```

40
41 //OUTPUT
42 mprintf('Minimum air required to burn 1 kg of coal
         is %3.2f kg \n Mass of air actually supplied per
         kg of coal is %3.3f kg \n Amount of excess air
         supplied per kg of coal burnt is %3.3f kg',m,mact
         ,me)
43
44
45
46
47
48 //=====END OF PROGRAM
=====

```

Scilab code Exa 1.6 Mass of air to be supplied and Mass of gaseous products

```

1 //Chapter-1, Illustration 6, Page 19
2 //Title: Fuels and Combustion
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 C=0.78; //Percentage composition of Carbon
9 H=0.03; //Percentage composition of Hydrogen
10 O=0.03; //Percentage composition of Oxygen
11 S=0.01; //Percentage composition of Sulphur
12 me=0.3; //Mass of excess air supplied
13
14 //CALCULATIONS
15 m=(11.5*C)+(34.5*(H-(O/8)))+(4.3*S); //Mass of air

```



```

    per kg of coal in kg
16 mec=me*m; //Excess air supplied per kg of coal in kg
17 mact=m+mec; //Actual mass of air supplied per kg of
    coal in kg
18 mCO2=(11*C)/3; //Mass of CO2 produced per kg of coal
    in kg
19 mHw=9*H; //Mass of H2O produced per kg of coal in kg
20 mSO2=2*S; //Mass of SO2 produced per kg of coal in kg
21 mO2=0.232*mec; //Mass of excess O2 produced per kg of
    coal in kg
22 mN2=0.768*mact; //Mass of N2 produced per kg of coal
    in kg
23
24 //OUTPUT
25 mprintf('Mass of air to be supplied is %3.2f kg \n
    Mass of CO2 produced per kg of coal is %3.2f kg \
    n Mass of H2O produced per kg of coal is %3.2f kg
    \n Mass of SO2 produced per kg of coal is %3.2f
    kg \n Mass of excess O2 produced per kg of coal
    is %3.2f kg \n Mass of N2 produced per kg of coal
    is %3.2f kg \n ',m,mCO2,mHw,mSO2,mO2,mN2)
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.7 Total mass of dry flue gases and Percentage composition of dry flue gases by volume

```

1 //Chapter –1, Illustration 7, Page 20
2 //Title: Fuels and Combustion
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  C=0.9; //Percentage composition of Carbon
9  H=0.033; //Percentage composition of Hydrogen
10 O=0.03; //Percentage composition of Oxygen
11 N=0.008; //Percentage composition of Nitrogen
12 S=0.009; //Percentage composition of Sulphur
13 M1=44; //Molecular mass of CO2
14 M2=64; //Molecular mass of SO2
15 M3=32; //Molecular mass of O2
16 M4=28; //Molecular mass of N2
17
18 //CALCULATIONS
19 m=(11.5*C)+(34.5*(H-(O/8)))+(4.3*S); //Minimum mass
    of air per kg of coal in kg
20 mCO2=(11*C)/3; //Mass of CO2 produced per kg of coal
    in kg
21 mHw=9*H; //Mass of H2O produced per kg of coal in kg
22 mSO2=2*S; //Mass of SO2 produced per kg of coal in kg
23 mt=11.5*1.5; //Total mass of air supplied per kg of
    coal in kg
24 me=mt-m; //Excess air supplied in kg
25 mO2=0.232*me; //Mass of excess O2 produced per kg of
    coal in kg
26 mN2=0.768*mt; //Mass of N2 produced per kg of coal in
    kg
27 mtN2=mN2+N; //Total mass of Nitrogen in exhaust in kg
28 md=mCO2+mSO2+mO2+mtN2; //Total mass of dry flue gases
    per kg of fuel in kg
29 CO2=(mCO2/md)*100; //Percentage composition of CO2 by
    mass in percent
30 SO2=(mSO2/md)*100; //Percentage composition of SO2 by
    mass in percent
31 O2=(mO2/md)*100; //Percentage composition of O2 by
    mass in percent

```

```

32 N2=(mN2/md)*100;//Percentage composition of N2 by
    mass in percent
33 c1=C02/M1;//Proportional volume of CO2
34 c2=S02/M2;//Proportional volume of SO2
35 c3=O2/M3;//Proportional volume of O2
36 c4=N2/M4;//Proportional volume of N2
37 c=c1+c2+c3+c4;//Total proportional volume of
    constituents
38 m1=c1/c;//Volume of CO2 in 1 (m^3) of flue gas
39 m2=c2/c;//Volume of SO2 in 1 (m^3) of flue gas
40 m3=c3/c;//Volume of O2 in 1 (m^3) of flue gas
41 m4=c4/c;//Volume of N2 in 1 (m^3) of flue gas
42 d1=m1*100;//Volume analysis of CO2
43 d2=m2*100;//Volume analysis of SO2
44 d3=m3*100;//Volume analysis of O2
45 d4=m4*100;//Volume analysis of N2
46
47 //OUTPUT
48 mprintf('Minimum mass of air required is %3.1f kg \n
    Total mass of dry flue gases per kg of fuel is
    %3.2f kg \n Percentage composition of CO2 by
    volume is %3.2f percent \n Percentage composition
    of SO2 by volume is %3.3f percent \n Percentage
    composition of O2 by volume is %3.1f percent \n
    Percentage composition of N2 by volume is %3.2f
    percent ',m,md,d1,d2,d3,d4)
49
50
51
52
53
54
55 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.8 Mass of air actually supplied and Percentage of excess air supplied

```
1 //Chapter -1, Illustration 8, Page 21
2 //Title: Fuels and Combustion
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 C=0.88; //Percentage composition of Carbon
9 H=0.036; //Percentage composition of Hydrogen
10 O=0.048; //Percentage composition of oxygen
11 CO2=0.109; //Volumetric composition of CO2
12 CO=0.01; //Volumetric composition of CO
13 O2=0.071; //Volumetric composition of Oxygen
14 N2=0.81; //Volumetric composition of Nitrogen
15 M1=44; //Molecular mass of CO2
16 M2=28; //Molecular mass of CO
17 M3=32; //Molecular mass of O2
18 M4=28; //Molecular mass of N2
19
20 //CALCULATIONS
21 m=(11.5*C)+(34.5*(H-(O/8))); //Theoretical air
    required in kg
22 c1=CO2*M1; //Proportional mass of CO2
23 c2=CO*M2; //Proportional mass of CO
24 c3=O2*M3; //Proportional mass of O2
25 c4=N2*M4; //Proportional mass of N2
26 c=c1+c2+c3+c4; //Total proportional mass of
    constituents
27 m1=c1/c; //Mass of CO2 per kg of flue gas in kg
28 m2=c2/c; //Mass of CO per kg of flue gas in kg
29 m3=c3/c; //Mass of O2 per kg of flue gas in kg
30 m4=c4/c; //Mass of N2 per kg of flue gas in kg
31 mC=((3*m1)/11)+((3*m2)/7); //Mass of carbon in kg
```

```

32 md=C/mC;//Mass of dry flue gas in kg
33 hc=H*9;//Hydrogen combustion in kg of H2O
34 mair=(md+hc)-(C+H+O);//Mass of air supplied per kg
    of coal in kg
35 me=mair-m;//Excess air per kg of coal in kg
36 mN2=m4*md;//Mass of nitrogen per kg of coal in kg
37 mact=mN2/0.768;//Actual mass of air per kg of coal
    in kg
38 pe=(me/m)*100;//Percentage excess air in percent
39
40 //OUTPUT
41 mprintf('Mass of air actually supplied per kg of
    coal is %3.2f kg \n Percentage of excess air is
    %3.2f percent',mact,pe)
42
43
44
45
46
47
48
49 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.9 Mass of excess air supplied and air fuel ratio

```

1 //Chapter -1, Illustration 9, Page 22
2 //Title: Fuels and Combustion
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA

```

```

8 C=0.84; //Percentage composition of Carbon
9 H=0.14; //Percentage composition of Hydrogen
10 O=0.02; //Percentage composition of oxygen
11 CO2=8.85; //Volumetric composition of CO2
12 CO=1.2; //Volumetric composition of CO
13 O2=6.8; //Volumetric composition of Oxygen
14 N2=83.15; //Volumetric composition of Nitrogen
15 M1=44; //Molecular mass of CO2
16 M2=28; //Molecular mass of CO
17 M3=32; //Molecular mass of O2
18 M4=28; //Molecular mass of N2
19 a=8/3; //O2 required per kg C
20 b=8; //O2 required per kg H2
21 mair=0.23; //Mass of air
22
23 //CALCULATIONS
24 c=C*a; //O2 required per kg of fuel for C
25 d=H*b; //O2 required per kg of fuel for H2
26 tO2=c+d+0; //Theoretical O2 required in kg/kg of
    fuel
27 tm=tO2/mair; //Theoretical mass of air in kg/kg of
    fuel
28 c1=CO2*M1; //Proportional mass of CO2 by Volume
29 c2=CO*M2; //Proportional mass of CO by Volume
30 c3=O2*M3; //Proportional mass of O2 by Volume
31 c4=N2*M4; //Proportional mass of N2 by Volume
32 c=c1+c2+c3+c4; //Total proportional mass of
    constituents
33 m1=c1/c; //Mass of CO2 per kg of flue gas in kg
34 m2=c2/c; //Mass of CO per kg of flue gas in kg
35 m3=c3/c; //Mass of O2 per kg of flue gas in kg
36 m4=c4/c; //Mass of N2 per kg of flue gas in kg
37 mC=((m1*12)/M1)+((m2*12)/M2); //Mass of carbon per kg
    of dry flue gas in kg
38 md=C/mC; //Mass of dry flue per kg of fuel in kg
39 p=(4*m2)/7; //Oxygen required to burn CO in kg
40 meO2=md*(m3-p); //Mass of excess O2 per kg of fuel in
    kg

```

```

41 me=me02/mair;//Mass of excess air in kg/kg fuel
42 mt=tm+me;//Total air required per kg fuel
43
44 //OUTPUT
45 mprintf('Mass of excess air supplied per kg of fuel
      burnt is %3.1f kg/kg of fuel \n Air-fuel ratio is
      %3.1f:1 ',me,mt)
46
47
48
49
50
51 //=====END OF PROGRAM
      =====

```

Scilab code Exa 1.10 Volume of air required

```

1 //Chapter-1, Illustration 10, Page 23
2 //Title: Fuels and Combustion
3 //
      =====

4 clc
5 clear
6
7 //INPUT DATA
8 H2=0.27;//Percentage composition of H2 by volume
9 CO2=0.18;//Percentage composition of CO2 by volume
10 CO=0.125;//Percentage composition of CO by volume
11 CH4=0.025;//Percentage composition of CH4 by volume
12 N2=0.4;//Percentage composition of N2 by volume
13
14 //CALCULATIONS
15 v=(2.38*(H2+CO))+(9.52*CH4);//Volume of air required
      for complete combustion in (m^3)

```

```

16
17 //OUTPUT
18 mprintf('Volume of air required for complete
      combustion is %3.3f (m^3)',v)
19
20
21
22
23
24
25 //=====END OF PROGRAM
      =====

```

Scilab code Exa 1.11 Air fuel ratio

```

1 //Chapter-1, Illustration 11, Page 24
2 //Title: Fuels and Combustion
3 //
      =====

4 clc
5 clear
6
7 //INPUT DATA
8 H2=0.5; //Percentage composition of H2 by volume
9 CO2=0.1; //Percentage composition of CO2 by volume
10 CO=0.05; //Percentage composition of CO by volume
11 CH4=0.25; //Percentage composition of CH4 by volume
12 N2=0.1; //Percentage composition of N2 by volume
13 pCO2=8; //Percentage volumetric analysis of CO2
14 pO2=6; //Percentage volumetric analysis of O2
15 pN2=86; //Percentage volumetric analysis of N2
16
17
18 //CALCULATIONS

```



```

19 v=(2.38*(H2+C0))+(9.52*CH4); //Volume of air required
    for complete combustion in (m^3)
20 vN2=v*0.79; //Volume of nitrogen in the air in m^3
21 a=C0+CH4+C02; //CO2 formed per m^3 of fuel gas burnt
22 b=vN2+N2; //N2 formed per m^3 of fuel gas burnt
23 vt=a+b; //Total volume of dry flue gas formed in m^3
24 ve=(p02*vt)/(21-p02); //Excess air supplied in m^3
25 V=v+ve; //Total quantity of air supplied in m^3
26
27 //OUTPUT
28 mprintf('Air-fuel ratio by volume is %3.3f:1',V)
29
30
31
32
33 //=====END OF PROGRAM
    =====

```

Scilab code Exa 1.12 Volume and analysis of products of combustion

```

1 //Chapter-1, Illustration 12, Page 24
2 //Title: Fuels and Combustion
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 H2=0.14; //Percentage composition of H2 by volume
9 C02=0.05; //Percentage composition of CO2 by volume
10 C0=0.22; //Percentage composition of CO by volume
11 CH4=0.02; //Percentage composition of CH4 by volume
12 O2=0.02; //Percentage composition of O2 by volume
13 N2=0.55; //Percentage composition of N2 by volume

```

```

14 e=0.4; //Excess air supplied
15 //CALCULATIONS
16 v=(2.38*(H2+C0))+(9.52*CH4)-(4.76*O2); //Volume of
    air required for complete combustion in (m^3)
17 ve=v*e; //Volume of excess air supplied in m^3
18 vtN2=v-(v*0.21); //Volume of N2 in theoretical air in
    m^3
19 veN2=ve-(ve*0.21); //Volume of N2 in excess air in m
    ^3
20 vt=vtN2+veN2; //Total volume of N2 in air supplied in
    m^3
21 vCO2=C0+CH4+C02; //CO2 formed per m^3 of fuel gas
22 vN2=vt+N2; //N2 formed per m^3 of fuel gas
23 veO2=ve*0.21; //Volume of excess O2 per m^3 of fuel
    gas
24 vT=vCO2+vN2+veO2; //Total volume of dry combustion
    products
25 pCO2=(vCO2*100)/vT; //Percentage volume of CO2
26 pN2=(vN2*100)/vT; //Percentage volume of N2
27 pO2=(veO2*100)/vT; //Percentage volume of O2
28
29 //OUTPUT
30 mprintf('Volume of air required for complete
    combustion is %3.3f (m^3) \n Volume of CO2 per m
    ^3 of gas fuel is %3.2f m^3/m^3 of gas fuel \n
    Volume of N2 per m^3 of gas fuel is %3.3f m^3/m^3
    of gas fuel \n Volume of excess O2 per m^3 of
    gas fuel is %3.2f m^3/m^3 of gas fuel \n Total
    volume of dry combustion products is %3.3f m^3/m
    ^3 of gas fuel \n Percentage volume of CO2 is %3
    .1f percent \n Percentage volume of N2 is %3.2f
    percent \n Percentage volume of O2 is %3.2f
    percent ',v,vCO2,vN2,veO2,vT,pCO2,pN2,pO2)
31
32
33
34
35

```

36

37

38

39

40 //=====END OF PROGRAM

=====

Chapter 2

Gas Power Cycles

Scilab code Exa 2.1 Maximum pressure and temperature of cycle and Cycle efficiency and Mean effective pressure

```
1 //Chapter-2, Illustration 1, Page 55
2 //Title: Gas Power Cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 P1=0.1; //Pressure of air supplied in MPa
9 T1=308; //Temperature of air supplied in K
10 rv=8; //Compression ratio
11 q1=2100; //Heat supplied in kJ/kg
12 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
13 Cv=0.718; //Specific heat at constant volume in kJ/kg
    -K
14 R=0.287; //Universal gas constant in kJ/kg-K
15
16 //CALCULATIONS
```

```

17 y=Cp/Cv;//Ratio of specific heats
18 n=(1-(1/(rv^(y-1))))*100;//Cycle efficiency
19 v1=(R*T1)/(P1*1000);//Specific volume at point 1 in
    (m^3)/kg
20 v2=v1/rv;//Specific volume at point 2 in (m^3)/kg
21 T2=T1*(rv^(y-1));//Temperature at point 2 in K
22 T3=(q1/Cv)+T2;//Temperature at point 3 in K
23 P2=P1*(rv^y);//Pressure at point 2 in MPa
24 P3=P2*(T3/T2);//Pressure at point 3 in MPa
25 wnet=(q1*n)/100;//Net workdone in J/kg
26 MEP=(wnet/(v1-v2))/1000;//Mean effective pressure in
    MPa
27
28 //OUTPUT
29 mprintf('Maximum pressure of the cycle is %3.3f MPa
    \n Maximum temperature of the cycle is %3.0f K \n
    Cycle efficiency is %3.1f percent \n Mean
    effective pressure is %3.3f MPa',P3,T3,n,MEP)
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.2 Relative efficiency of engine

```

1 //Chapter-2, Illustration 2, Page 57
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear

```

```

6
7 //INPUT DATA
8 d=80; //Bore in mm
9 L=85; //Stroke in mm
10 Vc=0.06; //Clearance volume in litre
11 n=0.22; //Actual thermal efficiency
12 y=1.4; //Ratio of specific heats
13
14 //CALCULATIONS
15 Vs=(3.147/4)*(d^2)*L; //Stroke volume in mm^3
16 Vt=Vs+(Vc*(10^6)); //Total volume in mm^3
17 rv=Vt/(Vc*(10^6)); //Compression ratio
18 ni=(1-(1/(rv^(y-1))))); //Ideal thermal efficiency
19 nr=(n/ni)*100; //Relative efficiency
20
21 //OUTPUT
22 mprintf('Relative efficiency of the engine is %3.1f
    percent ',nr)
23
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.3 Air standard efficiency

```

1 //Chapter-2, Illustration 3, Page 57
2 //Title: Gas Power Cycles
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  d=0.137; //Bore in m
9  L=0.13; //Stroke in m
10 Vc=280*(10^-6); //Clearance volume in m^3
11 y=1.4; //Ratio of specific heats
12
13 //CALCULATIONS
14 Vs=(3.147/4)*(d^2)*L; //Stroke volume in m^3
15 rv=(Vc/Vs)*100; //Compression ratio
16 rvf=(Vs+Vc)/Vc; //final compression ratio
17 n=(1-(1/rvf^(y-1)))*100; //Cycle efficiency
18
19 //OUTPUT
20 mprintf('Clearance volume is %3.1f percent of swept
          volume \n Otto cycle efficiency is %3.2f percent '
          ,rv,n)
21
22
23
24
25
26
27 //=====END OF PROGRAM
    
```

Scilab code Exa 2.4 Highest temperature and pressure in cycle and Amount of heat transferred and Thermal efficiency and Mean effective pressure

```

1 //Chapter -2, Illustration 4, Page 58
2 //Title: Gas Power Cycles
3 //
    
```

```

4  clc
5  clear
6
7  //INPUT DATA
8  rv=9.5; //Compression ratio
9  P1=100; //Air pressure in kPa
10 T1=290; //Air temperature in K
11 V1=600*(10^-6); //Volume of air in m^3
12 T4=800; //Final temperature in K
13 R=287; //Universal gas constan in J/kg.K
14 Cv=0.718; //Specific heat at constant volume in kJ/kg
    .K
15 y=1.4; //Ratio of specific heats
16
17 //CALCULATIONS
18 T3=T4*(rv^(y-1)); //Temperature at the end of
    constant volume heat addition in K
19 P2=P1*(rv^y); //Pressure at point 2 in kPa
20 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
21 P3=P2*(T3/T2); //Pressure at point 3 in kPa
22 m=(P1*1000*V1)/(R*T1); //Specific mass in kg/s
23 Q=m*Cv*(T3-T2); //Heat transferred in kJ
24 n=(1-(1/rv^(y-1)))*100; //Thermal efficiency
25 Wnet=(n*Q)/100; //Net workdone in kJ
26 MEP=Wnet/(V1*(1-(1/rv))); //Mean effective pressure
    in kPa
27
28 //OUTPUT
29 mprintf('Maximum pressure of the cycle is %3.2f kPa
    \n Maximum temperature of the cycle is %3.1f K \n
    Amount of heat transferred is %3.2f kJ \n
    Thermal efficiency is %3.1f percent \n Mean
    effective pressure is %3.1f kPa',P3,T3,Q,n,MEP)
30
31
32
33

```



```

34
35
36
37 //=====END OF PROGRAM
=====

```

Scilab code Exa 2.5 Pressure and Temperature at the end of heat addition process and Net work output and Thermal efficiency and Mean effective pressure

```

1 //Chapter-2, Illustration 5, Page 60
2 //Title: Gas Power Cycles
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 rv=8; //Compression ratio
9 P1=95; //Pressure at point 1 in kPa
10 T1=300; //Temperature at point 1 in K
11 q23=750; //Heat transferred during constant volume
    heat addition process in kJ/kg
12 y=1.4; //Ratio of specific heats
13 Cv=0.718; //Specific heat at constant volume in kJ/kg
    -K
14 R=287; //Universal gas constant in J/kg-K
15
16 //CALCULATIONS
17 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
18 P2=P1*(rv^y); //Pressure at point 2 in kPa
19 T3=(q23/Cv)+T2; //Temperature at point 3 in K
20 P3=P2*(T3/T2); //Pressure at point 3 in kPa
21 nth=(1-(1/(rv^(y-1))))*100; //Thermal efficiency

```

```

22 Wnet=(nth*q23)/100; //Net work output in kJ/kg
23 v1=(R*T1)/(P1*1000); //Speific volume at point 1 in (
    m^3)/kg
24 MEP=Wnet/(v1*(1-(1/rv))); //Mean effective pressure
    in kPa
25
26 //OUTPUT
27 mprintf('Pressure at the end of heat addition
    process is %3.1f kPa \n Temperature at the end of
    heat addition process is %3.1f K \n Net work
    output is %3.2f kJ/kg \n Thermal efficiency is %3
    .2f percent \n Mean effective pressure is %3.0f
    kPa ',P3,T3,Wnet,nth,MEP)
28
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.6 Air standard efficiency

```

1 //Chapter-2, Illustration 6, Page 61
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 rv=14; //Compression ratio

```

```

9  c=0.06; //Cut-off percentage
10 y=1.4; //Ratio of specific heats
11
12 //CALCULATIONS
13 rc=1.78; //Cut-off ratio
14 nth=(1-(((rc^y)-1)/((rv^(y-1))*y*(rc-1))))*100; //
    Thermal efficiency
15
16 //OUTPUT
17 mprintf('Air standard efficiency is %3.1f percent',
    nth)
18
19
20
21
22
23
24
25 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.7 Cutoff ratio and Heat supplied and Cycle efficiency and MEP

```

1 //Chapter -2, Illustration 7, Page 62
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 rv=16; //Compression ratio
9 P1=0.1; //Pressure at point 1 in MPa

```

```

10 T1=288; //Temperature at point 1 in K
11 T3=1753; //Temperature at point 3 in K
12 y=1.4; //Ratio of specific heats
13 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
14 R=0.287; //Universal gas constant in kJ/kg-K
15
16 //CALCULATIONS
17 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
18 rc=T3/T2; //Cut-off ratio
19 q1=Cp*(T3-T2); //Heat supplied in kJ/kg
20 nth=(1-(((rc^y)-1)/((rv^(y-1))*y*(rc-1))))*100; //
    Cycle efficiency
21 wnet=(q1*nth)/100; //Net work done in kJ/kg
22 v1=(R*T1)/(P1*1000); //Speific volume at point 1 in (
    m^3)/kg
23 v2=v1/rv; //Speific volume at point 2 in (m^3)/kg
24 MEP=wnet/(v1-v2); //Mean effective pressure in kPa
25
26 //OUTPUT
27 mprintf('Cut-off ratio is %3.2f \n Heat supplied is
    %3.1f kJ/kg \n Cycle efficiency is %3.1f percent
    \n Mean effective pressure is %3.2f kPa',rc,q1,
    nth,MEP)
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.8 Air standard efficiency and percentage loss in efficiency

1 //Chapter -2, Illustration 8, Page 64

```

2 //Title: Gas Power Cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 d=0.15; //Bore in m
9 L=0.25; //Stroke in m
10 Vc=400*(10^-6); //Clearance volume in m^3
11 V2=Vc; //Clearance volume in m^3
12 c1=0.05; //Cut-off percentage 1
13 c2=0.08; //Cut-off percentage 2
14 y=1.4; //Ratio of specific heats
15
16 //CALCULATIONS
17 Vs=(3.147/4)*(d^2)*L; //Stroke volume in m^3
18 V31=V2+(c1*Vs); //Volume at the point of cut-off in m
   ^3
19 rc1=V31/V2; //Cut-off ratio 1
20 rv=(Vc+Vs)/Vc; //Compression ratio
21 nth1=(1-(((rc1^y)-1)/((rv^(y-1))*y*(rc1-1))))*100; //
   Air standard efficiency 1
22 V32=V2+(c2*Vs); //Volume at the point of cut-off in m
   ^3
23 rc2=V32/V2; //Cut-off ratio 2
24 nth2=(1-(((rc2^y)-1)/((rv^(y-1))*y*(rc2-1))))*100; //
   Air standard efficiency 2
25 pl=nth1-nth2; //Percentage loss in efficiency
26
27 //OUTPUT
28 mprintf('Air standard efficiency at 5 percent cut-
   off is %3.2f percent \n Air standard efficiency
   at 8 percent cut-off is %3.2f percent \n
   Percentage loss in efficiency is %3.2f percent',
   nth1,nth2,pl)
29

```

```

30
31
32
33
34 //=====END OF PROGRAM
=====

```

Scilab code Exa 2.9 Maximum temperature and Thermal efficiency of cycle

```

1 //Chapter-2, Illustration 9, Page 65
2 //Title: Gas Power Cycles
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 e=7.5; //Expansion ratio
9 c=15; //Compression ratio
10 P1=98; //Pressure at point 1 in kN/(m^2)
11 P4=258; //Pressure at point 4 in kN/(m^2)
12 T1=317; //Temperature at point 1 in K
13 y=1.4; //Ratio of specific heats
14
15 //CALCULATIONS
16 T4=T1*(P4/P1); //Temperature at point 4 in K
17 T3=T4*(e^(y-1)); //Temperature at point 3 in K
18 t3=T3-273; //Temperature at point 3 in oC
19 T2=T1*(c^(y-1)); //Temperature at point 2 in K
20 n=(1-((T4-T1)/(y*(T3-T2))))*100; //Thermal efficiency
21
22 //OUTPUT
23 printf('Maximum temperature attained during the

```

```

    cycle is %3.1f oC \n Thermal efficiency of the
    cycle is %3.1f percent ',t3,n)
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.10 Thermal efficiency and MEP

```

1 //Chapter-2, Illustration 10, Page 66
2 //Title: Gas Power Cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 rv=20; //Compression ratio
9 P1=95; //Pressure at point 1 in kPa
10 T1=293; //Temperature at point 1 in K
11 T3=2200; //Temperature at point 3 in K
12 y=1.4; //Ratio of specific heats
13 R=287; //Universal gas constant in J/kg-K
14 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
15
16 //CALCULATIONS
17 P2=P1*(rv^y); //Pressure at point 2 in kPa
18 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
19 v2=(R*T2)/(P2*1000); //Specific volume at point 2 in

```

```

    (m^3)/kg
20 v3=v2*(T3/T2); // Specific volume at point 3 in (m^3)/
    kg
21 rc=v3/v2; // Cut-off ratio
22 nth=(1-(((rc^y)-1)/((rv^(y-1))*y*(rc-1))))*100; //
    Thermal efficiency
23 q23=Cp*(T3-T2); // Heat flow between points 2 and 3 in
    kJ/kg
24 wnet=(nth*q23)/100; // Net workdone in kJ/kg
25 MEP=wnet/(v2*(rv-1)); // Mean effective pressure in
    kPa
26
27 //OUTPUT
28 mprintf('Thermal efficiency is %3.1f percent \n Mean
    effective pressure is %3.2f kPa',nth,MEP)
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.11 Cutoff ratio and air standard efficiency

```

1 //Chapter -2, Illustration 11, Page 68
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA

```



```

8 rv=21; // Compression ratio
9 re=10.5; // Expansion ratio
10 y=1.4; // Ratio of specific heats
11
12 //CALCULATIONS
13 rc=rv/re; //Cut-off ratio
14 nth=(1-(((rc^y)-1)/((rv^(y-1))*y*(rc-1))))*100; // Air
    standard efficiency
15
16 //OUTPUT
17 mprintf('Cut-off ratio is %3.0f \n Air standard
    efficiency is %3.2f percent ',rc,nth)
18
19
20
21
22
23
24
25
26 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.12 Ideal efficiency of cycle

```

1 //Chapter-2, Illustration 12, Page 69
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 rv=16; //Compression ratio

```

```

9  rp=1.5; // Pressure ratio
10 y=1.4; // Ratio of specific heats
11 cp=8; // Cut-off percentage
12
13 // CALCULATIONS
14 rc=2.2; // Cut-off ratio
15 ntd=(1-(((rp*(rc^y)-1)/((rv^(y-1))*((rp-1)+(y*rp*(rc
    -1)))))))*100; // Dual cycle efficiency
16
17 // OUTPUT
18 mprintf('Ideal efficiency of engine is %3.1f percent
    ',ntd)
19
20
21
22
23
24
25 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.13 Ideal efficiency of engine

```

1 //Chapter-2, Illustration 13, Page 69
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 d=0.2; //Bore in m
9 L=0.5; //Stroke in m
10 c=0.06; //Cut-off percentage

```

```

11 y=1.4; //Ratio of specific heats
12 rv=15; //Compression ratio
13 rp=1.4; //Pressure ratio
14
15 //CALCULATIONS
16 Vs=(3.147/4)*(d^2)*L; //Stroke volume in m^3
17 DV=c*Vs; //Difference in volumes at points 4 and 3
18 V3=Vs/(rv-1); //Specific volume at point 3 in m^3
19 V4=V3+DV; //Specific volume at point 4 in m^3
20 rc=V4/V3; //Cut-off ratio
21 ntd=(1-(((rp*(rc^y)-1)/((rv^(y-1))*((rp-1)+(y*rp*(rc
    -1)))))))*100; //Ideal efficiency
22
23 //OUTPUT
24 mprintf('Ideal efficiency of the engine is %3.1f
    percent ',ntd)
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.14 Amount of heat added and rejected and Work done and Thermal efficiency

```

1 //Chapter -2, Illustration 14, Page 70
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6

```

```

7 //INPUT DATA
8 d=0.2; //Bore in m
9 L=0.3; //Stroke in m
10 c=0.04; //Cut-off percentage
11 y=1.4; //Ratio of specific heats
12 rv=8; //Compression ratio
13 P1=1; //Pressure at point 1 in bar
14 P3=60; //Pressure at point 3 in bar
15 T1=298; //Temperature at point 1 in K
16 R=287; //Universal gas constant in J/kg
17 Cv=0.718; //Speific heat at constant volume in kJ/kg-
    K
18 Cp=1.005; //Speific heat at constant pressure in kJ/
    kg-K
19
20 //CALCULATIONS
21 Vs=(3.147/4)*(d^2)*L; //Stroke volume in m^3
22 V2=Vs/(rv-1); //Specific volume at point 2 in m^3
23 V3=V2; //Specific volume at point 3 in m^3
24 V1=V2+Vs; //Specific volume at pont 1 in m^3
25 V5=V1; //Specific volume at pont 5 in m^3
26 P2=P1*(rv^y); //Pressure at point 2 in bar
27 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
28 T3=T2*(P3/P2); //Temperature at point 3 in K
29 V4=V3+(c*(V1-V2)); //Specific volume at point 4 in m
    ^3
30 T4=T3*(V4/V3); //Temperature at point 4 in K
31 T5=T4*((V4/V5)^(y-1)); //Temperature at point 5 in K
32 q1=(Cv*(T3-T2))+(Cp*(T4-T3)); //Heat added in kJ/kg
33 q2=Cv*(T5-T1); //Heat rejected in kJ/kg
34 nth=(1-(q2/q1))*100; //Thermal efficiency
35 m=(P1*V1*(10^5))/(R*T1); //Mass of air supplied in kg
36 W=m*(q1-q2); //Workdone in kJ/cycle
37
38 //OUTPUT
39 printf('Amount of heat added is %3.1f kJ/kg \n
    Amount of heat rejected is %3.2f kJ/kg \n
    Workdone per cycle is %3.2f kJ/cycle \n Thermal

```

```

    efficiency is %3.2f percent ',q1,q2,W,nth)
40
41
42
43
44
45
46
47 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.15 MEP and Thermal efficiency

```

1 //Chapter-2, Illustration 15, Page 72
2 //Title: Gas Power Cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1;//Pressure at point 1 in bar
9 P3=70;//Pressure at point 3 in bar
10 T1=310;//Temperature at point 1 in K
11 rv=10;//Compression ratio
12 y=1.4;//Ratio of specific heats
13 qin=2805;//Heat added in kJ/kg
14 m=1;//Mass of air in kg
15 R=287;//Universal gas constant in J/kg
16 Cv=0.718;//Speific heat at constant volume in kJ/kg-
    K
17 Cp=1.005;//Speific heat at constant pressure in kJ/
    kg-K
18

```

```

19 //CALCULATIONS
20 V1=(m*R*T1)/(P1*(10^5)); //Volume at point 1 in m^3
21 T2=T1*(rv^(y-1)); //Temperature at point 2 in K
22 P2=P1*(rv^y); //Pressure at point 2 in K
23 T3=T2*(P3/P2); //Temperature at point 3 in K
24 q23=Cv*(T3-T2); //Heat supplied at constant volume in
    kJ/kg
25 q34=qin-q23; //Heat supplied at constant pressure in
    kJ/kg
26 T4=(q34/Cp)+T3; //Temperature at point 4 in K
27 V2=V1/rv; //Volume at point 2 in m^3
28 V4=V2*(T4/T3); //Volume at point 4 in m^3
29 V5=V1; //Volume at point 5 in m^3
30 T5=T4*((V4/V5)^(y-1)); //Temperature at point 5 in K
31 qout=Cv*(T5-T1); //Heat rejected in kJ/kg
32 nth=(1-(qout/qin))*100; //Thermal efficiency
33 W=qin-qout; //Workdone in kJ/kg
34 Vs=V1*(1-(1/rv)); //Swept volume in (m^3)/kg
35 MEP=(W/Vs)/100; //Mean effective pressure in bar
36
37 //OUTPUT
38 fprintf('Mean effective pressure is %3.2f bar \n
    Thermal efficiency is %3.2f percent',MEP,nth)
39
40
41
42
43
44
45 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.16 Cycle efficiency and Heat supplied and rejected and Work output and Turbine exit temperature

```

1 //Chapter -2, Illustration 16, Page 74
2 //Title: Gas Power Cycles
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1; //Pressure at point 1 in bar
9 T1=298; //Temperature at point 1 in K
10 P2=3; //Pressure at point 2 in bar
11 T3=923; //Temperature at point 3 in K
12 y=1.4; //Ratio of specific heats
13 Cp=1.005; //Speific heat at constant pressure in kJ/
    kg-K
14
15 //CALCULATIONS
16 x=(y-1)/y; //Ratio
17 rp=P2/P1; //Pressure ratio
18 nth=(1-(1/(rp^x)))*100; //Cycle efficiency
19 T2=T1*(rp^x); //Temperature at point 2 in K
20 q1=Cp*(T3-T2); //Heat supplied in kJ/kg
21 Wout=(nth*q1)/100; //Work output in kJ/kg
22 q2=q1-Wout; //Heat rejected in kJ/kg
23 T4=T3*((1/rp)^x); //Temperature at point 4 in K
24
25 //OUTPUT
26 mprintf('Cycle efficiency is %3.2f percent \n Heat
    supplied to air is %3.1f kJ/kg \n Work available
    at the shaft is %3.2f kJ/kg \n Heat rejected in
    the cooler is %3.2f kJ/kg \n Turbine exit
    temperature is %3.2f K',nth,q1,Wout,q2,T4)
27
28
29
30
31

```

```

32
33 //=====END OF PROGRAM
=====

```

Scilab code Exa 2.17 Pressure ratio and Maximum net specific work output and Thermal efficiency and Work ratio and Carnot efficiency

```

1 //Chapter –2, Illustration 17, Page 75
2 //Title: Gas Power Cycles
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 T1=283; //Temperature at point 1 in K
9 T3=1353; //Temperature at point 3 in K
10 y=1.41; //Ratio of specific heats
11 Cp=1.007; //Specific heat constant pressure in kJ/kg–
    K
12
13 //CALCULATIONS
14 x=(y-1)/y; //Ratio
15 rpmax=((T3/T1)^(1/x)); //Maximum pressure ratio
16 rpopt=sqrt(rpmax); //Optimum pressure ratio
17 T2=T1*(rpopt^x); //Temperature at point 2 in K
18 T4=T2; //Maximum temperature at point 4 in K
19 Wmax=Cp*((T3-T4)-(T2-T1)); //Maximum net specific
    work output in kJ/kg
20 nth=(Wmax/(Cp*(T3-T2)))*100; //Thermal efficiency
21 WR=nth/100; //Work ratio
22 nc=((T3-T1)/T3)*100; //Carnot efficiency
23
24 //OUTPUT

```



```

25 mprintf('Optimum pressure ratio is %3.2f \n Maximum
    net specific work output %3.0f kJ/kg \n Thermal
    efficiency %3.0f percent \n Work ratio is %3.2f \
    n Carnot efficiency is %3.0f percent ',rpopt,Wmax,
    nth,WR,nc)
26
27
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.18 Maximum work output and Cycle efficiency and Comparison with carnot efficiency

```

1 //Chapter-2, Illustration 18, Page 76
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 Tmin=300;//Minimum temperature in K
9 Tmax=1073;//Maximum temperature in K
10 Cp=1.005;//Specific heat at constant pressure in kJ/
    kg-K
11
12 //CALCULATIONS
13 Wmax=Cp*((sqrt(Tmax)-sqrt(Tmin))^2);//Maximum work
    output in kJ/kg
14 nB=(1-sqrt(Tmin/Tmax))*100;//Brayton cycle

```

```

    efficiency
15 nC=(1-(Tmin/Tmax))*100;//Carnot efficiency
16 r=nB/nC;//Ratio of brayton cycle efficiency to
    carnot efficieny
17
18 //OUTPUT
19 mprintf('Maximum work per kg of air is %3.2f kJ/kg \
    n Cycle efficiency is %3.0f percent \n Ratio of
    brayton cycle efficiency to carnot efficieny is
    %3.3f ',Wmax ,nB ,r)
20
21
22
23
24
25
26 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.19 Net power output and Thermal efficiency and Work ratio

```

1 //Chapter -2, Illustration 19, Page 77
2 //Title: Gas Power Cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 T1=291;//Temperature at point 1 in K
9 P1=100;//Pressure at point 1 in kN/(m^2)
10 nC=0.85;//Isentropic efficiency of compressor
11 nT=0.88;//Isentropic efficiency of turbine

```

```

12 rp=8; //Pressure ratio
13 T3=1273; //Temperature at point 3 in K
14 m=4.5; //Mass flow rate of air in kg/s
15 y=1.4; //Ratio of speciifc heats
16 Cp=1.006; //Specific heat at constant pressure in kJ/
    kg-K
17
18 //CALCULATIONS
19 x=(y-1)/y; //Ratio
20 T2s=T1*(rp^x); //Temperature at point 2s in K
21 T2=T1+((T2s-T1)/nC); //Temperature at point 2 in K
22 t2=T2-273; //Temperature at point 2 in oC
23 T4s=T3*((1/rp)^x); //Temperature at point 4s in K
24 T4=T3-((T3-T4s)*nT); //Temperature at point 4 in K
25 t4=T4-273; //Temperature at point 4 in oC
26 W=m*Cp*((T3-T4)-(T2-T1)); //Net power output in kW
27 nth=((T3-T4)-(T2-T1))/(T3-T2)*100; //Thermal
    efficiency
28 WR=W/(m*Cp*(T3-T4)); //Work ratio
29
30 //OUTPUT
31 mprintf('Net power output of the turbine is %3.0f kW
    \n Thermal efficiency of the plant is %3.0f
    percent \n Work ratio is %3.3f',W,nth,WR)
32
33
34
35
36
37
38 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.20 Percentage increase in cycle efficiency

```

1 //Chapter -2, Illustration 20, Page 79
2 //Title: Gas Power Cycles
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 P1=0.1; //Pressure at point 1 in MPa
9 T1=303; //Temperature at point 1 in K
10 T3=1173; //Temperature at point 3 in K
11 rp=6; //Pressure ratio
12 nC=0.8; //Compressor efficiency
13 nT=nC; //Turbine efficiency
14 e=0.75; //Regenerator effectiveness
15 y=1.4; //Ratio of specific heats
16 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
17
18 //CALCULATIONS
19 x=(y-1)/y; //Ratio
20 T2s=T1*(rp^x); //Temperature at point 2s in K
21 T4s=T3/(rp^x); //Temperature at point 4s in K
22 DTa=(T2s-T1)/nC; //Difference in temperatures at
    point 2 and 1 in K
23 DTb=(T3-T4s)*nT; //Difference in temperatures at
    point 3 and 4 in K
24 wT=Cp*DTb; //Turbine work in kJ/kg
25 wC=Cp*DTa; //Compressor work in kJ/kg
26 T2=DTa+T1; //Temperature at point 2 in K
27 q1=Cp*(T3-T2); //Heat supplied in kJ/kg
28 nth1=((wT-wC)/q1)*100; //Cycle efficiency without
    regenerator
29 T4=T3-DTb; //Temperature at point 4 in K
30 T5=T2+(e*(T4-T2)); //Temperature at point 5 in K
31 q2=Cp*(T3-T5); //Heat supplied with regenerator in kJ
    /kg

```

```

32 nth2=((wT-wC)/q2)*100; //Cycle efficiency with
    regenerator
33 p=((nth2-nth1)/nth1)*100; //Percentage increase due
    to regeneration
34
35 //OUTPUT
36 mprintf('Percentage increase in the cycle efficiency
    due to regeneration is %3.2f percent',p)
37
38
39
40
41
42
43
44
45 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.21 Velocity of air leaving nozzle

```

1 //Chapter –2, Illustration 21, Page 80
2 //Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=1; //Pressure at point 1 in atm
9 P3=5; //Pressure at point 3 in atm
10 T1=288; //Temperature at point 1 in K
11 T4=1143; //Temperature at point 4 in K
12 y=1.4; //Ratio of specific heats

```

```

13 Cp=1.005; // Specific heat at constant pressure in kJ/
    kg-K
14
15 //CALCULATIONS
16 rp=P3/P1; // Pressure ratio
17 x=(y-1)/y; // Ratio
18 T3=T1*(rp^x); // Temperature at point 3 in K
19 T5=T4-(T3-T1); // Temperature at point 5 in K
20 T6=T4/(rp^x); // Temperature at point 6 in K
21 C6=sqrt(2000*Cp*(T5-T6)); // Velocity of air leaving
    the nozzle in m/s
22
23 //OUTPUT
24 mprintf('Velocity of air leaving the nozzle is %3.1f
    m/s ', C6)
25
26
27
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 2.22 Turbine exit pressure and Velocity of exhaust gases and Propulsive efficiency

```

1 //Chapter -2, Illustration 22, Page 81
2 // Title: Gas Power Cycles
3 //
    =====
4 clc
5 clear
6

```

```

7 //INPUT DATA
8 C1=280;//Velocity of aircraft in m/s
9 P1=48;//Pressure at point 1 kPa
10 T1=260;//Temperature at point 1 in K
11 rp=13;//Pressure ratio
12 T4=1300;//Temperature at point 4 in K
13 Cp=1005;//Specific heat at constant pressure in J/kg
14 y=1.4;//Ratio of specific heats
15
16 //CALCULATIONS
17 x=(y-1)/y;//Ratio
18 T2=T1+((C1^2)/(2*Cp));//Temperature at point 2 in K
19 P2=P1*((T2/T1)^(1/x));//Pressure at point 2 in kPa
20 P3=rp*P2;//Pressure at point 3 in kPa
21 P4=P3;//Pressure at point 4 in kPa
22 T3=T2*(rp^x);//Temperature at point 3 in K
23 T5=T4-T3+T2;//Temperature at point 5 in K
24 P5=P4*((T5/T4)^(1/x));//Pressure at point 5 in kPa
25 P6=P1;//Pressure at point 6 in kPa
26 T6=T5*((P6/P5)^x);//Temperature at point 6 in K
27 C6=sqrt(2*Cp*(T5-T6));//Velocity of air at nozzle
    exit in m/s
28 W=(C6-C1)*C1;//Propulsive power in J/kg
29 Q=Cp*(T4-T3);//Total heat transfer rate in J/kg
30 nP=(W/Q)*100;//Propulsive efficiency
31
32 //OUTPUT
33 mprintf('Pressure at the turbine exit is %3.1f kPa \
    n Velocity of exhaust gases are %3.1f m/s \n
    Propulsive efficiency is %3.1f percent ',P5,C6,nP)
34
35
36
37
38
39
40
41

```

42

43

44

45 //=====END OF PROGRAM

=====

Chapter 3

Internal Combustion Engines

Scilab code Exa 3.1 Air standard efficiency and Indicated Power and Indicated thermal efficiency

```
1 //Chapter-3, Illustration 1, Page 139
2 //Title: Internal Combustion Engines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 d=200;//diameter of cylinder in mm
9 L=300;//stroke of cylinder in mm
10 Vc=1.73;//Clearance volume in litres
11 imep=650;//indicated mean effective pressure in kN/(
    m2)
12 g=6.2;//gas consumption in (m3)/h
13 CV=38.5;//Calorific value in MJ/(m3)
14 y=1.4;//Ratio of specific heats
15 N=150;//No. of firing cycles per minute
16
17 //CALCULATIONS
```

```

18 Vs=((3.1415/4)*(d^2)*L)*(10^-6); //Stroke volume in
    litres
19 Vt=Vs+Vc; //Total volume in litres
20 rv=(Vt/Vc); //Compression ratio
21 n=(1-(1/rv^(y-1)))*100; //Air standard efficiency
22 IP=imep*(Vs*10^-3)*(N/60); //Indicated power in kW
23 F=(g*CV*1000)/3600; //Fuel energy input in kW
24 nT=(IP/F)*100; //Indicated thermal efficiency
25
26 //OUTPUT
27 mprintf('Air Standard Efficiency is %3.1f percent \n
    Indicated Power is %3.1f kW \n Indicated thermal
    efficiency is %3.0f percent ',n,IP,nT)
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 3.2 Relative efficiency of engine

```

1 //Chapter-3, Illustration 2, Page 140
2 //Title: Internal Combustion Engines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 Vs=0.0008; //Swept volume in m^3
9 Vc=0.00015; //Clearance volume in m^3
10 CV=38; //Calorific value in MJ/(m^3)
11 v=0.45; //volume in m^3

```

```

12 IP=81.5; //Indicated power in kW
13 y=1.4; //Ratio of specific heats
14
15 //CALCULATIONS
16 rv=(Vs+Vc)/Vc; //Compression ratio
17 n=(1-(1/rv^(y-1))); //Air standard efficiency
18 Ps=(v*CV*1000)/60; //Power supplied in kW
19 nact=IP/Ps; //Actual efficiency
20 nr=(nact/n)*100; //Relative efficiency
21
22 //OUTPUT
23 mprintf('Relative Efficiency is %3.2f percent ',nr)
24
25
26
27 //=====END OF PROGRAM
=====

```

Scilab code Exa 3.3 Indicated power and Brake power and and Brake thermal efficiency and Brake mean effective pressure and Mechanical efficiency and Brake specific fuel consumption

```

1 //Chapter-3, Illustration 3, Page 141
2 //Title: Internal Combustion Engines
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 n=6; //No. of cylinders
9 d=0.61; //Diameter in m
10 L=1.25; //Stroke in m
11 N=2; //No. of revolutions per second

```

```

12 m=340; //mass of fuel oil in kg
13 CV=44200; //Calorific value in kJ/kg
14 T=108; //Torque in kN-m
15 imep=775; //Indicated mean efective pressure in kN/(m
    ^2)
16
17 //CALCULATIONS
18 IP=(imep*L*3.1415*(d^2)*N)/(8); //Indicated power in
    kW
19 TotalIP=(n*IP); //Total indicated power in kW
20 BP=(2*3.1415*N*T); //Brake power in kW
21 PI=(m*CV)/3600; //Power input in kW
22 nB=(BP/PI)*100; //Brake thermal efficiency
23 bmep=(BP*8)/(n*L*3.1415*(d^2)*2); //Brake mean
    effective pressure in kN/(m^2)
24 nM=(BP/TotalIP)*100; //Mechanical efficiency
25 bsfc=m/BP; //Brake specific fuel consumption in kg/
    kWh
26
27 //OUTPUT
28 mprintf('Total Indicated Power is %3.1f kW \n Brake
    Power is %3.1f kW \n Brake thermal efficiency is
    %3.1f percent \n Brake mean effective pressure is
    %3.1f kN/(m^2) \n Mechanical efficiency is %3.1f
    percent \n Brake specific fuel consumption is %3
    .3f kg/kW.h', TotalIP, BP, nB, bmep, nM, bsfc)
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 3.4 Indicated power and Brake output and Mechanical efficiency and Overall energy balance

```

1 //Chapter-3, Illustration 4, Page 142
2 //Title: Internal Combustion Engines
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 Hm=21; //Mean height of indicator diagram in mm
9 isn=27; //indicator spring number in kN/(m^2)/mm
10 Vs=14; //Swept volume in litres
11 N=6.6; //Speed of engine in rev/s
12 Pe=77; //Effective brake load in kg
13 Re=0.7; //Effective vrake radius in m
14 mf=0.002; //fuel consumed in kg/s
15 CV=44000; //Calorific value of fuel in kJ/kg
16 mc=0.15; //cooling water circulation in kg/s
17 Ti=311; //cooling water inlet temperature in K
18 To=344; //cooling water outlet temperature in K
19 C=4.18; //specific heat capacity of water in kJ/kg-K
20 Ee=33.6; //Energy to exhaust gases in kJ/s
21 g=9.81; //Acceleration due to geravity in m/(s^2)
22
23 //CALCULATIONS
24 imep=isn*Hm; //Indicated mean efective pressure in kN
    /(m^2)
25 IP=(imep*Vs*N)/(2000); //Indicated Power in kW
26 BP=(2*3.1415*N*g*Pe*Re)/1000; //Brake Power in kW
27 nM=(BP/IP)*100; //Mechanical efficiency
28 Ef=mf*CV; //Energy from fuel in kJ/s
29 Ec=mc*C*(To-Ti); //Energy to cooling water in kJ/s
30 Es=Ef-(BP+Ec+Ee); //Energy to surroundings in kJ/s
31 p=(BP*100)/Ef; //Energy to BP in %
32 q=(Ec*100)/Ef; //Energy to coolant in %
33 r=(Ee*100)/Ef; //Energy to exhaust in %
34 w=(Es*100)/Ef; //Energy to surroundings in %
35

```

```

36 //OUTPUT
37 mprintf('Indicated Power is %3.1f kW \n Brake Power
is %3.0f kW \n Mechanical Efficiency is %3.0f
percent \n \nENERGY BALANCE                                kJ
/s      Percentage \nEnergy from fuel
              %3.0f          100\nEnergy to BP
              %3.0f          %3.0f\nEnergy
to coolant          %3.01f          %3.1f\
nEnergy to exhaust          %3.1f
%3.1f\nEnergy to surroundings , etc          %3.1f
              %3.1f ', IP , BP , nM , Ef , BP , p , Ec , q , Ee , r , Es , w)
38
39
40
41
42 //=====END OF PROGRAM
=====

```

Scilab code Exa 3.5 Brake power and Brake specific fuel consumption and Indicated thermal efficiency and Energy balance

```

1 //Chapter –3, Illustration 5, Page 143
2 //Title: Internal Combustion Engines
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 t=30;//duration of trial in minutes
9 N=1750;//speed in rpm
10 T=330;//brake torque in Nm
11 m=9.35;//mass of fuel in kg
12 CV=42300;//Calorific value in kJ/kg

```

```

13 mj=483; //jacket cooling water circulation in kg
14 Ti=290; //inlet temperature in K
15 T0=350; //outlet temperature in K
16 ma=182; //air consumption in kg
17 Te=759; //exhaust temperature in K
18 Ta=256; //atmospheric temperature in K
19 nM=0.83; //Mechanical efficiency
20 ms=1.25; //mean specific heat capacity of exhaust gas
    in kJ/kg-K
21 Cw=4.18; //specific heat capacity of water in kJ/kg-K
22
23 //CALCULATIONS
24 BP=(2*3.1415*T*N)/(60*1000); //Brake power in kW
25 sfc=(m*2)/BP; //specific fuel consumption in kg/kWh
26 IP=BP/nM; //Indicated power in kW
27 nIT=((IP*3600)/(m*CV*2))*100; //Indicated thermal
    efficiency
28 Ef=(m*CV)/t; //Energy from fuel in kJ/min
29 EBP=BP*60; //Energy to BP in kJ/min
30 Ec=(mj*Cw*(T0-Ti))/t; //Energy to cooling water in kJ
    /min
31 Ee=((ma+m)*ms*(Te-Ti))/30; //Energy to exhaust in kJ/
    min
32 Es=Ef-(EBP+Ec+Ee); //Energy to surroundings in kJ/min
33
34 //OUTPUT
35 mprintf('Brake power is %3.1f kW \n Specific fuel
    consumption is %3.3f kg/kWh \n Indicated thermal
    efficiency is %3.1f percent \n Energy from fuel
    is %3.0f kJ/min \n Energy to BP is %3.0f kJ/min \
    n Energy to cooling water is %3.0f kJ/min \n
    Energy to exhaust is %3.0f kJ/min \n Energy to
    surroundings is %3.0f kJ/min ',BP,sfc,nIT,Ef,EBP,
    Ec,Ee,Es)
36
37
38
39

```

```

40
41
42 //=====END OF PROGRAM
=====

```

Scilab code Exa 3.6 Indicated power and Mechanical efficiency of engine

```

1 //Chapter-3, Illustration 6, Page 144
2 //Title: Internal Combustion Engines
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 BP0=12; //Brake Power output in kW
9 BP1=40.5; //Brake Power in trial 1 in kW
10 BP2=40.2; //Brake Power in trial 2 in kW
11 BP3=40.1; //Brake Power in trial 3 in kW
12 BP4=40.6; //Brake Power in trial 4 in kW
13 BP5=40.7; //Brake Power in trial 5 in kW
14 BP6=40.0; //Brake Power in trial 6 in kW
15
16 //CALCULATIONS
17 BPALL=BP0+BP6; //Total Brake Power in kW
18 IP1=BPALL-BP1; //Indicated Power in trial 1 in kW
19 IP2=BPALL-BP2; //Indicated Power in trial 2 in kW
20 IP3=BPALL-BP3; //Indicated Power in trial 3 in kW
21 IP4=BPALL-BP4; //Indicated Power in trial 4 in kW
22 IP5=BPALL-BP5; //Indicated Power in trial 5 in kW
23 IP6=BPALL-BP6; //Indicated Power in trial 6 in kW
24 IPALL=IP1+IP2+IP3+IP4+IP5+IP6; //Total Indicated
    Power in kW
25 nM=(BPALL/IPALL)*100; //Mechanical efficiency

```



```

26
27 //OUTPUT
28 mprintf('Indicated Power of the engine is %3.1f kW \
    n Mechanical efficiency of the engine is %3.1f
    percent ',IPALL,nM)
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 3.7 Engine dimensions and Brake power

```

1 //Chapter-3, Illustration 7, Page 145
2 //Title: Internal Combustion Engines
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 n=2;//No. of cylinders
9 N=4000;//speed of engine in rpm
10 nV=0.77;//Volumetric efficiency
11 nM=0.75;//Mechanical efficiency
12 m=10;//fuel consumed in lit/h
13 g=0.73;//specific gravity of fuel
14 Raf=18;//air-fuel ratio
15 Np=600;//piston speed in m/min
16 imep=5;//Indicated mean efective pressure in bar
17 R=281;//Universal gas constant in J/kg-K
18 T=288;//Standard temperature in K
19 P=1.013;//Standard pressure in bar
20

```

```

21 //CALCULATIONS
22 L=Np/(2*N); //Piston stroke in m
23 mf=m*g; //mass of fuel in kg/h
24 ma=mf*Raf; //mass of air required in kg/h
25 Va=(ma*R*T)/(P*60*(10^5)); //volume of air required
    in (m^3)/min
26 D=sqrt((2*Va)/(nV*L*N*3.1415)); //Diameter in m
27 IP=(2*imep*100*L*3.1415*(D^2)*N)/(4*60); //Indicated
    Power in kW
28 BP=nV*IP; //Brake Power in kW
29
30 //OUTPUT
31 mprintf('Piston Stroke is %3.3f m \n Bore diameter
    is %3.4f m \n Brake power is %3.1f kW',L,D,BP)
32
33
34
35 //=====END OF PROGRAM
    =====

```

Chapter 4

Steam nozzles and Steam turbines

Scilab code Exa 4.1 Throat area and Exit area and Mach number at exit

```
1 //Chapter-4, Illustration 1, Page 161
2 //Title: Steam Nozzles and Steam Turbines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 P1=3.5; //Pressure at entry in MN/(m^2)
9 T1=773; //Temperature at entry in K
10 P2=0.7; //Pressure at exit in MN/(m^2)
11 ma=1.3; //mass flow rate of air in kg/s
12 y=1.4; //Ratio of specific heats
13 R=0.287; //Universal gas constant in KJ/Kg-K
14
15 //CALCULATIONS
16 c=y/(y-1); //Ratio
17 Pt=((2/(y+1))^c)*P1; //Throat pressure in MN/(m^2)
```

```

18 v1=(R*T1)/(P1*1000); // Specific volume at entry in (m
    ^3)/kg
19 Ct=((2*c*P1*v1*(1-((Pt/P1)^(1/c))))^0.5)*1000; //
    Velocity at throat in m/s
20 vt=v1*((P1/Pt)^(1/y)); // Specific volume at throat in
    (m^3)/kg
21 At=((ma*vt)/Ct)*(10^6); // Area of throat in (mm^2)
22 C2=((2*c*P1*v1*(1-((P2/P1)^(1/c))))^0.5)*1000; //
    Velocity at exit in m/s
23 v2=v1*((P1/P2)^(1/y)); // Specific volume at exit in (
    m^3)/kg
24 A2=((ma*v2)/C2)*(10^6); // Area of exit in (mm^2)
25 M=C2/Ct; // Mach number at exit
26
27 //OUTPUT
28 mprintf('Throat area is %3.0f (mm^2) \n Exit area is
    %3.0f (mm^2) \n Mach number at exit is %3.2f',At
    ,A2,M)
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.2 Increase in pressure and temperature and internal energy

```

1 //Chapter-4, Illustration 2, Page 163
2 //Title: Steam Nozzles and Steam Turbines
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  T1=273; //Temperature at section 1 in K
9  P1=140; //Pressure at section 1 in KN/(m^2)
10 v1=900; //Velocity at section 1 in m/s
11 v2=300; //Velocity at section 2 in m/s
12 Cp=1.006; //Specific heat at constant pressure in kJ/
    kg-K
13 Cv=0.717; //Specific heat at constant volume in kJ/kg
    -K
14 y=1.4; //Ratio of specific heats
15
16 //CALCULATIONS
17 c=y/(y-1); //Ratio
18 R=Cp-Cv; //Universal gas constant in KJ/Kg-K
19 T2=T1-(((v2)^2-(v1)^2)/(2000*c*R)); //Temperature at
    section 2 in K
20 DT=T2-T1; //Increase in temperature in K
21 P2=P1*((T2/T1)^c); //Pressure at section 2 in KN/(m
    ^2)
22 DP=(P2-P1)/1000; //Increase in pressure in MN/(m^2)
23 IE=Cv*(T2-T1); //Increase in internal energy in kJ/kg
24
25 //OUTPUT
26 mprintf('Increase in temperature is %3.0f K \n
    Increase in pressure is %3.2f MN/(m^2) \n
    Increase in internal energy is %3.0f kJ/kg',DT,DP
    ,IE)
27
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.3 Throat area and exit area and Degree of undercooling at exit

```

1 //Chapter-4, Illustration 3, Page 163
2 //Title: Steam Nozzles and Steam Turbines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 P1=2;//Pressure at entry in MN/(m^2)
9 T1=598;//Temperature at entry in K
10 P2=0.36;//Pressure at exit in MN/(m^2)
11 m=7.5;//mass flow rate of steam in kg/s
12 n=1.3;//Adiabatic gas constant
13 v1=0.132;//Volume at entry in (m^3)/kg from steam
    table
14 Ts=412.9;//Saturation temperature in K
15
16 //CALCULATIONS
17 c=n/(n-1);//Ratio
18 Pt=((2/(n+1))^c)*P1;//Throat pressure in MN/(m^2)
19 Ct=((2*c*P1*v1*(1-((Pt/P1)^(1/c))))^0.5)*1000;//
    Velocity at throat in m/s
20 vt=v1*((P1/Pt)^(1/n));//Specific volume at throat in
    (m^3)/kg
21 At=((m*vt)/Ct)*(10^6);//Area of throat in (mm^2)
22 C2=((2*c*P1*v1*(1-((P2/P1)^(1/c))))^0.5)*1000;//
    Velocity at exit in m/s
23 v2=v1*((P1/P2)^(1/n));//Specific volume at exit in (
    m^3)/kg
24 A2=((m*v2)/C2)*(10^6);//Area of exit in (mm^2)

```

```

25 T2=T1*((P2/P1)^(1/c)); //Temperature at exit in K
26 D=Ts-T2; //Degree of undercooling at exit in K
27
28 //OUTPUT
29 mprintf('Throat area is %3.0f (mm^2) \n Exit area is
          %3.0f (mm^2) \n Degree of undercooling at exit
          is %3.1f K',At,A2,D)
30
31
32
33
34 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.4 Throat and exit velocities and Throat and exit areas

```

1 //Chapter-4, Illustration 4, Page 165
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=2.2; //Pressure at entry in MN/(m^2)
9 T1=533; //Temperature at entry in K
10 P2=0.4; //Pressure at exit in MN/(m^2)
11 m=11; //mass flow rate of steam in kg/s
12 n=0.85; //Efficiency of expansion
13 h1=2940; //Enthalpy at entrance in kJ/kg from Moiller
    chart
14 ht=2790; //Enthalpy at throat in kJ/kg from Moiller
    chart
15 h2s=2590; //Enthalpy below exit level in kJ/kg from

```

```

    Moiller chart
16 vt=0.16; //Throat volume in (m^3)/kg
17 v2=0.44; //Volume at exit in (m^3)/kg
18
19 //CALCULATIONS
20 Ct=(2000*(h1-ht))^0.5; //Throat velocity in m/s
21 h2=ht-(0.85*(ht-h2s)); //Enthalpy at exit in kJ/kg
22 C2=(2000*(h1-h2))^0.5; //Exit velocity in m/s
23 At=((m*vt)/Ct)*(10^6); //Area of throat in (mm^2)
24 A2=((m*v2)/C2)*(10^6); //Area of exit in (mm^2)
25
26 //OUTPUT
27 mprintf('Throat velocity is %3.0f m/s \n Exit
    velocity is %3.0f m/s \n Throat area is %3.0f (mm
    ^2) \n Exit area is %3.0f (mm^2) \n',Ct,C2,At,A2)
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.5 Nozzle dimensions and Degree of undercooling and supersaturation and Loss in available heat and Increase in entropy and Ratio of mass flow rate

```

1 //Chapter-4, Illustration 5, Page 166
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA

```



```

8 P1=35; //Pressure at entry in bar
9 T1=573; //Temperature at entry in K
10 P2=8; //Pressure at exit in bar
11 Ts=443.4; //Saturation temperature in K
12 Ps=3.1; //Saturation pressure in bar
13 m=5.2; //mass flow rate of steam in kg/s
14 n=1.3; //Adiabatic gas constant
15 v1=0.06842; //Specific volume at entry in (m^3)/kg
    from steam table
16 v3=0.2292; //Specific volume at exit in (m^3)/kg from
    steam table
17 h1=2979; //Enthalpy in kJ/kg from Moiller chart
18 h3=2673.3; //Enthalpy in kJ/kg from Moiller chart
19
20 //CALCULATIONS
21 c=n/(n-1); //Ratio
22 C2=((2*c*P1*(10^5)*v1*(1-((P2/P1)^(1/c))))^0.5); //
    Velocity at exit in m/s
23 v2=v1*((P1/P2)^(1/n)); //Specific volume at exit in (
    m^3)/kg
24 A2=((m*v2)/C2)*(10^4); //Area of exit in (cm^2)
25 a=((A2/18)^0.5)*10; //Length in mm
26 b=3*a; //Breadth in mm
27 T2=T1*((P2/P1)^(1/c)); //Temperature at exit in K
28 D=Ts-T2; //Degree of undercooling in K
29 Ds=P2/Ps; //Degree of supersaturation
30 hI=h1-h3; //Isentropic enthalpy drop in kJ/kg
31 ha=(C2^2)/2000; //Actual enthalpy drop in kJ/kg
32 QL=hI-ha; //Loss in available heat in kJ/kg
33 DS=QL/Ts; //Increase in entropy in kJ/kg-K
34 C3=(2000*(h1-h3))^0.5; //Exit velocity from nozzle
35 mf=((A2*C3*(10^-4))/v3); //Mass flow rate in kg/s
36 Rm=m/mf; //Ratio of mass rate
37
38 //OUTPUT
39 printf('Cross section of nozzle is %3.1f mm * %3.1f
    mm \n Degree of undercooling is %3.1f K and
    Degree of supersaturation is %3.2f \n Loss in

```

```

    available heat drop due to irreversibility is %3
    .2f kJ/kg \n Increase in entropy is %3.5f kJ/kg-K
    \n Ratio of mass flow rate with metastable
    expansion to the thermal expansion is %3.3f',b,a,
    D,Ds,QL,DS,Rm)
40
41
42
43
44
45 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.6 Nozzle efficiency and Exit area and Throat velocity

```

1 //Chapter-4, Illustration 6, Page 169
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 m=14; //Mass flow rate of steam in kg/s
9 P1=3; //Pressure of Steam in MN/(m2)
10 T1=300; //Steam temperature in oC
11 h1=2990; //Enthalpy at point 1 in kJ/kg
12 h2s=2630; //Enthalpy at point 2s in kJ/kg
13 ht=2850; //Enthalpy at point t in kJ/kg
14 n=1.3; //Adiabatic gas constant
15 C2=800; //Exit velocity in m/s
16 v2=0.4; //Specific volume at exit in (m3)/kg
17
18 //CALCULATIONS

```

```

19 x=n/(n-1); //Ratio
20 Pt=((2/(n+1))^x)*P1; //Temperature at point t in MN/(
    m^2)
21 h2=h1-((C2^2)/2000); //Exit enthalpy in kJ/kg
22 nN=((h1-h2)/(h1-h2s))*100; //Nozzle efficiency
23 A2=((m*v2)/C2)*(10^6); //Exit area in (mm^2)
24 Ct=sqrt(2000*(h1-ht)); //Throat velocity in m/s
25
26 //OUTPUT
27 mprintf('Nozzle efficiency is %3.1f percent \n Exit
    area is %3.0f (mm^2) \n Throat velocity is %3.0f
    m/s ',nN,A2,Ct)
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.7 Areas at throat and exit and Steam quality at exit

```

1 //Chapter-4, Illustration 7, Page 170
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=10; //Pressure at point 1 in bar
9 P2=0.5; //Pressure at point 2 in bar
10 h1=3050; //Enthalpy at point 1 in kJ/kg
11 h2s=2480; //Enthalpy at point 2s in kJ/kg
12 ht=2910; //Enthalpy at throat in kJ/kg

```

```

13 n=1.3; //Adiabatic gas constant
14 r=0.1; //Total available heat drop
15 v1=0.258; //Specific volume at point 1 in (m^3)/kg
16 h2f=340.6; //Enthalpy for exit pressure from steam
    tables in kJ/kg
17 hfg=2305.4; //Enthalpy for exit pressure from steam
    tables in kJ/kg
18 m=0.5; //Mass flow rate in kg/s
19
20 //CALCULATIONS
21 x=n/(n-1); //Ratio
22 Pt=((2/(n+1))^x)*P1; //Temperature at throat in bar
23 h2=h2s+(r*(h1-h2s)); //Enthalpy at point 2 in kJ/kg
24 vt=((P1/Pt)^(1/n))*v1; //Specific volume at throat in
    (m^3)/kg
25 v2=((P1/P2)^(1/n))*v1; //Specific volume at point 2
    in (m^3)/kg
26 Ct=sqrt(2000*(h1-h2)); //Throat velocity in m/s
27 At=((m*vt)/Ct)*(10^6); //Throat area in (mm^2)
28 C2=sqrt(2000*(h1-h2)); //Exit velocity in m/s
29 A2=((m*v2)/C2)*(10^6); //Exit area in (mm^2)
30 x2=((h2-h2f)/hfg)*100; //Steam quality at exit
31
32 //OUTPUT
33 mprintf('Throat area is %3.0f (mm^2) \n Exit area is
    %3.0f (mm^2) \n Steam quality at exit is %3.0f
    percent ',At,A2,x2)
34
35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.8 Maximum discharge and Area of nozzle at exit

```
1 //Chapter-4, Illustration 8, Page 171
2 //Title: Steam Nozzles and Steam Turbines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 P1=3.5; //Dry saturated steam in bar
9 P2=1.1; //Exit pressure in bar
10 At=4.4; //Throat area in cm2
11 h1=2731.6; //Enthalpy at P1 in kJ/kg
12 v1=0.52397; //Specific volume at P1 in m3/kg
13 n=1.135; //Adiabatic gas constant
14 ht=2640; //Enthalpy at Pt in kJ/kg
15 vt=0.85; //Specific volume at throat in m3/kg
16 h2=2520; //Enthalpy at P2 in kJ/kg
17 v2=1.45; //Specific volume at P2 in m3/kg
18
19 //CALCULATIONS
20 x=n/(n-1); //Ratio
21 Pt=((2/(n+1))x)*P1; //Throat pressure in bar
22 Ct=sqrt(2000*(h1-ht)); //Throat velocity in m/s
23 mmax=((At*Ct*(10-4))/vt)*60; //Maximum discharge in
    kg/min
24 C2=sqrt(2000*(h1-h2)); //Exit velocity in m/s
25 A2=((mmax*v2)/(C2*60))*(106); //Exit area in mm2
26
27 //OUTPUT
28 mprintf('Maximum discharge is %3.3f kg/min \n Exit
    area is %3.2f mm2',mmax,A2)
```

```

29
30
31
32
33 //=====END OF PROGRAM
=====

```

Scilab code Exa 4.9 Type of nozzle and Minimum area of nozzle

```

1 //Chapter-4, Illustration 9, Page 172
2 //Title: Steam Nozzles and Steam Turbines
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=10; //Pressure at point 1 in bar
9 T1=200; //Temperature at point 1 in oC
10 P2=5; //Pressure at point 2 in bar
11 n=1.3; //Adiabatic gas constant
12 h1=2830; //Enthalpy at P1 in kJ/kg
13 ht=2710; //Enthalpy at point Pt in kJ/kg
14 vt=0.35; //Specific volume at Pt in m^3/kg
15 m=3; //Nozzle flow in kg/s
16
17 //CALCULATIONS
18 x=n/(n-1); //Ratio
19 Pt=((2/(n+1))^x)*P1; //Throat pressure in bar
20 Ct=sqrt(2000*(h1-ht)); //Throat velocity in m/s
21 At=(m*vt)/Ct; //Throat area in m^2
22
23 //OUTPUT
24 printf('Since throat pressure is greater than exit

```

```

    pressure ,nozzle used is convergent–divergent
    nozzle \n Minimum area of nozzle required is %3.5
    f m^2 ',At)
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.10 Throat velocity and Mass flow rate of steam

```

1 //Chapter–4, Illustration 10, Page 173
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=10.5; //Pressure at point 1 in bar
9 x1=0.95; //Dryness fraction
10 n=1.135; //Adiabatic gas constant
11 P2=0.85; //Pressure at point 2 in bar
12 vg=0.185; //Specific volume in m^3/kg
13
14
15 //CALCULATIONS
16 c=n/(n-1); // Ratio
17 Pt=((2/(n+1))^c)*P1; //Throat pressure in MN/(m^2)
18 v1=x1*vg; //Specific volume at point 1 in m^3/kg
19 Ct=sqrt((2*n*P1*v1*(10^5)/(n+1))); //Velocity at
    throat in m/s

```

```

20 vt=((P1/Pt)*(v1^n))^(1/1.135); // Specific volume at
    throat in m^3/kg
21 m=Ct/vt; // Mass flow rate per unit throat area in kg
    /(m^2)
22
23 //OUTPUT
24 mprintf('Throat velocity is %3.2f m/s \n Mass flow
    rate of steam is %3.2f kg/(m^2)',Ct,m)
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.11 Degree of undercooling and supersaturation

```

1 //Chapter-4, Illustration 11, Page 174
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=10; //Pressure at point 1 in bar
9 T1=452.9; //Temperature at point 1 in K
10 P2=4; //Pressure at point 2 in bar
11 n=1.3; //Adiabatic gas constant
12 Ps=0.803; //Saturation pressure at T2 in bar
13 Ts=143.6; //Saturation temperature at P2 in oC
14 //CALCULATIONS
15 x=(n-1)/n; //Ratio

```



```

16 T2=((P2/P1)^x)*T1;//Temperature at point 2 in K
17 Ds=P2/Ps;//Degree of supersaturation
18 Du=Ts-(T2-273);//Degree of undercooling
19
20 //OUTPUT
21 mprintf('Degree of supersaturation is %3.2f \n
          Degree of undercooling %3.0f oC',Ds,Du)
22
23
24
25
26
27
28
29 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.12 Quantity of steam used and Exit velocity of steam

```

1 //Chapter-4, Illustration 12, Page 174
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=9;//Pressure at point 1 in bar
9 P2=1;//Pressure at point 2 in bar
10 Dt=0.0025;//Throat diameter in m
11 nN=0.9;//Nozzle efficiency
12 n=1.135;//Adiabatic gas constant
13 h1=2770;//Enthalpy at point 1 in kJ/kg
14 ht=2670;//Throat enthalpy in kJ/kg

```

```

15 h3=2400; //Enthlapy at point 2 in kJ/kg
16 x2=0.96; //Dryness fraction 2
17 vg2=0.361; //Specific volume in m^3/kg
18
19 //CALCULATIONS
20 x=n/(n-1); //Ratio
21 Pt=((2/(n+1))^x)*P1; //Throat pressure in bar
22 Ct=sqrt(2000*(h1-h3)*nN); //Throat velocity in m/s
23 At=(3.147*2*(Dt^2))/4; //Throat area in m^2
24 vt=x2*vg2; //Specific volume at throat in m^3/kg
25 m=(At*Ct)/vt; //Mass flow rate of steam in kg/s
26 hact=nN*(h1-h3); //Actual enthalpy drop in kJ/kg
27 C2=sqrt(2000*hact); //Exit velocity of steam in m/s
28
29 //OUTPUT
30 mprintf('Quantity of steam used per second is %3.3f
          kg/s \n Exit velocity of steam is %3.2f m/s',m,C2
          )
31
32
33
34
35 //=====END OF PROGRAM
=====

```

Scilab code Exa 4.13 Blade angles and Tangential force and Axial thrust and Diagram power and Diagram efficiency

```

1 //Chapter-4, Illustration 13, Page 202
2 //Title: Steam Nozzles and Steam Turbines
3 //
=====
4 clc
5 clear

```

```

6
7 //INPUT DATA
8 C1=1000; //Steam velocity in m/s
9 a1=20; //Nozzle angle in degrees
10 U=400; //Mean blade speed in m/s
11 m=0.75; //Mass flow rate of steam in kg/s
12 b1=33; //Blade angle at inlet from the velocity
    triangle in degrees
13 b2=b1; //Blade angle at exit from the velocity
    triangle in degrees
14 Cx=1120; //Change in whirl velocity from the velocity
    triangle in m/s
15 Ca=0; //Change in axial velocity from the velocity
    triangle in m/s
16
17 //CALCULATIONS
18 Fx=m*Cx; //Tangential force on blades in N
19 Fy=m*Ca; //Axial thrust in N
20 W=(m*Cx*U)/1000; //Diagram power in kW
21 ndia=((2*U*Cx)/(C1^2))*100; //Diagram efficiency
22
23 //OUTPUT
24 mprintf('Blade angles are %3.0f degrees,%3.0f
    degrees \n Tangential force on blades is %3.0f N
    \n Axial thrust is %3.0f \n Diagram power is %3.0
    f kW \n Diagram efficiency %3.1f percent ',b1,b2,
    Fx,Fy,W,ndia)
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.14 Power developed and Blade efficiency and Steam consumption

```
1 //Chapter-4, Illustration 14, Page 203
2 //Title: Steam Nozzles and Steam Turbines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 D=2.5; //Mean diameter of blade ring in m
9 N=3000; //Speed in rpm
10 a1=20; //Nozzle angle in degrees
11 r=0.4; //Ratio blade velocity to steam velocity
12 Wr=0.8; //Blade friction factor
13 m=10; //Steam flow in kg/s
14 x=3; //Sum in blade angles in degrees
15 b1=32.5; //Blade angle at inlet from the velocity
    triangle in degrees
16 W1=626.7; //Relative velocity at inlet from the
    velocity triangle in m/s
17 Cx=967; //Change in whirl velocity from the velocity
    triangle in m/s
18
19 //CALCULATIONS
20 U=(3.147*D*N)/60; //Blade velocity in m/s
21 C1=U/r; //Steam velocity in m/s
22 b2=b1-x; //Blade angle at exit in degrees
23 W2=Wr*W1; //Relative velocity at outlet from the
    velocity triangle in m/s
24 W=(m*Cx*U)/1000; //Power developed in kW
25 ndia=((2*U*Cx)/(C1^2))*100; //Blade efficiency
26 sc=(m*3600)/W; //Steam consumption in kg/kWh
27
28 //OUTPUT
29 mprintf('Power developed is %3.0f kW \n Blade
```

```

    efficiency is %3.1f percent \n Steam consumed is
    %3.2f kg/kWh',W,ndia,sc)
30
31
32
33
34 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.15 Blading efficiency and Blade velocity coefficient

```

1 //Chapter-4, Illustration 15, Page 204
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 m=3; //Mass flow rate of steam in kg/s
9 C1=425; //Steam velocity in m/s
10 r=0.4; //Ratio of blade speed to jet speed
11 W=170; //Stage output in kW
12 IL=15; //Internal losses in kW
13 a1=16; //Nozzle angle in degrees
14 b2=17; //Blade angle at exit in degrees
15 W1=265; //Relative velocity at inlet from the
    velocity triangle in m/s
16 W2=130; //Relative velocity at outlet from the
    velocity triangle in m/s
17
18 //CALCULATIONS
19 U=C1*r; //Blade speed in m/s
20 P=(W+IL)*1000; //Total power developed in W

```

```

21 Cx=P/(m*W); //Change in whirl velocity in m/s
22 ndia=((2*U*Cx)/(C1^2))*100; //Blading efficiency
23 Wr=W2/W1; //Blade velocity co-efficient
24
25 //OUTPUT
26 mprintf('Blading efficiency is %3.1f percent \n
          Blade velocity co-efficient is %3.2f',ndia,Wr)
27
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.16 Blade angles and Turbine power

```

1 //Chapter-4, Illustration 16, Page 205
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 C1=375; //Steam velocity in m/s
9 a1=20; //Nozzle angle
10 U=165; //Blade speed in m/s
11 m=1; //Mass flow rate of steam in kg/s
12 Wr=0.85; //Blade friction factor
13 Ca1=130; //Axial velocity at inlet from the velocity
          triangle in m/s
14 Ca2=Ca1; //Axial velocity at outlet in m/s
15 W1=230; //Relative velocity at inlet from the
          velocity triangle in m/s

```

```

16 Cx=320;//Change in whirl velocity from the velocity
    triangle in m/s
17
18 //CALCULATIONS
19 b2=41;//Blade angle at exit from the velocity
    triangle in degrees
20 b1=34;//Blade angle at exit from the velocity
    triangle in degrees
21 W=(m*Cx*U)/1000;//Power developed by turbine in kW
22
23 //OUTPUT
24 mprintf('Blade angles assumed are %3.0f degrees,%3.0
    f degrees \n Power developed by turbine is %3.1f
    kW',b1,b2,W)
25
26
27
28
29 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.17 Nozzle angle and Blade angle at entry and exit

```

1 //Chapter-4, Illustration 17, Page 206
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 m=2;//Mass flow rate of steam in kg/s
9 W=130;//Turbine power in kW
10 U=175;//Blade velocity in m/s

```

```

11 C1=400; //Steam velocity in m/s
12 Wr=0.9; //Blade friction factor
13 W1=240; //Realtive velocity at inlet from the
    velocity triangle in m/s
14
15 //CALCULATIONS
16 Cx1=(W*1000)/(m*U); //Whirl velocity at inlet in m/s
17 W2=Wr*W1; //Realtive velocity at outlet from the
    velocity triangle in m/s
18 a1=19; //Nozzle angle from the velocity triangle in
    degrees
19 b1=33; //Blade angle at inlet from the velocity
    triangle in degrees
20 b2=36; //Blade angle at outlet from the velocity
    triangle in degrees
21
22 //OUTPUT
23 mprintf('Nozzle angle is %3.0f degrees \n Blade
    angles are %3.0f degrees,%3.0f degrees',a1,b1,b2)
24
25
26
27
28
29 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.18 Diagram efficiency

```

1 //Chapter-4, Illustration 18, Page 207
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc

```



```

5  clear
6
7  //INPUT DATA
8  U=150; //Blade speed in m/s
9  m=3; //Mass flow rate of steam in kg/s
10 P=10.5; //Pressure in bar
11 r=0.21; //Ratio blade velocity to steam velocity
12 a1=16; //Nozzle angle in first stage in degrees
13 b2=20; //Blade angle at exit in first stage in
    degrees
14 a3=24; //Nozzle angle in second stage in degrees
15 b4=32; //Blade angle at exit in second stage in
    degrees
16 Wr=0.79; //Blade friction factor for first stage
17 Wr2=0.88; //Blade friction factor for second stage
18 Cr=0.83; //Blade velocity coefficient
19 W1=570; //Relative velocity at inlet from the
    velocity triangle for first stage in m/s
20 C2=375; //Velocity in m/s
21 W3=185; //Relative velocity at inlet from the
    velocity triangle for second stage in m/s
22
23 //CALCULATIONS
24 C1=U/r; //Steam speed at exit in m/s
25 W2=Wr*W1; //Relative velocity at outlet for first
    stage in m/s
26 C3=Cr*C2; //Steam velocity at inlet for second stage
    in m/s
27 W4=Wr2*W3; //Relative velocity at exit for second
    stage in m/s
28 DW1=W1+W2; //Change in relative velocity for first
    stage in m/s
29 DW2=275; //Change in relative velocity from the
    velocity triangle for second stage in m/s
30 ndia=((2*U*(DW1+DW2))/(C1^2))*100; //Diagram
    efficiency
31
32 //OUTPUT

```

```

33 mprintf('Diagram efficiency is %3.1f percent ',ndia)
34
35
36
37
38 //=====END OF PROGRAM
=====

```

Scilab code Exa 4.19 Blade speed and Blade tip angles and Diagram efficiency

```

1 //Chapter-4, Illustration 19, Page 208
2 //Title: Steam Nozzles and Steam Turbines
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 b1=30;//Blade angle at inlet in first stage in
   degrees
9 b2=30;//Blade angle at exit in first stage in
   degrees
10 b3=30;//Blade angle at inlet in second stage in
   degrees
11 b4=30;//Blade angle at exit in second stage in
   degrees
12 t1=240;//Temperature at entry in oC
13 P1=11.5;//Pressure at entry in bar
14 P2=5;//Pressure in wheel chamber in bar
15 v1=10;//Loss in velocity in percent
16 h=155;//Enthalpy at P2 in kJ/kg
17 W4=17.3;//Relative velocity at exit from the
   velocity triangle for second stage in m/s

```

```

18 a4=90; //Nozzle angle in second stage in degrees
19 C3=33; //Steam velocity at inlet from the velocity
    triangle for second stage in m/s
20 W2=49; //Relative velocity at outlet from the
    velocity triangle for first stage in m/s
21 x=15; //Length of AB assumed for drawing velocity
    triangle in mm
22 y=67; //Length of BC from the velocity triangle in mm
23
24 //CALCULATIONS
25 C1=sqrt(2000*h); //Velocity of steam in m/s
26 W3=W4/0.9; //Relative velocity at inlet for second
    stage in m/s
27 C2=C3/0.9; //Velocity in m/s
28 W1=W2/0.9; //Relative velocity at inlet for first
    stage in m/s
29 C1n=C1/y; //Velocity of steam in m/s
30 U=x*C1n; //Blade speed in m/s
31 a3=17; //Nozzle angle in second stage from the
    velocity triangle in degrees
32 a2=43; //Nozzle angle from the velocity triangle in
    degrees
33 DW1=731.5; //Change in relative velocity from the
    velocity triangle for first stage in m/s
34 DW2=257.5; //Change in relative velocity from the
    velocity triangle for second stage in m/s
35 ndia=((2*U*(DW1+DW2))/(C1^2))*100; //Diagram
    efficiency
36
37 //OUTPUT
38 mprintf('Blade speed is %3.1f m/s \n Blade tip
    angles of the fixed blade are %3.0f degrees and
    %3.0f degrees \n Diagram efficiency is %3.1f
    percent ',U,a3,a2,ndia)
39
40
41
42

```

```

43
44
45
46 //=====END OF PROGRAM
=====

```

Scilab code Exa 4.20 Blade speed and Turbine power

```

1 //Chapter-4, Illustration 20, Page 210
2 //Title: Steam Nozzles and Steam Turbines
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 C1=600;//Steam velocity in m/s
9 b1=30;//Blade angle at inlet in first stage in
   degrees
10 b2=30;//Blade angle at exit in first stage in
   degrees
11 b3=30;//Blade angle at inlet in second stage in
   degrees
12 b4=30;//Blade angle at exit in second stage in
   degrees
13 a4=90;//Nozzle angle in second stage in degrees
14 m=3;//Mass of steam in kg/s
15 x=15;//Length for drawing velocity triangle in mm
16 y=56;//Length of BC from the velocity triangle in mm
17
18 //CALCUALTIONS
19 C1n=C1/y;//Velocity of steam in m/s
20 U=x*C1n;//Blade speed in m/s
21 l=103;//Length from velocity triangle in mm

```

```

22 P=(m*1*C1n*U)/1000; //Power developed in kW
23
24 //OUTPUT
25 mprintf('Blade speed is %3.1f m/s \n Power developed
        by the turbine is %3.2f kW',U,P)
26
27
28
29
30
31
32 //=====END OF PROGRAM
        =====

```

Scilab code Exa 4.21 Mean diameter of drum and Volume of steam

```

1 //Chapter-4, Illustration 21, Page 211
2 //Title: Steam Nozzles and Steam Turbines
3 //
        =====
4 clc
5 clear
6
7 //INPUT DATA
8 N=400; //Speed in rpm
9 m=8.33; //Mass of steam in kg/s
10 P=1.6; //Pressure of steam in bar
11 x=0.9; //Dryness fraction
12 W=10; //Stage power in kW
13 r=0.75; //Ratio of axial flow velocity to blade
        velocity
14 a1=20; //Nozzle angle at inlet in degrees
15 a2=35; //Nozzle angle at exit in degrees
16 b1=a2; //Blade tip angle at exit in degrees

```

```

17 b2=a1;//Blade tip angle at inlet in degrees
18 a=25;//Length of AB from velocity triangle in mm
19 vg=1.091;//Specific volume of steam from steam
    tables in (m^3)/kg
20
21 //CALCULATIONS
22 Cx=73.5;//Change in whirl velocity from the velocity
    triangle by measurement in mm
23 y=Cx/a;//Ratio of change in whirl velocity to blade
    speed
24 U=sqrt((W*1000)/(m*y));//Blade speed in m/s
25 D=((U*60)/(3.147*N))*1000;//Mean diameter of drum in
    mm
26 v=m*x*vg;//Volume flow rate of steam in (m^3)/s
27
28 //OUTPUT
29 mprintf('Mean diameter of drum is %3.0f mm \n Volume
    of steam flowing per second is %3.2f m^3/s',D,v)
30
31
32
33
34
35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.22 Drum diameter and Blade height

```

1 //Chapter-4, Illustration 22, Page 212
2 //Title: Steam Nozzles and Steam Turbines

```

```

3 //


---




---


4 clc
5 clear
6
7 //INPUT DATA
8 N=300; //Speed in rpm
9 m=4.28; //Mass of steam in kg/s
10 P=1.9; //Pressure of steam in bar
11 x=0.93; //Dryness fraction
12 W=3.5; //Stage power in kW
13 r=0.72; //Ratio of axial flow velocity to blade
    velocity
14 a1=20; //Nozzle angle at inlet in degrees
15 b2=a1; //Blade tip angle at inlet in degrees
16 l=0.08; //Tip leakage steam
17 vg=0.929; //Specific volume of steam from steam
    tables in (m^3)/kg
18
19 //CALCULATIONS
20 mact=m-(m*l); //Actual mass of steam in kg/s
21 a=(3.147*N)/60; //Ratio of blade velocity to mean dia
22 b=r*a; //Ratio of axial velocity to mean dia
23 c=46; //Ratio of change in whirl velocity to mean dia
24 D=sqrt((W*1000)/(mact*c*a)); //Mean dia in m
25 Ca=b*D; //Axial velocity in m/s
26 h=((mact*x*vg)/(3.147*D*Ca))*1000; //Blade height in
    mm
27 D1=D-(h/1000); //Drum dia in m
28
29 //OUTPUT
30 mprintf('Drum diameter is %3.3f m \n Blade height is
    %3.0f mm',D1,h)
31
32
33
34

```

```

35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.23 Rotor blade angles and Flow coefficient and Blade loading coefficient and Power developed

```

1 //Chapter-4, Illustration 23, Page 214
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P0=800; //Steam pressure in kPa
9 P2=100; //Pressure at point 2 in kPa
10 T0=973; //Steam temperature in K
11 a1=73; //Nozzle angle in degrees
12 ns=0.9; //Steam efficiency
13 m=35; //Mass flow rate in kg/s
14 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
15 y=1.4; //Ratio of specific heats
16
17 //CALCULATIONS
18 b1=atand(tand(a1)/2); //Blade angle at inlet in
    degrees
19 b2=b1; //Blade angle at exit in degrees
20 p=2/tand(a1); //Flow coefficient

```



```

21 s=p*(tand(b1)+tand(b2)); //Blade loading coefficient
22 Dh=ns*Cp*T0*(1-((P2/P0)^((y-1)/y))); //Difference in
    enthalpies in kJ/kg
23 W=(m*Dh)/1000; //Power developed in MW
24
25 //OUTPUT
26 mprintf('Rotor blade angles are %3.2f degrees and %3
    .2f degrees \n Flow coefficient is %3.3f \n Blade
    loading coefficient is %3.0f \n Power developed
    is %3.1f MW',b1,b2,p,s,W)
27
28
29
30
31
32
33
34
35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.24 Rotor blade angles and Power developed and Final state of steam and Blade height

```

1 //Chapter-4, Illustration 24, Page 215
2 //Title: Steam Nozzles and Steam Turbines
3 //
    =====
4 clc

```

```

5 clear
6
7 //INPUT DATA
8 P0=100; //Steam pressure in bar
9 T0=773; //Steam temperature in K
10 a1=70; //Nozzle angle in degrees
11 ns=0.78; //Steam efficiency
12 m=100; //Mass flow rate of steam in kg/s
13 D=1; //Turbine diameter in m
14 N=3000; //Turbine speed in rpm
15 h0=3370; //Steam enthalpy from Moiller chart in kJ/kg
16 v2=0.041; //Specific volume at P2 from steam tables
    in (m^3)/kg
17 v4=0.05; //Specific volume at P4 from steam tables in
    (m^3)/kg
18
19 //CALCULATIONS
20 U=(3.147*D*N)/60; //Blade speed in m/s
21 C1=(2*U)/sind(a1); //Steam speed in m/s
22 b1=atand(tand(a1)/2); //Blade angle at inlet for
    first stage in degrees
23 b2=b1; //Blade angle at exit for first stage in
    degrees
24 b3=b1; //Blade angle at inlet for second stage in
    degrees
25 b4=b2; //Blade angle at exit for second stage in
    degrees
26 Wt=(4*m*(U^2))/(10^6); //Total workdone in MW
27 Dh=(2*(U^2))/1000; //Difference in enthalpies in kJ/
    kg
28 Dhs=Dh/ns; //Difference in enthalpies in kJ/kg
29 h2=h0-Dh; //Enthalpy at point 2 in kJ/kg
30 h2s=h0-Dhs; //Enthalpy at point 2s in kJ/kg
31 Dh2=(2*(U^2))/1000; //Difference in enthalpies in kJ/
    kg
32 Dh2s=Dh2/ns; //Difference in enthalpies in kJ/kg
33 h4=h2-Dh2; //Enthalpy at point 4 in kJ/kg
34 h4s=h2-Dh2s; //Enthalpy at point 4s in kJ/kg

```

```

35 Ca=C1*cosd(a1); // Axial velocity in m/s
36 hI=(m*v2)/(3.147*D*Ca); // Blade height at first stage
    in m/s
37 hII=(m*v4)/(3.147*D*Ca); // Blade height at second
    stage in m/s
38
39 //OUTPUT
40 mprintf('Rotor blade angles for first stage are %3.2
    f degrees and %3.2f degrees \n Rotor blade angles
    for second stage are %3.2f degrees and %3.2f
    degrees \n Power developed is %3.2f MW \n Final
    state of steam at first stage is %3.2f kJ/kg \n
    Final state of steam at second stage is %3.2f kJ/
    kg \n Blade height at first stage is %3.4f m \n
    Blade height at second stage is %3.4f m',b1,b2,b3
    ,b4,Wt,h2s,h4s,hI,hII)
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.25 Rotor blade angles and Power developed and Final state of steam and Blade height

```

1 //Chapter-4, Illustration 25, Page 218
2 //Title: Steam Nozzles and Steam Turbines
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 P0=100; //Steam pressure in bar
9 T0=773; //Steam temperature in K
10 a1=70; //Nozzle angle in degrees
11 ns=0.78; //Steam efficiency
12 m=100; //Mass flow rate of steam in kg/s
13 D=1; //Turbine diameter in m
14 N=3000; //Turbine speed in rpm
15 h0=3370; //Steam enthalpy from Moiller chart in kJ/kg
16 P4=27; //Pressure at point 4 in bar
17 T4=638; //Temperature at point 4 in K
18 v4=0.105; //Specific volume at P4 from mollier chart
    in (m^3)/kg
19 ns=0.65; //Stages efficiency
20
21 //CALCULATIONS
22 U=(3.147*D*N)/60; //Blade speed in m/s
23 C1=(4*U)/sind(a1); //Steam speed in m/s
24 Ca=C1*cosd(a1); //Axial velocity in m/s
25 b1=atand((3*U)/Ca); //Blade angle at inlet for first
    stage in degrees
26 b2=b1; //Blade angle at exit for first stage in
    degrees
27 b4=atand(U/Ca); //Blade angle at exit for second
    stage in degrees
28 b3=b4; //Blade angle at inlet for second stage in
    degrees
29 WI=m*6*(U^2); //Power developed in first stage in MW
30 WII=m*2*(U^2); //Power developed in second stage in
    MW

```

```

31 W=(WI+WII)/(10^6); //Total power developed in MW
32 Dh=(W*1000)/100; //Difference in enthalpies in kJ/kg
33 Dhs=(W*1000)/(ns*100); //Difference in enthalpies in
    kJ/kg
34 h4=h0-Dh; //Enthalpy at point 4 in kJ/kg
35 h4s=h0-Dhs; //Enthalpy at point 4s in kJ/kg
36 h=(m*v4)/(3.147*D*Ca); //Rotor blade height in m
37
38 //OUTPUT
39 mprintf('Rotor blade angles for first stage are %3.2
    f degrees and %3.2f degrees \n Rotor blade angles
    for second stage are %3.2f degrees and %3.2f
    degrees \n Power developed is %3.2f MW \n Final
    state of steam at first stage is %3.1f kJ/kg \n
    Final state of steam at second stage is %3.2f kJ/
    kg \n Rotor blade height is %3.4f m',b1,b2,b3,b4,
    W,h4,h4s,h)
40
41
42
43
44
45
46
47
48
49
50 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.26 Rotor blade angles

```

1 //Chapter-4, Illustration 26, Page 221
2 //Title: Steam Nozzles and Steam Turbines
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  a1=30; //Nozzle angle in degrees
9  Ca=180; //Axial velocity in m/s
10 U=280; //Rotor blade speed in m/s
11 R=0.5; //Degree of reaction
12
13 //CALCULATIONS
14 a1n=90-a1; //Nozzle angle measured from axial
    direction in degrees
15 Cx1=Ca*tand(a1n); //Whirl velocity in m/s
16 b1=atand((Cx1-U)/Ca); //Blade angle at inlet in
    degrees
17 b2=a1n; //Blade angle at exit in degrees
18
19 //OUTPUT
20 mprintf('Blade angle at inlet is %3.0f degrees \n
    Blade angle at exit is %3.0f degrees ',b1,b2)
21
22
23
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 4.27 Rotor blade angles and Power developed and Isentropic enthalpy drop

```
1 //Chapter-4, Illustration 27, Page 222
2 //Title: Steam Nozzles and Steam Turbines
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 P0=800; //Steam pressure in kPa
9 T0=900; //Steam temperature in K
10 a1=70; //Nozzle angle in degrees
11 ns=0.85; //Steam efficiency
12 m=75; //Mass flow rate of steam in kg/s
13 R=0.5; //Degree of reaction
14 U=160; //Blade speed in m/s
15
16 //CALCULATIONS
17 C1=U/sind(a1); //Steam speed in m/s
18 Ca=C1*cosd(a1); //Axial velocity in m/s
19 b1=0; //Blade angle at inlet from velocity triangle
    in degrees
20 b2=a1; //Blade angle at exit in degrees
21 a2=b1; //Nozzle angle in degrees
22 W=(m*(U^2))/(10^6); //Power developed in MW
23 Dhs=(W*1000)/(ns*m); //Isentropic enthalpy drop in kJ
    /kg
24
25 //OUTPUT
26 mprintf('Rotor blade angles are %3.0f degrees and %3
    .0f degrees \n Power developed is %3.2f MW \n
    Isentropic enthalpy drop is %3.2f kJ/kg',b1,b2,W,
    Dhs)
27
28
```

29

30

31

32

33

34

35 //=====END OF PROGRAM

=====

Chapter 5

Air Compressors

Scilab code Exa 5.1 Indicated power and Mass of air and Temperature delivered by compressor

```
1 //Chapter-5, Illustration 1, Page 250
2 //Title: Air Compressors
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 D=0.2; //Cylinder diameter in m
9 L=0.3; //Cylinder Stroke in m
10 P1=1; //Pressure at entry in bar
11 T1=300; //Temperature at entry in K
12 P2=8; //Pressure at exit in bar
13 n=1.25; //Adiabatic gas constant
14 N=100; //Speed in rpm
15 R=287; //Universal gas constant in J/kg-K
16
17 //CALCULATIONS
18 x=(n-1)/n; //Ratio
```

```

19 V1=(3.147*L*(D^2))/4; //Volume of cylinder in m^3/
    cycle
20 W=(P1*(10^5)*V1*(((P2/P1)^x)-1))/x; //Work done in J/
    cycle
21 Pc=(W*100)/(60*1000); //Indicated power of compressor
    in kW
22 m=(P1*(10^5)*V1)/(R*T1); //Mass of air delivered in
    kg/cycle
23 md=m*N; //Mass delivered per minute in kg
24 T2=T1*(((P2/P1)^x)); //Temperature of air delivered in
    K
25
26 //OUTPUT
27 mprintf('Indicated power of compressor is %3.2f kW \
    n Mass of air delivered by compressor per minute
    is %3.2f kg \n Temperature of air delivered is %3
    .1fK ',Pc,md,T2)
28
29
30
31
32 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.2 Size of cylinder

```

1 //Chapter -5, Illustration 2, Page 251
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA

```

```

 8 IP=37; //Indicated power in kW
 9 P1=0.98; //Pressure at entry in bar
10 T1=288; //Temperature at entry in K
11 P2=5.8; //Pressure at exit in bar
12 n=1.2; //Adiabatic gas constant
13 N=100; //Speed in rpm
14 Ps=151.5; //Piston speed in m/min
15 a=2; //For double acting compressor
16
17 //CALCULATIONS
18 L=Ps/(2*N); //Stroke length in m
19 x=(n-1)/n; //Ratio
20 r=(3.147*L)/4; //Ratio of volume to bore
21 D=sqrt((IP*1000*60*x)/(N*a*r*P1*(10^5)*(((P2/P1)^x)
    -1))); //Cylinder diameter in m
22
23 //OUTPUT
24 mprintf('Stroke length of cylinder is %3.4f m \n
    Cylinder diameter is %3.4f m',L,D)
25
26
27
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.3 Cylinder dimensions

```

1 //Chapter-5, Illustration 3, Page 251
2 //Title: Air Compressors
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  IP=11; //Indicated power in kW
9  P1=1; //Pressure at entry in bar
10 P2=7; //Pressure at exit in bar
11 n=1.2; //Adiabatic gas constant
12 Ps=150; //Piston speed in m/s
13 a=2; //For double acting compressor
14 r=1.5; //Stroke to bore ratio
15
16 //CALCULATIONS
17 x=(n-1)/n; //Ratio
18 y=3.147/(4*(r^2)); //Ratio of volume to the cube of
    stroke
19 z=(P1*(10^2)*y*(((P2/P1)^x)-1))/x; //Ratio of
    workdone to the cube of stroke
20 L=(sqrt(IP/(z*Ps)))*1000; //Stroke in mm
21 D=(L/r); //Bore in mm
22
23 //OUTPUT
24 mprintf('Stroke length of cylinder is %3.0f mm \n
    Bore diameter of cylinder is %3.0f mm',L,D)
25
26
27
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.4 Volumetric efficiency and Volumetric efficiency referred to atmospheric conditions and Work required

```

1 //Chapter -5, Illustration 4, Page 252
2 //Title: Air Compressors
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 x=0.05; //Ratio of clearance volume to swept volume
9 P1=1; //Pressure at point 1 in bar
10 T1=310; //Temperature at point 1 in K
11 n=1.2; //Adiabatic gas constant
12 P2=7; //Pressure at point 2 in bar
13 Pa=1.01325; //Atmospheric pressure in bar
14 Ta=288; //Atmospheric temperature in K
15
16 //CALCULATIONS
17 V1=1+x; //Ratio of volume of air sucked to stroke
    volume
18 V4=((P2/P1)^(1/n))/20; //Ratio of volume delivered to
    stroke volume
19 DV=V1-V4; //Difference in volumes
20 nv1=DV*100; //Volumetric efficiency
21 V=(P1*DV*Ta)/(T1*Pa); //Ratio of volumes referred to
    atmospheric conditions
22 nv2=V*100; //Volumetric efficiency referred to
    atmospheric conditions
23 W=(n*0.287*T1*((P2/P1)^((n-1)/n)-1))/(n-1); //Work
    required in kJ/kg
24
25 //OUTPUT
26 mprintf('Volumetric efficiency is %3.1f percent \n
    Volumetric efficiency referred to atmospheric
    conditions is %3.1f percent \n Work required is
    %3.1f kJ/kg',nv1,nv2,W)
27
28

```

```

29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.5 Theoretical volume of air taken

```

1 //Chapter-5, Illustration 5, Page 253
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 D=0.2; //Bore in m
9 L=0.3; //Stroke in m
10 lc=0.015; //Linear clearance in m
11 P1=1; //Pressure at point 1 in bar
12 P2=7; //Pressure at point 2 in bar
13 n=1.25; //Adiabatic gas constant
14
15 //CALCULATIONS
16 V3=(3.147*(D^2)*lc)/4; //Clearance volume in m^3
17 Vs=(3.147*(D^2)*L)/4; //Stoke volume in m^3
18 C=V3/Vs; //Clearance ratio
19 nv=(1+C-(C*((P2/P1)^(1/n))))*100; //Volumetric
    efficiency
20 DV=(nv*Vs)/100; //Volume of air taken in (m^3)/stroke
21

```

```

22 //OUTPUT
23 mprintf('Theoretical volume of air taken in per
        stroke is %3.6f (m^3)/stroke',DV)
24
25
26
27
28
29
30
31
32 //=====END OF PROGRAM
        =====

```

Scilab code Exa 5.6 Mean effective pressure and Power required

```

1 //Chapter-5, Illustration 6, Page 254
2 //Title: Air Compressors
3 //
        =====

4 clc
5 clear
6
7 //INPUT DATA
8 D=0.2; //Bore in m
9 L=0.3; //Stroke in m
10 r=0.05; //Ratio of clearance volume to stroke volume
11 P1=1; //Pressure at point 1 in bar
12 T1=293; //Temperature at point 1 in K
13 P2=5.5; //Pressure at point 2 in bar
14 n=1.3; //Adiabatic gas constant
15 N=500; //Speed of compressor in rpm
16
17 //CALCULATIONS

```

```

18 x=(n-1)/n; //Ratio
19 Vs=(3.147*L*(D^2))/4; //Stroke volume in m^3
20 Vc=r*Vs; //Clearance volume in m^3
21 V1=Vc+Vs; //Volume at point 1 in m^3
22 V4=Vc*((P2/P1)^(1/n)); //Volume at point 4 in m^3
23 EVs=V1-V4; //Effective swept volume in m^3
24 W=(P1*(10^5)*EVs*((P2/P1)^x)-1)/x; //Work done in J
    /cycle
25 MEP=(W/Vs)/(10^5); //Mean effective pressure in bar
26 P=(W*N)/(60*1000); //Power required in kW
27
28 //OUTPUT
29 mprintf('Mean effective pressure is %3.2f bar \n
    Power required is %3.2f kW',MEP,P)
30
31
32
33
34
35
36 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.7 Free air delivered and Volumetric efficiency and Delivery temperature and Cycle power and Isothermal efficiency

```

1 //Chapter -5, Illustration 7, Page 255
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA

```



```

8 D=0.2; //Bore in m
9 L=0.3; //Stroke in m
10 r=0.05; //Ratio of clearance volume to stroke volume
11 P1=97; //Pressure at entry in kN/(m^2)
12 P4=P1; //Pressure at point 4 in kN/(m^2)
13 T1=293; //Temperature at point 1 in K
14 P2=550; //Compression Pressure in kN/(m^2)
15 P3=P2; //Pressure at point 3 in kN/(m^2)
16 n=1.3; //Adiabatic gas constant
17 N=500; //Speed of compressor in rpm
18 Pa=101.325; //Air pressure in kN/(m^2)
19 Ta=288; //Air temperature in K
20
21 //CALCULATIONS
22 x=(n-1)/n; //Ratio
23 DV=(3.147*L*(D^2))/4; //Difference in volumes in m^3
24 V3=r*DV; //Clearance volume in m^3
25 V1=V3+DV; //Volume at point 1 in m^3
26 V4=V3*((P3/P4)^(1/n)); //Volume at point 4 in m^3
27 Vs=V1-V4; //Effective swept volume in m^3
28 EVs=Vs*N; //Effective swept volume per min
29 Va=(P1*EVs*Ta)/(Pa*T1); //Free air delivered in (m^3)
    /min
30 nV=((V1-V4)/(V1-V3))*100; //Volumetric efficiency
31 T2=T1*((P2/P1)^x); //Air delivery temperature in K
32 t2=T2-273; //Air delivery temperature in oC
33 W=(n*P1*(V1-V4)*(((P2/P1)^x)-1))*N/((n-1)*60); //
    Cycle power in kW
34 Wiso=P1*V1*(log(P2/P1)); //Isothermal workdone
35 P=(n*P1*V1*(((P2/P1)^x)-1))/(n-1); //Cycle power
    neglecting clearance
36 niso=(Wiso/P)*100; //Isothermal efficiency
37
38 //OUTPUT
39 mprintf('Free air delivered is %3.3f (m^3)/min \n
    Volumetric efficiency is %3.0f percent \n Air
    delivery temperature is %3.1f oC \n Cycle power
    is %3.0f kW \n Isothermal efficiency is %3.1f

```

```

percent ',Va,nV,t2,W,niso)
40
41
42
43
44
45
46
47
48
49
50 //=====END OF PROGRAM
=====

```

Scilab code Exa 5.8 Mean effective pressure and Brake power

```

1 //Chapter-5, Illustration 8, Page 257
2 //Title: Air Compressors
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 Ve=30;//Volume of air entering compressor per hour
   in m^3
9 P1=1;//Presure of air entering compressor in bar
10 N=450;//Speed in rpm
11 P2=6.5;//Pressure at point 2 in bar
12 nm=0.8;//Mechanical efficiency
13 nv=0.75;//Volumetric efficiency
14 niso=0.76;//Isothermal efficiency
15
16 //CALCULATIONS

```

```

17 Vs=Ve/(nv*3600); //Swept volume per sec in (m^3)/s
18 V=(Vs*60)/N; //Swept volume per cycle in m^3
19 V1=(Ve*60)/(3600*N); //Volume at point 1 in m^3
20 Wiso=P1*100*V1*log(P2/P1); //Isothermal workdone per
    cycle
21 Wact=Wiso/niso; //Actual workdone per cycle on air
22 MEP=(Wact/V)/100; //Mean effective pressure in bar
23 IP=(Wact*N)/60; //Indicated power in kW
24 BP=IP/nm; //Brake power in kW
25
26 //OUTPUT
27 mprintf('Mean effective pressure is %3.3f bar \n
    Brake power is %3.2f kW',MEP,BP)
28
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.9 Cylinder dimensions

```

1 //Chapter-5, Illustration 9, Page 258
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 Va=15; //Volume of air in (m^3)/min

```

```

 9 Pa=1.01325; // Pressure of air in bar
10 Ta=302; // Air temperature in K
11 P1=0.985; // Pressure at point 1 in bar
12 T1=313; // Temperature at point 1 in K
13 r=0.04; // Ratio of clearance volume to swept volume
14 y=1.3; // Ratio of stroke to bore diameter
15 N=300; // Speed in rpm
16 n=1.3; // Adiabatic gas constant
17 P2=7.5; // Pressure at point 2 in bar
18
19 //CALCULATIONS
20 x=((P2/P1)^(1/n))-1; // Ratio of volume at point 4 to
    clearance volume
21 a=x*r; // Ratio of volume at point 4 to swept volume
22 nv=1-a; // Volumetric efficiency
23 V1=(Pa*Va*T1)/(Ta*P1); // Volume at point 1 in (m^3)/
    min
24 Vs=V1/(nv*N*2); // Swept volume in m^3
25 D=((Vs*4)/(3.147*y))^(1/3); // Bore in m
26 L=y*D; // Stroke in m
27
28 //OUTPUT
29 mprintf('Cylinder bore in %3.3f m \n Cylinder stroke
    %3.3f m',D,L)
30
31
32
33
34
35
36 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.10 Volumetric efficiency and Indicated power and Isothermal efficiency of compressor

```

1 //Chapter -5, Illustration 10, Page 259
2 //Title: Air Compressors
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 P1=0.98; //Pressure at point 1 in bar
9 P4=P1; //Pressure at point 4 in bar
10 P2=7; //Pressure at point 2 in bar
11 P3=P2; //Pressure at point 3 in bar
12 n=1.3; //Adiabatic gas constant
13 Ta=300; //Air temperature in K
14 Pa=1.013; //Air pressure in bar
15 T1=313; //Temperature at point 1 in K
16 c=0.04; //Ratio of clearance volume to swept volume
17 Va=15; //Volume of air delivered in m^3
18 R=0.287; //Universal gas constant in kJ/kg-K
19
20 //CALCULATIONS
21 x=(n-1)/n; //Ratio
22 r=(P2/P1)^(1/n); //Ratio of volumes
23 a=r*c; //Ratio of volume at point 4 to swept volume
24 DV=1+c-a; //Difference in volumes
25 V=(P1*DV*Ta)/(T1*Pa); //Volume of air delivered per
    cycle
26 nv=V*100; //Volumetric efficiency
27 DV1=(Pa*Va*T1)/(Ta*P1); //Difference in volumes
28 T2=T1*((P2/P1)^x); //Temperature at point 2 in K
29 ma=(Pa*100*Va)/(R*Ta); //Mass of air delivered in kg/
    min
30 IP=(ma*R*(T2-T1))/(x*60); //Indicated power in kW
31 Piso=(ma*R*T1*log(P2/P1))/60; //Isothermal indicated
    power in kW
32 niso=(Piso/IP)*100; //Isothermal efficiency
33

```

```

34 //OUTPUT
35 mprintf('Volumetric efficiency is %3.1f percent \n
        Indicated power is %3.2f kW \n Isothermal
        efficiency is %3.0f percent ',nv,IP,niso)
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.11 Power required

```

1 //Chapter-5, Illustration 11, Page 261
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 V1=7*(10^-3); //Volume of air in (m^3)/s
9 P1=1.013; //Pressure of air in bar
10 T1=288; //Air temperature in K

```

```

11 P2=14; // Pressure at point 2 in bar
12 n=1.3; // Adiabatic gas constant
13 nm=0.82; // Mechanical efficiency
14
15 //CALCULATIONS
16 x=(n-1)/n; // Ratio
17 W=(P1*100*V1*(((P2/P1)^x)-1))/x; // Work done by
    compressor in kW
18 P=W/nm; // Power required to drive compressor in kW
19
20 //OUTPUT
21 mprintf('Power required to drive compressor is %3.2f
    kW', P)
22
23
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.12 Theoretical volume efficiency and Volume of air delivered and Power of compressor

```

1 //Chapter -5, Illustration 12, Page 261
2 //Title: Air Compressors
3 //
    =====
4 clc
5 clear
6

```

```

7 //INPUT DATA
8 L=0.15;//Stroke in mm
9 D=0.15;//Bore in mm
10 N=8;//Speed in rps
11 P1=100;//Pressure at point 1 in kN/(m^2)
12 P2=550;//Pressure at point 2 in kN/(m^2)
13 n=1.32;//Adiabatic gas constant
14 C=0.06;//Ratio of clearance volume to swept volume
15
16 //CALCULATIONS
17 x=(n-1)/n;//Ratio
18 nv=(1+C-(C*((P2/P1)^(1/n))))*100;//Volumetric
    efficiency
19 DV=(3.147*(D^2)*L)/4;//Difference in volumes at
    points 1 and 3
20 DV1=(nv*DV)/100;//Difference in volumes at points 1
    and 4
21 V2=DV1*((P1/P2)^(1/n))*N;//Volume of air delivered
    per second
22 W=(P1*DV1*(((P2/P1)^x)-1))*N/x;//Power of compressor
    in kW
23
24 //OUTPUT
25 mprintf('Theoretical volume efficiency is %3.1f
    percent \n Volume of air delivered is %3.5f (m^3)
    /s \n Power of compressor is %3.3f kW',nv,V2,W)
26
27
28
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.13 Minimum indicated power and Maximum temperature and Heat to be removed and Mass of cooling water

```

1 //Chapter -5, Illustration 13, Page 262
2 //Title: Air Compressors
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 V=16; //Volume of air compressed in m3
9 P1=1; //Pressure at point 1 in bar
10 P3=10.5; //Pressure at point 3 in bar
11 T1=294; //Temperature at point 1 in K
12 Tc=25; //Temperature of cooling water in oC
13 n=1.35; //Adiabatics gas constant
14 R=0.287; //Universal gas constant in kJ/kg-K
15 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
16 Cw=4.187; //Specific heat of water in kJ/kg-K
17
18 //CALCULATIONS
19 x=(n-1)/n; //Ratio
20 P2=sqrt(P1*P3); //Pressure at point 2 in bar
21 W1=(2*P1*100*V*(((P2/P1)^x)-1))/(x*60); //Indicated
    power of compressor from P1 to P2 in kW
22 W2=(P1*100*V*(((P3/P1)^x)-1))/(x*60); //Indicated
    power of compressor from P1 to P3 in kW
23 T4=T1*(((P2/P1)^x)); //Maximum temperature for two
    stage compression in K
24 T2=T1*(((P3/P1)^x)); //Maximum temperature for single
    stage compression in K

```

```

25 m=(P1*100*V)/(R*T1); //Mass of air compressed in kg/
    min
26 Q=m*Cp*(T4-T1); //Heat rejected by air in kJ/min
27 mc=Q/(Cw*Tc); //Mass of cooling water in kg/min
28
29 //OUTPUT
30 mprintf('Minimum indicated power required for 2
    stage compression is %3.1f kW \n Power required
    for single stage compression is 18 percent more
    than that for two stage compression with perfect
    intercooling \n Maximum temperature for two stage
    compression is %3.1f K \n Maximum temperature
    for single stage compression is %3.1f K \n Heat
    rejected by air is %3.1f kJ/min \n Mass of
    cooling water required is %3.1f kg/min ',W1,T4,T2,
    Q,mc)
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.14 Intermediate pressure and Total volume of each cylinder and Cycle power

```

1 //Chapter -5, Illustration 14, Page 264
2 //Title: Air Compressors
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 V=0.2; //Air flow rate in (m3)/s
9 P1=0.1; //Intake pressure in MN/(m2)
10 P3=0.7; //Final pressure in MN/(m2)
11 T1=289; //Intake temperature in K
12 n=1.25; //Adiabatic gas constant
13 N=10; //Compressor speed in rps
14
15 //CALCULATIONS
16 x=(n-1)/n; //Ratio
17 P2=sqrt(P1*P3); //Intermediate pressure in MN/(m2)
18 V1=(V/N)*1000; //Total volume of LP cylinder in
    litres
19 V2=((P1*V1)/P2); //Total volume of HP cylinder in
    litres
20 W=((2*P1*V*((P2/P1)^x)-1))/x)*1000; //Cycle power in
    kW
21
22 //OUTPUT
23 mprintf('Intermediate pressure is %3.3f MN/(m2) \n
    Total volume of LP cylinder is %3.0f litres \n
    Total volume of HP cylinder is %3.1f litres \n
    Cycle power is %3.0f kW',P2,V1,V2,W)
24
25
26
27
28
29
30

```

```
31
32 //=====END OF PROGRAM
    =====
```

Scilab code Exa 5.15 Power of compressor

```
1 //Chapter-5, Illustration 15, Page 265
2 //Title: Air Compressors
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1; //Pressure at point 1 in bar
9 T1=290; //Temperature at point 1 in K
10 P3=60; //Pressure at point 3 in bar
11 P2=8; //Pressure at point 2 in bar
12 T2=310; //Temperature at point 2 in K
13 L=0.2; //Stroke in m
14 D=0.15; //Bore in m
15 n=1.35; //Adiabatic gas constant
16 N=200; //Speed in rpm
17
18 //CALCULATIONS
19 x=(n-1)/n; //Ratio
20 V1=(3.147*(D^2)*L)/4; //Volume at point 1 in m^3
21 V2=(P1*V1*T2)/(T1*P2); //Volume of air entering LP
    cylinder in m^3
22 W=((P1*(10^5)*V1*((P2/P1)^x)-1)/x)+((P2*(10^5)*V2
    *(((P3/P2)^x)-1))/x); //Workdone by compressor per
    cycle in J
23 P=(W*N)/(60*1000); //Power of compressor in kW
24
```

```

25 //OUTPUT
26 mprintf('Power of compressor is %3.2f kW',P)
27
28
29
30
31
32
33
34
35 //=====END OF PROGRAM
=====

```

Scilab code Exa 5.16 Heat rejected and Diameter of HP cylinder and Power required

```

1 //Chapter-5, Illustration 16, Page 265
2 //Title: Air Compressors
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 N=220;//Speed of compressor in rpm
9 P1=1;//Pressure entering LP cylinder in bar
10 T1=300;//Temperature at point 1 in K
11 Dlp=0.36;//Bore of LP cylinder in m
12 Llp=0.4;//Stroke of LP cylinder in m
13 Lhp=0.4;//Stroke of HP cylinder in m
14 C=0.04;//Ratio of clearance volumes of both
    cylinders
15 P2=4;//Pressure leaving LP cylinder in bar
16 P5=3.8;//Pressure entering HP cylinder in bar

```

```

17 T3=300; //Temperature entering HP cylinder in K
18 P6=15.2; //Discharge pressure in bar
19 n=1.3; //Adiabatic gas constant
20 Cp=1.0035; //Specific heat at constant pressure in kJ
    /kg-K
21 R=0.287; //Universal gas constant in kJ/kg-K
22 T5=T1; //Temperature at point 5 in K
23
24 //CALCULATIONS
25 x=(n-1)/n; //Ratio
26 Vslp=(3.147*(Dlp^2)*Llp*N*2)/4; //Swept volume of LP
    cylinder in m^3/min
27 nv=1+C-(C*((P2/P1)^(1/n))); //Volumetric efficiency
28 V1=nv*Vslp; //Volume of air drawn at point 1 in (m^3)
    /min
29 m=(P1*100*V1)/(R*T1); //Mass of air in kg/min
30 T2=T1*((P2/P1)^x); //Temperature at point 2 in K
31 QR=m*Cp*(T2-T5); //Heat rejected in kJ/min
32 V5=(m*R*T5)/(P5*100); //Volume of air drawn in HP
    cylinder M^3/min
33 P1p=P2/P1; //Pressure ratio of LP cylinder
34 P1hp=P6/P5; //Pressure ratio of HP cylinder
35 Vshp=V5/nv; //Swept volume of HP cylinder in m^3/min
36 Dhp=sqrt((Vshp*4)/(3.147*Lhp*N*2)); //Bore of HP
    cylinder in m
37 P=(m*R*(T2-T1))/(x*60); //Power required for HP
    cylinder in kW
38
39 //OUTPUT
40 mprintf('Heat rejected in intercooler is %3.1f kJ/
    min \n Diameter of HP cylinder is %3.4f m \n
    Power required for HP cylinder is %3.0f kW',QR,
    Dhp,P)
41
42
43
44
45

```

```
46
47
48
49
50 //=====END OF PROGRAM
    =====
```

Scilab code Exa 5.17 Ratio of cylinder diameters

```
1 //Chapter -5, Illustration 17, Page 267
2 //Title: Air Compressors
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1;//Pressure at point 1 in bar
9 P3=30;//Pressure at point 3 in bar
10 T1=300;//Temperature at point 1 in K
11 n=1.3;//Adiabatics gas constant
12
13 //CALCULATIONS
14 P2=sqrt(P1*P3);//Intermediate pressure in bar
15 rD=sqrt(P2/P1);//Ratio of cylinder diameters
16
17 //OUTPUT
18 mprintf('Ratio of cylinder diameters is %3.2f',rD)
19
20
21
22
23
24
```

```

25
26 //=====END OF PROGRAM
=====

```

Scilab code Exa 5.18 Delivery pressures and Ratio of cylinder volumes and Temperature and Heat rejected in intercooler and Total indicated power

```

1 //Chapter -5, Illustration 18, Page 268
2 //Title: Air Compressors
3 //
=====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1.013; //Pressure at point 1 in bar
9 T1=288; //Temperature at point 1 in K
10 v1=8.4; //free air delivered by compressor in m^3
11 P4=70; //Pressure at point 4 in bar
12 n=1.2; //Adiabatic gas constant
13 Cp=1.0035; //Specific heat at constant pressure in kJ
    /kg-K
14
15 //CALCULATIONS
16 x=(n-1)/n; //Ratio
17 P2=P1*((P4/P1)^(1/3)); //LP cylinder delivery
    pressure in bar
18 P3=P2*((P4/P1)^(1/3)); //IP cylinder delivery
    pressure in bar
19 r=P2/P1; //Ratio of cylinder volumes
20 r1=P3/P2; //Ratio of cylinder volumes
21 r2=r*r1; //Ratio of cylinder volumes
22 V3=1; //Volume at point 3 in m^3
23 T4=T1*((P2/P1)^x); //Three stage outlet temperature

```



```

    in K
24 QR=Cp*(T4-T1); //Heat rejected in intercooler in kJ/
    kg of air
25 W=((3*P1*100*v1*(((P4/P1)^(x/3))-1))/(x*60)); //Total
    indicated power in kW
26
27 //OUTPUT
28 mprintf('LP cylinder delivery pressure is %3.3f bar
    \n IP cylinder delivery pressure is %3.2f bar \n
    Ratio of cylinder volumes is %3.2f:%3.1f:%3.0f \n
    Temperature at end of each stage is %3.2f K \n
    Heat rejected in each intercooler is %3.1f kJ/kg
    of air \n Total indicated power is %3.2f kW',P2,
    P3,r2,r1,V3,T4,QR,W)
29
30
31
32
33
34
35
36
37
38
39
40
41 //=====END OF PROGRAM
    =====

```

Scilab code Exa 5.19 Intermediate pressures and Effective swept volume and Temperature and volume of air delivered and Workdone

```

1 //Chapter-5, Illustration 19, Page 269
2 //Title: Air Compressors
3 //

```

```

4  clc
5  clear
6
7  //INPUT DATA
8  D=0.45; //Bore in m
9  L=0.3; //Stroke in m
10 C=0.05; //Ratio of clearance volume to swept volume
11 P1=1; //Pressure at point 1 inn bar
12 T1=291; //Temperature at point 1 in K
13 P4=15; //Pressure at point 4 in bar
14 n=1.3; //Adiabatic gas constant
15 R=0.29; //Universal gas constant in kJ/kg-K
16
17 //CALCULATIONS
18 x=(n-1)/n; //Ratio
19 k=(P4/P1)^(1/3); //Pressure ratio
20 P2=k*P1; //Pressure at point 2 in bar
21 P3=k*P2; //Pressure at point 1 in bar
22 Vs1p=(3.147*(D^2)*L)/4; //Swept volume of LP cylinder
23 V7=C*Vs1p; //Volume at point 7 in m^3
24 V1=Vs1p+V7; //Volume at point 1 in m^3
25 V8=V7*(k^(1/n)); //Volume at point 8 in m^3
26 EVs=(V1-V8)*1000; //Effective swept volume in litres
27 T4=T1*(k^x); //Temperature at point 4 in K
28 t4=T4-273; //Delivery temperature in oC
29 DV=((P1*T4*(V1-V8))/(P4*T1))*1000; //Delivery volume
    per stroke in litres
30 W=(3*R*T1*((k^x)-1))/x; //Workdone per kg of air in
    kJ
31
32 //OUTPUT
33 mprintf('Intermediate pressures are %3.3f bar and %3
    .3f bar \n Effective swept volume of LP cylinder
    is %3.2f litres \n Temperature of air delivered
    per stroke is %3.1f oC \n Volume of air delivered
    per stroke is %3.2f litres \n Work done per kg

```

```

    of air is %3.1f kJ',P2,P3,EVs,t4,DV,W)
34
35
36
37
38
39
40
41
42
43
44
45 //=====END OF PROGRAM
=====

```

Scilab code Exa 5.20 Number of stages and Exact stage pressure ratio and Intermediate pressures

```

1 //Chapter-5, Illustration 20, Page 271
2 //Title: Air Compressors
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=1;//Pressure at point 1 in bar
9 Pns=100;//Maximum pressure in bar
10 p=4;//Pressure ratio
11
12 //CALCULATIONS
13 Ns=log(Pns)/log(p);//Number of stages
14 y=ceil(Ns);//Rounding off to next higher integer
15 ps=(Pns/P1)^(1/y);//Exact stage pressure ratio

```

```
16 P2=ps*P1; //Pressure at point 2 in bar
17 P3=ps*P2; //Pressure at point 3 in bar
18 P4=ps*P3; //Pressure at point 4 in bar
19
20 //OUTPUT
21 mprintf('Number of stages are %3.0f \n Exact stage
    pressure ratio is %3.3f \n Intermediate pressures
    are %3.3f bar,%3.2f bar,%3.2f bar ',y,ps,P2,P3,P4
    )
22
23
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====
```

Chapter 6

Refrigeration Cycles

Scilab code Exa 6.1 Claim is correct or not

```
1 //Chapter-6, Illustration 1, Page 308
2 //Title: Refrigeration cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 COP=8.5; //Co-efficient of performance
9 T1=300; //Room temperature in K
10 T2=267; //Refrigeration temperature in K
11
12 //CALCULATIONS
13 COPmax=T2/(T1-T2); //Maximum COP possible
14
15 //OUTPUT
16 mprintf('Maximum COP possible is %3.2f \n Since the
        COP claimed by the inventor is more than the
        maximum possible COP his claim is not correct',
        COPmax)
```

```

17
18
19
20 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.2 Weight of ice formed and Minimum power required

```

1 //Chapter-6, Illustration 2, Page 309
2 //Title: Refrigeration cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 TL=268;//Low temperature in K
9 TH=293;//High temperature in K
10 t=24;//time in hrs
11 C=2100;//Capacity of refrigerator in kJ/s
12 Tw=10;//Water temperature in oC
13 L=335;//Latent heat of ice in kJ/kg
14
15 //CALCULATIONS
16 COP=TL/(TH-TL);//Co-efficient of performance
17 Pmin=C/COP;//Minimum power required in kW
18 Qr=(4.187*(Tw-0))+L;//Heat removed from water in kJ/
    kg
19 m=C/Qr;//mass of ice formed in kg/s
20 W=(m*t*3600)/1000;//Weight of ice formed in tons
21
22 //OUTPUT
23 mprintf('Minimum power required is %3.2f kW \n
    Weight of ice formed in 24 hours is %3.2f tons',

```

```

    Pmin ,W)
24
25
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.3 Mass of ice formed

```

1 //Chapter -6, Illustration 3, Page 309
2 //Title: Refrigeration cycles
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 TL=-10;//Temperature of brine in oC
9 TH=20;//Temperature of water in oC
10 L=335;//Latent heat of ice in kJ/kg
11
12 //CALCULATIONS
13 Qr=(4.187*(TH-0))+L;//Heat removed from water in kJ/
    kg
14 COP=(TL+273)/(TH-TL);//Co-efficient of performance
15 mi=(COP*3600)/Qr;//mass of ice formed per kWh in kg
16
17 //OUTPUT
18 mprintf('Mass of ice formed per kWh is %3.1f kg',mi)
19
20

```

```

21
22
23
24
25 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.4 Rate of heat removed and Power input to compresor and Rate of heat rejection to environment and Coefficient of performance

```

1 //Chapter-6, Illustration 4, Page 310
2 //Title: Refrigeration cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=1.2; //Pressure at point 1 in bar
9 P2=7; //Pressure at point 2 in bar
10 m=0.05; //mass flow rate of refrigerant in kg/s
11 h1=340.1; //Enthalpy at point 1 from refrigerant-12
    tables in kJ/kg
12 s1=1.57135; //Entropy at point 1 from refrigerant-12
    tables in kJ/kg-K
13 s2=1.57135; //Entropy at point 2 from refrigerant-12
    tables in kJ/kg-K
14 h2=372; //Enthalpy at point 2 from refrigerant-12
    tables in kJ/kg
15 h3=226.575; //Enthalpy at point 3 from refrigerant-12
    tables in kJ/kg
16 h4=226.575; //Enthalpy at point 4 from refrigerant-12
    tables in kJ/kg
17

```



```

18 //CALCULATIONS
19 Q2=m*(h1-h4); //Rate of heat removed from the
    refrigerated space in kW
20 W=m*(h2-h1); //Power input to the compressor in kW
21 Q1=m*(h2-h3); //Rate of heat rejection to the
    environment in kW
22 COP=Q2/W; //Co-efficient of performance
23
24 //OUTPUT
25 mprintf('Rate of heat removed from the refrigerated
    space is %3.2f kW \n Power input to the
    compressor is %3.3f kW \n Rate of heat rejection
    to the environment is %3.2f kW \n Co-efficient of
    performance is %3.2f',Q2,W,Q1,COP)
26
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.5 COP of system

```

1 //Chapter-6, Illustration 5, Page 311
2 //Title: Refrigeration cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 T2=40; //Temperature at point 2 in oC
9 T1=-10; //Temperature at point 1 in oC
10 h2=367.155; //Enthalpy at point 2 from refrigerant -12

```

```

    tables in kJ/kg
11 s2=1.54057; //Entropy at point 2 from refrigerant -12
    tables in kJ/kg-K
12 s1=1.54057; //Entropy at point 1 from refrigerant -12
    tables in kJ/kg-K
13 sg=1.56004; //Entropy from refrigerant -12 tables in
    kJ/kg-K
14 sf=0.96601; //Entropy from refrigerant -12 tables in
    kJ/kg-K
15 hf=190.822; //Enthalpy from refrigerant -12 tables in
    kJ/kg-K
16 hfg=156.319; //Enthalpy from refrigerant -12 tables in
    kJ/kg-K
17 h3=238.533; //Enthalpy at point 3 from refrigerant -12
    tables in kJ/kg-K
18 h4=h3; //Enthalpy at point 4 from refrigerant -12
    tables in kJ/kg-K
19
20 //CALCULATIONS
21 x1=(s1-sf)/(sg-sf); //Quality factor
22 h1=hf+(x1*hfg); //Enthalpy at point 1 from
    refrigerant -12 tables in kJ/kg
23 COP=(h1-h4)/(h2-h1); //Co-efficient of performance
24
25 //OUTPUT
26 fprintf('COP of the system is %3.2f',COP)
27
28
29
30 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.6 Capacity of refrigeration plant and Mass flow rate of refrigerant and Discharge temperature and Cylinder dimensions and Power of compressor and Theoretical and actual COP

```

1 //Chapter-6, Illustration 6, Page 311
2 //Title: Refrigeration cycles
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 Tc=35; //Temperature of condenser in oC
9 Te=-15; //Temperature of evaporator in oC
10 m=10; //Mass of ice per day in tons
11 Tw=30; //Temperature of water in oC
12 Ti=-5; //Temperature of ice in oC
13 nv=0.65; //Volumetric efficiency
14 N=1200; //Speed in rpm
15 x=1.2; //Stroke to bore ratio
16 na=0.85; //Adiabatic efficiency
17 nm=0.95; //Mechanical efficiency
18 S=4.187; //Specific heat of water in kJ/kg
19 L=335; //Latent heat of ice in kJ/kg
20 h1=1667.24; //Enthalpy at Te from Ammonia chart in kJ
    /kg
21 h2=1925; //Enthalpy at Te from Ammonia chart in kJ/kg
22 h4=586.41; //Enthalpy at Tc from Ammonia chart in kJ/
    kg
23 v1=0.508; //Specific humidity at Te from Ammonia
    chart in (m^3)/kg
24
25 //CALCULATIONS
26 Qr=((m*1000)/24)*((S*(Tw-0))+L+(1.94*(0-Ti)))
    /3600; //Refrigerating capacity in kW
27 mr=Qr/(h1-h4); //Refrigerant mass flow rate in kg/s
28 T2=112; //Discharge temperature in oC
29 D=((mr*v1*4*60)/(nv*3.14*x*N))^(1/3); //Cylinder
    diameter in m
30 L=x*D; //Stroke length in m
31 W=(mr*(h2-h1))/(na*nm); //Compressor motor power in

```

```

    kW
32 COPth=(h1-h4)/(h2-h1); // Theoretical COP
33 COPact=Qr/W; // Actual COP
34
35 //OUTPUT
36 mprintf('Refrigerating capacity of plant is %3.2f kW
    \n Refrigerant mass flow rate is %3.4f kg/s \n
    Discharge temperature is %3.0f oC \n Cylinder
    diameter is %3.3f m \n Stroke length is %3.3f m \n
    n Compressor motor power is %3.2f kW \n
    Theoretical COP is %3.2f \n Actual COP is %3.2f',
    Qr ,mr ,T2 ,D ,L ,W ,COPth ,COPact)
37
38
39
40
41 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.7 Circulation rate of ammonia and Power required and COP

```

1 //Chapter-6, Illustration 7, Page 313
2 //Title: Refrigeration cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 T1=-5; //Temperature at point 1 in oC
9 T2=30; //Temperature at point 2 in oC
10 m=13500; //mass of ice per day in kg
11 Tw=20; //Temperature of water in oC

```

```

12 COP=0.6; //Co-efficient of performance
13 h2=1709.33; //Enthalpy at point 2 in kJ/kg
14 s2=6.16259; //Entropy at point 2 in kJ/kg-K
15 s1=6.16259; //Entropy at point 1 in kJ/kg-K
16 sf=1.8182; //Entropy in kJ/kg-K
17 sg=6.58542; //Entropy in kJ/kg-K
18 hf=400.98; //Enthalpy in kJ/kg
19 hfg=1278.35; //Enthalpy in kJ/kg
20 h4=562.75; //Enthalpy at point 4 in kJ/kg
21 S=4.187; //Specific heat of water in kJ/kg
22 L=336; //Latent heat of ice in kJ/kg
23
24 //CALCULATIONS
25 x1=(s1-sf)/(sg-sf); //Quality factor
26 h1=hf+(x1*hfg); //Enthalpy at point 1 from
    refrigerant-12 tables in kJ/kg
27 COPi=(h1-h4)/(h2-h1); //Ideal COP
28 COPact=COP*COPi; //Actual COP
29 Qr=((m*S*(Tw-0))+(m*L))/(24*3600); //Total amount of
    heat removed in kJ/s
30 mr=Qr/(h1-h4); //Circulation rate of ammonia in kg/s
31 W=mr*(h2-h1); //Power required in kW
32
33 //OUTPUT
34 mprintf('Circulation rate of ammonia is %3.3f kg/s \
    n Power required is %3.3f kW \n COP is %3.3f',mr,
    W,COPact)
35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.8 Refrigerating effect and Mass flow rate of refrigerant and Theoretical power and COP and Theoretical bore and stroke of compressor

```

1 //Chapter-6, Illustration 8, Page 314
2 //Title: Refrigeration cycles
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 Tc=20; //Temperature of condenser in oC
9 Te=-25; //Temperature of evaporator in oC
10 m=15; //Mass of ice per day in tons
11 Ts=5; //Subcooled temperature in oC
12 Tsh=10; //Superheated temperature in oC
13 n=6; //No. of cylinders
14 N=950; //Speed of compressor in rpm
15 x=1; //Stroke to bore ratio
16 h1=402; //Enthalpy at point 1 from R-22 tables in kJ/
    kg
17 h2=442; //Enthalpy at point 2 from R-22 tables in kJ/
    kg
18 h3=216; //Enthalpy at point 3 from R-22 tables in kJ/
    kg
19 h4=216; //Enthalpy at point 4 from R-22 tables in kJ/
    kg
20 v1=2.258; //Specific volume at point 1 in (m^3)/min
21
22 //CALCULATIONS
23 Re=h1-h4; //Refrigerating effect in kJ/kg
24 mr=(m*14000)/(Re*60); //Mass flow of refrigerant in
    kg/min
25 Pth=(mr*(h2-h1))/60; //Theoretical power in kW
26 COP=(h1-h4)/(h2-h1); //Co-efficient of performance
27 Dth=v1/n; //Theoretical displacement per cylinder

```

```

28 D=(((Dth*4)/(3.147*N))^(1/3))*1000; // Theoretical
    bore of compressor in mm
29 L=D; // Theoretical stroke of compressor in mm
30
31 //OUTPUT
32 mprintf('Refrigerating effect is %3.0f kJ/kg \n Mass
    flow of refrigerant per minute is %3.2f kg/min \
    n Theoretical input power is %3.2f kW \n COP is
    %3.2f \n Theoretical bore of compressor is %3.2f
    mm \n Theoretical stroke of compressor is %3.2f
    mm', Re ,mr ,Pth ,COP ,D ,L)
33
34
35
36
37
38
39
40 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.9 COP when there is no subcooling and when there is subcooling

```

1 //Chapter-6, Illustration 9, Page 316
2 //Title: Refrigeration cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 T2=40; //Temperature at point 2 in oC
9 T1=-5; //Temperature at point 1 in oC

```

```

10 h2=367.155; //Enthalpy at point 2 from F-12 tables in
    kJ/kg
11 sg=1.55717; //Entropy from F-12 tables in kJ/kg-K
12 s1=1.54057; //Entropy at point 1 from F-12 tables in
    kJ/kg-K
13 sf=0.98311; //Entropy from F-12 tables in kJ/kg-K
14 hf=195.394; //Enthalpy from F-12 tables in kJ/kg
15 hfg=153.934; //Enthalpy from F-12 tables in kJ/kg
16 h4=238.533; //Enthalpy at point 4 from F-12 tables in
    kJ/kg
17 h4s=218; //Enthalpy at point 4 with subcooling from F
    -12 tables in kJ/kg
18
19 //CALCULATIONS
20 x1=(s1-sf)/(sg-sf); //Quality factor
21 h1=hf+(x1*hfg); //Enthalpy at point 1 from
    refrigerant -12 tables in kJ/kg
22 COPns=(h1-h4)/(h2-h1); //Co-efficient of performance
    with no subcooling
23 COPs=(h1-h4s)/(h2-h1); //Co-efficient of performance
    with subcooling
24
25 //OUTPUT
26 mprintf('COP with no subcooling is %3.3f \n COP with
    subcooling is %3.3f',COPns,COPs)
27
28
29
30
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.10 Ideal COP of system

```
1 //Chapter-6, Illustration 10, Page 309
2 //Title: Refrigeration cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 Tg=470; //Heating temperature in K
9 T0=290; //Cooling temperature in K
10 TL=270; //Refrigeration temperature in K
11
12 //CALCULATIONS
13 COP=((Tg-T0)/Tg)*(TL/(T0-TL)); //Ideal COP of
    absorption refrigeration system
14
15 //OUTPUT
16 mprintf('Ideal COP of absorption refrigeration
    system is %3.2f',COP)
17
18
19
20 //=====END OF PROGRAM
    =====
```

Scilab code Exa 6.11 Maximum and minimum temperature in cycle and COP and Rate of refrigeration

```
1 //Chapter-6, Illustration 11, Page 317
```

```

2 //Title: Refrigeration cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 T1=-18;//Temperature at point 1 in oC
9 T3=27;//Temperature at point 3 in oC
10 rp=4;//Pressure ratio
11 m=0.045;//mass flow rate in kg/s
12 y=1.4;//Ratio of specific heats
13 Cp=1.005;//Specific heat at constant pressure in kJ/
    kg-K
14
15 //CALCULATIONS
16 x=(y-1)/y;//Ratio
17 T2=(rp^x)*(273+T1);//Temperature at point 2 in K
18 Tmax=T2-273;//Maximum temperature in oC
19 T4=((1/rp)^x)*(273+T3);//Temperature at point 4 in K
20 Tmin=T4-273;//Minimum temperature in oC
21 qL=Cp*(T1-Tmin);//Heat rejected
22 Wcin=Cp*(Tmax-T1);//Compressor work
23 Wtout=Cp*(T3-Tmin);//Turbine work
24 Wnet=Wcin-Wtout;//Net work done
25 COP=qL/Wnet;//Co-efficient of performance
26 Qref=m*qL;//Rate of refrigeration in kW
27
28 //OUTPUT
29 mprintf('Maximum temperature in the cycle is %3.0foC
    \n Minimum temperature in the cycle is %3.0foC \
    n COP is %3.2f \n Rate of refrigeration is %3.2f
    kW',Tmax,Tmin,COP,Qref)
30
31
32
33

```

```
34
35 //=====END OF PROGRAM
=====
```

Scilab code Exa 6.12 Work developed and Refrigerating effect and COP

```
1 //Chapter-6, Illustration 12, Page 318
2 //Title: Refrigeration cycles
3 //
=====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=1; //Pressure at point 1 in bar
9 T1=268; //Temperature at point 1 in K
10 P2=5; //Pressure at point 2 in bar
11 T3=288; //Temperature at point 3 in K
12 n=1.3; //Adiabatic gas constant
13 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
14
15 //CALCULATIONS
16 x=(n-1)/n; //Ratio
17 T2=((P2/P1)^x)*T1; //Temperature at point 2 in K
18 T4=((P1/P2)^x)*T3; //Temperature at point 4 in K
19 W=Cp*(T3-T4); //Work developed per kg of air in kJ/kg
20 Re=Cp*(T1-T4); //Refrigerating effect per kg of air
    in kJ/kg
21 Wnet=Cp*((T2-T1)-(T3-T4)); //Net work output in kJ/kg
22 COP=Re/Wnet; //Co-efficient of performance
23
24 //OUTPUT
25 printf('Work developed per kg of air is %3.3f kJ/kg
```

```

    \n Refrigerating effect per kg of air is %3.3 f
    kJ/kg \n COP of the cycle is %3.2 f',W,Re,COP)
26
27
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.13 COP of refrigerator and Driving power required and Air mass flow rate

```

1 //Chapter-6, Illustration 13, Page 319
2 //Title: Refrigeration cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 T1=277; //Temperature at point 1 in K
9 T3=328; //Temperature at point 3 in K
10 P1=0.1; //Pressure at point 1 in MPa
11 P2=0.3; //Pressure at point 2 in MPa
12 nc=0.72; //Isentropic efficiency of compressor
13 nt=0.78; //Isentropic efficiency of turbine
14 y=1.4; //Adiabatic gas constant
15 Cp=1.005; //Specific heat at constant pressure in kJ/
    kg-K
16 m=3; //Cooling load in tonnes
17
18 //CALCULATIONS
19 x=(y-1)/y; //Ratio

```

```

20 T2s=T1*((P2/P1)^x); //Temperature at point 2s in K
21 T2=((T2s-T1)/nc)+T1; //Temerature at point 2 in K
22 T4s=T3*((P1/P2)^x); //Temperature at point 4s in K
23 T4=T3-((T3-T4s)*nt); //Temperature at point 4 in K
24 Re=Cp*(T1-T4); //Refrigerating effect in kJ/kg
25 Wnet=Cp*((T2-T1)-(T3-T4)); //Net work output in kJ/kg
26 COP=Re/Wnet; //Co-efficient of performance
27 P=(m*3.52)/COP; //Driving power required in kW
28 ma=(m*3.52)/Re; //Mass flow rate of air in kg/s
29
30 //OUTPUT
31 mprintf('COP of refrigerator is %3.2f \n Driving
    power required is %3.0f kW \n Mass flow rate of
    air is %3.3f kg/s',COP,P,ma)
32
33
34
35
36 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.14 Theoretical COP and Net cooling produced

```

1 //Chapter-6, Illustration 14, Page 321
2 //Title: Refrigeration cycles
3 //
    =====
4 clc
5 clear
6
7 //INPUT DATA
8 P1=2.5; //Pressure at point 1 in bar
9 P3=9; //Pressure at point 3 in bar
10 COPr=0.65; //Ratio of actual COP to the theoretical

```

```

COP
11 m=5; //Refrigerant flow in kg/min
12 T1=309; //Temperature at point 1 in K
13 T2s=300; //Temperature at point 2s in K
14 h1=570.3; //Enthalpy at P1 from the given tables in
    kJ/kg
15 h4=456.4; //Enthalpy at P3 from the given tables in
    kJ/kg
16 h2g=585.3; //Enthalpy at P3 from the given tables in
    kJ/kg
17 s2=4.76; //Entropy at P1 from the given tables in kJ/
    kg-K
18 s2g=4.74; //Entropy at P3 from the given tables in kJ
    /kg-K
19 Cp=0.67; //Specific heat at P3 in kJ/kg-K
20
21 //CALCULATIONS
22 T2=(2.718^((s2-s2g)/Cp))*T2s; //Temperature at point
    2 in K
23 h2=h2g+(Cp*(T2-T2s)); //Enthalpy at point 2 in kJ/kg
24 COPR=(h1-h4)/(h2-h1); //Refrigerant COP
25 COPact=COPr*COPR; //Actual COP
26 qL=COPact*(h2-h1); //Heat rejected in kJ/kg
27 QL=((m*qL*60)/3600)/3.516; //Cooling produced per kg
    of refrigerant in tonnes of refrigeration
28
29 //OUTPUT
30 mprintf('Theoretical COP is %3.2f \n Net cooling
    produced per hour is %3.2f TR',COPR,QL)
31
32
33
34
35 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.15 Theoretical COP of machine

```
1 //Chapter-6, Illustration 15, Page 322
2 //Title: Refrigeration cycles
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 T2=298; //Temperature at point 2 in K
9 T1=268; //Temperature at point 1 in K
10 hf1=-7.54; //Liquid Enthalpy at T1 in kJ/kg
11 x1=0.6; //Quality factor 1
12 hfg1=245.3; //Latent heat at T1 in kJ/kg
13 sf1=0.251; //Liquid Entropy at T1 in kJ/kg-K
14 s1=0.507; //Entropy at point 1 in kJ/kg-K
15 hfg2=121.4; //Latent heat at T2 in kJ/kg
16 hf2=81.3; //Liquid Enthalpy at T2 in kJ/kg
17 h4=hf2; //Enthalpy at point 4 in kJ/kg
18
19 //CALCULATIONS
20 h1=hf1+(x1*hfg1); //Enthalpy at point 1 in kJ/kg
21 x2=((s1-sf1)*T2)/hfg2; //Quality factor 2
22 h2=hf2+(x2*hfg2); //Enthalpy at point 2 in kJ/kg
23 COP=(h1-h4)/(h2-h1); //COP of the machine
24
25 //OUTPUT
26 mprintf('COP of the machine is %3.2f',COP)
27
28
29
30
```

```
31 //=====END OF PROGRAM
    =====
```

Scilab code Exa 6.16 Theoretical COP of refrigerator and Capacity of refrigerator

```
1 //Chapter-6, Illustration 16, Page 323
2 //Title: Refrigeration cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 P1=25; //Pressure at point 1 in bar
9 P2=60; //Pressure at point 2 in bar
10 h2=208.1; //Vapour enthalpy at P2 in kJ/kg
11 h3=61.9; //Liquid enthalpy at P2 in kJ/kg
12 h4=h3; //Liquid enthalpy at P2 in kJ/kg
13 s2=0.703; //Vapour entropy at P2 in kJ/kg-K
14 sf1=-0.075; //Liquid entropy at P1 in kJ/kg-K
15 sfg1=0.971; //Entropy in kJ/kg-K
16 hf1=-18.4; //Liquid Enthalpy at P1 in kJ/kg
17 hfg1=252.9; //Latent heat at P1 in kJ/kg
18 m=5; //Refrigerant flow in kg/min
19
20 //CALCULATIONS
21 x1=(s2-sf1)/sfg1; //Quality factor 1
22 h1=hf1+(x1*hfg1); //Enthalpy at point 1 in kJ/kg
23 COP=(h1-h4)/(h2-h1); //Co-efficient of performance
24 QL=(m*(h1-h4))/60; //Capacity of the refrigerator in
    kW
25
26 //OUTPUT
```



```

27 mprintf('COP of refrigerator is %3.2f \n Capacity of
    refrigerator is %3.2f kW',COP,QL)
28
29
30
31 //=====END OF PROGRAM
    =====

```

Scilab code Exa 6.17 COP and Theoretical power required

```

1 //Chapter-6, Illustration 17, Page 324
2 //Title: Refrigeration cycles
3 //
    =====

4 clc
5 clear
6
7 //INPUT DATA
8 T1=271;//Temperature at point 1 in K
9 T=265;//Temperature at point 1' in K
10 Ta=303;//Temperature at point 2' in K
11 Cpv=0.733;//Specific heat of vapour in kJ/kg
12 Cpl=1.235;//Specific heat of liquid in kJ/kg
13 h=184.07;//Liquid enthalpy at T in kJ/kg
14 s=0.7;//Entropy at point 1' in kJ/kg-K
15 sa=0.685;//Vapour entropy at Ta in kJ/kg-K
16 ha=199.62;//Enthalpy at point 2' in kJ/kg
17 hfb=64.59;//Liquid enthalpy at Ta in kJ/kg
18 DT3=5;//Temperature difference in oC
19 Q=2532;//Refrigeration capacity in kJ/min
20
21 //CALCULATIONS
22 s2=s+(Cpv*((log(T1/T))/(log(2.718)))); //Entropy at
    point 1 in kJ/kg-K

```

```

23 h1=h+(Cpv*(T1-T)); //Enthalpy at point 1 in kJ/kg-K
24 T2=(2.718^((s2-sa)/Cpv))*Ta; //Temperature at point 2
    in K
25 h2=ha+(Cpv*(T2-Ta)); //Enthalpy at point 2 in kJ/kg
26 h4=hfb-(Cpl*DT3); //Enthalpy at point 4 in kJ/kg
27 COP=(h1-h4)/(h2-h1); //Co-efficient of performance
28 m=Q/(h1-h4); //Mass flow rate of refrigerant in kJ/
    min
29 P=(m*(h2-h1))/(60*12); //Power required in kW/TR
30
31 //OUTPUT
32 mprintf('COP is %3.2f \n Theoretical power required
    per tonne of refrigeration is %3.3f kW/TR',COP,P)
33
34
35
36
37
38
39
40
41
42
43
44
45 //=====END OF PROGRAM
    =====

```

Chapter 7

Air Conditioning

Scilab code Exa 7.1 Heating capacity of coil and Surface temperature and Capacity

```
1 //Chapter-7, Illustration 1, Page 345
2 //Title: Air Conditioning
3 //


---


4 clc
5 clear
6
7 //INPUT DATA
8 DBTo=10; //Out door Dry bulb temperature in oC
9 WBTto=8; //Out door Wet bulb temperature in oC
10 DBTi=20; //In door Dry bulb temperature in oC
11 RH=0.6; //Re-Heat factor
12 a=0.3; //amount of air circulated in (m^3)/min/person
13 S=50; //Seating capacity of office
14 BPF=0.32; //ByPass factor
15 ha=25; //Enthalpy at point a from Psychrometric chart
    shown in Page 346 in kJ/kg
16 hb=42.5; //Enthalpy at point b from Psychrometric
    chart shown in Page 346 in kJ/kg
```

```

17 hc=42.5; //Enthalpy at point c from Psychrometric
    chart shown in Page 346 in kJ/kg
18 Wa=0.006; //Specific humidity at point a from
    Psychrometric chart shown in Page 346 in kg/kg
    dry air
19 Wc=0.009; //Specific humidity at point c from
    Psychrometric chart shown in Page 346 in kg/kg
    dry air
20 Tb=27; //Temperature at point b in oC
21 na=0.81; //Specific Volume from Psychrometric chart
    shown in page 346 in (m^3)/kg
22
23 //CALCULATIONS
24 ma=(a*S)/(na*60); //mass of air circulated per second
    in kg/s
25 Hc=ma*(hb-ha); //Heating capacity of coil in kW
26 Ts=(Tb-(BPF*DBTo))/(1-BPF); //Heating coil surface
    temperature in oC
27 C=(ma*3600)*(Wc-Wa); //Capacity of humidifier in kg/
    hr
28
29 //OUTPUT
30 mprintf('Heating capacity of coil is %3.2f kW \n
    Surface temperature of coil is %3.0f oC \n
    Capacity of humidifier is %3.2f kg/hr ',Hc,Ts,C)

```

Scilab code Exa 7.2 Capacity of coils and Amount of water vapour removed and by pass factor

```

1 //Chapter -7, Illustration 2, Page 346
2 //Title: Air Conditioning
3 //


---


4 clc

```

```

5  clear
6
7  //INPUT DATA
8  S=60; //No. of staff
9  DBTo=30; //Out door Dry bulb temperature in oC
10 RHo=0.7; //Re-Heat factor at out-door
11 a=0.4; //amount of air circulated in (m^3)/min/person
12 DBTi=20; //In door Dry bulb temperature in oC
13 RHi=0.6; //Re-Heat factor at indoor
14 Td=25; //Heating coil surface temperature in oC
15 ha=82.5; //Enthalpy at point a from Psychrometric
    chart shown in Page 347 in kJ/kg
16 hb=34.5; //Enthalpy at point b from Psychrometric
    chart shown in Page 347 in kJ/kg
17 hc=42.5; //Enthalpy at point c from Psychrometric
    chart shown in Page 347 in kJ/kg
18 Wa=0.020; //Specific humidity at point a from
    Psychrometric chart shown in Page 347 in kg/kg
    dry air
19 Wb=0.009; //Specific humidity at point b from
    Psychrometric chart shown in Page 347 in kg/kg
    dry air
20 Tb=12; //Temperature at point b in oC
21 na=0.89; //Specific Volume from Psychrometric chart
    shown in page 346 in (m^3)/kg
22
23 //CALCULATIONS
24 ma=(a*S)/(na*60); //mass of air circulated per second
    in kg/s
25 Hc=(ma*(ha-hb))/3.5; //Heating capacity of cooling
    coil in tonnes
26 Hh=ma*(hc-hb); //Heating capacity of heating coil in
    kW
27 W=(ma*3600)*(Wa-Wb); //Amount of water vapour removed
    per hour in kg/hr
28 BPF=(Td-DBTi)/(Td-Tb); //By-Pass factor
29
30 //OUTPUT

```

```

31 mprintf('Capacity of cooling coil is %3.2f tonnes \n
    Capacity of heating coil is %3.1f kW \n Amount
    of water vapour removed per hour is %3.2f kg/hr \
    n Bypass factor is %3.3f',Hc,Hh,W,BPF)

```

Scilab code Exa 7.3 Supply air condition and Refrigeration load and Total refrigeration capacity and Quantity of fresh air supplied

```

1 //Chapter-7, Illustration 3, Page 347
2 //Title: Air Conditioning
3 //

```

```

4 clc
5 clear
6
7 //INPUT DATA
8 RSH=10; //Room sensible heat in kW
9 RLH=10; //Room latent heat in kW
10 td1=25; //Inside temperature in oC
11 RH1=0.5; //Inside Re-Heat factor
12 h1=50.4; //Enthalpy at point 1 in kJ/kg
13 td2=35; //Out door Dry bulb temperature in oC
14 tw2=28; //Out door Wet bulb temperature in oC
15 CR=4; //Cooling coil ratio
16 BPF=0.1; //Cooling coil bypass factor
17 tADP=10; //Apparatus dew point temperature in oC
18 RH3=0.55; //Re-Heat factor at point 3
19 h3=58.2; //Enthalpy at point 3 in kJ/kg
20 RH4=0.95; //Re-Heat factor at point 4
21 h4=32.2; //Enthalpy at point 4 in kJ/kg
22 RH5=0.81; //Re-Heat factor at point 5
23 h5=36.8; //Enthalpy at point 5 in kJ/kg
24 RH6=0.54; //Re-Heat factor at point 6
25 h6=43.1; //Enthalpy at point 5 in kJ/kg

```

```

26 td6=22; //Temperature at point 6 in oC
27
28 //CALCULATIONS
29 td3=((td2-td1)/5)+td1; //Temperature at point 3 from
    Psychrometric chart shown in Page 348 in oC
30 td4=(BPF*(td3-tADP))+tADP; //Temperature at point 4
    from Psychrometric chart shown in Page 348 in oC
31 td5=td4+((td1-td4)/5); //Temperature at point 5 from
    Psychrometric chart shown in Page 348 in oC
32 RSHF=RSH/(RSH+RLH); //Room Sensible Heat Factor
33 QR=h1-h6; //Total heat removed in kJ/kg
34 S=(RSH+RLH)/QR; //Supply air quantity in kg/s
35 R=(S*(h6-h5))/3.5; //Refrigeration load due to reheat
    in ton
36 D=(S*4)/5; //Dehumidified air quantity in kg/s
37 T=(D*(h3-h4))/3.5; //Total refrigerating capacity in
    ton
38 Q=(D/5)/1.2; //Quantity of fresh air supplied in (m
    ^3)/s
39
40 //OUTPUT
41 mprintf('Supply air condition to the room is %3.2f
    kg/s \n Refrigeration load due to reheat is %3.2f
    ton \n Total refrigerating capacity is %3.2f ton
    \n Quantity of fresh air supplied is %3.3f (m^3)
    /s ',S,R,T,Q)

```
